

Serum Selenium Levels in Ischaemic Heart Disease

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Abstract

Selenium is an essential element and a cofactor required to maintain glutathione peroxidase activity. Its deficiency may induce modification in the cellular antioxidant status and the appearance of different diseases such as cardiovascular diseases and cancer. This study was conducted to determine whether an association exists between serum selenium levels and ischaemic heart disease. Serum selenium concentrations of the test and control groups were measured with hydride generation atomic absorption spectrometry. Statistical analysis indicated that there is a significant difference between the serum selenium levels of two groups.

Key Words: Serum selenium, ischaemic heart disease, cellular antioxidants.

Introduction

The relation between trace elements and human health has been scarcely studied. With respect to cardiovascular diseases and hypertension, attention has mostly focused on arsenic, cobalt, copper, chromium, fluorine, manganese, vanadium, zinc, selenium, silicon, cadmium and lead. Environmental contamination can influence organ concentrations through long-term, low-level effects. Attention is paid to interpretation problems due to the complexity of biochemical interactions with proteins of various sorts which determine metabolic processes and to the occurrence of detoxification mechanisms in which trace elements interact. This can also lead to strong variations in individual vulnerability. In general the elements selenium, copper, zinc, chromium and manganese seem to counteract the development of cardiovascular diseases.

Depending on its concentration, selenium can either be beneficial or toxic to humans and certain plants and animals.

The key role of selenium in mammalian metabolism is attributed to the presence of four selenocysteine residues in the active site of the enzyme glutathione peroxidase. This enzyme catalyzes the reduction of organic hydroperoxides- in particular, hydrogen peroxide and thus as an endogenous antioxidant is important for intracellular defense. Selenium, functioning as a part of glutathione peroxidase, has been recognized as a cellular antioxidant (1). In addition,

the element is also known as a protective agent against heavy metal toxicity (2-4), cancer (5-7), and cardiovascular diseases (8-11). Lack of selenium or disturbance in its metabolism by metals may promote free radical production (12). Interest in the biochemical role of selenium has noticeably increased in the last decades and many epidemiological studies have been performed to investigate its potential protective role in preventing carcinogenesis and other chronic diseases (13-24). Low blood levels of the element have been reported in individuals from several countries, and its deficiency has been associated with several pathological conditions (25-32). Dietary selenium deficiency in people in certain areas of China is associated with an endemic cardiomyopathy called "Keshan disease", which affects primarily children and women of childbearing age (33,34). In the industrialized West, dietary selenium deficiency is thought to be associated with cardiovascular diseases. A prospective epidemiological study done in Finland revealed the selenium concentration in serum to be inversely related to the risk of cardiovascular disease (35,36). However, other prospective epidemiological studies found no such association (37-40).

Materials and Methods

Sample Preparation

Following an overnight starvation, 10 ml of venous blood was collected from the patients and the control group and centrifuged at 3500 rpm for 15 min. Serum samples stored in polystyrene tubes were prevented from metal contamination and kept in deep-freezer at -20°C. Biochemical blood parameters of two groups were recorded using an autoanalyzer (Roche, Cobas Mira Plus).

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Selenium Analysis

1 ml of serum sample was added into a test tube and was digested by wet acid ashing using 2 ml of concentrated (65%) nitric acid. Tubes were placed in the acid digestion oven and the temperature was raised gradually up to 140°C within 15 minutes. After keeping 25 min in the oven and cooling down to room temperature, a mixture of concentrated sulphuric acid and 60% perchloric acid (1 : 0.4 ml) was added to each tube and the temperature was raised step by step up to 310°C within 2 h. The clear, colourless samples were removed from the oven and redissolved in 5 ml of 37% hydrochloric acid and kept 20 min in digestion oven at 90 °C. Upon cooling, the total volume of each tube was completed to 40 ml with bidistilled water.

Analytical Instrumentation

Ultrapure grade chemicals (Merck and Sigma) were used throughout. Water was purified by ion-exchange and double-distillation.

High purity Se was stock material and used for preparation of standard. Working standards and a standard blank were prepared with the same procedures.

Measurements were carried out on a Varian Spectra AA-30/40 absorption spectrophotometer equipped with Varian VGA-76 hydride generation system. The instrument was calibrated for the Se analysis using standards containing 20, 40, 60, 80 and 100 µg Se per litre (41).

Hydride generation of selenium was based on the reaction of acidic digest with a sodium tetraborohydride/ perchloric acid reaction system. The hydride was transferred into a reaction cell which was heated by an ethylene- air flame for the decomposition of the hydride. Absorbance values were measured at 196 nm.

The mean absorbances produced by the standards (corrected for the standard blank) were plotted vs the concentrations of the Se in the standards. The concentrations of the analyte in the original samples were obtained using the standard curve in units of µg Se /L.

Student -t test was employed for the statistical evaluation of the results while r-correlation test was used for the correlation analysis (42).

Results and Discussion

In this study of 20 patients with ischaemic ECG findings at exercise having mean age of 49 years were

under investigation and their serum Se concentrations had no association with plasma HDL cholesterol ($r= 0.2260$; $p= 0.338$). On the other hand, they had a lower mean serum selenium than others ($35.05 \mu\text{g/L}$ vs $61.39 \mu\text{g/L}$, $p=0.046$ for difference). Because of $p<0.05$, it was concluded that there is a difference between the variances of two groups.

Our data is in agreement with those of Oster et al. (43) and Salonen et al. (8, 35, 36). In another study effect of selenium supplementation was evaluated in 81 patients with acute myocardial infarction in a double-blind, placebo-controlled study. The results encourage further studies to evaluate the efficiency of antioxidants in the prevention and therapy of myocardial infarction (9).

Selenium is a naturally occurring trace element that is essential for animal and human nutrition. But range between dietary requirements and toxic levels is relatively narrow. Therefore, further epidemiological studies related to selenium levels in different tissues throughout the Turkish population should be carried out. In addition to these, quantitative status of trace elements together with selenium should be investigated in soil, major foods and crops like ones done in some other countries such as Saudi Arabia (44, 45) because of the very limited data according to the intensity of the counter effects of selenium deficiency on public health.

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