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# Packing Chromatic Number of Bismuth Tri-iodide and First Type Nanostar Dendrimers 

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#### Abstract

The packing chromatic number $\chi_{\rho}(G)$ of a connected graph $G$ is the smallest number $m$ for which a function $g: V(G) \rightarrow\{1,2, \ldots, m\}$ exists, such that if $g(a)=g(b)=j$, then $d(a, b)>j$. Here, we determine the packing chromatic numbers of bismuth tri-iodide and first type nanostar dendrimers.


## 1. INTRODUCTION

A packing $m$-coloring of a connected graph $G$ is a function $g: V(G) \rightarrow\{1,2, \ldots, m\}$ such that if $g(a)=g(b)=j$, then $d(a, b)>j$. The packing chromatic number $\chi_{\rho}(G)$ of $G$ is the smallest number $m$ for which $G$ has packing $m$-coloring. The idea of packing coloring comes from the zone of frequency assignment in wireless networks and was proposed by Goddard et al. [1] underneath the title broadcast coloring. The name packing chromatic number was coined by Bresar et al. [2].

Dendrimers are novel artificial polymeric structures having extended bodily and chemical properties due to their special three dimensional architecture and they also have a nicely defined shape, size and molecular weight. These are adaptable with drug moieties as properly as bioactive molecules like DNA, Liquaemin and different polyanions [3,4].

A crystal structure is made out of a unit cell, a set of atoms arranged particularly; repeated on a lattice periodically in three dimensions. The crystal structure plays a crucial role in deciding several physical properties, like electronic band structure, cleavage, and optical transparency [5]. Here, we determine the packing coloring of first type nanostar dendrimers and bismuth tri-iodide.

## 2. SOME FAMILIES OF NANOSTAR DENDRIMERS

The first type of nanostar dendrimers $D_{1}(n)$ are shown in Figures 1,2 and 3, where $n$ is the stage of growth. The order and size of $D_{1}(n)$ nanostar dendrimers are $24+36(n-1)$ and $27+42(n-1)$ respectively. In the following Theorems, by $c(v), d(u, v)$ and $d(u,\{v, w\})$ we mean the color of a vertex $v$, the distance between vertices $u$ and $v$, and the distances between $u$ and $v$ and $u$ and $w$.

Proposition 2.1. [1] Let $H$ be a subgraph of $G$. Then $\chi_{\rho}(H) \leq \chi_{\rho}(G)$

Proposition 2.2. [1] For a cycle graph $C_{n}$ of length $n$, we have

$$
\chi_{\rho}\left(C_{n}\right)=\left\{\begin{array}{l}
3 \text { when } n \text { is divisible by } 4 \\
4 \text { otherwise }
\end{array}\right.
$$

Theorem 2.3. For $D_{1}(1), \chi_{\rho}\left(D_{1}(1)\right)=5$.
Proof: Assume for logical inconsistency this is not valid. Then $\chi_{\rho}\left(D_{1}(1)\right)=4$. Obviously there must be a vertex to which color 1 or 2 or 3 or 4 is assigned. Label $D_{1}(1)$ as it is shown in Figure 1(a). Let $x$ be a vertex in $D_{1}(1)$ with $c(x)=2$ or $c(x)=1$ or $c(x)=3$ or $c(x)=4$. Now we will discuss these four cases in detail.

Case 1: For $c(x)=1$, note that each of the neighbors of $x$ have pairwise various colors and let $c\left(u_{1}\right)=2$, $c\left(v_{1}\right)=3$ and $c\left(w_{1}\right)=4$. Now vertices $u_{2}$ and $u_{6}$ can't be colored except if $c\left(u_{2}\right)=c\left(u_{6}\right)=1$. Since $d\left(w_{1},\left\{u_{3}, u_{5}\right\}\right)=4$ and $d\left(v_{1},\left\{u_{3}, u_{5}\right\}\right)=4$, either $u_{3}$ or $u_{5}$ can be colored 3. But if $c\left(u_{3}\right)=3, u_{5}$ can't be colored except if a color more than 4 is used. Similarly, if $c\left(u_{5}\right)=3, u_{3}$ can't be colored except if a color more than four is used.

Case 2: For $c(x)=2$, let $c\left(u_{1}\right)=c\left(v_{1}\right)=c\left(w_{1}\right)=1$. Consider the vertices $u_{2}, u_{6}$ and $v_{2}$. Since $c(x)=2$ and $d\left(v_{2},\left\{u_{2}, u_{6}\right\}\right)=4$, vertices $u_{2}$ and $v_{2}$ can be colored 3 , which also implies that $c\left(w_{2}\right)=3$ because $d\left(w_{2},\left\{u_{2}, u_{6}\right\}\right)=4$. Now for vertices $u_{6}, v_{6}$ and $w_{6}$, the only choice is to get color 4 . Without loss of all inclusive statement, let $c\left(u_{6}\right)=4$. Then $v_{6}$ and $w_{6}$ can't be colored except if a color more than four is used.

Case 2.1: For $c(x)=2$, let $c\left(v_{1}\right)=3$ and $c\left(w_{1}\right)=4$ or vice versa. This forces $c\left(u_{1}\right)=1$. Since $d\left(v_{1},\left\{u_{2}, u_{6}\right\}\right)=3, d\left(w_{1},\left\{u_{2}, u_{6}\right\}\right)=3$ and $d\left(x,\left\{u_{2}, u_{6}\right\}\right)=2$, vertices $u_{2}$ and $u_{6}$ can't be colored except if a color more than 4 is used.

Case 2.2: For $c(x)=2$, let $c\left(u_{1}\right)=c\left(v_{1}\right)=1$. Then $c\left(w_{1}\right) \in\{3,4\}$. Since $d\left(w_{1},\left\{v_{2}, v_{6}\right\}\right)=3$, if $c\left(w_{1}\right)=3$, vertex $v_{6}$ can be colored 4 but $v_{2}$ needs a color greater than 4 . Similarly, if $c\left(w_{1}\right)=4$ instead of $c\left(w_{1}\right)=3$ , vertex $v_{6}$ can be colored 3 but $v_{2}$ needs a color more than 4 .

(a)

(b)

Figure 1. a) Labeling of $D_{1}(1)$ b) Packing 5-coloring of $D_{1}(1)$

Case 3: For $c(x)=3$, let $c\left(u_{1}\right)=c\left(v_{1}\right)=c\left(w_{1}\right)=1$. Consider the vertices $u_{6}, v_{6}$ and $w_{6}$. Since $d\left(u_{6},\left\{v_{6}, w_{6}\right\}\right)=4$ and $d\left(v_{6}, w_{6}\right)=4, u_{6}, v_{6}$ and $w_{6}$ can be colored 2 . Now for vertex $v_{2}$, the only choice
is to get color 4. But if $c\left(v_{2}\right)=4$, vertices $u_{2}$ and $w_{2}$ can't be colored except if a color more than four is used.

Case 3.1: For $c(x)=3$, let $c\left(v_{1}\right)=2$ and $c\left(w_{1}\right)=4$. This forces $c\left(u_{1}\right)=1$. Since $d\left(v_{1},\left\{u_{2}, u_{6}\right\}\right)=3$, if $c\left(u_{2}\right)=2, u_{6}$ can't be colored except if a color greater than 4 is used. Similarly, if $c\left(u_{6}\right)=2$ instead of $c\left(u_{2}\right)=2, u_{2}$ can't be colored except if a color more than four is used.

Case 3.2: For $c(x)=3$, let $c\left(v_{1}\right)=c\left(w_{1}\right)=1$. Then $c\left(u_{1}\right) \in\{2,4\}$. Since $d\left(u_{1},\left\{u_{6}, w_{6}\right\}\right)=3, u_{1}, u_{6}$ and $w_{6}$ can be colored 2. This forces $c\left(w_{2}\right)=4$. But if $c\left(w_{2}\right)=4, v_{2}$ can't be colored except if a color greater than four is used and suppose if $c\left(u_{1}\right)=4$ instead of $c\left(u_{1}\right)=2$, vertex $v_{2}$ should receive a color more than four.

Case 4: For $c(x)=4$, let $c\left(u_{1}\right)=1$. Then $c\left(u_{2}\right)$ and $c\left(u_{6}\right) \in\{2,3\}$. Without loss of all inclusive statement, let $c\left(u_{2}\right)=2$ and $c\left(u_{6}\right)=3$. This forces $c\left(u_{3}\right)=c\left(u_{5}\right)=1$. But for vertex $u_{4}$, the only choice is to get a color more than four.

Case 4.1: For $c(x)=4$, let $c\left(u_{1}\right) \in\{2,3\}$. To minimize the packing coloring, vertices $u_{2}$ and $u_{6}$ can be colored 1. When $c\left(u_{1}\right)=2$, either $c\left(u_{3}\right)=3$ or $c\left(u_{5}\right)=3$. Without loss of all inclusive statement let $c\left(u_{3}\right)=3$. Then $u_{5}$ needs a color more than four. Similarly, when $c\left(u_{1}\right)=3$ instead of $c\left(u_{1}\right)=2$, either $c\left(u_{3}\right)=2$ or $c\left(u_{5}\right)=2$. Without loss of generality let $c\left(u_{3}\right)=2$. Then $u_{5}$ needs a color more than four. Hence from above four Cases, $\chi_{\rho}\left(D_{1}(1)\right) \geq 5$. From Figure $1(\mathrm{~b})$, we conclude that $\chi_{\rho}\left(D_{1}(1)\right)=5$.

Theorem 2.4. For $D_{1}(n), n=2,3, \chi_{\rho}\left(D_{1}(n)\right)=5$.
Proof: Since $D_{1}(1)$ is a subgraph of $D_{1}(n), n=2,3$, by Propositions 2.1 and $2.2, \chi_{\rho}\left(D_{1}(n)\right) \geq 5$. From Figures 2 and 3, the upper bound is $\chi_{\rho}\left(D_{1}(n)\right) \leq 5$. Hence for $D_{1}(n), n=2,3, \chi_{\rho}\left(D_{1}(n)\right)=5$.


Figure 2. Packing 5-coloring of $D_{1}(2)$
Open Problem 2.5. By a comparable, yet considerably more monotonous investigation as in the above verification one can set up $\chi_{\rho}\left(D_{1}(n)\right)=5$, for $n \geq 4$.

## 3. THE STRUCTURE OF BISMUTH TRI-IODIDE

Bismuth tri-iodide (BiI3) is one of the inorganic compounds. The result of the response of bismuth and iodine is used to be of attractive for subjective inorganic investigation. Bismuth tri-iodide is additionally shaped by the activity of ethyl iodide on bismuth trichloride within the sight of ethyl chloride. Layered BiI3 crystal is viewed as 3-layered stacking structure. The occasional stacking of 3 layers frames rhombohedral BiI3 crystal with R-3 symmetry. The progressive stacking of one I-Bi-I layer frames hexagonal structure with symmetry [6]. Figure 4(a) demonstrates one unit of bismuth tri-iodide. An $m$-bismuth chain is obtained by linearly arranging $m$ unit cells.


Figure 3. Packing 5-coloring of $D_{1}(3)$

Theorem 3.1. Let $G$ be an $m$-bismuth chain. Then $\chi_{\rho}(G)=3$.

Proof: Since $C_{4}$ is a subgraph of $G$, by Propositions 2.1 and $2.2, \chi_{\rho}(G) \geq 3$. The upper bound pursues from the coloring of an $m$-bismuth chain with 3 colors, whose pattern is shown in Figure 4(b). Thus, from Figure 4(b), $\chi_{\rho}(G) \leq 3$. Hence $\chi_{\rho}(G)=3$.


Figure 4. a) One unit of bismuth b) Packing 3-coloring of 3-bismuth chain c) Packing 3-coloring of subdivided 3-bismuth chain

By subdividing each edge of a graph $G$, subdivision graph $S(G)$ is acquired.

Theorem 3.2. Let $S(G)$ be a subdivided $m$-bismuth chain. Then $\chi_{\rho}(S(G))=3$.

Proof: Since $C_{8}$ is a subgraph of $S(G)$, by Propositions 2.1 and 2.2, $\chi_{\rho}(S(G)) \geq 3$. Then upper bound pursues from the labelling of a subdivided $m$-bismuth chain with 3 colors, whose pattern is shown in Figure 4(c). In Figure 4(c), it is expected that the vertices which are not colored get color 1. Thus, from

Figure 4(c), $\chi_{\rho}(S(G)) \leq 3$. Hence $\chi_{\rho}(S(G))=3$.

## 4. THE STRUCTURE OF LEAD CHLORIDE

Lead chloride is a salt crystal which is utilized in the creation of infra red disseminating glass. It is additionally used as an intermediate in refinement of bismuth (Bi) ore. The structure of lead chloride is orthorhombic dipyramidal [6]. The diagram of one unit of lead chloride is gotten from that of bismuth triiodide by joining only one two degree vertex of every one of the 4-cycles to another vertex. As on account of bismuth tri-iodide, chains and sheets of lead chloride are characterized. See Figure 5(a).


Figure 5. a) One unit of lead chloride b) Packing 5-coloring of 4-lead chloride chain c) Packing 3coloring of subdivided 4-lead chloride chain

Theorem 4.1. Let $G$ be an $m$-lead chloride chain. Then $4 \leq \chi_{\rho}(G) \leq 5$.
Proof: Since $C_{6}$ is a subgraph of $G$, by Propositions 2.1 and $2.2, \chi_{\rho}(G) \geq 4$. Then upper bound pursues from the labelling of an $m$-lead chloride chain with 5 colors, whose pattern is shown in Figure 5(b). In Figure 5(b), it is expected that the vertices which are not colored get color 1. Thus, from Figure 5(b), $\chi_{\rho}(G) \leq 5$. Hence $4 \leq \chi_{\rho}(G) \leq 5$.
Theorem 4.2. Let $S(G)$ be a subdivided $m$-lead chloride chain. Then $\chi_{\rho}(S(G))=3$.
Proof: Since $C_{8}$ is a subgraph of $S(G)$, by Propositions 2.1 and $2.2, \chi_{\rho}(S(G)) \geq 3$. Then upper bound pursues from the labelling of a subdivided $m$-lead chloride chain with 3 colors, whose pattern is shown in Figure 5(c). In Figure 5(c), it is expected that the vertices which are not colored get color 1. Thus, from Figure 5(c), $\chi_{\rho}(S(G)) \leq 3$. Hence $\chi_{\rho}(S(G))=3$.

## CONFLICTS OF INTEREST

No conflict of interest was declared by the authors.

## REFERENCES

[1] Goddard, W., Hedetniemi, S.M., Hedetniemi, S.T., Harris, J. M., Rall, D.F., "Broadcast Chromatic Numbers of Graphs", Ars Combin , 86: 33-49, (2008).
[2] Brešar, B., Klavžar, S., Rall, D.F., "On the packing chromatic number of cartesian products, hexagonal lattice and trees", Discrete Appl. Math., 155 (17): 2303-2311, (2007).
[3] Azari, M., Iranmanesh, M., Diudea, M.V., "Vertex-eccentricity descriptors in dendrimers", Studia Univ. Babes Bolyai Chem., 62(1): 129-142, (2017).
[4] Tripathy, S., Das, M.K., "Dendrimers and their applications as novel drug delivery carriers", Journal of Applied Pharmaceutical Science., 3(09): 142-149, (2013).
[5] https://en.wikipedia.org/wiki/Crystal_structure.
[6] Krishnan, S., and Rajan, B., "Fault-tolerant resolvability of certain crystal structures", Applied Mathematics, 7: 599-604, (2016).

