TEKSTİL VE KONFEKSİYON



Simultaneous Coloring and Antibacterial Finishing to Cotton Through Environmentally-Friendly Way

Burcu Sancar Beşen¹ ⁽ⁱ⁾ 0000-0001-5120-268X Pınar Parlakyiğit¹ ⁽ⁱ⁾ 0000-0001-9235-6239 Erkan Yılmaz² ⁽ⁱ⁾ 0000-0002-0947-3096

¹Adıyaman University, Faculty of Engineering, Department of Textile Engineering, Adıyaman/Turkiye ²Adıyaman University, Faculty of Pharmacy, Department of Pharmacognosy, Adıyaman/Turkiye

Corresponding Author: Burcu Sancar Beşen, bbesen@adiyaman.edu.tr

ABSTRACT

The present paper aimed to color 100% cotton fabrics while also ensuring antibacterial activity in an eco-friendly manner. For this purpose, cationized and pre-mordanted fabrics were dyed by using everlasting flower extract as a natural dye source. After dyeing process, antibacterial activity, CIEL*a*b* color values, color strength (K/S), and fastness properties (washing, perspiration, rubbing) of the dyed fabrics were investigated, as well as the bursting strength of the dyed and undyed fabrics was also examined. The results showed that it was possible to ensure simultaneous coloring and antibacterial activity on the cotton fabrics by using a methanol extract of the everlasting flower as a natural dye source. In addition, applying the cationizing and mordanting processes to cotton fabrics before dyeing provided higher antibacterial activity, better fastness properties, and a darker color.

ARTICLE HISTORY

Received: 11.11.2022 Accepted: 24.03.2023

KEYWORDS

Everlasting flower (*Helichrysum armenium* subsp. *araxinum*), natural dye, antibacterial activity, cationizing, mordanting

1. INTRODUCTION

Due to the public's enhanced environmental awareness and health concerns, sustainable, ecologic, nontoxic, and biodegradable products have recently gained more popularity [1]. Normally, textile dyeing and finishing processes include a number of steps in which hazardous chemicals are used. Scientists and textile industry firms have labored to reduce pollution caused by synthetic dyes and toxic chemicals currently used in textile applications [2,3]. Interest in the potential use of natural dyes has been growing, because of being safe, non-toxic, noncarcinogenic and biodegradable, having high compatibility with the environment, and being available in a range of natural shades as compared with synthetic dyes [3-6].

Natural dyes mainly contain various natural bioactive compounds, such as phenolic acids, flavonoids, alkaloids, terpenoids, essential oils, and natural color. These bioactive compounds have antimicrobial, antioxidant, anti-inflammatory activities and they can be used for hygiene, medical and pharmacological applications [2,4,7].

Antimicrobial textile materials have gained considerable popularity all over the world [4,8], since the textile materials provide suitable environment for growth and multiplication of pathogenic microbes, leading to unpleasant odor, dermal infection, weakening of the substrate, discoloration, allergies, and other related diseases [7]. Various methods have been developed or are under development to give antimicrobial activity to textiles [5]. Although synthetic antimicrobial agents such as triclosan, metal and their salts, organometallics, phenols, etc. show good inhibition against bacteria, using natural eco-friendly agents such as natural dyes can eliminate their environmental risks [4]. For the reasons mentioned, with an ecological view, extract of everlasting flower (*Helichrysum armenium* DC. subsp. *araxinum* (Kirp.) Takht) was used as natural dye and antimicrobial agent in the present study.

The genus *Helichrysum*, belonging to the family of Asteraceae, consists of a few hundred species that are widespread throughout the world [9]. Plants of the genus *Helichrysum* are fertile producers of a host of secondary metabolites, including flavonoids, acetophenones, phloroglucinol, pyrones, triterpenoids and sesquiterpenes [10]. *Helichrysum* species are commonly used in Turkey and other parts of the world for their various biological properties, such as anti-inflammatory, antioxidant, and antimicrobial activities

To cite this article: Sancar Beşen B, Parlakyiğit P, Yılmaz E. 2024. Simultaneous Coloring and Antibacterial Finishing to Cotton Through Environmentally-Friendly Way. *Tekstil ve Konfeksiyon*, 34(1), 44-50.



[11]. Albayrak et.al (2010) showed the antibacterial activities of methanol extract of *Helichrysum armenium* subsp. *araxinum* against *Aeromonas hydrophila*, *Bacillus brevis*, *B. cereus*, *Klebsiella pneumoniae*, *Pseudomonas aeruginosa*, and *Staphylococcus aureus* bacteria [11].

In the literature, there are a lot of studies about coloring of some textile materials with natural sources. When the studies are focused, it can be seen that most of them have interested in dyeing of cationic textile materials such as wool, silk and polyamide [1,3-8,12-14] since the cationic fibers can be dyed easier than anionic ones. Although cotton is the most preferred natural fiber [15], there are limited studies about natural dyeing of cotton materials [16,17]. A cationization process should be applied to the cotton materials before natural dyeing in order to achieve better dyeing.

The most common methodology for cationization is to introduce amino groups into the fiber [18]. There have been various reported cationic agents for the modification of cotton in the literature, such as polyamide-based epichlorohydrin type of polymers, dendritic polymers, quaternary ammonium-based compounds, glycidyltrimethyl ammonium chloride (Glytac), choline chloride, Nmethylolacrylamide, and biopolymers like chitosan [18,19], starch and their derivatives, keratin hydrolysate from chicken feather, horn and hoofs [19]. In the present study, chitosan, one of the biopolymers, was preferred for cationization. A renewable polysaccharide-based cationic biopolymer, chitosan (CH), is the deacetylated derivative of chitin [20] and is an important alternative to commonly used compounds to functionalize cotton because of its protonated amino groups [21]. Chitosan is inexpensive, abundant, non-toxic, biocompatible, and eco-friendly material [21,22].

Due to the low affinity of most natural dyes to textile fibers, mordants are used to increase their affinity and color fastness [12]. Mordants are commonly metallic salts such as tannic acid, alum, chrome alum, sodium chloride [24] which can be fixed on the cationic fibers and interact with dye through coordination bonds [4,12]. Mordanting can be achieved in three ways as pre-mordanting (before dyeing), simultaneously mordanting, and post-mordanting (after dyeing). In this study, the pre-mordanting process was carried out with using aluminum potassium sulphate as mordant agent due to being non-harmful to the environment.

In the experimental study, it was aimed to ensure simultaneous coloring and antibacterial activity to 100% cotton fabrics in an eco-friendly way. For this purpose, the methanol extract of the everlasting flower was used as a natural dye source. The cationized, pre-mordanted 100% cotton knitted fabrics were dyed with the extract. After the dyeing process, CIEL*a*b* color values, color strength values, fastness properties (washing, perspiration, rubbing), and antibacterial activity of the dyed fabrics were

investigated, as well as the bursting strength tests were also applied to dyed and undyed fabrics.

2. MATERIAL AND METHOD

2.1. Material

Chitosan (medium molecular weight, Sigma-Aldrich), aluminium potassium sulphate dodecahydrate (KAl(SO₄)₂.12H₂O, Merck), Rucon LFF (cross-linker, Rudolf Duraner) methanol (Isolab), rotatory evaporator (Scilogex), ultrasonic bath (Isolab), and water bath (Isolab) were used. The KAl(SO₄)₂ was chosen as mordanting agent since it is known as green agent [25]. The chemicals were used without any further purification.

The aerial parts of the everlasting plant used in the study were collected from Bingöl Metan Mountain in Turkey on June, 2021. The photo of the plant can be seen in Figure 1. The collected sample were dried in the shade and stored in a clean and dry environment. Plant identity was verified by Dr. Ömer Kılıç and the prepared herbarium sample is kept in Hacettepe University Faculty of Pharmacy Herbarium with the number HUEF-21042.



Figure 1. Helichrysum armenium subsp. araxinum in its habitat

100% cotton knitted fabric (Ne 30/1, 1*1 rib knit, 212 g/m^2) was used in the experiments. The fabric was scoured and bleached by the supplier (İskur Group, Kahramanmaraş). A laboratory-type fulard (Termal), an IR dyeing machine (Termal), and a steamer (Ataç) were also used.

2.2. Method

2.2.1. Preparation of everlasting flower extract

The powder of the everlasting flower was extracted with methanol via the maceration process. The ratio of plant:solvent was determined as approximately 1:10 (w/v) in the water bath at 40 °C. The maceration process was



repeated three times for three hours for each one, and the combined solutions were filtered through filter paper, and they were then evaporated to dryness under vacuum at 40 °C with a rotary evaporator. In order to preserve the structure of such chemical substances in the plant, the temperature was not exceeded by 40 °C during both extraction and evaporation.

The photo of the produced extract is given in Figure 2. The yield of the extract was calculated as 13.79%.



Figure 2. The methanol extract of everlasting flower

A stock solution of the everlasting flower extract was prepared in distilled water at the concentration of 20 mg/ml by using ultrasonic bath. The prepared stock solution was stored in the freezer at -24 °C for further usage.

2.2.2. Cationization process of the cotton fabrics

The cotton fabrics were cationized with chitosan. The concentration of chitosan was chosen at two different levels as 10 g/L and 20 g/L and it was applied to cotton samples by pad-dry-cure method. In order to prepare the application solution, the chitosan was dissolved in a %1 (w/v) acetic acid solution, and 50 g/L cross-linker agent was added to the solution. The fabric samples were squeezed to 96% pick up at fulard, then dried at 100 °C and cured for 2 minutes at steamer at 110 °C. Some of the fabric samples were saved without cationizing to investigate the effect of the cationization process on the results.

2.2.3. Mordanting process

The pre-mordanting process was carried out with 5 g/L concentration of KAl(SO₄)₂ at 98 °C for 60 minutes with a liquor ratio of 1:10 by using IR dyeing machine. After the process, the samples were rinsed with cold water and dried at 100 °C. In order to research the effect of the mordanting process on the results, some of the fabric samples were saved without mordanting.

2.2.4. Dyeing process

The different pretreated fabric samples (only cationized, only mordanted, both cationized and mordanted, and neither cationized nor mordanted) were dyed. The dyed fabric samples can be seen in Table 1. The dyeing processes were carried out with the stock extract solution with a liquor ratio of 1:5 at 40 °C for 30 minutes by using an IR dyeing machine. After the dyeing, the samples were washed with cold water for 10 minutes and then dried at the room temperature.

Table 1. The dyed fabric samples

Sample Number	Chitosan concentration (g/l)	Mordant concentration (g/l)
1	0	0
2	10	0
3	20	0
4	0	5
5	10	5
6	20	5

2.3. The research methods

Bursting strength: The test was applied to dyed and undyed samples according to the ISO 13938-2:2019 standard [26] as three replications and the average of the results was calculated.

CIEL*a*b* color value and color strength (K/S) value: The CIEL*a*b* values and the reflectance (R) values at wavelengths ranging from 400 to 700 nm of the dyed samples were measured. The measurements were carried out under D65 daylight and an aspect of 10° with a spectrophotometer (Datacolor SF 600 model). The results were measured over three replications, and the average of the results was calculated. The K/S values were calculated using the Kubelka-Munk equation by using the R-value at the wavelength (410 nm) at which each fabric sample had maximum absorption.

Color fastness test: In order to investigate the color fastness results of the dyed samples, fastness tests to washing [27], acidic and alkaline perspiration [28], and wet and dry rubbing [29] were carried out.

Antibacterial activity: The antibacterial activities of the everlasting flower extract and the fabric samples were investigated against Gram-negative bacteria (Escherichia ATCC coli 8739) and Gram-positive bacteria (Staphylococcus aureus ATCC 6538), since they are the major microorganisms responsible for 25% of hospital infections and popular test organisms which are resistant to common antimicrobial agents [30]. The tests were carried out in Ege University's Microbiological Analysis Laboratory (Egemikal). While the antibacterial activity of the extract was investigated through the disc diffusion method, it was researched via the AATCC 100 method for fabrics. The tests were applied as three replications.

3. RESULTS AND DISCUSSION

The antibacterial activity results of the stock solution of everlasting flower extract assessed according to the disc



diffusion method against *S. aureus* and *E. coli* bacteria are given in Figure 3. The inhibition zones formed by the extract solution against *S. aureus* and *E. coli* are 11.2 and 8 mm, respectively. Thus, it is possible to say that the prepared stock extract solution had antibacterial activity against both tested bacteria species.

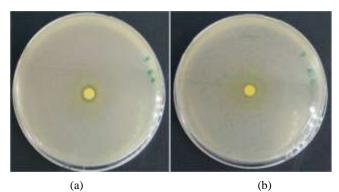


Figure 3. The antibacterial activity results of everlasting extract

The antibacterial activity results of the dyed samples assessed according to the AATCC 100 method against *E. coli* and *S. aureus* bacteria can be found in Table 2. The results show that the fabrics gained antibacterial activity against both *E. coli* and *S. aureus* bacteria. The fabric sample that was not subjected to cationizing and mordanting processes had less activity against *E. coli* than other samples since it had less dye. As *E. coli* is a more resistant bacteria than *S. aureus*, the result was observed more specifically for *E. coli*. As a result, it can be said that it is possible to provide both coloring and antibacterial activity simultaneously to a cotton fabric with a natural dyeing processes before dyeing.

The bursting strength results of the fabric samples before and after dyeing can be seen in Figure 4. The bursting strength of the untreated raw fabric was 390,8 KPa, as shown in Figure 4, and the strength of the fabrics decreased after the cationizing and mordanting processes. However, the decline was not crucial and occurred at 14%. Generally, the increase in the chitosan concentration had no negative effect on the strength property. In addition, the dyeing process had a negligible effect on the strength properties of the samples, since no harsh chemicals were used in the natural dyeing process.

Table 3 shows the CIEL*a*b* color values as well as the color strength (K/S) values of the dyed samples. It is commonly known that while the L* value indicates the lightness and darkness, the b^{*} value gives information about the yellowness and blueness, and the a* value is about the reddish and greenish tones of the fabric samples in the CIEL*a*b* color system. The L* value changes between 0 and 100, and the color becomes lighter as the L* value increases. The increase in the a^{*} and b^{*} values indicates that the color is becoming more reddish and yellowish, respectively. In addition, the increase in the K/S value means that the fabric is darker. When the results in Table 3 are focused, it can be clearly seen that both the application of the cationizing process and the use of mordants increased the dye uptake. Furthermore, the increase in the mordant concentration also increased the depth of color. In addition, the b* value increased significantly, especially when the mordant was used. This circumstance shows that the use of mordant had an effect on the shade of the fabric samples. The application of the mordant process after the chitosan treatments caused the color to have greater depth and strength. Thus, it is possible to say that in order to have a naturally colored fabric with a darker shade, it is necessary to apply both the cationizing and mordanting processes.

Sample number		E. coli		S. aureus					
	0.hour	24. hours	% Reduction	0.hour	24. hours	% Reduction			
Untreated fabric	560000	Reproduction	-	263000	Reproduction	-			
1	560000	45000	91.96	263000	100	99.96			
2	560000	610	99.89	263000	50	99.98			
3	560000	3380	99.40	263000	970	99.63			
4	560000	1430	99.74	263000	75	99.97			
5	560000	650	99.88	263000	80	99.97			
6	560000	50	99 99	263000	300	99.89			

 Table 2. % Reduction in bacteria after 24 hours

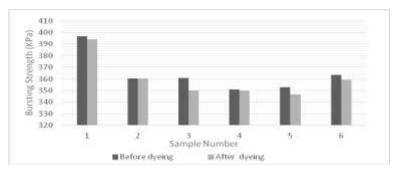


Figure 4. The bursting strength results of the fabric samples before and after dyeing



The results of washing, perspiration (acidic and alkaline), and rubbing (dry and wet) fastness of dyed samples are given in Table 4. Generally, the wash fastness of the fabric samples with dark colors has shown more staining than those with light colors. For that reason, the highest fastness results were obtained for the samples to which neither the cationizing nor mordanting processes were applied, since they had less dye. Besides, the washing and perspiration fastness results got better when the mordanting process was applied to the fabric samples before the dyeing process.

Sample number	Photos	\mathbf{L}^{*}	a*	b*	K/S
Undyed fabric		90.44	-0.07	1.87	-
1	1000	79.02	-3.91	25.95	3.19
2		71.09	-3.79	27.98	6.97
3		70.35	-2.67	33.07	9.89
4		77.43	-3.95	45.67	7.96
5		71.13	-4.51	40.91	11.22
6		67.84	-2.95	42.89	11.84

Table 4. The results of color fastness of dyed samples

ber	Washing							Perspiration									Rubbing				
number	50					- \			Acidic				Alkaline								
Sample	Fading	\mathbf{As}	Co	ΡA	PES	PAC	OM	As	Co	PA	PES	PAC	OM	As	Co	PA	PES	PAC	OM	Wet	Dry
1	5	4/5	3/4	4/5	4/5	4/5	4/5	4/5	3/4	3/4	4/5	4	4	4/5	2/3	3/4	4	4	4	4	4/5
2	4	4	2/3	4	4	4	4	4	3	3	4/5	4	4	4	2/3	3	4	4	3/4	3/4	4
3	3/4	4	2/3	4	4	4	4	4/5	3	3	4	4	3	4	1/2	2/3	4	4	3	3/4	4
4	4	4/5	3	4/5	4/5	4/5	4/5	4/5	3	3	4	4	4	4	2/3	3	3/4	3/4	3/4	3/4	4
5	4/5	4	3	4	4/5	4/5	4/5	4/5	3	3	4/5	4/5	4	4	2/3	3	4	4	3/4	3	4
6	4	4/5	3	4	4/5	4/5	4	4/5	3	3	4	4	3/4	4	2/3	2/3	3	3	3	3	4

4. CONCLUSION

The subject of applying dyeing process, which has crucial importance among the textile finishing process, with natural dyes and eco-friendly way has been more significant with increasing environmental awareness. For this reason, researchers have studied on coloring of the textile materials with different natural dye source. Furthermore, the vast majority of natural sources have antimicrobial activity, allowing for simultaneous coloring and antibacterial activity in textiles. In the present study, it was aimed to provide simultaneous coloring and antibacterial activity to 100% cotton knitted fabrics with the everlasting flower extract. For this purpose, firstly, a methanol extract of the everlasting flower was prepared, and the antibacterial activity of the extract was examined. Then, the fabrics were cationized by using chitosan as a natural biopolymer and pre-mordanted with KAl(SO₄)₂ as a green mordanting agent. The treated fabrics were dyed with the prepared extract. Following the dyeing process, the fabrics' antibacterial activity, CIEL*a*b* color values, and color fastness to washing, perspiration, and rubbing were



evaluated. In addition, the bursting strength of the dyed and undyed fabrics was also measured. The results can be summarized as given below.

- The methanol extract of everlasting flower has antibacterial activity against both *S. aureus* and *E. coli* bacteria species.
- The bursting strength results are negatively affected by the cationizing and mordanting processes. However, the decline in the results is lower than 15% and not crucial. The increase in chitosan concentration has no crucial effect on the strength property, and the usage of both chitosan and mordant together have the most effect on the results.
- The applications of cationizing and mordanting processes increase the dye uptake, and treated fabrics dye with darker shade. In addition, the mordanting

5. REFERENCES

- Koçak, ÖF, Yılmaz, F. 2022. Use of Alpinia Officinarum Rhizome in Textile Dyeing and Gaining Simultaneous Antibacterial Properties *Journal of Natural Fibers* 19(5), 1925–1936.
- Rehan, M, Ibrahim, GE, Mashaly, HM, Hasanin, M, Rashad, HG, Mowaf, S. 2022. Simultaneous dyeing and multifunctional finishing of natural fabrics with Hibiscus flowers extract *Journal of Cleaner Production* 374 https://doi.org/10.1016/j.jclepro.2022.133992.
- 3. Zhang, B, Wang, L, Luo, L, King, MW. 2014. Natural dye extracted from Chinese gall e the application of color and antibacterial activity to wool fabric *Journal of Cleaner Production* 80, 204-210
- Ghaheh, FS., Mortazavi, SM, Alihosseini, F, Fassihi, A, Nateri, AS, Abedi, D. 2014. Assessment of antibacterial activity of wool fabrics dyed with natural dyes *Journal of Cleaner Production* 72, 139-145
- Mirjalili, M, Karimi, L. 2013a. Antibacterial Dyeing of Polyamide Using Turmeric as a Natural Dye, *AUTEX Research Journal*, 13(2) 51-56
- Safapour, S, Sadeghi-Kiakhani, M, Eshaghloo-Galugahi, S. 2018. Extraction, Dyeing, and Antibacterial Properties of Crataegus Elbursensis Fruit Natural Dye on Wool Yarn *Fibers and Polymers* 19(7), 1428-1434.
- Mirjalili, M, Karimi, L. 2013b. Extraction and Characterization of Natural Dye from Green Walnut Shells and Its Use in Dyeing Polyamide: Focus on Antibacterial Properties *Hindawi Publishing Corporation Journal of Chemistry* 2013, http://dx.doi.org/10. 1155/2013/375352.
- Shahid, M; Islam, S, Rather, LJ, Manzoor, N, Mohammad, F. 2019. Simultaneous shade development, antibacterial, and antifungal functionalization of wool using Punica granatum L. Peel extract as a source of textile dye *Journal of Natural Fibers* 16:4, 555-566.
- Bahşi, M, Emre, MY, Emre, İ, Kurşat, M, Yılmaz, Ö. 2020. The determination of some biochemical contents of Helichrysum armenium DC. subsp. araxinum (Kirp.) Takht *Biological Diversity* and Conservation 13(1), 96-101.
- 10. Oji, K, Shafaghat, A. 2012. Constituents and Antimicrobial Activity of the Essential Oils from Flower, Leaf and Stem of *Helichrysum armenium*, *Natural Product Communications* 7(5), 671-674.
- Albayrak, S, Aksoy, A, Sağdıç, O, Budak, Ü. 2010. Phenolic compounds and antioxidant and antimicrobial properties of Helichrysum species collected from eastern Anatolia, Turkey *Turkish Journal of Biology* 34(4), 463-473.
- Ghoreishian, SM, Maleknia, L, Mirzapour, H, Norouz, M. 2013. Antibacterial Properties and Color Fastness of Silk Fabric Dyed with Turmeric Extract *Fibers and Polymers* 14(2), 201-207

process have positive effect on the fastness properties of the samples.

- The dyed samples after cationizing and mordanting processes gain high antibacterial activity against both *S. aureus* and *E. coli* bacteria.

As a conclusion, the %100 cotton fabrics can be colored with everlasting flower extract with gaining antibacterial activity. In order to achieve dyeing process with high antibacterial activity, good fastness, and dark color, the cationizing and mordanting processes are recommended before dyeing.

Acknowledgements

We are grateful to Prof. Dr. Ömer Kılıç (Adiyaman University Faculty of Pharmacy) for helpful assistance in botanical description.

- Haji, A. 2012. Antibacterial Dyeing of Wool with Natural Cationic Dye Using Metal Mordants *Materials Science* 18(3), 267-270
- 14. Shabbir, M, Rather, LJ, Azam, M, Haque, QMR, Khan, MA, Mohammad, F. 2020. Antibacterial Functionalization and Simultaneous Coloration of Wool Fiber with the Application of Plant-Based Dyes *Journal of Natural Fibers* 17(3), 437-449
- Gedik, G, Yavaş, A, Avinç, O, Şimşek, Ö. 2013. Cationized Natural Dyeing of Cotton Fabrics with Corn Poppy (Papaver rhoeas) and Investigation of Antibacterial Activity. *Asian Journal of Chemistry* 25(15), 8475-8483
- Haji, A. 2013. Eco-Friendly Dyeing and Antibacterial Treatment of Cotton Cellulose Chem. Technol. 47 (3-4), 303-308
- Kamel MM, Helmy HM, Meshaly HM, Abou-Okeil A. 2015. Antibacterial Activity of Cationised Cotton Dyed with Some Natural Dyes J Textile Sci Eng 5: 180. doi:10.4172/2165-8064.1000180
- Correia, J, Rainert, KT, Oliveira, FR, Valle, RCSC, Valle, JAB. 2020. Cationization of cotton fiber: an integrated view of cationic agents, processes variables, properties, market and future prospects *Cellulose* 27, 8527–8550
- Arivithamani, N., Dev, V.R.G. 2017. Cationization of Cotton for Industrial Scale Salt-Free Reactive Dyeing Of Garments *Clean Techn Environ Policy* 19, 2317–2326
- Şahan, G, Demir, A. 2016. A Green Application of Nano Sized Chitosan in Textile Finishing *Tekstil ve Konfeksiyon* 26(4), 414-420
- Yavuz, G, Zille, A, Seventekin, N, and Souto, AP. 2017. Structural Coloration of Chitosan-Cationized Cotton Fabric Using Photonic Crystals. *IOP Conf. Series: Materials Science and Engineering* 254, 102012 doi:10.1088/1757-899X/254/10/102012
- Altınkaya E. 2021. Development of Multifunctional Bio-Based Cotton Composite *Tekstil ve Konfeksiyon* 31(4), 264-273
- 23. Arık, B, Seventekin, N. 2011. Evaluation Of Antibacterial and Structural Properties Of Cotton Fabric Coated By Chitosan/Titania and Chitosan/Silica Hybrid Sol-Gel Coatings *Tekstil ve Konfeksiyon* 2, 107-115
- Brahma, S., Islam, M.R., Shimo, S.S., Dina, R.B., 2019. Influence of Natural and Artificial Mordants on the Dyeing Performance of Cotton Knit Fabric with Natural Dyes *IOSR Journal of Polymer and Textile Engineering* 6(1), pp.1-6.
- 25. Karabulut, K. 2015. Giving Color, UV Protection and Antibacterial Activity In A Single Step To Knitted Cotton Fabrics Via Dyeing With Natural Dyes, MSc. Thesis, Namık Kemal University Graduate School of Natural and Applied Sciences, Tekirdağ, p.39.



- 26. ISO 13938-2:2019: Textiles Bursting properties of fabrics- Part 2: Pneumatic Method for Determination of Bursting Strength and Bursting Distension
- 27. TS EN ISO 105-C06: Textiles Tests for Colour Fastness Part C06: Colour Fastness to Domestic and Commercial Laundering
- 28. TS EN ISO 105-E04: Textiles Tests for Colour Fastness Part E04: Colour Fastness to Perspiration
- 29. TS EN ISO 105-X12: Textiles. Tests for Colour Fastness Colour Fastness to Rubbing
- 30. Sancar Beşen, B. 2020. Tea Tree Oil/Ethyl Cellulose Microcapsule Loaded Antimicrobial Textiles AATCC Journal of Research 7(2), 1-6

