

Study on Thiazolyl-Hydrazone Derivatives as Acetylcholinesterase Inhibitors

Yusuf Özkay¹, Leyla Yurttaş¹, Usama Abu Mohsen², Belgin Sever¹, Weiam Hussein¹, Ömer Öztürk¹, Begüm Nurpelin Sağlık¹, Ulviye Acar¹, Özge Nurhan Erdoğan¹, Ayşen Pekbağ¹, Zafer Asım Kaplancıklı^{1,3}

¹Anadolu University, Faculty of Pharmacy, Department of Pharmaceutical Chemistry, Eskişehir - Turkey

²Al-Azhar University, Faculty of Pharmacy, Department of Pharmaceutical Chemistry, Gaza - Palestine

³Anadolu University, Graduate School of Health Sciences, Department of Pharmaceutical Chemistry, Eskişehir - Turkey

Yazışma Adresi / Address reprint requests to: Zafer Asım Kaplancıklı

Anadolu University, Faculty of Pharmacy, Department of Pharmaceutical Chemistry, Eskişehir - Turkey

Elektronik posta adresi / E-mail address: zakaplan@anadolu.edu.tr

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ÖZET

Asetilkolinesteraz inhibitörü olarak tiyazolil-hidrazon türevleri üzerine çalışma

Amaç: Bu çalışmada, tiyazolün bazı hidrazon türevlerinin sentezlerinin ve antikolinesteraz aktivitelerinin araştırılması amaçlandı.

Yöntem: Piro-2-karboksaldehidler tiyosemikarbazit ile etanol içinde direkt olarak reaksiyona tabi tutuldu ve oluşan tiyosemikarbazon, α -bromoasetofenon türevleri ile (Hantzsch reaction) kondanse edilerek, 1-sübstitüe pirol-2-karboksaldehid (4-(4-sübstitüe fenil)-1,3-tiyazol-2-il)hidrazon türevlerini verdi. Bileşiklerin kimyasal yapıları, IR, ¹H-NMR ve FAB+-MS spektral verileri ve elementel analiz verileri ile aydınlatıldı. Modifiye edilmiş Ellman spektrofotometrik metodu kullanılarak, elde edilen tüm bileşiklerin asetilkolinesteraz (AChE) inhibisyonları incelendi.

Bulgular: Bileşik 1, AChE üzerinde %64.10 (1 mM) and 33.00 (0.1 mM) inhibisyon oranları ve IC₅₀=0.59 mM değeri ile en aktif antikolinesteraz molekül olarak belirlenmiştir.

Sonuçlar: Fenil halkası üzerinde para konumundan ve pirol halkasının 1. konumundan sübstitüsyonlar antikolinesteraz etkiyi negatif yönde etkilemiştir.

Anahtar sözcükler: Tiyazol, hidrazon, pirol, antikolinesteraz aktivite

ABSTRACT

Study on thiazolyl-hydrazone derivatives as acetylcholinesterase inhibitors

Objective: In this study we aimed to synthesize some hydrazone derivatives of thiazole and to evaluate their anticholinesterase activities.

Method: Pyrrole-2-carboxaldehydes were reacted directly with thiosemicarbazide in ethanol and then obtained thiosemicarbazones were condensed with α -bromoacetophenone derivatives (Hantzsch reaction) to give 1-substituted pyrrole-2-carboxaldehyde (4-(4-substituted phenyl)-1,3-thiazol-2-yl) hydrazones. The structures of the obtained compounds were elucidated by using IR, ¹H-NMR and FAB+-MS spectral data and elemental analyses results. In the pharmacological studies, anticholinesterase activities of these compounds have been evaluated by using modified Ellman's spectrophotometric method.

Results: The compound (1) can be identified as the most active anticholinesterase molecule due to its inhibitory effect on acetylcholinesterase with inhibition percentages of 64.10 (1 mM) and 33.00 (0.1 mM) % and also IC₅₀ value of 0.59 mM.

Conclusion: The substitutions on phenyl ring at para position and at first position of pyrrole ring have negatively affected anticholinesterase activity.

Key words: Thiazole, hydrazone, pyrrole, anticholinesterase activity

INTRODUCTION

Alzheimer's disease (AD) is the most common form of dementia, a serious brain disorder that impacts daily living through memory loss and cognitive changes. Although not all memory loss indicates AD, one in ten people over 65 years of age, and over half of those over 85 have AD.

Currently, it affects about 4.5 million Americans and close to 27 million people worldwide. AD destroys neurons in the cortex and limbic structures of the brain, areas that are responsible for learning, memory, behavior, emotion, and reasoning in humans. The devastating effects of AD include memory loss, dementia, and ultimately death (1-3). AD is mainly characterized by the pre-synaptic decrease of

acetylcholine (ACh) due to damage of cholinergic neurons in some special parts of the brain such as hippocampus and cortex (cholinergic hypothesis) (4,5). Concerning the cholinergic hypothesis one of the rational and effective approaches to treat the AD's symptoms, is raising the ACh through inhibition of acetylcholinesterase (AChE) that is responsible for hydrolysis of ACh in presynaptic areas (6,7).

In the last few decades, the chemistry of thiazoles and their fused heterocyclic derivatives has received considerable attention due to their synthetic and biological importance. Compounds bearing thiazole moiety have been reported to exhibit a wide spectrum of biological effects including anticholinesterase activity (8-14).

The literature survey reveals that, hydrazone group plays an important role for anticholinesterase activity (15,16).

Some researchers also carried out considerable research for novel cholinesterase inhibitors bearing pyrrole moiety previously (17-19).

In previous study (20), we identified the synthesis of title compounds and antimicrobial activity profile. In extending our work, motivated by the above data, we synthesized 1-substituted pyrrole-2-carboxaldehyde

(4-(4-substituted phenyl)-1,3-thiazol-2-yl)) hydrazones (1-14) again, confirmed the structures of the compounds and focused on their potential acetylcholinesterase inhibitor activities.

MATERIALS AND METHODS

Chemistry

All chemicals were purchased from Sigma-Aldrich Chemical Co. All melting points (m.p.) were determined by Electrothermal 9100 digital melting point apparatus and were uncorrected. Spectroscopic data were recorded with the following instruments: ¹HNMR, Bruker 400 MHz spectrometer; MS-FAB, VG Quattro Mass spectrometer and elemental analyses were performed on a Perkin Elmer EAL 240 elemental analyzer (Perkin-Elmer, Norwalk, CT, USA).

General procedure for synthesis of the compounds

Preparation of 1-substituted pyrrol-2-carbaldehyde thiosemicarbazones (A, B)

Thiosemicarbazide (5 mmol) and 2-formylpyrrole or N-methyl-2-formylpyrrole (5 mmol) was refluxed for 3 h in ethanol (40 mL). The resulting mixture was cooled, filtered

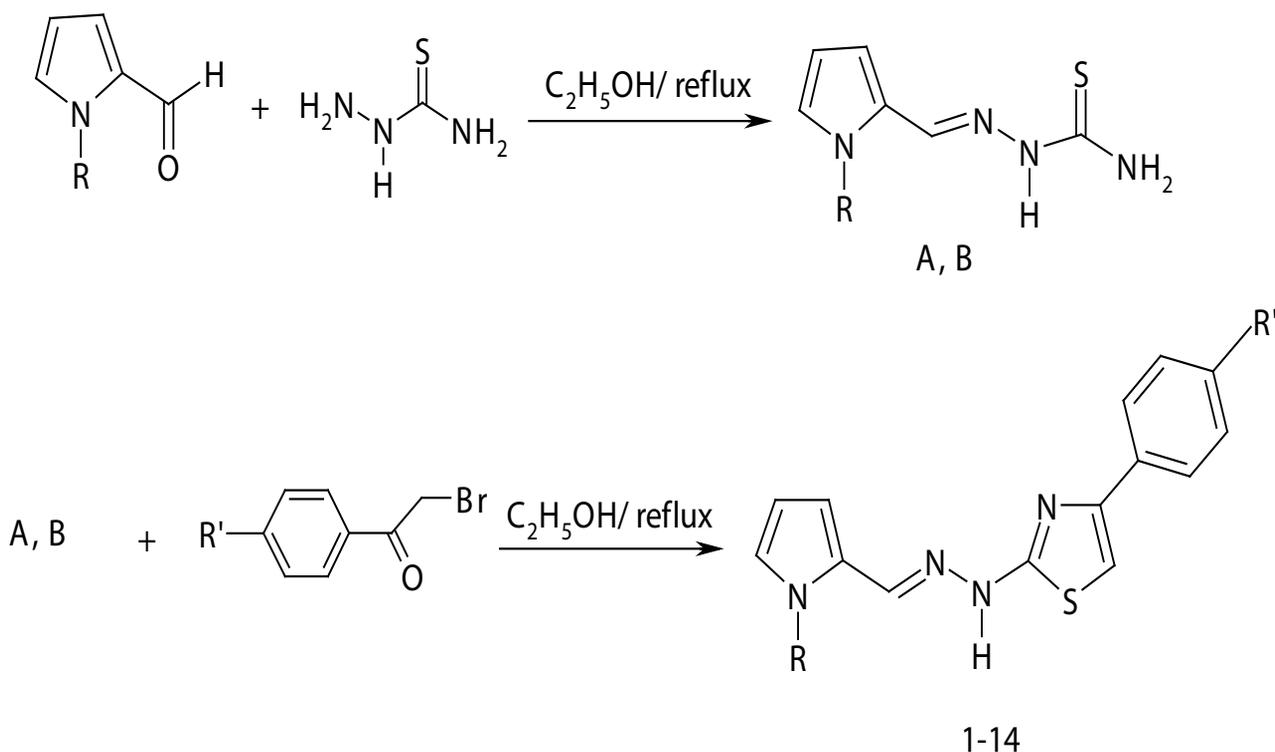


Figure 1: The synthetic protocol of the compounds

and recrystallized from ethanol (21).

Preparation of 1-substituted pyrrole-2-carboxaldehyde (4-(4-substituted phenyl)-1,3-thiazol-2-yl) hydrazones (1-14)

A or B (0.3 mmol) and α -bromoacetophenone (0.3 mmol) were stirred at room temperature in ethanol (10 mL). The precipitate was filtered and recrystallized from ethanol to give title compounds (Table 1) (Figure 1) (20).

Pyrrole-2-carboxaldehyde (4- phenyl-1,3-thiazol-2-yl) hydrazone (1):

IR (KBr) ν_{\max} (cm^{-1}): 3375-3110 (N-H), 3045 (aromatic C-H), 1621-1470 (C=C and C=N). $^1\text{H NMR}$ (DMSO- d_6) δ ppm: 6.10 (brs, 1H, pyrrole-H), 6.55 (brs, 1H, pyrrole-H), 6.95 (brs, 1H, pyrrole-H), 7.35 (s, 1H, C₅-H thiazole), 7.40–7.49 (m, 3H, Ar-H), 7.79–7.85 (m, 2H, Ar-H), 8.10 (s, 1H, -CH=N-N), 11.30 (s, 1H, pyrrole N-H), 11.85 (brs, 1H, C=N-NH). For C₁₄H₁₂N₄S

calculated: 62.66 % C, 4.51 % H, 20.88 % N; found: 62.62 % C, 4.50 % H, 20.92 % N.

MS (FAB) [M+1]⁺: m/z 269.

Pharmacology

AChE Inhibition

All compounds were subjected to a slightly modified method of Ellman's test (22) in order to evaluate their potency to inhibit the AChE. The spectrophotometric method is based on the reaction of released thiocholine to give a colored product with a chromogenic reagent 5,5-dithio-bis (2-nitrobenzoic acid) (DTNB). AChE, (E.C.3.1.1.7 from Electric Eel, 500 units), and Donepezil hydrochloride were purchased from Sigma–Aldrich

Table 1: Some characteristics of the compounds

Comp.	R	R'	Mp (°C)	Yield %	Mol. Formula
1	H	H	125	85	C ₁₄ H ₁₂ N ₄ S
2	H	CH ₃	157	80	C ₁₅ H ₁₄ N ₄ S
3	H	OCH ₃	155	80	C ₁₅ H ₁₄ N ₄ OS
4	H	Br	171	75	C ₁₄ H ₁₁ BrN ₄ S
5	H	Cl	200	82	C ₁₄ H ₁₁ ClN ₄ S
6	H	F	165	90	C ₁₄ H ₁₁ FN ₄ S
7	H	NO ₂	201	82	C ₁₄ H ₁₁ N ₅ O ₂ S
8	CH ₃	H	224	70	C ₁₅ H ₁₄ N ₄ S
9	CH ₃	CH ₃	183	75	C ₁₆ H ₁₆ N ₄ S
10	CH ₃	OCH ₃	210	75	C ₁₆ H ₁₆ N ₄ OS
11	CH ₃	Br	214	70	C ₁₅ H ₁₃ BrN ₄ S
12	CH ₃	Cl	202	85	C ₁₅ H ₁₃ ClN ₄ S
13	CH ₃	F	222	73	C ₁₅ H ₁₃ FN ₄ S
14	CH ₃	NO ₂	199	72	C ₁₅ H ₁₃ N ₅ O ₂ S

Table 2: % AChE inhibition of the compounds and IC₅₀ values

Comp.	AChE Inhibition (%)		
	1 mM	0.1 mM	IC ₅₀ (mM)
1	64.10±4.13**	33.00±2.32**	0.59±3.64
2	ND	ND	ND
3	ND	3.05±2.36**	> 1
4	ND	4.22±3.52**	> 1
5	ND	ND	ND
6	ND	ND	ND
7	ND	ND	ND
8	17.23±4.56**	12.93±3.45**	> 1
9	19.74±5.42**	15.14±3.47**	> 1
10	18.05±2.54**	22.52±4.87**	> 1
11	17.41±3.69**	11.89±2.78**	> 1
12	ND	ND	ND
13	ND	ND	ND
14	ND	ND	ND
Donepezil	99.01±4.89	95.52±5.01	0.054±0.002 (μM)

ND: Not determined. **p< 0.01 (unpaired Student's t test between test compound and Donepezil)

(Steinheim, Germany). Potassium dihydrogen phosphate, DTNB, potassium hydroxide, sodium hydrogen carbonate, gelatine and acetylthiocholine iodide (ATC) were obtained from Fluka (Buchs, Switzerland). Spectrophotometric measurements were performed on a 1700 Shimadzu UV-1700 UV-Vis spectrophotometer. Cholinesterase activity of the compounds (1-14) was measured in 100 mM phosphate buffer (pH 8.0) at 25 °C, using ATC as substrates. DTNB (10 mM) was used in order to observe absorbance changes at 412 nm. Donepezil hydrochloride was used as a positive control (Table 2) (23).

Enzymatic assay

Enzyme solutions were prepared in gelatin solution (1%), at a concentration of 2.5 units/mL. AChE and compound solution (50 µL) which is prepared in 2% DMSO at 0.1 and 1 mM concentrations were added to 3.0 mL phosphate buffer (pH 8±0.1) and incubated at 25 °C for 5 min. The reaction was started by adding DTNB (50 µL) and ATC (10 µL) to the enzyme-inhibitor mixture. The production of the yellow anion was recorded for 10 min at 412 nm. As a control, an identical solution of the enzyme without the inhibitor is processed following the same protocol. The blank reading contained 3.0 mL buffer, 50 µL 2% DMSO, 50 µL DTNB and 10 µL substrate. All processes were assayed in triplicate. The inhibition rate (%) was calculated by the following equation:

$$\text{Inhibition \%} = (A_c - A_i) / A_c \times 100$$

Where A_i is the absorbance in the presence of the inhibitor, A_c is the absorbance of the control and A_b is the absorbance of blank reading. Both of the values were corrected with blank-reading value. SPSS for Windows 15.0 was used for statistical analysis. Student's t- test was used for all statistical calculations. Data were expressed as Mean±SD.

RESULTS AND DISCUSSION

In this work, we synthesized fourteen thiazolyl-hydrazone derivatives, which had been synthesized previously in order to check their antimicrobial activity as

described in the introduction section. Structure elucidation of those compounds was explained in detail in the referenced literature and these data were used to check the characteristics of the compounds synthesized again in this study (20). It was observed that, all the spectral data were in agreement with the spectral data of the previous study. IR, ¹H-NMR and FAB⁺-MS spectral data and elemental analyses results were given only for a sample compound (1).

The anticholinesterase effects of the compounds (1-14) were determined by modified Ellman's spectrophotometric method (Table 2). Among these compounds (1-14), compound 1 including non-substituted phenyl moiety in its structure was found as the most active compound with inhibition values of 64.10 and 33.00% at 1 and 0.1 mM concentrations, respectively. The IC₅₀ value was calculated as 0.59 mM only for this compound. The inhibition percentages were not determined for compounds 2-7, 12-14 and these compounds were evaluated as inactive at two tested concentrations. Compound 10 bearing 1-methyl pyrrole and 4-methoxy phenyl moieties exhibited moderate anticholinesterase activity with 18.05 and 22.52% inhibition values. The other compounds 8, 9, 11 showed relatively weak activity and the inhibition values were found less than 19.74%. Standard drug Donepezil was studied at lower concentrations for the purpose of finding IC₅₀ value and it was determined as 0.054 µM. None of the compounds showed comparable activity with Donepezil and significant anticholinesterase activity contrary to expectations.

CONCLUSION

In conclusion, a series of thiazolyl-hydrazone derivatives were synthesized and studied for their anticholinesterase activities. According to the activity test results, compound 1 exhibited the highest anticholinesterase activity with a value of IC₅₀=0.59 mM and none of the compounds showed activity as much as standard drug Donepezil. The results indicate that the substitutions on phenyl ring at para position and at first position of pyrrole ring have negatively affected anticholinesterase activity.

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