

# Nanocomposite Nanofibers of Polyacrylonitrile (PAN) and Silver Nanoparticles (AgNPs) Electrospun from Dimethylsulfoxide

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## ABSTRACT

In this study, DMSO was used as the solvent and composite nanofibers of PAN with 1 and 3w% AgNO<sub>3</sub> content were electrospun. Then silver nanoparticles were in situ synthesized by chemical reduction. The effect of silver nitrate amount on the morphology, conductivity and mechanical properties of PAN/AgNPs composite nanofibers were investigated. Beadless and uniform composite nanofibers, the diameters of which were in the 499-515 nm range, were successfully electrospun. The breaking stress and breaking elongation of PAN/Ag composite nanofibers were higher than the neat PAN nanofibers. The conductivity was improved to around 10-8S/cm with the incorporation silver nanoparticles.

**Keywords:** Conductive, electrospinning, nanocomposite, nanofiber, polyacrylonitrile, silver nanoparticles.

## I. INTRODUCTION

Polymer composites containing metal nanoparticles have attracted great attention because of their unique optical, electrical, and catalytic properties. The properties of these nanocomposites strongly depend on the amount, size and dispersion quality of the metal nanoparticles [1]. The mechanical mixing of metal nanoparticles into the polymer solution is an easy method with the disadvantage of agglomeration of the particles [2]. Recently in situ synthesis of metal nanoparticles has attracted more attention. Silver nanoparticles can be synthesized in polymers using many different methods such as photo reduction, heat treatment, chemical reduction using aqueous solution of sodium borohydride and hydrazinium hydroxide [3-6].

In literature, there are many studies about the synthesis of silver nanoparticles in polyacrylonitrile in many of which N,N-dimethylformamide (DMF) is used as the solvent [3-6] due to its ability to reduce Ag ions to the metallic silver even at room temperature without the use of any external reducing agent [1]. In this study, DMSO was used as the solvent and composite nanofibers of PAN with 1 and 3w% AgNO<sub>3</sub> content were electrospun. Then silver nanoparticles were in situ synthesized by chemical reduction. The effect of silver nitrate amount on the morphology, conductivity and

mechanical properties of PAN/AgNPs composite nanofibers were investigated. Beadless and uniform composite nanofibers, the diameters of which were in the 499-515 nm range, were successfully electrospun. The breaking stress and breaking elongation of PAN/Ag composite nanofibers were higher than the neat PAN nanofibers. The conductivity was improved to around 10-8S/cm with the incorporation silver nanoparticles.

## II. MATERIALS AND METHOD

### 2.1. Materials

Polyacrylonitrile (PAN) (Sigma Aldrich, 181315, average Mw: 150.000g/mol), silver nitrate (AgNO<sub>3</sub>, Alfa Aesar Premion, 10858) and dimethylsulfoxide (DMSO) were used as received.

### 2.2. Methods

#### 2.2.1. Preparation of the solutions

1w% and 3w% AgNO<sub>3</sub> (with respect to the weight of PAN) were added to the required amount of DMSO and homogenized with ultrasonic tip for 10 minutes and with ultrasonic bath for 45 min. Then PAN was added to the dispersion. Magnetic stirrer was used to mix the solution

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at 40°C for 3 hours. The concentration of PAN was kept constant as 7 w%.

### 2.2.2. Electrospinning

Electrospinning was performed using equipment in which a horizontal electric field was generated between the nozzle and the rotating collector. It contained syringe pump and a grounded rotating collector. The electrospinning solution was fed using a syringe of 10mL through a capillary tip with a diameter of 1.25 mm. A high voltage power supply (0–50 kV) was used to apply a voltage of 15kV and a syringe pump was used to feed the electrospinning solution at a constant rate of 1mL/h. The distance between the nozzle and the collector was set as 10 cm.

### 2.2.3. In-situ synthesis of silver nanoparticles

Chemical reduction process using hydrazinium hydroxide was applied for the in-situ synthesis of silver nanoparticles. For this, piece of the as-spun nanowebs was immersed into the aqueous solution of hydrazinium hydroxide (1:20 hydrazinium hydroxide: distilled water) for 30 min at room temperature, then washed with 100 mL distilled water two times and dried in room temperature to obtain composite nanofibers containing Ag nanoparticles.

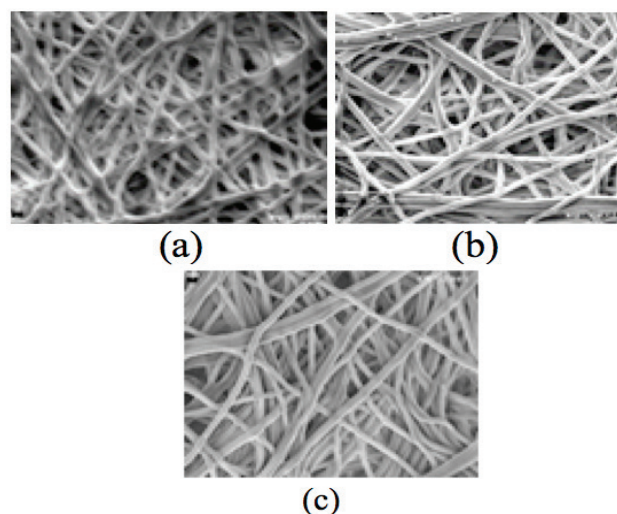
### 2.2.4. Characterization

Scanning electron microscopy (SEM; EVO MA 10) was used to take the images of pure PAN and composite nanowebs. The diameters of at least 50 randomly selected nanofibers were measured using Image Analysis Software. Mechanical properties of the nanowebs were measured using a tensile tester with a 100N load cell at a crosshead speed of 20 mm/min. The specimens were cut as 35mm in length and 5mm in width. The gage length was 15 mm. The thicknesses of the specimens were measured with a Mitutoyo digital micrometre. Resistance measurements were performed using a two-probe system connected to Microtest 6370 LCR meter with a four-wire system and conductivity in S/cm was calculated using the volume resistance value measured and the geometric dimensions of the samples.

## III. RESULTS AND DISCUSSION

### 3.1. Morphology

The SEM images of the pure PAN and composite nanowebs taken with 10.0kX magnification can be seen in Figure 1.



**Figure 1.** SEM images of (a) pure PAN, (b)1w% AgNO<sub>3</sub>/PAN, (c)3w% AgNO<sub>3</sub>/PAN nanowebs.

The nanofibers had uniform structures. On addition of AgNO<sub>3</sub> and application of the reduction process, the average diameter of the nanofibers didn't change significantly at AgNO<sub>3</sub> contents of 1 and 3w%. While the diameter of pure PAN nanofibers was 515.34nm, 1 w% and 3w% AgNO<sub>3</sub> addition followed by the chemical reduction process resulted in diameters of 499 nm and 514 nm, respectively. The additions of nanoparticles are expected to result in an increase in nanofiber diameter due to the increase in solution viscosity. On the other hand, the chemical reduction process is observed to result in a decrease in nanofiber diameter while the increase in net charge density is known to have thinning effect. The combination of all these factors resulted in a compensation in nanofiber diameter for the AgNO<sub>3</sub> contents of 1w%and 3w% and no significant variation was observed in nanofiber diameter.

### 3.2. Mechanical Properties

Table 1 shows the mechanical properties of pure PAN and nanocomposite nanofibers containing AgNPs.

**Table 1.** Mechanical properties of composite nanofibers.

Samples	Breaking Strength (MPa)	Breaking Elongation (%)	E-modulus (MPa)
PAN-DMSO	8.64	8.95	100.59
1w% AgNO <sub>3</sub> /PAN (reduced)	13.81	36.41	73.00
3w% AgNO <sub>3</sub> /PAN (reduced)	11.24	18.69	39.23

The breaking strength and breaking elongation values increased with addition of 1w% AgNO<sub>3</sub> and then decreased with the increase in AgNO<sub>3</sub> content from 1w% to 3w% still being higher than the pure PAN nanoweb. The improvements in the breaking strength and the breaking elongation were attributed to the increased compactness of the nanoweb as a result of the chemical reduction process. Besides; the nanosilver as an inorganic filler material might have had a positive effect on the breaking strength. The decrease in breaking strength with the increase in the additive amount might have been due to the agglomeration of the nanoparticles. There was a decrease in E-modulus values which showed that stiffness of the nanoweb decreased with the addition of AgNO<sub>3</sub> and subsequent reduction process.

### 3.3. Conductivity

Table 2 shows the conductivity values of nanocomposite nanofibers produced with the addition of AgNPs.

**Table 2.** Conductivity values of nanocomposite nanofibers.

Samples	Conductivity S/cm	Standard deviation	Coefficient of variation
1w% AgNO <sub>3</sub> / PAN	4.56*10 <sup>-8</sup>	1.53*10 <sup>-8</sup>	33.5
3w% AgNO <sub>3</sub> / PAN	3.79 *10 <sup>-8</sup>	1.12*10 <sup>-10</sup>	29.6

Pure PAN nanoweb has a reported conductivity value of 10-12 S/cm [7]. The in-situ synthesis of silver nanoparticles resulted in an increase in the conductivity of PAN nanofibers. The conductivity was measured as 10-8 S/cm for the composite nanowebs produced. With the conductivity of 10-8S/cm, they can be utilized in antistatic applications [8].

### III. CONCLUSION

Polyacrylonitrile nanofibers with AgNO<sub>3</sub> from electrospinning solutions prepared with DMSO were successfully electrospun at the loadings of 1 and 3w%. It was demonstrated in this study that dimethylsulfoxide was a suitable solvent for the production of PAN/AgNPs nanowebs. While the diameter of the nanofibers didn't change significantly with addition of AgNO<sub>3</sub> followed by the chemical reduction process at loadings of 1w% and 3w%, the mechanical properties improved with the addition of 1w%AgNO<sub>3</sub>. Increase in the content of AgNO<sub>3</sub> resulted in a decrease in breaking strength and breaking elongation while they were still higher than the mechanical properties

of pure polyacrylonitrile nanoweb. The composite nanowebs showed a conductivity on the order of 10-8S/cm which was in the suitable range for antistatic applications. It was concluded that AgNO<sub>3</sub> could be used as a filler developing antibacterial properties while improving both mechanical and electrical properties.

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