

Determination of optimum reaction and process control parameters of starch conversion in maltose syrup production

Maltoz şurubu üretiminde nişasta dönüşümünün optimum reaksiyon ve proses kontrol parametrelerinin belirlenmesi

Sema Nur ÇİNÇİK¹, Fatih BALCI^{2*}, Mustafa BAYRAM³

^{1,2,3} Gaziantep University. Faculty of Engineering. Department of Food Engineering. 27310. Gaziantep. Turkey. ¹https://orcid.org/0000-0003-3944-2482; ²https://orcid.org/0000-0002-9651-2064; ³https://orcid.org/0000-0001-6705-5899

ABSTRACT

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Address for Correspondence: Fatih BALCI e-mail: fbalci@gantep.edu.tr

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This work is licensed under a Creative Commons Attribution-Non Commercial 4.0 International License. In maltose syrup production, one of the critical processing stages is the starch conversion process. During this process, the reaction time and enzyme concentrations are two important parameters to obtain the standard sugar spectrum. The purpose of this study is; i) to find optimum reaction time and enzyme concentrations during the starch conversion process, ii) to determine process control and dynamic parameters during the starch conversion process in the maltose syrup production. The different amounts of beta and alpha-amylase enzymes (0.10, 0.15, 0.20 and 0.25 ml of β -amylase; 0.03, 0.05, 0.07 and 0.09 ml of α -amylase) were used to determine the optimum concentrations and time. pH, Brix and the concentrations of sugars (dextrose, maltose, maltotriose (DP3) and high sugars (DPN)) were determined. It was found that the enzyme concentration, ratios of the enzyme used and reaction time significantly affect the starch conversion process. The mixture containing 0.20 ml β -amylase and 0.05 ml α -amylase was determined as the optimum value (P≤0.05). It was found that the maximum process gains were obtained at 0.1 ml β -amylase and 0.03 ml α -amylase, 0.25 ml β -amylase and 0.03 ml α -amylase, 0.2 ml β -amylase and 0.03 ml α -amylase for dextrose, maltose, DP3 and DPN, respectively.

Key Words: Process control, Gain value, Starch conversion, Corn maltose syrup, α -amylase, β -amylase

ÖZ

Maltoz şurubu üretiminde kritik aşamalardan biri de nişasta dönüştürme işlemidir. Proses esnasında standart şeker spektrumunu elde etmek için iki önemli parametre reaksiyon zamanı ve enzim konsantrasyonlarıdır. Bu çalışmanın amacı maltoz şurubu üretiminde; i) Nişasta dönüşümü esnasında optimum reaksiyon ve enzim konsantrasyonunu bulmak, ii) Proses kontrol ve dinamik parametrelerinin tanımlanmasıdır. Optimum konsantrasyon ve zamanı belirlemek için farklı miktarlardaki alfa ve beta amilaz enzimleri (α -amilaz: 0.03, 0.05, 0.07 ve 0.09 ml ve β -amilaz: 0.10, 0.15, 0.20 ve 0.25 ml) kullanılmıştır. Ph, briks ve şeker konsantrasyonları (dekstroz, maltoz, maltotrioz (DP3) ve yüksek şekerler (DPN) tanımlanmıştır. Bu çalışmada açıkça görülmüştür ki enzim konsantrasyonu, kullanılan enzim oranları ve reaksiyon zamanı nişastanın maltoza dönüşümünde önemli ölçüde etkilidir. Maltoz şurubunun optimum şeker değerlerine ulaşması için en ideal enzim karışım 0.20 ml β -amilaz ve 0.03 ml α -amilazdır. Maksimum proses kazanımları dekstroz, maltoz, 0.2 ml β -amilaz ve 0,03 ml α -amilaz ve 0,03 ml

Anahtar Kelimeler: Proses kontrolü, Kazanç değeri, Nişasta dönüşümü, Mısır maltoz şurubu, α -amilaz, β -amilaz

Introduction

Starch is one of the most common foods consumed by humans in the world. It is generally derived from corn, rice, wheat, potato and tapioca, (Johnson et al., 2009). Starch is found as a carbohydrate in the seeds and roots of many plants (Zeeman et al., 2010). Starch-containing foods are an important element of nutrition and among the foods consumed by humans. Also, starch is used as a different product e.g. chemically or enzymatically processed such as glucose, fructose syrup, etc.

Starch is accepted as a polymer of dextrose; consists of two noticeable fractions amylose and amylopectin. Despite differ depending on the starch type, it is generally in the range of 20-25% amylose and 75-80% amylopectin. Amylose and amylopectin have a similar molecular structure (Gough et al., 2020)

The amylose fraction comprises dextrose molecules combined at the 1-4 linkages to form a long, spiral chain. A chain of dextrose molecules which in the amylopectin fraction, combined at the 1-4 linkages, but with an additional straightchain attached at the 1-6 linkages. When using enzymes in maltose syrup production 1-4 and 1-6 linkages are very crucial because some enzymes can bind with this region. Therefore, the right choice of enzyme ratio is crucial for a successful process (Hull, 2010).

The starch industry uses α and β amylase, isomerase, glucoamylase, pullulanase as enzymes. the α -amylase enzyme binds to 1-4 linkages in gelled starch at dextrose and maltose production. β -amylase attacks the linkages to produce maltose, with little amounts of dextrose, just as α amylase (Hull, 2010).

Maltose syrup (called also "corn syrup") is produced using the technique of the hydrolysis of starch. Corn starch is the main raw material for maltose syrup production (Pontoh and Low, 1995). Moreover, maltose syrup is get from wheat, rice, cassava and barley (Hull, 2010). At the same time, It can be defined as maltose, glucose and other saccharides solutions obtained from edible starch (Eke-Ejiofor, 2015). Corn syrup gives foods to soften texture, add volume, prevent sugar from crystallization and enhancing aroma. Furthermore, they are used in pharmaceuticals, brewery and bakery products. Based on the hydrolysis technique of starch and to what extent the hydrolysis reaction is allowed to proceed, the different maltose syrups with different properties and uses are produced (Norman et al., 2001).

Industrially, corn maltose syrup is produced by enzymatic conversion of starch. Type 2-yellow dent-corn, which is removed from the cob is used for maltose syrup (Öktem et al., 2013). Approximately, 70% (d.b.) of kernel is starch and 10% (d.b.) is protein. During steeping, corn absorbs water. Soluble protein passes to water. The moisture of corn increase from 15 to 40% and the kernel swell and then soften. After steeping, corn (40%) is milled and corn gluten meal, corn germ, corn protein and starch are separated from each other. The maltose process sets out the liquefaction step. Liquefaction step is done by using jet cooking, feed has 30-35% dry solid starch slurry (Van Der Maarel et al, 2013). Starch is passed from jet cooking at 105 °C for 10 min and hold at the liquefaction step to work α amylase enzyme. Then, the β -amylase enzyme is added for the conversion of sugar. This step is called starch conversion and/or reaction. In this study, this step was evaluated based on enzyme concentration, reaction time, process control and dynamic (dead time, process time constant and gain).

For the further step in the production, there are filtration, ion exchange and evaporation. Corn protein is not completely separated from sugar therefore rotary vacuum filter is used for separating corn protein from sugar. Maltose syrup is passed through the ion exchange step for removing anions and cations. Finally, the concentration of maltose syrup is increased at the evaporation step. For commercial purposes, corn syrups are subdivided into four types based on dextrose equivalent; 20-38 Dextrose Equivalent (DE), 38-58 DE, 58-73 DE and 73 DE or above (Blanchard, 1992). Sugar spectrum is determined by using HPLC (Polat et al., 2016).

As mentioned previously, the enzymatic conversion/reaction stage is very important for maltose syrup production. Since, polysaccharide chains are broken down to monomers, disaccharide and higher sugars during the reaction process. Temperature, enzyme type, substrate concentration and reaction time affect the maltose reaction. But, the information is very limited in the literature about the reaction (starch conversion) step.

In the starch industry, the reaction time is generally decided by the operators, in practice. The reaction time and initial enzyme concentration are the main decisive factors on the cost of the operation and product quality. If the reaction time is decreased, it will cause an increase in capacity and a decrease in energy cost. Additionally, process control and dynamics are critical for this step to obtain smooth production and product quality. Therefore, process gain (change in output/change in input), reaction time constant and dead time are the most important values for the process control of the dynamic systems (Altmann, 2005). In order to determine these values, the process reaction curve technique is maybe the best generally used method. It is simple to perform and provides an adequate model for many applications.

The purpose of this study was i) to find optimum reaction time and enzyme concentrations (α -amylase and β -amylase), ii) to determine process control and dynamic parameters which are, dead time values, process time constant and process gain during the starch conversion process in the maltose syrup production.

Material and Methods

Materials

The starch slurry was obtained from a controlled production line of Beşan Starch Plant (Gaziantep, Turkey). The moisture and starch

contents of corn (*Pioneer-2013*) used in the starch production were measured as 15 and 70% (w.b.) (AOAC, 1990), respectively. β -amylase (E1), (1280 U/g Spezyme DBA-F, Genencor, Dupont, USA) and α -amylase (E2), (32518 U/g Spezyme LT 300, Genencor, Dupont, USA) were used for the enzymatic conversion of starch during the reaction step.

Experimental set-up

The experimental set-up is given in Figure 1. The study was comprised of two parts as the plant and laboratory sections. The starch was produced and then cooked by using a jet-cooking under the controlled conditions (105-106 °C, 10 mins). The starch content of slurry in the jetcooking operation was 37%. The pH of the slurry was 5.7-6.0. After the jet cooking operation, the slurry was transferred to the liquefaction unit. In this step, dextrose equivalent (DE) and pH were reached 13 and 5.7, respectively. In order to stop the liquefaction operation, pH was adjusted to 2.0 by adding hydrochloric acid (Akay Kimya, Turkey). Then, the samples were stored for the experiments in a deep freezer (Beko, Turkey) at -5 °C.

For the reaction/starch conversion stage, 2 liters of the sample was used for each run. pH and temperature were adjusted to 5.50-6.20 and 60 °C, respectively. The pH adjustment was made by using 45% of NaOH (Akay Kimya, Turkey). The reaction was carried out in an agitator (RW 16 Basic, IKA, Japan) and a modified controllable water bath (BW20G, Jeo Tech, Korea).

The different amounts of enzymes (E1 and E2) were added for the reaction/starch conversion process (Table 1). The samples were collected every one hour for the analysis of dextrose, maltose, maltotriose (DP3), high-sugars (DPN), Brix, DE and pH.





Figure 1. Experimental Set-up

Brix was measured using a refractometer (NAR-3T, Atago, Japan) at 20 °C. The samples were filtered before the measurements.

Determination of sugar profile

The amount of dextrose, maltose, maltotriose (DP3) and high sugars (DPN) were determined by using HPLC (DGU-20A3, SIL-20 AHT, Reservoir tray, RID-10A, CTO-20A, Shimadzu, Japan) according to the method of Corn Refiner Associations, E-61 method (CRA, 2010). The sample (Brix=10) was prepared in a 1 ml vial. After 20 minutes of resting, the sugar spectrum was determined. DPN, DP3, maltose and dextrose were determined at 7, 8, 9 and 11th mins, respectively. The pump flow rate and oven temperature were 0.5 ml/min and 80 °C, respectively (CRA, 2010).

Determinations of pH

The Brix of the sample was adjusted to 40 before the measurement of pH. pH was measured by using a pH meter (Orion Star Series, Thermo Scientific, Singapore).

Determination of dextrose equivalent (DE)

Dextrose equivalent (DE) was measured by using Lane and Eynon method (CRA, 2010).

Determination of process control and dynamic parameters (process gain, process time constant and dead time)

To determine the process control and dynamic variables (process gain, process time constant and dead time), the process reaction curve technique was used (Marlin, 2000). For this technique, there are two methods such as Method I and Method II, which both use graphical solutions. The graphical solution for the process reaction technique decides the parameters for a first-order with a dead-time model. The version defines X(s) denoting the input and Y(s) denoting the output (Eq. 1). Both X and Y expressed in aberration factors;

$$\frac{Y(s)}{X(s)} = \frac{Kpe^{-\theta s}}{\tau s + 1}$$
(1)

where Kp, θ , τ and s are process gain, dead time, process time constant and Laplace frequency value, respectively.

The first method (Method I) concerted by Ziegler et al. (1942) uses graphical calculations. The average rates decided from the plot are the value of the input change (δ), the steady-state change in the output (Δ) and the maximum slope of the output-versus-time plot (S). The values from the plot can be described to the model parameters about the subsequent relationships for a first-order-with-dead-time model. The common model for a step within the input with t $\geq \theta$ is;

$$Y'(t) = Kp * \delta[1 - e^{-\frac{(t-\theta)}{\tau}}]$$
⁽²⁾

can be designated to be;

$$\frac{dY'(t)}{dt} = \frac{d}{dt} \left\{ Kp * \delta \left[1 - e^{-\frac{t-\theta}{\tau}} \right] \right\} = \frac{\Delta}{\tau} e^{-(t-\theta)/\tau}$$
(3)

The max slope at t = θ . That, S = Δ/τ . Thus, the variables can be described as;

$$Kp = \Delta/S \tag{4}$$
$$\tau = \Delta/S \tag{5}$$

$$\Delta/S$$
 (5)

where θ is the intercept of maximum slope with initial value, Y2(t) is output value versus time.

The second method, Method II also uses the graphical solution. The average rates examined from the graph are the steady-state change in the output (Δ), the significance of the input change (δ) and the times where the output levels up 28% and 63% of its final value. The chart can be associated with the model parameters using the common definition Eq. 2. Choice of any two values of time can examine the unexplained specifications, θ and τ . The ordinary times are preferred where the transient response is changing rapidly so that the model parameters can be exactly stated despite measurement noise (Smith, 1972). The definitions are;

$$Y(\theta + \tau) = \Delta(1 - e^{-1}) = 0.632\Delta$$
(6)
$$Y(\theta + \tau/3) = \Delta(1 - e^{-1/3}) = 0.283\Delta$$
(7)

Thus, the values of time at which the output reaches 28.3 and 63.2 percent of its final value were used to calculate the parameters.

$$t_{28\%} = \theta + \frac{\tau}{3} \qquad t_{63\%} = \theta + \tau \quad (8)$$

$$\tau = 1.5 (t_{63\%} - t_{28\%}) \qquad \theta = t_{63\%} - \tau \quad (9)$$

To sketch the graphs, the experimental data was modeled by using the Sigmoidal model (Sigmoid, 3 Parameter) (Eq. 10) (Sigmaplot Software, Sigmaplot Co., Cornwall, Canada).

$$f = a/(1+exp(-(x-x_0)/b))$$
 (10)

where f, a, b, x and x_o are parameters of the model.

The slope for this response at any time $t \ge \theta$

Statistical analysis

A standard procedure of the analysis of variance (One-way Anova) ($\alpha \le 0.05$) was performed to the data to assess the influence of the amount of enzyme and time on the sugar spectrum. Statistically significant correlations were decided as regards the Pearson coefficient (P ≤ 0.01) All statistical analyses were applied using SPSS software (ANOVA, SPSS Inc., Chicago, IL). The experiments were two replicated and the measurements were duplicated.

Results and Discussion

In the starch industry, the optimum maltose sugar spectrum is normally reached within 6 hours in the reaction step. Enzymes used during this step are quite costly. To reduce this cost, it is necessary to determine the optimum enzyme mixture. In the present study, the optimum enzyme mixture of β -amylase and α -amylase was determined by using different concentrations shown in Table 1. The experimental and statistical data obtained in the study conducted to find the ratio of the ideal mixture of these two enzymes to determine the optimum enzyme mixture are

presented in Table 2. Besides, during the conversion of starch to maltose in the reaction step, it was determined the dextrose, maltose, maltotriose (DP3), higher sugars (DPN), Brix and pH for each enzyme mixture for each hour. The overall experimental results are shown in Figure 6-9. The correlations among the parameters were analyzed statistically as Pearson coefficients and shown in Table 2.

Table 1. Concentration of enzymes (E1: β-amylase, E	2: α–
amylase)	

Dur Na		E2 (m)	
Run No	EI (MI)	E2 (MI)	Ratio (E2/E1)
1	0.10	0.03	0.30
2	0.10	0.05	0.50
3	0.10	0.07	0.70
4	0.10	0.09	0.90
5	0.15	0.03	0.20
6	0.15	0.05	0.33
7	0.15	0.07	0.47
8	0.15	0.09	0.60
9	0.20	0.03	0.15
10	0.20	0.05	0.25
11	0.20	0.07	0.35
12	0.20	0.09	0.45
13	0.25	0.03	0.12
14	0.25	0.05	0.20
15	0.25	0.07	0.28
16	0.25	0.09	0.36

	E1	E2	Time (hr)	Dextrose (%, g/g)	Maltose (%, g/g)	DP3 (%, g/g)	DPN (%, g/g)						
E1 (ml)	1												
E2 (ml)	0.00	1											
Time (hr)	0.00	0.000	1										
Dextrose (%, g/g)	-613**	0.156**	0.476**	1									
Maltose (%, g/g)	0.550**	0.007	0.748**	-0.006	1								
DP3 (%, g/g)	-0.225**	0.271**	0.765**	0.540**	0.556**	1							
DPN (%, g/g)	-0.371**	-0.080	-0.830**	-0.170**	-0.961**	-0.725**	1						
** Correlation is signi	ficant at the 0.0	1 level (2-taile	d). * Correlatio	n is significant at	the 0.05 level (2	-tailed).							

Table 2. Pearson correlation test coefficients

** Correlation is significant at the 0.01 level (2-tailed). * Correlation is significant at the 0.05 level (2-tailed).
 E1: β–amylase, E2: α–amylase, DP3: maltotriose, DPN: degree of polymerization high sugars

Changes in the amount of dextrose, maltose, DP3 and DPN during the starch conversion reaction

The starch from the different sources could be converted into glucose, maltose and maltodextrins. One of the most widely used in the food industry is maltose, especially in the brewing industry (Piddocke et al., 2009). The starch chain can be converted into dextrin by a few groups of amylases (BeMiller and Huber, 2007; Synowiecki, 2007). The α -amylases hydrolyze the starch molecule into smaller molecules. It is used to advance the gelatinization process, in enzymatic starch extraction, to decrease medium viscosity and in saccharification processes (BeMiller et al., 2007; Synowiecki, 2007). β -amylases (EC 3.2.1.2) are liable for partial hydrolysis of starch.

E1 is the β -amylase enzyme responsible for the

hydrolysis of starch. E2 is an α -amylase enzyme that decreases medium viscosity. When the changes in the amount of dextrose, maltose, DP3 and DPN were analyzed during the starch conversion stage, it was found that the amount of dextrose significantly (P≤0.05) changed. The minimum (0.30%) and maximum (2.74%) amount of dextrose were obtained at Run 5 (E1=0.15+E2=0.03) (Figure 7) and Run 3 (Figure 6) (E1=0.10+E2=0.07), respectively. Figure 2 shows that not only E1 and E2 enzymes affect the dextrose amount and time also affects the dextrose concentration.



Figure 2. Three-dimensional illustration of effect of different enzyme concentrations (E1:E2) on dextrose

Statistical analysis showed that E1xE2, E2xTime and E1xE2xTime interactions had a significant (P≤0.05) effect on the amount of dextrose. According to the results of the Pearson correlation test (P≤0.01), the amount of dextrose changed by using alpha (within the range 0.03-0.09) and beta-amylases (within the range 0.1-0.25) during the starch conversion operation. Besides, the correlations between E1, E2 and time were significant (P≤0.01) as shown in Table 2. As a of both analyses, the result dextrose concentration was affected by alpha and betaamylase enzymes and time.

When the change in the amount of maltose was evaluated during the starch conversion stage by using the different enzyme concentrations and ratios (Figure 3), it was determined that there was a significant ($P \le 0.05$) increase in the amount of

maltose during the starch conversion. According to the Pearson correlation analysis, the amount of maltose increased with E1 concentration; however, E2 concentration had not significant (P>0.05) effect (Table 2) on the amount of maltose. The maximum amount of maltose was obtained at E1=0.25+E2=0.03 (Run 13) (Figure 9) while the minimum amount of maltose was obtained at E1=0.1+E2=0.07 (Run 3). It was also found that E1, time, E1xtime and E1xE2 were significantly (P≤0.05) effective on the maltose concentration. It is obvious that the E1 enzyme and time were effective in the maltose concentration. The effectiveness of E1 can be explained that it is beta-amylase as a debranching enzyme that converts polysaccharides to monosaccharides based on Hull (2010).



Figure 3. Three-dimensional illustration of effect of different enzyme concentrations (E1:E2) on maltose

In a similar study on the conversion of barley starch, MacGregor et al. (1999) applied response surface methodology to examine α and β amylase levels and dextrinase enzymes. These enzymes are required for adequate modification of starch to fermentable sugars during the mashing process. Micro-scale mashes clarified from barley starch and malt enzymes. They were treated in a mash bath and solubilized starch, reducing sugars (neocuproine assay) and fermentable sugars (anion exchange HPLC) analyses were done for mash liquors. They found that fermentable sugars in the mash liquor were exactly related to reducing sugars and the percentage of starch solubilized during mashing. The addition of limit dextrinase to the mashes caused an important rising in levels of fermentable sugars and limit dextrinase displayed a synergistic effect with combined high levels of beta-amylase in

increasing levels of maltose in the mash liquor.

The obtained data and literature information support each other. As the β -amylase level increases in the reaction, the number of maltose increases. Similar to dextrose and maltose, the amount of DP3 significantly (P≤ 0.05) changed (Figure 4). It was found that E1, E2 and time had a significant effect (P≤0.01) on the change of DP3 (Table 2). Additionally, there was a correlation between dextrose, maltose and DP3. The maximum and minimum amount of DP3 were 21.86 and 6.78 % at 3rd hour (E1 = 0.15 + E2=0.07) and at 1st hour (E1=0.20+E2=0.03), respectively. According to the statistical analysis, there was a significant interaction (P≤0.05) such as E1x Time, E1xE2, E2xTime and E1xE2xTime. An increase in the concentration of E1 and E2, and also reaction time had a reverse effect on the concentration of DP3 (Table 2).

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Figure 4. Three-dimensional illustration of effect of different enzyme concentrations (E1:E2) on DP3

The minimum amount of DPN (high sugar) was also found as 28.82% at the 6th hour (0.25 ml E1+0.07 ml E2 enzyme) and its maximum amount was determined as 77.74% at the first hour (0.1 ml E1+0.07 E2) (Figure 5). E1, E2, time, E2xTime, E2xE1, E1xTime, E1xE2xTime had a significant (P \leq 0.05) effect on the high sugar concentration. E1 and E2 enzymes had a reverse effect on DPN concentration according to Pearson correlation. Besides, time, dextrose, maltose and DP3 concentrations have a reverse effect on DPN concentrations (P \leq 0.05).



Figure 5. Three-dimensional illustration of effect of different enzyme concentrations (E1:E2) on DPN

In the present study, additionally, besides the enzyme amounts, the enzyme ratios (E2/E1) (Table 1) were evaluated to find their effect on the amount of dextrose, maltose, DP3 and DPN. The minimum amount of dextrose was found at 0.12 of enzyme ratio (E2/E1) in 1st hour (Run 13; 0.25 ml E1+0.03 ml E2) (Table 1). However, the maximum amount of dextrose was found at 0.70 of enzyme ratio (E2/E1) at the 6th hour (Run 3; 0.10 ml E1+0.07 ml E2) (Table 1). It indicated

while increasing the E1 enzyme that the amount of dextrose did not increase. On the other hand, the E2 enzyme was important for increasing the dextrose concentration. Also, time proportionally affected the amount of dextrose.

The minimum and maximum amounts of maltose were determined at 0.7 and 0.12 of enzyme ratios (E2/E1), (Run 3; 0.10 ml E1+0.07 ml E2. Run 13: 0.25 ml E1+0.03 ml E2) at 1st and 6th hours, respectively (Table 1). E1 enzyme and time had a positive effect on the maltose concentration. On contrary, the E2 enzyme had no important effect on the maltose. Besides, the minimum and maximum amounts of DP3 were determined at 0.15 and 0.47 of enzyme ratios (E2/E1), (Run 9; 0.20 E1+0.03 E2, Run 7; 0.15 E1+0.09 E2) at 1st and 6th hours (Figure 7-8), respectively. The minimum and maximum amounts of DPN were determined at 0.28 and 0.47 enzyme ratios (E2/E1), (Run 15; 0.25 ml E1+0.07 ml E2, Run 3; 0.10 ml E1+0.07 ml E2) at 1st and 6th hours (Figure 6 and 9), respectively. All of these data indicate that time positively affects dextrose, maltose and DP3 concentrations, but affects DPN reverse on concentration. Additionally, E1 and E2 enzymes show a positive effect on all of the sugars.

There are different starch sources in the industry and they are nearly used the same production method. The results of the present study are also helpful for the other starch processing. In order to evaluate the suitability of this study to the other starch applications, the study of Lin et al. (2013) is important. In their studies, the rice starch sample was converted into maltose syrup for comparison with the corn starch sample. A amylase enzyme was used for conversion at the liquefaction step (EC 3.2.1.1). Also, pullulanase (EC 3.2.1.41) and β -amylase (EC 3.2.1.2) were used at the saccharification step. Scanning electron micrographs were used for determining the structure of the rice and corn starch samples. The final data showed that the granule range of the rice starch was considerably small and the optimal liquefaction time of rice starch short according to the corn starch sample.

The main impurities of the rice starch sample which are isomaltose and higher oligosaccharides were considerably low according to the sample of corn (P \leq 0.05) and starch of rice to maltose conversion was similar to corn starch. It was showed that rice starch was used as an alternative to corn starch. Therefore, the results of the present study can also be better used in rice starch production.

Chemical, thermal and enzymatic treatment can be a new process for hulled barley as an applicable starch source. These processes were developed by Kim et al. (2011). The aim of this treatment is both starch and lignocellulosic polysaccharides in hulled barley grains convert into fermentable sugars at the same time no need for milling and bran separation. In that study, hulled barley grains were processed in a 63-ml flow-through packed-bed stainless steel reactor at 0.1 and 1.0 (wt.) % sulfuric acid and 110 - 170 °C. After this step, conversion of starch and lignocellulose in the barley into sugar was carried by using alpha-amylase, glucoamylase, out cellulase and beta-glucosidase. Barley grains in the pre-treated which have starch and nonstarch polysaccharides were performed to fermentable sugars. Hulled barley grains were converted into fermentable sugars with the available glucose and xylose. The 1.0 (wt.) % sulfuric acid 110 °C conditions were used to obtain maximum efficiency from fermentable sugars.



Figure 6. Different E2 concentrations; (a): 0.03 ml, (b): 0.05 ml, (c): 0.07 ml and (d): 0.09 ml mixed with E1: 0.10



Figure 7. Different E2 concentrations. (a): 0.03 ml, (b): 0.05 ml, (c): 0.07 ml and (d): 0.09 ml mixed with E1: 0.15 ml



Figure 8. Different E2 concentrations. (a): 0.03 ml, (b): 0.05 ml, (c): 0.07 ml and (d): 0.09 ml mixed with E1:0.20 ml





Change in brix and pH during the starch conversion reaction

To follow the changes in Brix and pH values, their results were also analyzed. According to the statistical analysis, it was found that E1, E2, E1xE2 had a significant (P \leq 0.05) effect on Brix and pH according to the Duncan test. Time, E2xTime, E1xTime, E1xE2xTime had no significant (P>0.05) effect on the values.

Process control and dynamic parameters (process gain, process time constant and dead time)

To determine the process control variables, two methods (Methods I and II) were used (Marlin, 2000; Smith, 1972; Ziegler and Nichols, 1942). Process gain (Kp, the effect of the change in the input to the change in the output), process time constant (τ , "how fast" the process will respond after it starts moving) and dead time (θ , "how long" it will take a process to react to a change in the control output) were determined and they were used to analyze the dynamics of the starch conversion (Table 3).

According to the process gain values (Kp), both methods (Methods I and II) use the same formula, so their values were the same. Gain value means that the changes in the enzyme concentrations (change in input) affect the change in the amount of dextrose, maltose, DP3 and DPN (as the change in output). According to the results, there was a positive gain for dextrose, maltose and DP3. DPN had negative gain concerning the enzyme concentrations. The maximum process gains were obtained at Run 1, 13, 9 and 13 for dextrose, maltose, DP3 and DPN, respectively. According to these results, the amount of E2 needed to be 0.03 ml at each maximum value of all Runs.

Another important result was obtained such as; when the amount of E2 was increased at constant E1, the gain values decreased systematically for maltose, DP3 and DPN. This result is very important during the controlling process and the automation of the system.

It should be noted that the control of the starch conversion system can be made by using these gain values. Especially, if small or big changes are required for dextrose, maltose, DP3 and DPN, these small or big gain values can be used to determine the enzyme concentrations of E1 and E2.

Another effective usage of these data is to determine the processing time constant and dead time for the starch conversion stage. At the required gain or conversion values, the processing time constant and dead time can be determined from the results obtained. As a note, in the literature, Method II values for both processing time constant and dead time are preferred in the process control studies. Practically, in the process control system, there are processes, transmitters, controllers and actuators. The values obtained from the study can be used for the controller the process control during as dynamic parameters. According to gain, process time constant and dead time values, the controller will send a signal to the actuator to manipulate the enzyme dosage pump during the starch conversion process.

The processing time constant defines "how fast" the process will respond after it starts moving, as explained previously. Therefore, when the enzyme is added, the processing time constant value gives the how fast the starch conversion occurs during the reaction. According to Table 3, the processing time constant at maximum gain values was 1.009, 24.005, 6.608 and |-31.912| hours for dextrose, maltose, DP3 and DPN, respectively. It was higher for maltose and DPN than dextrose and DPN.

Table 3. Process control and dynamic parameters (process gain, process time constant and dead time)

Enzyme			yme	Model Parameters					Model 1							Model 2						
Comn	Run	со	nc.		would Par	ameters					Juer I						WOO					
comp.	No.	E1	52	-	h	Vo	D7	Smax	Δ	δ	Θ	τ	Кр	S	Δ	t	t	δ	Θ	τ	Кр	
		C1	EZ	d	U	70	ΝZ	(g/hr)	(g)	(ml)	(hr)	(hr)	(g/ml)	(g/hr)	(g)	(63%)	(28%)	(ml)	(hr)	(hr)	(g/ml)	
	1	0.10	0.03	2.424	1.28	1.905	0.952	0.473	1.922	0.030	1	4.066	64.067	0.473	1.922	1.211	0.538	0.03	0.202	1.009	64.067	
	2	0.10	0.05	1.689	4.805	6.135	0.92	0.739	1.375	0.050	1	1.860	27.500	0.739	1.375	0.866	0.385	0.05	0.144	0.722	27.500	
	3	0.10	0.07	5.145	2.635	5.642	0.945	0.813	2.240	0.070	1	2.754	32.000	0.813	2.24	1.411	0.627	0.07	0.235	1.176	32.000	
	4	0.10	0.09	1.636	4.959	7.075	0.996	1.232	1.335	0.090	1	1.084	14.833	1.232	1.335	0.841	0.374	0.09	0.140	0.701	14.833	
	5	0.15	0.03	9.832	1.958	1.834	0.985	0.240	0.611	0.030	1	2.545	20.367	0.24	0.611	0.385	0.171	0.03	0.064	0.321	20.367	
	6	0.15	0.05	1.721	8.553	9.338	0.95	2.744	1.305	0.050	1	0.476	26.100	2.744	1.305	0.822	0.365	0.05	0.137	0.685	26.100	
a)	7	0.15	0.07	9.408	1.078	6.295	0.973	0.444	0.624	0.070	1	1.406	8.914	0.444	0.624	0.393	0.175	0.07	0.066	0.328	8.914	
OSE	8	0.15	0.09	2.764	2.787	3.23	0.827	1.994	1.434	0.090	1	0.719	15.932	1.994	1.434	0.903	0.402	0.09	0.151	0.753	15.933	
ext	9	0.20	0.03	1.036	1.823	1.764	0.986	0.301	0.671	0.030	1	2.227	22.367	0.301	0.671	0.423	0.188	0.03	0.070	0.352	22.367	
Δ	10	0.20	0.05	8.852	2.038	1.401	0.938	0.824	0.524	0.050	1	0.636	10.480	0.824	0.524	0.330	0.147	0.05	0.055	0.275	10.480	
	11	0.20	0.07	3.049	2.493	5.795	0.975	0.928	1.317	0.070	1	1.419	18.814	0.928	1.317	0.830	0.369	0.07	0.138	0.691	18.814	
	12	0.20	0.09	1.037	1.138	1.153	0.974	1.720	0.758	0.090	1	0.441	8.422	1.72	0.758	0.478	0.212	0.09	0.080	0.398	8.422	
	13	0.25	0.03	7.711	1.864	8.593	0.946	0.469	0.441	0.030	1	0.941	14.700	0.469	0.441	0.278	0.123	0.03	0.046	0.232	14.700	
	14	0.25	0.05	1.08	7.509	6.602	0.979	2.318	0.789	0.050	1	0.340	15.780	2.318	0.789	0.497	0.221	0.05	0.083	0.414	15.780	
	15	0.25	0.07	9.986	8.705	5.613	0.958	0.820	0.696	0.070	1	0.848	9.943	0.82	0.696	0.438	0.195	0.07	0.073	0.365	9.943	
	16	0.25	0.09	9.908	7.463	6.261	0.985	1.105	0.725	0.090	1	0.656	8.056	1.105	0.725	0.457	0.203	0.09	0.076	0.381	8.056	
	1	0.10	0.03	4.637	1.470	2.634	0.991	7.114	35.156	0.030	1	4.942	1171.860	7.114	35.156	22.148	9.844	0.030	3.691	18.457	1171.860	
	2	0.10	0.05	3.893	9.864	1.691	0.991	1.230	32.614	0.050	1	26.510	652.272	1.230	32.614	20.547	9.132	0.050	3.424	17.122	652.272	
	3	0.10	0.07	3.896	1.144	2.010	0.992	0.888	32.148	0.070	1	36.212	459.256	0.888	32.148	20.253	9.001	0.070	3.376	16.878	459.256	
	4	0.10	0.09	3,853	9.831	1.604	0.985	1.144	31,880	0.090	1	27.872	354,220	1,144	31,880	20.084	8.926	0.090	3.347	16.737	354,220	
	5	0.15	0.03	4 867	8 269	1 378	0.982	1 368	40 931	0.030	1	29 924	1364 380	1 368	40 931	25 787	11 461	0.030	4 298	21 489	1364 380	
	6	0.15	0.05	4 326	8 795	1 481	0.982	0.849	36.099	0.050	1	42 509	721 988	0.849	36.099	22 743	10 108	0.050	3 790	18 952	721 988	
	7	0.15	0.05	4.920	6 269	1 1 2 3	0.902	1 319	42 574	0.050	1	32 272	608 201	1 319	42 574	26 822	11 921	0.050	4 470	22 351	608 201	
ose	, 8	0.15	0.07	4.370	7 468	1 316	0.909	0.818	36 777	0.070	1	14 948	408 631	0.818	36 777	23 169	10 298	0.070	3 862	19 308	408 631	
alto	a	0.15	0.03	4.527	8 250	1 / 53	0.909	1 030	30.777	0.030	1	38 608	1325 /33	1 030	30.777	25.105	11 13/	0.030	1 175	20.876	1325 /33	
Σ	10	0.20	0.05	4.005	8 186	1 / 25	0.382	1.050	<i>JJ.703</i> <i>J</i> 1 778	0.050	1	30.000	835 564	1.050	<i>JJ.703</i> <i>J</i> 1 778	25.051	11.134	0.050	4.175	20.870	835 564	
	11	0.20	0.05	4.517 1 711	8 9 2 8	1 / 58	0.907	0.906	38 882	0.030	1	12 038	555 /61	0.906	38 882	20.520	10.887	0.050	4.083	21.334	555 /61	
	12	0.20	0.07	4.714	7 912	1 360	0.382	1 090	J0.002 //1 050	0.070	1	37 663	JJJ.401 156 111	1 090	J0.002 /1 050	24.450	11 /0/	0.070	4.005	20.415	<i>4</i> 56 111	
	12	0.20	0.03	5 203	5 1/3	9 / 21	0.387	1 2/9	41.030	0.030	1	36.618	152/ 133	1 2/19	41.000	23.802	12 803	0.030	4.310	24.005	152/ 133	
	14	0.25	0.05	5.225	J.14J 4 104	9.796	0.382	1.245	47.076	0.050	1	17 225	050 526	1.245	47.076	20.000	12.005	0.050	5 029	24.005	050 526	
	15	0.25	0.05	1 001	6.047	0.200 9.750	0.984	0.857	47.570	0.030	1	47.22J	646 910	0.857	47.970	28 520	12 670	0.030	1 755	23.100	646 910	
	16	0.25	0.07	4.334 5 377	1 022	0.7JZ	0.929	1 157	40.204	0.070	1	12.013	540.056	0.857	40.204	20.525	12.075	0.070	5 1 9 0	25.774	540.510	
	10	0.25	0.09	5.277	4.055	0.230	0.987	1.157	49.413	0.090	1	42.094	545.050	1.157	49.415	51.151	13.830	0.090	5.185	23.943	549.050	
	1	0 10	0.03	2 012	2 413	1 238	0 998	1 947	10 354	0.030	1	5 317	345 127	1 947	10 354	6 5 2 3	2 899	0.030	1 087	5 436	345 127	
	2	0.10	0.05	1 857	1 334	6.011	0.998	1 552	11 104	0.050	1	7 156	222 072	1 552	11 104	6 995	3 109	0.050	1 166	5 829	222 072	
	2	0.10	0.05	1 941	1 800	8 4 2 9	0.900	0.821	10 938	0.050	1	13 315	156 253	0.821	10 938	6 891	3 063	0.050	1 148	5 742	156 253	
	4	0.10	0.07	1 971	1 147	5 476	0.999	1 431	12 107	0.070	1	8 460	134 526	1 431	12 107	7 628	3 390	0.070	1 271	6 356	134 526	
P3	5	0.15	0.03	2 150	3 101	1 711	0.907	0 601	9 975	0.030	1	16 / 26	370 157	0 601	9 875	6 221	2 765	0.020	1 027	5 1 9 /	270 157	
D	6	0.15	0.03	1 762	1 1/0	3 2 2 2 4	0.917	1 202	9.075	0.050	1	7 625	199 /30	1 202	9.075 9.072	6 282	2.705	0.050	1.037	5 225	100 /20	
	7	0.15	0.05	1 032	8 762	6.052	0.555	1 539	13 012	0.050	1	8 /61	185 880	1 529	13 012	0.202 8 107	3 6/2	0.050	1 366	6 8 3 1	185 880	
	, 8	0.15	0.07	2.026	1 /121	6 809	0.001	1.556	12 297	0.070	1	10 5 2 2	137 637	1.330	12 287	7 804	3 169	0.070	1 300	6 502	137 627	
	0	0.13	0.09	2.020	7 200	0.000	0.900	0.034	12.30/	0.090	1 1	19.320	110 522	0.054	12.30/	7.004	2 524	0.090	1 2 2 2	6 600	110 522	
	9	0.20	0.05	2.293	2.333	2.152	0.920	0.003	12.300	0.050	т	10.420	413.333	0.003	12.300	1.929	5.524	0.050	1.322	0.000	413.333	

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CONTINUE	u

	Run	Enz	yme nc.	Model Parameters				Model 1							Model 2							
Comp.	No.	E1	E2	а	b	Хо	R2	Smax (g/hr)	Δ (g)	δ (ml)	Θ (hr)	τ (hr)	Kp (g/ml)	S (g/hr)	Δ (g)	t (63%)	t (28%)	δ (ml)	Θ (hr)	τ (hr)	Kp (g/ml)	
	10	0.20	0.05	1 97/	1 / 16	1 100	0 896	1 11/	13 160	0.050	1	11 820	263 372	1 11/	13 160	8 296	3 687	0.050	1 3 8 3	6 91/	263 372	
	11	0.20	0.05	2 2 4 2	2 149	1 545	0.050	0.955	12 753	0.050	1	13 357	182 187	0.955	12 753	8 034	3.507	0.050	1 3 3 9	6 695	182 187	
	12	0.20	0.07	2.242	1 635	9 6/8	0.965	1 115	12.733	0.070	1	11 227	130 1/1	1 115	12 5 2 3	7 8 8 9	3 506	0.070	1 3 1 5	6 574	130 1/1	
	13	0.20	0.05	1 377	1.000	-3 522	0.960	0.521	5 963	0.030	1	11 439	198 777	0.521	5 963	3 757	1 670	0.030	0.626	3 131	198 777	
	14	0.25	0.05	1 483	1 193	-6 024	0.900	1 323	7 211	0.050	1	5 451	144 228	1 323	7 211	4 543	2 019	0.050	0.020	3 786	144 228	
	15	0.25	0.05	1 534	9 412	-1 580	0.995	1 1 2 3	7 760	0.050	1	6 910	110 851	1 1 2 3	7 760	4.545	2.013	0.050	0.757	4 074	110 851	
	16	0.25	0.07	1 490	8 748	-6 867	0.907	0.966	7 291	0.070	1	7 548	81 006	0.966	7 291	4 593	2.173	0.070	0.015	3 828	81 006	
	10	0.23	0.05	1.150	0.210	0.007	0.557	0.500	7.231	0.050	-	7.510	01.000	0.500	7.231	1.555	2.011	0.000	0.700	5.020	01.000	
	1	0.10	0.03	2.095	-5.021	-1.819	0.992	-8.754	-51.939	0.030	1	5.933	-1731.303	-8.754	-51.939	-32.722	-14.543	0.030	-5.454	-27.268	-1731.303	
	2	0.10	0.05	2.457	-7.074	-4.016	0.974	1.323	-49.907	0.050	1	-37.711	-998.138	1.323	-49.907	-31.441	-13.974	0.050	-5.240	-26.201	-998.138	
	3	0.10	0.07	1.146	-7.431	-3.629	0.993	0.887	-50.025	0.070	1	-56.415	-714.649	0.887	-50.025	-31.516	-14.007	0.070	-5.253	-26.263	-714.649	
	4	0.10	0.09	2.928	-6.913	-4.055	0.955	1.204	-49.771	0.090	1	-41.339	-553.016	1.204	-49.771	-31.356	-13.936	0.090	-5.226	-26.130	-553.016	
	5	0.15	0.03	4.064	-5.435	-3.362	0.969	1.137	-59.094	0.030	1	-51.973	-1969.800	1.137	-59.094	-37.229	-16.546	0.030	-6.205	-31.024	-1969.800	
	6	0.15	0.05	3.384	-5.779	-3.467	0.969	0.951	-56.664	0.050	1	-59.583	-1133.276	0.951	-56.664	-35.698	-15.866	0.050	-5.950	-29.748	-1133.276	
	7	0.15	0.07	8.417	-4.265	-2.957	0.918	1.315	-68.445	0.070	1	-52.049	-977.784	1.315	-68.445	-43.120	-19.165	0.070	-7.187	-35.934	-977.784	
Nd	8	0.15	0.09	4.756	-5.303	-3.367	0.933	0.753	-60.003	0.090	1	-79.737	-666.699	0.753	-60.003	-37.802	-16.801	0.090	-6.300	-31.502	-666.699	
	9	0.20	0.03	2.202	-5.353	-2.964	0.981	1.000	-59.924	0.030	1	-59.925	-1997.460	1.000	-59.924	-37.752	-16.779	0.030	-6.292	-31.460	-1997.460	
	10	0.20	0.05	3.346	-4.952	-2.957	0.978	0.978	-62.077	0.050	1	-63.481	-1241.546	0.978	-62.077	-39.109	-17.382	0.050	-6.518	-32.591	-1241.546	
	11	0.20	0.07	2.837	-5.231	-3.036	0.985	0.999	-60.161	0.070	1	-60.213	-859.441	0.999	-60.161	-37.901	-16.845	0.070	-6.317	-31.584	-859.441	
	12	0.20	0.09	4.318	-4.858	-3.030	0.974	1.098	-62.552	0.090	1	-56.945	-695.017	1.098	-62.552	-39.407	-17.514	0.090	-6.568	-32.840	-695.017	
	13	0.25	0.03	8.325	-4.695	-3.266	0.899	1.097	-60.785	0.030	1	-55.421	-2026.157	1.097	-60.785	-38.294	-17.020	0.030	-6.382	-31.912	-2026.157	
	14	0.25	0.05	9.664	-4.478	-3.186	0.886	1.033	-62.094	0.050	1	-60.086	-1241.886	1.033	-62.094	-39.119	-17.386	0.050	-6.520	-32.600	-1241.886	
	15	0.25	0.07	9.254	-4.395	-3.099	0.898	1.021	-64.961	0.070	1	-63.641	-928.014	1.021	-64.961	-40.925	-18.189	0.070	-6.821	-34.105	-928.014	
	16	0.25	0.09	1.116	-4.338	-3.156	0.847	1.010	-61.840	0.090	1	-61.204	-687.114	1.010	-61.840	-38.959	-17.315	0.090	-6.493	-32.466	-687.114	
E1: β–	amyla	ase, E	2: α–	amylas	e, a and	b: inpu	ıt variat	oles, Xo: ini	tial inpu	t varia	ble, R	^{2:} Regres	sion value,	Kp: Proces	s gain, δ:	magnitu	de of the	e input	change,	Δ: the m	nagnitude	

of the steady-state change in the output, S: the maximum slope of the output-versus-time plot, θ : intercept of maximum slope with initial value, τ is absolute. The negative values show the trend of the change.

There is a great need to develop their use in a different form and various industries like medical, food. Moreover, some alternative technological changes develop the enzyme's practicability of cost-effectiveness. There is a different solution in the literature, for example, surface functionalization of calixarene has been used for the effectiveness of immobilization of α amylase. α amylase was covalently immobilized with a glutaraldehyde-containing amino group functionalized calixarene. This technique was studied by Veesar et al. (2015). In this technique, imide bonds are formed between amino groups on the protein aldehyde groups on the calixarene surface. The result of different preparation conditions on the immobilized alpha-amylase process like immobilization time, enzyme concentration, temperature and pН were determined by these researchers. The result of hydrogen ion concentration and temperature changes on the activity of free and immobilized alpha-amylase was researched by using starch. The optimum reaction temperature and pH value were catalyzed by the immobilized alpha-amylase at 25 °C and 7 °C, respectively in the enzymatic conversion. Compared to the free enzyme, immobilized alpha-amylase retained 85% of its original activity, also showed thermal stability and excellent durability.

Further, another research was made by Talekar et al. (2013) that a tri-enzyme biocatalyst which name is combi-CLEAs with starch hydrolytic activity was set from pullulanase, alpha-amylase and glucoamylase. These enzymes are aggregating enzymes with ammonium sulfate which are cross bonding formed aggregates for 4.5 h with 40 mM glutaraldehyde. The biocatalyst was identified. Cross-linking and precipitant type were examined. Optimum pH and temperature changes from 6 to 7 and from 65 to 75 °C were examined after the co-immobilization of enzymes. Afterwards starch hydrolysis reaction in batch, separate CLEAs, combi-CLEAs and free enzyme mixtures were used for examining 60, 100 and 40% conversions. Furthermore, thermal stability increased with of enzymes were coimmobilization. Lastly, the catalytic activity of enzymes is preserved during starch hydrolysis up to 5 cycles without performance change in combi-CLEAs.

In the literature, there are also some different operations to determine their effect on starch conversion. Buckow et al. (2007) were studied in the barley malt, the effect of temperature and high hydrostatic pressure on the stability and catalytic activity of alpha-amylase were observed. Inactivation operations with alpha-amylase which include with and without calcium ions were done under 0.1-800 MPa pressure-and 30-75 °C temperature range. Ca²⁺ ions have a stabilizing effect on the enzyme at all pressure-temperature ranges. According to kinetic analysis, aberrations of simple first-order reactions were based on the existence of isoenzyme fractions.

Conclusion

Maltose syrup is a value-added product and it is characterized by having 50% of maltose content and less than 5% of dextrose. The exact sugar spectrum of maltose syrup varies from one producer to another and varies with the demand of customers and the experience of operators. Due to this fluctuation, the optimum enzyme concentration is accurately not determined during production.

The starch conversion process of the maltose syrup production was analyzed to determine the optimum enzyme concentrations, process control parameters and dynamics. Practically, in industrial production, the fluctuation in the enzyme and time consumption increase the maltose syrup production cost. The ideal operational and control parameters were determined for the desired product specification

In this study, maltose concentration is obtained as 50 % in this study by using; 0.15 ml E1+0.03 ml E2 enyzmes at sixth hour, 0.15 ml E1+0.07 ml E2 enyzmes at sixth hour, 0.2 ml E1+0.03 ml E2 enyzmes at sixth hour, 0.2 ml E1+0.05 ml E2 enyzmes at sixth hour, 0.2 ml E1+0.09 ml E2 enyzmes at sixth hour and 0.25 ml E1+0.03 ml E2 enzymes at third hour. As a result, optimum concentration of enzyme is 0.20 ml E1+0.05 ml E2 enzymes at sixth hour.

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Author Contribution: Sema Nur Çinçik made preparation of manuscript, experimental analysis and data analysis. Fatih Balcı made experimental design, statistical analysis, coordination and management of the paper. Mustafa Bayram made the process control studies and overall evaluation of the data.

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