RESEARCH ARTICLE

The Protective Effect of Amino-guanidine, an Inducible Nitric Oxide Synthase Inhibitor, on Aluminium Sulphate Neuro-toxicity in the Rat (Wistar albino) Cerebellar Purkinje Cells: Stereological Study

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Abstract

Objective: Aluminium (Al) is quite abundant in nature and humans are frequently exposed to Al in daily life. Aluminium salts can exist in different forms and they may have toxic impacts on several tissues including brain. In this study, potential preventive effects of amino-guanidine (AG) (100 mg/kg, i.p.), an inducible nitric oxide synthase inhibitor, on neuron damage to be created by aluminium sulphate (3 mg/kg, i.c.v.) in cerebellar Purkinje cells were determined.

Methods: 24 female Wistar albino rats were divided into 4 groups with 6 rats in each: Control (C), Sham (S), Aluminium sulphate ($Al_2(SO_4)_3$), Aluminium sulphate + Amino-guanidine ($Al_2(SO_4)_3$ +AG). A single aluminium sulphate (3 mg/kg) dose dissolved in 0.9% NaCl was injected intracerebroventricularly to aluminium sulphate and aluminium sulphate + amino-guanidine groups at the beginning of experiments. Following aluminium sulphate injection, amino-guanidine (100 mg/kg) dissolved in distilled water was injected to aluminium sulphate + amino-guanidine group intraperiteonally for 15 days. Nothing was administered to control group, a single dose of 0.9% (3 mg/kg, i.c.v.) sodium chloride (NaCl) was administered to sham group at the beginning of experiments. Cerebellum tissues of the rats were removed 15 days after treatments and they were assessed histopathologically and stereologically.

Results: Stereological optic fractionation method revealed cerebellar total number of Purkinje cells as 417615±16238,8 in control group; 378650±20171,6 in Sham group; 272945±15499,5 in Aluminium sulphate group; 324581±16324,8 in Aluminium sulphate + Amino-guanidine group.

Conclusion: It was concluded based on present findings that amino-guanidine reduced aluminium induced Purkinje cell loss through nitric oxide synthase (NOS) inhibition.

Key words: Aluminium, Amino-guanidine, Cerebellum, Purkinje cell, Stereology

	Introduction
Address for correspondence/reprints:	Chemical pollutants not only pollute
Burcu Demirel Yılmaz	environment, but also they have long-standing impacts on cellular development (Önger et al., 2011). Aluminium (Al) always exists in environment
Telephone number: +90 (452) 611 25 55 / 4905	and humans are continuously exposed to aluminium (Yavuz et. al., 2013). It doesn't have a known
E-mail: bioburcudemirel@gmail.com	biological function and it's the third abundant element in nature (Oğuz et al., 2012). Al can be
DOI: 10.19127/mbsjohs.322015	found in broad range of items from drugs to tools, from kitchenware to electric industry, from vehicles

to cosmetics (Yavuz et. al., 2013). Al can get into the human body through oral, penetration, inhalation, skin etc. (Yokel et al., 2008). Just because of abundance in nature, it can accumulate in tissues of organisms to certain extend (Oğuz et al., 2008), may have neurotoxic impacts resulting in Alzheimer and Parkinson-like neurological disorders (Kamal and Kamal, 2013).

Clinical studies with animals revealed that brain was a significant target organ in Al neurotoxicity (Kamal and Kamal, 2013). Al can accumulate in all sections of the brain and may have maximum accumulation in hippocampus. Al accumulation in tissues induces the formation of reactive oxygen species and then these species result in protein, lipid and DNA oxidation. Neural membranes of central nervous system are quite rich in polyunsaturated fatty acids, have low antioxidant enzyme levels and significant quantities need of oxygen for metabolism. Therefore, they are quite more prone to oxidative damage than the other systems. Al has a cytotoxic impact on brain through inhibition of Ca2+ ATPaz responsible for sustaining quite low Ca₂+ levels in cells by pumping Ca₂+ in brain out of the cells. Increase in neural Ca2+ activates various protease enzymes of caspase family and may result in irreversible neural damage. Al also increases the quantities of glutamate, an excitatory neurotransmitter existing in about 40% of synapse in brain. When the Al around a neuron reached to a certain level, it annihilates the neuron through apoptosis. An uncontrolled increase in glutamate quantity of synaptic gap activates N-methyl Daspartic acid (NMDA), increases Na+ and Ca₂+ ion concentrations along the cell membrane and aggravate neurological damage (Cabus, 2012).

Together with recently increased atmospheric pollution, nitric oxide (NO) is another remarkable compound. NO is known as a basic precursor molecule and called as a free radical molecule because of unpaired electron. While the other free radicals are harmful to cells at any concentrations, NO play a role in quite significant physiological functions at low concentrations. However, excessive and uncontrolled NO synthesis may be harmful for cells (Özgüneş and Atasayar, 2009). In case of excessive production, NO confronts us as a neurotoxin in various nervous system diseases (Satarug et al., 2000). Nitric oxide (NO) is synthesized from L-arginine by nitric oxide synthase (NOS) (Buraimoh et al., 2014). NOS exist in brain tissues of humans and animals at varying concentrations (Türköz and Özerol, 1997). There are three NOS isoforms in brain as of endothelial

(eNOS), neural (nNOS) and inducible (iNOS) (Stevanović et al., 2010). iNOS is an enzyme able to produce NO to a level with toxic effects (Gross and Volin, 1995). Cytokines and stimulant substances induced by acute inflammatory changes may also contribute to iNOS enzyme activation just as in immunological or inflammatory stimulations and may produce toxic NO levels which can be expressed as nanomole throughout subsequent couple days of pathological stimulation and ultimately aggravate the damage which may end up with cell death (Önger et al., 2011).

Amino-guanidine (AG), used to reduce toxic impacts of Al on Purkinje cells, structurally resembles to L-Arginine amino acid, inhibits iNOS and thus result in reduced NO formation (Budavari et al., 1989). It was reported in a previous study that AG prevented inflammation in hippocampus as a selective iNOS inhibitor and reduced neuron damage (Anaeigoudari, 2016). It was also reported that AG had scavenging activity in various tissue damage models through scavenging hydrogen peroxide (H₂O₂) derivative hydroxyl radicals (OH-) produced from NO and superoxide (O₂-) (Polat et al., 2006). In this sense, AG prevents lipid peroxidation and formation of reactive oxygen species (ROS) and thus reduces toxicity through the impacts like as an antioxidant agent (Babu et al., 1995).

Cerebellum, over which toxic effects of Al were investigated, is one the most complex sections of central nervous system of mammalian (Tunc et al., 2007). Nearly 50% of all neurons of the brain are located in cerebellum, which takes up only 10% of the total brain volume and receives nearly 200 million afferent fibers (Eweka and Om'Iniabohs, 2007). Cerebellum is responsible for the control of motor movements through comparing information coming to brain from various receptors in periphery with the responds of brain to this information. While performing this function, Purkinje cells are the only output of cerebellum cortex. Therefore, Purkinje cells have a great place in cerebellum functions. These cells are also quite sensitive to alcohol toxicity and ischemia-like pathologic cases (Kozan et al., 2009). High number of NMDA receptors creating neurotoxic effects of NO in cerebellum also aggravates NOS activity (Sefil et al., 2009). It was reported in previous Al studies that with increasing NOS levels in brain tissue (Flora et al., 2003). Al created oxidative stress in cortex and cerebellum (Esparza et al., 2005) and had neurotoxicity on cerebellar Purkinje cells (Buraimoh et al., 2014).

Although brain is the primary target organ in Al toxicity, effects of Al on cerebellum were

investigated in limited number of studies. Therefore, the present study was designed to elucidate the preventive effect of amino-guanidine (AG) as a specific nitric oxide synthase (NOS) inhibitor against potential damage of Al to be created on cerebellum Purkinje cells through stereological optic fractionation method with systematic randomized sampling and unbiased counting.

Methods

Experimental Procedure

Experiments were conducted with Wistar albino rats grown in Laboratory Animals Implementation and Research Center of Ondokuz Mayıs University. About 200 \pm 250g 24 same-generation rats were selected and they were randomly divided into four groups with 6 rats in each. These groups were control (C), Sham (S), Aluminium sulphate (Al₂(SO₄)₃), Aluminium sulphate + Amino-guanidine (Al₂(SO₄)₃), Aluminium sulphate + Amino-guanidine (Al₂(SO₄)₃) +AG). In this study, 3 mg/kg of aluminum sulphate (Çabuş et al. 2014) and 100 mg/kg of aminoguanidine (Önger et al., 2011) were applied to experimental groups. Rats were placed in plastic cages at 20 \pm 22°C temperature, 50% relative humidity and 12 hours' light/dark periods. Feed and water was supplied ad libitum.

Chemicals and Method of Administration

Aluminium sulphate was supplied from Sigma-Aldrich as aluminium sulphate hydrate (in powder, pure form), Amino-guanidine was supplied from again Sigma-Aldrich as amino-guanidine hydrochloride (in powder, pure form).

Rats were weighted before the injections and they were anesthetized through intraperitoneal ketamine (100 mg/kg) and xylazine (10 mg/kg) administration (Sefil et al., 2009). Rats were than fixed to streotaxy device and their scalp was opened 2 cm with an electrical cautery (Ellman Surgitron) from the midsection along rostro-caudal direction. Tendon and fascia over cranium were removed to see Bregma clearly. As to comply with left lateral ventricle, 1 mm hole was opened at 2 mm lateral and 0.6 mm posterior of Bregma (Kozan et al., 2009). From there, Hamilton micro-injector was introduced and 2 µl aluminium sulphate solution was injected intracerebroventricularly (i.c.v.) to a depth of 4.2 mm at 0.5 µl/min flow rate and 3 mg/kg dose. Nothing was administered to control group. A single dose of 0.9% (3 mg/kg, i.c.v.) sodium chloride (NaCl) was administered to sham group at the beginning of experiments. A single aluminium sulphate (3 mg/kg, i.c.v. Al₂(SO₄)₃) dose dissolved in 0.9% NaCl was administered to Al2(SO4)3 and

Al₂(SO₄)₃+AG groups at the beginning of experiments. Amino-guanidine (AG) was injected to Al₂(SO₄)₃+AG group for 15 days at a dose of 100 mg/kg, i.p. After 15 days following the last injection, rats were perfused intracardially under ketamine (75 mg/kg, i.p.) and xylazine (10 mg/kg, i.p.) anesthesia. Rat brains were put into 10% neural-tamponed formalin solution for stereological examination.

Sampling via Optical Fractionation

Total number of Purkinje cell in the cerebellum was performed by the optical fractionation method, which is a combination of two stereological applications (fractionation and optical dissector) (Korkmaz et al., 1996; Korkmaz and Tumkaya, 1997; Korkmaz et al., 2000). Systematic random sampling was performed from cerebellum tissue immersed into paraffin in accordance with fractionation principles. The issue to be considered in sampling was to have a proper error coefficient (0.05 or less) and coefficient of variation (0.10). The calculation results below these two values prove the accuracy of the sampling process (Gundersen and Jensen, 1987). Present samplings were performed by considering error coefficients proper and of variation. In this coefficients direction. cerebellum tissues of the all group rats were sectioned from back to front at 30 µm intervals, leaving no tissue unused. Section sampling fraction in systematic random sampling was determined as (f1) 1/7 and the first cross-section based on randomness principle was taken randomly from the 1-4th sections, the other cross-sections were taken from the 7th section through by passing 6th section over the initially selected section. In this way, about 25-28 cerebellum sections were obtained from each rat. The selected sections after sampling were stained with cresyl violet. For section sampling fraction (F2), the approach in West et al., (1991) was tried to be applied. F2 was calculated by dividing small counting area (frame) with step area (X, Y step). Accordingly, F2 was calculated as $40x40\mu m^2/220x220\mu m^2 = 1/12=0.033$. Finally, cerebellum Purkinje cell layer was scanned within the range defined over X-Y axis with systematic random sampling. For cross-section thickness and optic dissector height measurements, 'micro-screw calibration' method developed by Korkmaz and Tümkaya (1997) was used. In this way, thickness sampling fraction (F3) was calculated for each rat separately by dividing average dissector height (hort) with average cross-section thickness (tort)

The Rat Cerebellar Purkinje Cells: Stereological Study

Cerebellum Purkinje cell counts and calculations

The method developed by Korkmaz et al. (2000) was used for Purkinje cell counts. For this method, Purkinje profile of each preparate was imaged with 10X objective, then acetate template including X, Y steps (large squares) and unbiased counting frames (small squares) were placed over the monitor (Fig 1 and 2). In this magnification, in systematic randomly sampled every seventh area, cross-section thickness (t) was determined with 100X objective (numerical aperture 1.25). In these areas, upper surface of the cross-section was focused and the location of pointer over the scale was determined. Then the lower surface of the cross-section was focused and the unit movement of the pointer with respect to initial position was recorded. The distance covered by a unit of scale connected to micro-screw along Z-axis (1-degree movement of micro-screw) corresponded to $0.27 \,\mu\text{m}$. Cross-sections were scanned in steps and with optic dissector, optically 11 units (2.7 µm, $0.27 \times 11 = 3 \ \mu m$) were moved downward from the upper surface of the cross-section at Purkinje cell count phase. Cross-section scanning was performed at this level. The parameters obtained in this study were placed in equation of N (Total) = ($\sum Q$ -) x (1/ F1) x (1/F2) x (1/F3) (West et al., 1991). In this way, average number of cerebellum Purkinje cells of each group was calculated.

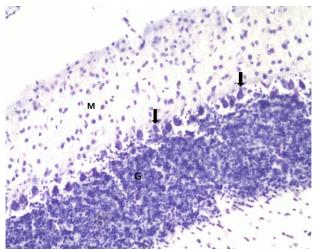


Figure 1. Photomicrograph of cerebellum from control group. G: Granular layer, M: Molecular layer, Purkinje cells (arrows), cresyl violet, 20x

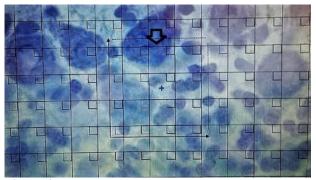


Figure 2. Systematic random sampling area fraction on the sampled section. X, Y steps (large squares) and unbiased counting frames (small squares), Dissector granule (arrow), cresyl violet, 100x.

Data Analysis

Statistical analyses were performed with SPSS (Statistical Package for Social Sciences for Windows) Version 20. One-Way ANOVA was used to compare treatment groups. Differences between the groups means was assessed with post hoc Tukey test.

Results

Average number of Purkinje cells present in cerebellum of rats is provided in Table 1. Coefficient of variation was calculated for each group (a value to be less than 0.10) to prove accuracy of the study. As compared to the control group (C), 34.6% cellular loss was observed in Al₂(SO₄)₃ group and 27.9% lose was observed in S group. In Al₂(SO₄)₃ +AG group, amino-guanidine reduced cellular loss by 22.3% as compared to C group and by 14.3% as compared to S group (Table 1). The differences in Purkinje cell counts of C and S groups and the differences in Purkinje cell counts of Al₂(SO₄)₃ groups were found to be significant (p<0.01).

Table 1. Average number of Purkinje cells in cerebellum, standard error of the mean (SEM) and coefficients of variance (CV).

Groups	Average number of Purkinje cells ± SEM	Coefficient variation (CV)
С	417615±16238,8	0,06
S	378650±20171,6	0,06
Al ₂ (SO ₄) ₃	272945±15499,5	0,06
Al ₂ (SO ₄) ₃ +AG	324581±16324,8	0,08

Post hoc Tukey test revealed that differences in Purkinje cell counts of the experimental groups were significant (p<0.01). One-Way ANOVA test revealed that $Al_2(SO_4)_3$ group was significantly different from C and S groups (p<0.01). There were significant differences also between $Al_2(SO_4)_3$ group and $Al_2(SO_4)_3 + AG$ group (p<0.01) (Fig 3).

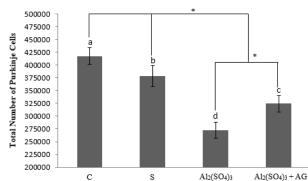


Figure 3. Cerebellum Purkinje cell counts for C, S, Al2(SO4)3 and Al2(SO4)3 + AG groups. a, b, c, d express different groups of post hoc Tukey test. * p<0.01 significance level used to compare the groups with one-way ANOVA.

Discussion

Accumulation and damages of different Al forms on target organ and in body tissues were reported in previous studies (Kutlubay et al., 2007a; Oğuz et al., 2008, 2012). Exposure to high Al levels results in neural degeneration characterized by programmed and selective neuron loss (Stevanović et al., 2010). Al has neurotoxic impacts through inhibiting DNA repair enzymes (Bharathi et al., 2008), destroying cell membrane integrity (Zatta et al., 2002; Stevanović et al., 2010), increasing production of reactive oxygen species (ROS) (Ogasawara et al., 2003) and reducing neurotransmitter biosynthesis (Cheng et al., 2014).

Al is a source of oxidative stress and results in

lipid peroxidation through the abundance of polyunsaturated fatty acids in cerebellum (Chaudhary et al., 2014). Al actualizes such impacts through increasing cerebral NOS levels (Bondy et al., 1998). In such cases, reduced forms of excessively produced NO increases. Excessive reduced NO reacts with superoxide (O2-) and thus increases proxy nitrite (ONOO-) and free radical formation. Increased free radicals eject electrons from cell membrane lipids and results in lipid peroxidation. Lipid peroxidation then accelerates cell deaths (Marangoz, 1996). Sharma et al. (2007) treated rats orally with 172.5 mg/kg-day aluminium chloride for 10 weeks and reported significant decreases in SOD, GSPx and CAT antioxidant enzyme levels and significant increases in lipid peroxidation levels. Similar results were also reported in another study carried out with aluminium lactate-treated rats (Ogasawara et al., 2003). There are evidences that Al induced apoptosis-mediated neural deaths. Cabuş et al. (2014) carried out a study on rats treated with aluminium sulphate (3 mg/kg i.p.) and assessed the neurons in stratum pyramidale layer of left-hemisphere with TUNEL method and reported significant decreases in number of neurons of aluminium-treated rats, then indicated aluminium as a neurotoxic agent and also indicated that neural death mechanisms might have been resulted from apoptosis. It was also showed that Al might result in cell damages in different tissues like thyroid follicles. It was reported in a previous study that 5% aluminium chloride supplementation of drinking water created significant damages on thyroid follicles of rats (p<0.05) (Aktaç and Bakar, 2002). Malekshah et al. (2005) reported that aluminium chloride treatments (150 mg/kg, i.p.) for 10, 11 and 12 days created anomalies in fetus and reduced body weight of pregnant rats. Al may also reduce DNA and RNA synthesis and thus inhibit protein synthesis (Darbre, 2006). Buraimoh et al. (2014) in a study investigating the effects of Al on Purkinje cells, treated rats with aluminium chloride (40 mg/kg) for 4 weeks and reported significant decreases in Purkinje cell counts of treated rats through neurodegeneration as compared to untreated control rats.

In present stereological study with systematic random sampling, there was 34.6% decrease in number of Purkinje cells in cerebellum of Al (3 mg/kg, i.c.v.) treated group as compared to control group. Such a case revealed that Al had neuro-toxic impacts on Purkinje cells and the case may be attributed to increased NOS production (Esparza et al., 2005; Kozan et al, 2009; Buraimoh et al., 2014). It was reported in previous studies that toxic impacts

of aluminium may reduce antioxidants like vitamin E (Kutlubay et al., 2007b), AG (Stevanović et al., 2010), taurine (Kozan et al., 2009). The AG used in this study to reduce the toxic impacts of Al was proved to reduce neurotoxic impacts resulted from increased aluminium sulphate-induced NOS levels (Önger et al., 2011). Sefil et al. (2009) reported that AG reduced iron-induced Purkinje cell loss from 25% to 12%. It was reported in another study carried out on Wistar albino rats that AlCl3 treatments increased NO production in CA1 region of hippocampus 3 hours after the treatments, subsequently initiated neuro degeneration process, but combined AlCl3 + AG treatments rapidly reduced NO production 3 hours after treatments (Stevanović et al., 2010). In another study carried out with AG, neuron damage resulted from excessive iNOS induced NO production was eliminated with AG (100 mg/kg, i.p., day) treatments (Lu et al., 2002).

It was observed in present study, investigating the preventive effects of AG against cerebellar Purkinje cell loss, that Al2(SO4)3 + AG treatments reduced Purkinje cell loss by 22.3% and such a reduction revealed the preventive effects of AG against neuron damage. These findings support the earlier findings of researchers indicating AG as a specific iNOS inhibitor and may have preventive effects against neurotoxicity (Eroğlu et al., 2008; Gökçe et al., 2011). Neuron preventive effects of AG are realized through inhibiting the formation reactive oxygen radicals (ROR), inhibiting lipid peroxidation (LPO) in cell and tissues, scavenging hydrogen peroxide derivative hydroxyl radicals and prevention from oxidant-induced apoptosis (Özgüneş and Atasayar, 2009).

Conclusion

It was concluded based on present findings that AG, also known as an iNOS inhibitor, had preventive effects against neuron damage in cerebellum Purkinje cells resulted from Al-induced increased NOS levels. Nowadays, the use of Al in many areas especially on a sectoral basis causes people to be directly or indirectly exposed to toxic effects of Al. Therefore, the present study shows that amino-guanidine can be used to reduce the toxic effects of Al, and it contributes to the literature in this issue. Ethics Committee Approval: All experiments were conducted in accordance with the guidelines for care and use of laboratory animals and protocols were approved by the local ethical committee on experimental animals (2014/14), Ondokuz Mayıs University, Samsun, Turkey.

Peer-review: Externally peer-reviewed.

Author Contributions: Concept HG; Design BB; Supervision HG, BDY; Materials HG, BB; Data collection and/or Processing BB; Analysis and/or Interpretation BB, BDY; Literature Review BE, BDY, BB; Writing BDY; Critical Review BE, BDY. Conflict of Interest: No conflict of interest was declared by the authors.

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