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Radon concentration measurements in surface water samples from Van Lake, Turkey using CR-39 detectors

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ABSTRACT

In this study, radon concentration was examined in surface water samples taken from Van Lake in spring and autumn. The samples were taken along the coastline from Tatvan, where the active fault line lies, until Erciş. Analyses were performed by using CR-39 solid nuclear track detectors and RadoSYS radon measurement system. Radon measurement parameters (C_{Rn} , E_s , E_m and EC_{Rn}) were calculated with the data obtained. The results of this study were compared with the limit values given for radon in water and the results obtained in similar studies.

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1. Introduction

Radon is a noble gas with a half-life of 3.82 days and it occurs as a result of the radioactive conversion of ^{226}Ra (Radium) located in the decay chain of ^{238}U (Uranium). Radioactive radon gas, which is colorless, odorless, tasteless and short-lived decay products in the atmosphere account for about 50 % of the radiation that people are exposed to from natural sources (UNSCEAR, 2000). ^{222}Rn , which is approximately seven times heavier than air, is collected in hollows and dissolved in water (Alkan and Göksel, 1975). It is not possible to find water in its pure form in the natural environment. Radiation pollutes water and threatens human health. The contamination of water by radiation results from the natural radioisotopes in the atmosphere and soil, reactor accidents, nuclear weapon tests and medical radioactive wastes (Yarımış, 1985). Besides the solubility of radon gas in water there is also a tendency to escape into the environment where the dissolved radon is located. As a result, high radon concentrations in groundwater and thermal waters, it poses a

great danger not only for people who drink water, but also for people who breathe it (UNSCEAR, 2000).

One of the significant factors that need to be determined in terms of water usability is the level of radioactivity. For this reason, the conformity of drinking and domestic water to standards should be investigated with analyses on the radioisotopes available in water (Varol, 2011). The natural radioactivity of the water arises from the radioactive masses or minerals, which they pass through or are in contact with. Radon concentration in water changes depending on some factors such as the emission of radon from the rocks that the water contacts, temperature, pressure, precipitation and seismic activities (Ilani et al., 2006; Sannappa et al., 2006). Some rock types contain uranium in concentrations higher than 5 ppm. These are granites, syenites, pegmatites, acid volcanic rocks and gneisses (Shashikumar et al., 2009; UNSCEAR, 2000).

Radon measurement techniques divide into two groups as active and passive measurement techniques. In this study, the

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passive measurement technique was employed to detect radon gas concentrations in water. CR-39 solid track detectors are frequently used in passive measurement technique (Kayakökü et al., 2016; Turhan et al., 2018; Büyüksulu et al., 2018; Alali et al., 2019).

The aim of present research study was to determine the radon concentration, effective radium content, the surface and mass diffusion rates of radon in surface water samples taken from Van Lake in spring and autumn. For this purpose, CR-39 solid nuclear track detectors were used. Radon analyzes were done using RadoSYS system.

2. Material and Method

2.1. Research area

Van Lake is located in the east of Turkey and the GPS coordinates of its geographical location are $38^{\circ} 38' 27''$ north and $42^{\circ} 48' 45''$ east. It was formed as a result of the explosion of the Nemrut volcanic mountain, which is located within the borders of the province of Bitlis. Van Lake is a volcanic embankment with a surface area of 3790 km^2 as an enclosed lake. The water of this lake is an alkaline lake with a salinity rate of 0.19% and pH of 9.8 on average.

Water samples were collected using a Hydro bios ruttner water sampler from 24 stations within Van Lake. 15 ml of HNO_3 (nitric acid) was added to each 1 liter of sample. (Bohus-Saja et al., 1997). In this way, $\text{pH} \leq 2$ was obtained for water samples, and it was ensured that the elements in the samples would not precipitate and stick on the surface of the sample container (Bohus-Saja et al., 1997).



Figure 1. Research area and sampling points

In present study, the radon concentrations, exhalation rates and radium contents were calculated using CR-39 detectors for forty eight surface water samples. After the water samples were placed in the sample containers, they were left for a month to ensure radioactive equilibrium. CR-39 detectors were prepared in pieces of $1 \text{ cm} \times 1 \text{ cm}$ dimensions. After a one-month waiting period, CR-39 detectors were placed in sample containers. Likewise, after a one-month waiting period, the detectors were removed from the containers.

The cylindrical polyethylene sample containers, in which the sample and the CR-39 detector are placed, are shown in Figure 2.

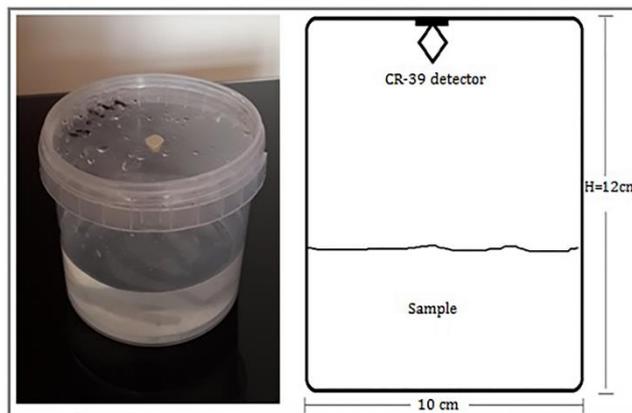


Figure 2. The structure cylindrical polyethylene sample container used for radon measurements

After the one-month waiting period was over, the sample containers were opened, and the detectors were removed from the containers. During this time, the chemical etching process was employed in order to make the traces of the alpha particles left on the detectors apparent. For this purpose, a 30% NaOH solution was prepared and distributed into the beakers equally. After the detectors were placed in the solution, the beakers were kept in the oven at 60°C for 18 hours. Afterwards, the detectors were removed from the beakers which were taken out of the oven and then washed with pure water. Following the completion of the washing process, the detectors were kept aside for about 1 hour to dry. After drying up, the detectors were ready for the tracking procedure.

Tracks of alpha particles hitting Cr-39 detectors during the waiting period were shown in Figure 3.

Radon analysis operations were carried out using the Passive Radon Detector System (RadoSYS system) in Bitlis Eren University Nuclear Physics Research Laboratory. The radon tracking system (Figure 3) consists of a $500\times$ magnification microscope and software connected to the computer. The program works with the LINUX operating system. First, CR-39 detectors were placed on the slides. After the detectors on the slides were divided into 144 equal parts by the optical tracking system, the traces on the detectors were counted. After the periods of keeping the detectors in the dishes containing the

samples were entered on the program, radon density and radon activity concentration were separately calculated by the program for each sample.

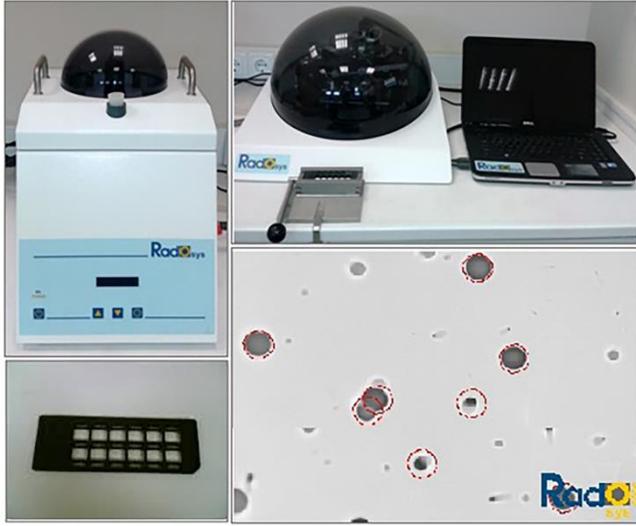


Figure 3. Radon track reading system and the tracks of alpha particles in the CR-39 detector

In Figure 3, the tracks of alpha particles hitting the CR-39 detector can be seen. While calculating radon concentrations, it is necessary to detect the background value of the detector placed in the dish when there is no sample. For this, the detectors in the sample dishes and the detector in the empty dish were tracked on the detector tracking system in the same way, and the net track densities were acquired by the subtraction of the track densities from each other.

2.2. Radon concentration calculations

After specifying the net trace densities formed on the CR-39 detectors, the radon concentrations were calculated by Equation (1) using the obtained net trace densities, the periods the detectors were kept in the sample containers and the calibration factor.

$$C_{Rn} = \frac{\rho \times f}{\Delta t} \quad (1)$$

where C_{Rn} is radon concentration (Bq/m^3), ρ is net track density ($track/mm^2$), f is calibration factor and Δt is detector exposure time (in hour) (Tokonami et al., 2005). Calibration factor for CR-39 solid nuclear trace detectors is $44.47 (kBq/m^3)/(track/mm^2)$.

The equation used to calculate the effective radium content (EC_{Ra}) was shown in Equation 2.

$$EC_{Ra} = \frac{\rho \times h \times A \times f}{V \times T_{eff}} \quad (2)$$

2.3. Effective radium content (EC_{Ra})

where h is distance between detector and sample (m), A is the surface area of sample container (m^2), T is total exposure time (h), T_{eff} is effective exposure time [$T_{eff} = T + 1/\lambda (e^{\lambda T} - 1)$], V is sample volume (m^3) (Jönsson et al., 1999).

2.4. Radon surface and mass exhalation rates (E_s, E_m)

Radon surface exhalation rates (E_s) and radon mass exhalation rates (E_m) were calculated using Equation (3) and Equation (4), respectively.

$$E_s = \frac{\rho \times f \times h \times \lambda}{T_{eff}} \quad (3)$$

$$E_m = E_s \times \frac{A}{M} \quad (4)$$

where, λ (2.1×10^{-6}) is ^{222}Rn decay constant, h is height of the sample container (m), A is surface area of the sample in the container (m^2) and M is the mass of the sample in the sample container (kg).

3. Results and discussion

Radon concentrations (C_{Rn}), radon surface exhalation rates (E_s), radon mass exhalation rates (E_m) and effective radium contents (EC_{Ra}) for surface water samples taken in spring and autumn periods were given in Table 1 and Table 2, respectively. While the comparison of the radon concentration values in the surface water samples of the spring and autumn period with the limit value recommended by USEPA (United States Environmental Protection Agency, 2012) was shown in Figure 4 and Figure 5, respectively, the seasonal comparison of the radon concentration values was shown in Figure 6.

$$EC_{Ra} = \frac{\rho \times h \times A \times f}{V \times T_{eff}}$$

Table 1. The C_{Rn} , E_s , E_m and EC_{Ra} values of surface water sample in spring period (Kayakökü and Doğru, 2019).

Sample ID	Sample locations		C_{Rn} (Bq/m ³)	E_s (Bq/m ² .h)×10 ⁻¹	E_m (Bq/m ³ .h)×10 ⁻³	EC_{Ra} (Bq/m ³)
	Latitude	Longitude				
Y.S-1	38° 29' 04.3"	42° 29' 22.5"	10.00±0.92	0.12±0.01	0.14±0.01	0.08±0.01
Y.S-2	38° 30' 02.8"	42° 23' 39.8"	18.30±1.69	0.14±0.01	0.21±0.02	0.15±0.01
Y.S-3	38° 30' 03.1"	42° 18' 03.4"	43.41±4.00	0.32±0.03	1.31±0.12	0.72±0.07
Y.S-4	38° 33' 17.7"	42° 22' 11.4"	78.60±7.25	0.58±0.05	2.15±0.20	1.30±0.12
Y.S-5	38° 36' 26.5"	42° 24' 30.9"	33.85±3.12	0.25±0.02	1.02±0.09	0.56±0.05
Y.S-6	38° 39' 05.3"	42° 28' 14.7"	50.00±4.61	0.37±0.03	0.48±0.04	0.35±0.03
Y.S-7	38° 42' 56.9"	42° 26' 29.8"	31.07±2.8	0.23±0.02	0.94±0.09	0.51±0.05
Y.S-8	38° 44' 55.7"	42° 32' 03.1"	45.20±4.17	0.33±0.03	0.44±0.04	0.31±0.03
Y.S-9	38° 46' 15.3"	42° 36' 44.6"	10.75±0.99	0.08±0.01	0.33±0.03	0.18±0.02
Y.S-10	38° 47' 31.5"	42° 43' 30.5"	32.00±2.95	0.24±0.02	0.31±0.03	0.22±0.02
Y.S-11	38° 46' 22.4"	42° 49' 41.6"	80.85±7.45	0.60±0.06	2.30±0.21	1.34±0.12
Y.S-12	38° 47' 00.4"	42° 54' 52.7"	16.24±1.50	0.12±0.01	0.16±0.01	0.11±0.01
Y.S-13	38° 46' 01.1"	42° 59' 53.7'	41.82±3.86	0.31±0.03	0.48±0.04	0.35±0.03
Y.S-14	38° 46' 43.2"	43° 03' 41.9"	58.24±5.37	0.43±0.04	0.56±0.05	0.41±0.04
Y.S-15	38° 48' 47.6"	43° 08' 04.7"	49.29±4.54	0.33±0.03	1.35±0.12	0.73±0.07
Y.S-16	38° 52' 19.1"	43° 11' 12.9"	62.80±5.79	0.37±0.03	0.66±0.06	0.47±0.04
Y.S-17	38° 54' 24.9"	43° 10' 04.9"	56.24±5.19	0.42±0.04	0.54±0.05	0.39±0.03
Y.S-18	38° 51' 45.3"	43° 06' 14.9"	60.00±5.53	0.44±0.04	0.58±0.05	0.42±0.04
Y.S-19	38° 55' 16.7"	43° 03' 42.0"	25.53±2.35	0.12±0.01	0.49±0.04	0.26±0.01
Y.S-20	38° 56' 23.5"	43° 08' 05.3"	28.28 2.61	0.21±0.03	0.86±0.08	0.47±0.03
Y.S-21	38° 57' 08.6"	43° 14' 30.1"	18.65±1.72	0.14±0.01	0.18±0.02	0.13±0.01
Y.S-22	38° 56' 50.8"	43° 17' 39.1"	30.10±2.78	0.22±0.02	0.29±0.03	0.21±0.02
Y.S-23	38° 57' 41.5"	43° 19' 27.5"	68.45±6.31	0.51±0.05	2.07±0.19	1.49±0.14
Y.S-24	38° 58' 46.7"	43° 21' 23.8"	20.55±1.89	0.15±0.01	0.20±0.02	0.14±0.01
	Average		40.43±3.73	0.29±0.03	0.75±0.07	0.47±0.04
	Maximum		80.85±7.45	0.60±0.06	2.30±0.21	1.49±0.14
	Minimum		10.00±0.92	0.08±0.01	0.14±0.01	0.08±0.01

Table 2. The C_{Rn} , E_s , E_m and EC_{Ra} values of surface water sample in autumn period (Kayakökü and Doğru, 2019).

Sample ID	Sample locations		C_{Rn} (Bq/m ³)	E_s (Bq/m ² .h)×10 ⁻¹	E_m (Bq/m ³ .h)×10 ⁻³	EC_{Ra} (Bq/m ³)
	Latitude	Longitude				
Y.S-1	38° 29' 04.3"	42° 29' 22.5"	a	a	a	a
Y.S-2	38° 30' 02.8"	42° 23' 39.8"	a	a	a	a
Y.S-3	38° 30' 03.1"	42° 18' 03.4"	48.59±4.48	0.36±0.03	1.47±0.14	0.80±0.07
Y.S-4	38° 33' 17.7"	42° 22' 11.4"	125.33±11.60	1.15±0.11	4.70±0.43	2.57±0.24
Y.S-5	38° 36' 26.5"	42° 24' 30.9"	28.76±2.65	0.45±0.04	0.66±0.06	0.55±0.05
Y.S-6	38° 39' 05.3"	42° 28' 14.7"	65.00±5.99	0.48±0.04	2.00±0.18	1.18±0.11
Y.S-7	38° 42' 56.9"	42° 26' 29.8"	86.43±7.97	0.64±0.06	2.61±0.24	1.43±0.13
Y.S-8	38° 44' 55.7"	42° 32' 03.1"	46.50±4.29	0.34±0.03	1.45±0.13	0.84±0.08
Y.S-9	38° 46' 15.3"	42° 36' 44.6"	a	a	a	a
Y.S-10	38° 47' 31.5"	42° 43' 30.5"	54.92±5.06	0.47±0.04	1.96±0.18	1.08±0.10
Y.S-11	38° 46' 22.4"	42° 49' 41.6"	171.17±15.80	1.27±0.12	5.19±0.48	3.28±0.26
Y.S-12	38° 47' 00.4"	42° 54' 52.7"	45.14±4.16	0.32±0.03	1.84±0.17	1.10±0.10
Y.S-13	38° 46' 01.1"	42° 59' 53.7'	60.55±5.53	0.45±0.04	0.89±0.08	1.06±0.10
Y.S-14	38° 46' 43.2"	43° 03' 41.9"	a	a	a	a
Y.S-15	38° 48' 47.6"	43° 08' 04.7"	a	a	a	a
Y.S-16	38° 52' 19.1"	43° 11' 12.9"	44.21±4.08	0.33±0.03	1.36±0.13	0.80±0.07
Y.S-17	38° 54' 24.9"	43° 10' 04.9"	a	a	a	a
Y.S-18	38° 51' 45.3"	43° 06' 14.9"	60.00±5.53	0.44±0.04	1.85±0.17	1.09±0.10
Y.S-19	38° 55' 16.7"	43° 03' 42.0"	72.45±6.68	0.53±0.05	2.23±0.21	1.32±0.12
Y.S-20	38° 56' 23.5"	43° 08' 05.3"	28.28±2.61	0.21±0.02	0.38±0.04	0.42±0.04
Y.S-21	38° 57' 08.6"	43° 14' 30.1"	36.80±3.39	0.27±0.02	0.50±0.05	0.58±0.05
Y.S-22	38° 56' 50.8"	43° 17' 39.1"	a	a	a	a
Y.S-23	38° 57' 41.5"	43° 19' 27.5"	a	a	a	a
Y.S-24	38° 58' 46.7"	43° 21' 23.8"	a	a	a	a
Average			64.94±5.99	0.51±0.05	1.94±0.18	1.21±0.11
Maximum			171.17±15.80	1.27±0.12	5.19±0.48	3.28±0.26
Minimum			28.28±2.61	0.21±0.02	0.38±0.04	0.42±0.04

^a Water samples could not be taken

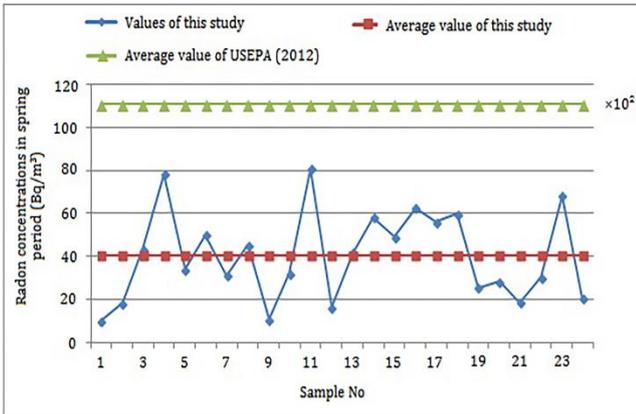


Figure 4. Comparison of radon concentration values in surface water samples of spring period with the value recommended by United States Environmental Protection Agency (USEPA, 2012).

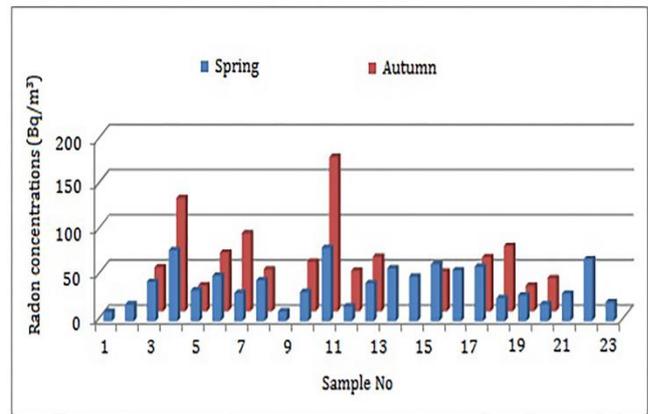


Figure 6. Seasonal variation of radon concentration in samples

Finally, in all water samples, EC_{Ra} value was determined to be the lowest with $(0.08 \pm 0.01) \times 10^{-1} \text{ Bq/m}^3$ at Y.S-23 sample in spring period and the highest with $(3.28 \pm 0.26) \times 10^{-1} \text{ Bq/m}^3$ at Y.S-9 sample in autumn period.

Comparison of radon concentration averages obtained for surface water samples in this study with the results obtained in similar studies was given in Table 3.

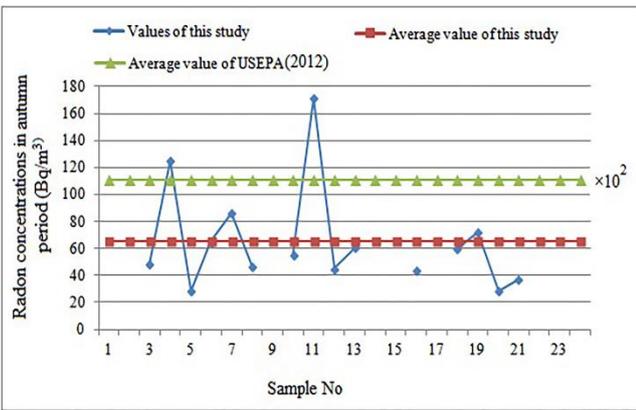


Figure 5. Comparison of radon concentration values in surface water samples of autumn period with the value recommended by United States Environmental Protection Agency (USEPA, 2012).

The highest C_{Rn} , E_s , E_m values in spring water samples were found in the sample Y.S-11. Similarly, the highest C_{Rn} , E_s , E_m and EC_{Ra} values in autumn water samples were found in the sample Y.S-11. While the lowest C_{Rn} value obtained for water samples belonged to the spring period Y.S-1 sample with $10.00 \pm 0.92 \text{ Bq/m}^3$, the highest C_{Rn} value belonged to the autumn period Y.S-11 sample with $171.17 \pm 15.80 \text{ Bq/m}^3$. While the lowest E_s value belonged to the spring period Y.S-9 sample with $(0.08 \pm 0.01) \times 10^{-1} \text{ Bq/m}^2 \cdot \text{h}$, the highest E_s value belonged to the autumn period Y.S-11 sample with $(1.27 \pm 0.12) \times 10^{-1} \text{ Bq/m}^2 \cdot \text{h}$. E_m value was calculated as the lowest with $(0.14 \pm 0.01) \times 10^{-3} \text{ Bq/m}^3$ at Y.S-1 sample in spring period and the highest with $(5.19 \pm 0.48) \times 10^{-3} \text{ Bq/m}^3$ at Y.S-11 sample in autumn period.

Table 3. Comparison of radon concentration averages obtained for water samples in this study with the results obtained in similar studies

Countries	Concentration of ^{222}Rn (Bq/m ³)	Reference
England (English Lake District)	53.7±8.15-1130.74±35.92	Al-Masri and Blackburn, 1999
Turkey (Yeşilirmak River)	280±40-1080±300	Oner et al., 2009
Egypt (Manzala Lake)	(1.720 – 6.400) ×10 ³	Yousef et al., 2017
China (Yunlong Lake) (average)	2.154 ×10 ³	Yang et al., 2020
India (Varahi River)	(2.07±0.84) ×10 ³	Somashekar and Ravikumar, 2010
India (Markandeya River) (average)	(9.30±1.45) ×10 ³	Somashekar and Ravikumar, 2010
UNSCEAR	10×10 ³	UNSCEAR, 2000
USEPA	11.1×10 ³	USEPA, 2012
EPA	11×10 ³	EPA, 2000
WHO	100×10 ³	WHO, 2008
Turkey (Van Lake) (average)	40.43±3.73 (spring) 64.94±5.99 (autumn)	Present Study

Baykara and Dođru (2006) collected 14 water samples from the region of the North and East Anatolian Active Faults. By using the passive radon measurement method, they calculated the highest radon concentration, effective radium content and radon diffusion rate in water samples as 3319.3 Bq/m³, 270.5 Bq/m³ and 0.470 Bq/m².h, respectively (Baykara and Dođru, 2006).

Bal and Dođru (2013), passively examined radon concentrations in the water samples they received from the radon monitoring stations they established in the Sivrice Fault Zone. As a result of the study, it was seen that the radon concentration values of the water samples taken in autumn and spring varied between 2750±736 Bq/m³ and 4624±1251 Bq/m³ with 4464±1524 Bq/m³ and 7163±1941 Bq/m³, respectively (Bal and Dođru, 2013).

According to the study conducted by Zorer et al., (2013), the radon concentration in lake water around Van Lake varied between 47.80 Bq/m³ and 354.86 Bq/m³ (Zorer et al., 2013).

United States Environmental Protection Agency (USEPA) recommends the maximum value of the radon activity level as

11.1 Bq/L (11.1×10³ Bq/m³) in drinking water (USEPA, 2012). The United Nations Scientific Committee on the Effects of Atomic Radiations (UNSCEAR) has suggested the radon concentration in drinking water must lie between 4 and 40 Bq/L (UNSCEAR, 2008).

In present study, the radon concentration values calculated in all samples were lower than 100 Bq/L (100×10³ Bq/m³) recommended for drinking water by European Commission (EU, 2001) and World Health Organization (WHO, 2004).

The results obtained in water samples of this study were found to be lower than the results obtained in the above studies and the limit values given for water.

Considering Table 1, Table 2 and Figure 6, when the C_{Rn}, E_s, E_m and EC_{Ra} values obtained for the surface water samples taken in spring and autumn were compared seasonally, it was observed that the C_{Rn}, E_s, E_m and EC_{Ra} values in spring were lower than those from autumn. This may result from the high rainfall in spring and excessive flows from other water sources into the lake.

4. Conclusions

In this study, the seasonal variation in the C_{Rn}, E_s, E_m and EC_{Ra} values in the surface water samples taken in spring and autumn from 24 different points of Van Lake was investigated. Measurements were made using the Passive Radon Detector System (RadoSYS system).

At the end of the study, the highest C_{Rn}, E_s, E_m and EC_{Ra} values were calculated for the S-11 point. The Suphan fault line passes through the region where S-11 is located, and this area densely contains volcanic sediments and late Miocene-Quaternary. In a study, Ramola et al. (2006) discovered that the radon level was higher in the areas comprised of granite, quartz-porphry, schist and phyllite slates, and lower in the areas with sedimentary rocks.

The radon measurement results obtained in this study were found to be lower than the limit values permitted by USEPA (2012), UNSCEAR (2008), EU (2001) and WHO (2004) and results obtained in similar studies.

When the results obtained for the C_{Rn}, E_s, E_m and EC_{Ra} parameters were compared seasonally, it was seen that the results obtained in the autumn period were higher than those in the spring period.

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