ASSSESSMENT OF GROUNDWATER METAL-METALLOID CONTENT USING GEOSTATISTICAL METHODS IN KARABAĞLAR POLJE (MUĞLA, TURKEY)

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Research Article

ABSTRACT

In this research, heavy metals (Ti, Cr, Mn, Fe, Co, Ni, Cu, Zn, Mo, Cd, Ba, Pb) and a metalloid (As) sampled from 84 wells used for drinking water and irrigation in Karabağlar Karstic Polje (Muğla, Turkey) were analyzed. The results were evaluated by different statistical methods in order to investigate the interaction between elements. Ti, Mn, Fe, Zn, Mo and Pb were detected in many wells. According to findings, the strongest correlation is between As and Cu (R=0.832). As-Ni (R=0.789) and Cu-Ni (R=0.776) are the other strong correlations. The relationship between these elements were also shown by Cluster Analysis (CA). With respect to CA, the closest proximity distance matrix are found for these three elements. Also, elements examined composed 3 main clusters in dendrogram created from CA result matrix. These clusters match up with the findings of Principal Component Analysis (PCA) too. PCA gathers all elements up in 3 components: As, Cu and Ni at the first component; Mo and Cd at the second; Cr and Ba at the third. The accumulation of the elements studied were investigated by Kruskal-Wallis and Mann-Whitney U tests. According to the results, the least and the most common elements are Cu and Ba, respectively. The differences in the amount of all elements except Ni and Mo were statistically significant (p<0.05). In addition, both the amount of the elements and some water quality parameters were compared with EPA (U. S. Environmental Protection Agency) and Turkish inland water quality classes and no hazardous situation has been found.

1. Introduction

Since surface waters have very low resistance to external factors, the use of groundwater in many parts of the world is increasing day by day. The fact that city network water cannot reach everywhere and that people can use groundwater free of charge by opening wells to their land increases the importance of underground water resources. For these reasons, groundwater has become an alternative to surface waters as a source of drinking water (Girish et al., 2013). People use groundwater daily and in field irrigation. This raises the importance of knowing the pollution status, especially heavy metal pollution and the quality of groundwater. Studies have shown that metal pollution in groundwater, particularly in urban areas, is often not noticed and remains confidential (Huang et al., 2014). Metal contamination of waters can occur both by natural means and by human origin. Many metals such as Cu, Fe, Zn, Mn and Ni are essential for the survival of life. Metals such as Cd, Pb and As are not essential, but they are highly poisonous even at low concentrations. However, whether or not it is essential or not, each metal has a toxic effect above a certain threshold value (Pavlovic et al., 2014). The accumulation properties, permanence and toxic effects of metals make them the most important common contaminants to the world’s genera (Okbah et al., 2014).

Karabağlar Plate (polje), which is selected as the study area, is located in Muğla province (Turkey) (Figure 1). The study area is about 40 km² and the altitude ranges from 606 to 717 meters except for the surrounding hills. Karabağlar Plate is an active settlement area located in the center of Menteşe district of Muğla province. Typical Mediterranean climate prevails in the region, with hot and dry summers and warm and abundant rains in the winters, with an average annual rainfall of 1185 mm which is above the Turkish average. These rains cause floods and material damage every year in the spring season. In addition to being one of the most important tourism centers of Turkey, Muğla also contributes very much to the regional economy through its wide agricultural land. The dense population and intensive agricultural activities in the
Figure 1- Geological map of Karabağlar Polje and the studied wells (MTA, 2002).
study area cause the underground waters to be used extensively in the direction of irrigation and domestic purposes and cause undesirable pollution to be found in the groundwater. For this reason, the importance of determining and regularly monitoring the metal content of the groundwater in the region is increasing. Because these elements accumulate intensively in the plants that grow on agricultural land, which is irrigated with heavy metals contaminated waters. An example is the arsenic pollution that occurs in rice fields in Bangladesh (Alam et al., 2003, Polizzotto et al., 2013). This accumulation in the plants poses a danger to all living beings consuming these plants (Kafadar and Saygıdeğer, 2010). Determination of the amount of metal in the region was carried out by Yıldıztekin and Tuna (2011). This study was conducted seasonally at 20 wells in February-November 2006 and it was researched whether these wells were suitable for irrigation purposes. Yıldıztekin and Tuna (2011) have shown that the wells investigated do not pose a risk for irrigation, but the importance of long run monitoring including the other wells, has been emphasized.

The studied wells which are water samples taken from are shallow and the depths vary between 8-20 meters. The geology of the study area is given in figure 1 and all wells sampled are within the Quaternary alluvium unit. The thickness of this unit varies between 80 and 100 m (Atalay, 1980). Alluvial deposits cover the Jurassic Yılanlı Formation (Jkmu) in much of the land. Kurttaş (1997) stated that this formation consists of dolomite-dolomitic limestone and limestone. It is estimated that the Yılanlı Formation is the most important karstic unit in the region and that the most important karstic structures in the region have developed in this unit (Ekmekçi et al., 2012; Açıkel, 2003). As a result of the field studies, water sinks were determined and it was also determined that these structures drained the flood waters accumulated in the field during the high flow period in the spring to the karst system (Figure 1). Ekmekçi et al. (2012) and Açıkel (2003) have formed a detailed hydrogeological conceptual model of the region and the broader region with the studies they have conducted and have shown that the studied area is in the drainage basin of Gökova sources.

Tertiary-Miocene Köprüçay formation (Tk) is found in the south of the study area. Ekmekçi et al. (2012) stated that this formation consists of conglomerate and limestone formed by very large limestone pebbles bonding with a carbonate cement. Tertiary-Pliocene Yatağan formation (Tya) is found in the northwest of the study area. This unit consists of sandstone, conglomerate, siltstone, marl, claystone, limestone and coal deposits (Ekmekçi et al., 2012).

Examination of metal content of groundwater, investigation of the interaction between the metals by different statistical methods, determination whether the current situation poses a risk in terms of ecosystem and public health and determination of possible sources of pollution were aimed in the present study carried out in Karabağlar Polje, Muğla.

2. Method

Within the scope of the field study conducted in April 2013, water samples were taken from 84 different wells to perform chemical analyzes. In addition, the specific electrical conductance (SPC), total dissolved solid (TDS), salinity, pH and ammonium (NH₄-N) values of the samples were determined.

2.1. Chemical Analyzes

Elemental analyzes were performed for ⁴⁷Ti, ⁵²Cr, ⁵⁵Mn, ⁵⁶Fe, ⁶⁰Co, ⁶⁰Ni, ⁶⁵Cu, ⁶⁶Zn, ⁷⁵As, ⁹⁵Mo, ¹⁰¹Cd, ¹³⁷Ba and ²⁰⁸Pb by using ICP-MS (Inductively Coupled Plasma – Mass Spectrometer). The values of ⁴⁷Ti, ⁵⁵Mn, ⁹⁰Fe, ⁶⁶Zn, ⁹⁵Mo and ²⁰⁸Pb were not included in the results because they were below the detection limits in most of the analyzes performed. The detection limits for the elements are 0.1, 0.5, 0.06, 0.9, 0.1, 0.1, 0.01, 1.4, 1.2, 0.04, 0.04, 3 and 0.04 ppb, respectively. ICP-MS measurements were performed at Bilkent UNAM Laboratories with the devices; Thermo Scientific X-Series II and Cetac Axs-260 Autosampler. Water samples were first filtered (pore size 0.22 µm) and then acid treated (%65 HNO₃) to both increase mobility and solubility of elemental ions and equate acidity of the samples to the medium of the instrument (which is about 2% HNO₃). Ultrapure distilled water was used. High Purity Standards branded QCS-27 series standard reference material with 27 elements was used in order to preparing of calibration curves. Calibration curves for all elements were determined as the correlation coefficient is higher than 0.99 by taking account of the element concentrations of samples and keeping the searching
metal concentration on the calibration curve. 10 ppb 
$^{209}\text{Bi}$ was used as internal standard. Number of main 
run was assigned 3 as the analyses parameters. Sample 
drawing and washing periods were decided as 60 
seconds by the consideration of tubing length.

2.2. Statistical Analyzes

All the statistical analyses were carried out by 
SPSS 19.0 (IBM, USA).

2.2.1. Comparison of Quantity Difference Between 
Metals (Mann-Whitney U)

This test was used to determine which of the 
studied metals in the region had accumulated more. 
Shapiro-Wilk test was used to determine whether the 
data normally distributed or not. No data appeared 
to be normally distributed. On this, nonparametric 
Kruskal Wallis and Mann-Whitney U tests were 
preferred in the comparison of the averages (Tunca et 
al., 2013a).

2.2.2. Correlation Analysis

Correlation Analysis was used to determine 
whether there were correlations between amounts of 
metal quantity change in all samples collected during 
the study. Spearman’s Correlation Analysis was chosen 
as the correlation analysis to be applied, because it was 
determined that the data were not normally distributed 
with the Shapiro-Wilk Test (Tunca et al., 2013b).

2.2.3. Principal Component Analysis (PCA)

PCA, one of the most frequently used factor 
analyzes, is used to collect data in common 
components, depending on the relationships of the data 
with each other. For his study, this means collecting 
elements whose variabilities are statistically similar 
to each other in a common component. The PCA 
applied within the scope of this study was performed 
according to Varmuza and Filzmoser (2009) method.

2.2.4. Cluster Analysis (CA)

CA is an analysis that classifies similar features in 
dense data by proximity matrix. Moreover, the created 
dendrogram makes it easier to understand the relations 
visually. From here, it was desired to cluster samples 
according to the variations of elements quantities. 
In the proximity matrix, the shorter the distance 

between the elements, it means that the variations 
in the amounts of those elements are so similar. The 
analysis was carried out with the “Z-score” correction 
in the Euclidean distance according to Ward’s method 
(Lopez et al., 2004).

3. Results and Discussion

In this study carried out at the Karabağlar Polje 
in Muğla province, water samples from a total of 84 
wells were taken for metal-metalloid analysis and 
some principal parameters (SPC, TDS, salinity, pH and 
ammonium) of these wells were determined (Figure 
2). The metal-metalloid values of 23 from these wells 
were below the measurement detection limits and 
were therefore excluded from the study. In this study; 
$^{47}\text{Ti}$, $^{52}\text{Cr}$, $^{55}\text{Mn}$, $^{56}\text{Fe}$, $^{59}\text{Co}$, $^{60}\text{Ni}$, $^{65}\text{Cu}$, $^{66}\text{Zn}$, $^{75}\text{As}$, $^{95}\text{Mo}$, 
$^{111}\text{Cd}$, $^{137}\text{Ba}$ and $^{208}\text{Pb}$ parameters were determined. 
$^{47}\text{Ti}$, $^{55}\text{Mn}$, $^{56}\text{Fe}$, $^{66}\text{Zn}$, $^{95}\text{Mo}$ and $^{208}\text{Pb}$ were not included 
because their values were below the detection limits 
for the majority of the wells investigated. The spatial 
distributions of the assessed elements and the water 
quality parameters are given in figure 3.

Statistical relationships of elements with each 
other were investigated by Correlation Analysis, CA 
and PCA. These are generally very reliable methods 
which are frequently used in studies related to both 
the environment and the earth sciences, to clarify the 
relationships between variables (metal-metalloid in 
this case). Therefore, they are frequently preferred 
in many studies (Tsai et al., 2003; Gergen and 
Harmanescu, 2012; Pandey et al., 2014).

According to Correlation Analysis, the strongest 
correlation is between As and Cu (R=0.832) (Table 
1). From these elements, arsenic can be mixed with 
groundwater by natural processes, pesticides, industrial 
activities and industrial wastes. As is carcinogenic to 
humans as well as can lead to chronic-acute diseases, 
kidney and liver disorders and anemia (Fatrorini et 
al., 2006; Thorsen et al., 2009; Rodriguez-Sosa et al., 
2013). Copper found in groundwater can be originated 
from metal coatings, industrial & domestic waste and 
mining. Cu is essential for all known organisms and 
serves as the cofactor for many different enzymes 
(Franco et al., 2009). It even enters the structure of the 
hemocyanin, the oxygen carrier protein, in the 
crustaceans (Tunca et al., 2013a, b). Cu can cause 
stomach, intestinal, liver, kidney and anemia diseases. 
Although copper is an extremely vital element, there
are many studies showing toxic effects in different organisms; in mice (Wang et al., 2014), in fungus (Klimek and Niklinska, 2007), in bacteria (Wang et al., 2009), in aquatic floating plants (Üçüncü et al., 2013), in rooted plants (Wodala et al., 2012), in crustaceans (Luis Gama-Flores et al., 2009), in mollusks (Ramakritinan et al., 2012) and in mammals (Matos et al., 2010).

Correlation of As-Ni is also very strong (R=0.789). Ni is one of the vital elements like Cu (Serafim et al., 2012). Nickel can be found naturally in soil, groundwater and surface water, and is used in the coating of stainless steel and alloy products. Nickel can be toxic to mammals and can lead to heart and liver disorders (Gathwan et al., 2013).

Other strong correlations were found between Cu-Ni (R = 0.776) and Mo-Cu (R = 0.726). Besides these, moderate correlations were observed between Mo-As and Mo-Ni. Molybdenum is not only naturally found in low concentrations in groundwater, but it can also cause acute diseases in humans, although it is one of the vital elements that can be sourced from long-term domestic waste storage areas, mine tailings and sewerage (Alonso et al., 2004). But its role in metabolism and the interaction with other elements are not yet fully understood.

![Figure 2a](image-url) Figure 2a- Analysis and in-situ measurement results; The amounts of 75As, 137Ba, 111Cd, 52Cr, 65Cu and 95Mo.
Figure 2 b- Analysis and in-situ measurement results; The amount of $^{60}$Ni and the values of water quality parameters; ammonium, SPC, TDS, salinity and pH.
Figure 3 a- Spatial display of analysis and in-situ measurement results; The values of $^{75}$As, $^{137}$Ba, $^{111}$Cd, $^{52}$Cr, $^{65}$Cu and $^{95}$Mo.
Figure 3 b: Spatial display of analysis and in-situ measurement results; The values of 60Ni and water quality parameters; ammonium, SPC, TDS, salinity and pH.
These strong correlations between Cu, Ni and As observed in Correlation Analysis are also noted in other methods in which relations between elements are investigated. The Euclidean distance between these 3 elements is closest to each other in the proximity matrix in CA (Table 2). However, unlike the Correlation Analysis, the nearest in the matrix, that is, the strongest relationship, appears between As and Ni (As-Ni=5.154 Euclidean distance). Then Cu-As (5.357 Euclidean distance) and Ni-Cu (5.762 Euclidean distance) stand out.

PCA findings also support Correlation Analysis and CA outcomes (Figure 4 and Table 3). The elements assessed in this study were collected in 3 different groups according to PCA results. These 3 components cover 77.89% of all data. The first component contains Cu, Ni and As, while the second component contains Mo and Cd. Cadmium dissolves in rocks with acidic water and can mix with groundwater at low concentrations. Other sources are industrial-mining waste, metal waste, water pipes, batteries and paints. Cd is one of the elements that are not essential for living things, which can cause liver, kidney, anemia and hypertension diseases in humans (Gallego et al., 2012). In addition, EPA has incorporated the Cd into the B1 class, which includes possible carcinogens for humans. The third component consists of Cr and Ba. The source of chromium content in groundwater can be represented by old mining operations, fossil fuel consumption and cement plant waste. Cr is an element which can be found in different valences from -2 to +6 but trivalent and hexavalent valences are more common (Ergül-Ülger et al., 2014). Especially, hexavalent form is extremely important. Hexavalent Cr is not vital for the living creatures and has a high toxic effect (Bankar et al., 2009). Chromium is mutagenic and carcinogenic to humans and can lead to liver, kidney, internal bleeding, respiratory, skin inflammation and ulcer disorders (Mishra and Doble, 2008). Barium is an element that can be found naturally in some limestones, sandstones and soils, and whose concentration in the soil can be relatively higher than other metals (Suwa et al., 2008). It does not have a known biological function, but certain forms may cause heart, digestive and hypertension disorders, while showing a highly toxic effect (Llugany et al., 2000).

Table 1- Correlations of elements.

<table>
<thead>
<tr>
<th></th>
<th>As</th>
<th>Ba</th>
<th>Cd</th>
<th>Cr</th>
<th>Cu</th>
<th>Mo</th>
<th>Ni</th>
</tr>
</thead>
<tbody>
<tr>
<td>As</td>
<td>1.000</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Ba</td>
<td>0.049</td>
<td>1.000</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cd</td>
<td>-0.003</td>
<td>0.067</td>
<td>1.000</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cr</td>
<td>0.011</td>
<td>0.399**</td>
<td>0.312*</td>
<td>1.000</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cu</td>
<td>0.832**</td>
<td>-0.013</td>
<td>0.252</td>
<td>-0.048</td>
<td>1.000</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Mo</td>
<td>0.679**</td>
<td>-0.003</td>
<td>0.412**</td>
<td>0.079</td>
<td>0.726**</td>
<td>1.000</td>
<td></td>
</tr>
<tr>
<td>Ni</td>
<td>0.789**</td>
<td>0.259</td>
<td>-0.093</td>
<td>0.092</td>
<td>0.776**</td>
<td>0.606**</td>
<td>1.000</td>
</tr>
</tbody>
</table>

* Correlation is significant at 0.05 (2-tailed).
** Correlation is significant at 0.01 (2-tailed).

Table 2- CA proximity matrix.

<table>
<thead>
<tr>
<th></th>
<th>As</th>
<th>Ba</th>
<th>Cd</th>
<th>Cr</th>
<th>Cu</th>
<th>Mo</th>
<th>Ni</th>
</tr>
</thead>
<tbody>
<tr>
<td>As</td>
<td>0.000</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Ba</td>
<td>9.827</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cd</td>
<td>10.284</td>
<td>10.167</td>
<td>0.000</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cr</td>
<td>11.280</td>
<td>9.656</td>
<td>9.552</td>
<td>0.000</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cu</td>
<td>5.357</td>
<td>10.628</td>
<td>9.328</td>
<td>11.526</td>
<td>0.000</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Mo</td>
<td>8.238</td>
<td>10.257</td>
<td>6.877</td>
<td>9.967</td>
<td>8.405</td>
<td>0.000</td>
<td></td>
</tr>
<tr>
<td>Ni</td>
<td>5.154</td>
<td>9.375</td>
<td>11.338</td>
<td>10.836</td>
<td>5.762</td>
<td>8.841</td>
<td>0.000</td>
</tr>
</tbody>
</table>

Table 3- PCA rotated component matrix.

<table>
<thead>
<tr>
<th></th>
<th>Components</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>1</td>
</tr>
<tr>
<td>As</td>
<td>0.911</td>
</tr>
<tr>
<td>Cu</td>
<td>0.847</td>
</tr>
<tr>
<td>Ni</td>
<td>0.920</td>
</tr>
<tr>
<td>Cd</td>
<td>0.916</td>
</tr>
<tr>
<td>Mo</td>
<td>0.803</td>
</tr>
<tr>
<td>Ba</td>
<td>0.822</td>
</tr>
<tr>
<td>Cr</td>
<td>0.674</td>
</tr>
</tbody>
</table>
The PCA results also coincide with the CA results. In the dendrogram formed by CA matrix, it is seen that the elements constitute 3 main clusters (Figure 5). The element distributions of these clusters are similar to those of PCA.

The accumulation of the elements has also been revealed (Figure 6). Nonparametric Kruskal Wallis and Mann-Whitney U tests were used to compare the amount of the elements, since the data was not normally distributed. This test combination is one of the most frequently used non-parametric comparison tests (Blazewicz et al., 2013; Farmaki et al., 2014). According to the results, Cu is the least common element and Ba is the most common element in the study area (p <0.05). The amounts of all the elements except Ni and Mo are statistically significant. There is no statistically significant difference only between the amounts of Ni and Mo (p> 0.005).

When the relations of the elements studied in this paper are evaluated as holistic, the highest statistical relationship was found between As and Cu, followed by As-Ni, Ni-Cu, Cd-Mo, Cu-Mo, As-Mo, Mo-Ni and Cr-Ba. When the statistical relationships and the spatial distributions shown in figure 3 are evaluated together in order to determine the common sources of the elements studied, the main sources of the elements in the region are the mine and the industrial activities. In addition, household waste, agricultural activities and rock dissolution can be considered as potential sources.

The values of all metal and metalloids studied do not exceed EPA’s limits for drinking water (EPA, 2012). The total dissolved matter (TDS), one of the water quality parameters, is compared by considering EPA and Turkish inland water quality classifications. The upper limit determined by EPA as 500 mg/l corresponds to the 2nd class water quality group from 4 different level found in the Turkish inland water quality classification. (ÇOB, 2004). The TDS values of the water in the examined wells were generally less than 500 mg/l and were included in the 1st and 2nd grade Turkish quality groups. A few values above 500 mg/l are point-based and do not represent a specific area. The TDS measurements determined above this value are in 3rd grade group according to the Turkish inland water quality classifications. Again according to this classification, no wells in 4th grade class were found in the study area. The range determined by EPA for pH is 6.5-8.5. This range corresponds to the 1st and 2nd degree quality groups according to the Turkish inland water quality classifications. Only 3 of the wells
examined in the study are not within this range. In terms of ammonium, most of the water samples are in the 1º degree class according to Turkish classification. Only a few of them are at the limits of the 2º degree class. No water samples in 3º or 4º degree classes were found. None of the parameters examined constitute health risk by the standards of the United States and Turkey and no pollution could be mentioned for the groundwater in the region. In addition, the quantities of the parameters studied do not exceed the natural limits for the water (Hem, 1985).

Yıldıztekin and Tuna (2011) obtained similar findings in the same study area. However, due to the fact that Yıldıztekin and Tuna (2011) did not specify the locations of the sampling wells in the study they performed, time-based and well-based comparisons could not be done, only general comparisons were performed.

Sampling and in situ measurements made in April, the rainy season for the region, are more likely to result in rare values due to the high water content from the rainfall. For this reason, it is recommended that the same field studies be carried out again in August, which is an arid period for the region, in order to be able to monitor and evaluate each parameter more accurate.

If the values of the elements and the other parameters examined in the study increase to the level that can be considered as “pollution”, the source or sources of this pollution should be determined urgently and necessary precautions should be taken. In this context, actively operated waste water treatment plant, cement plant, sand and gravel quarry, industrial site, swimming pool and agricultural activities can be predicted as potential sources of pollution. Furthermore, it is important that the region be monitored in a stable manner, as the probability of a certain and continuous source of pollutants will increase pollution over time.

4. Conclusion

Within the scope of the study, heavy metal and metalloid analyzes of 84 water wells used for drinking and irrigation in Karabağlar Polje (Muğla, Turkey) were made and water quality parameters were measured. No evidence has been identified to threaten human health and ecosystem in terms of metals and metalloids studied. Water quality parameters measured in place do not indicate the presence of any pollution in the area. However, the high interaction and correlations found in the context of statistical evaluation of the relationship between metal and metalloids have shown that some elements may have common sources.

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