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SYNTHESIS OF NANO AND MICRO PARTICLES CuSbSe₂ - ONE OF HIGH RESPECTIVE SOLAR CELL COMPONENTS

NANO VE MİKRO PARÇACIKLAR CuSbSe₂ SENTEZİ - YÜKSEK İLGİLİ GÜNEŞ HÜCRE BİLEŞENLERİNDEN BİRİ

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ABSTRACT

The mix potassium antimonytitrate with chloride of copper (I) is dissolved in ethylene-glycol and to it increases as seleniruyushchy reagent selenidesulphate sodium. The experimental ware in a teflon ditch is located in the microwave electric oven. Test within 10 hours at 180⁰C to temperature remains in the furnace. The received deposit is filtered via the glass filter, washed out by the diluted solution of hydrochloric acid, ultrapure water, at last, ethyl alcohol, is dried up at 60-70⁰C in vacuum. The exit makes 92-94%. Chemical, thermographic and morphological analyses CuSbSe₂ are executed, and is established that crystals of connection are presented in the form of nanocores.

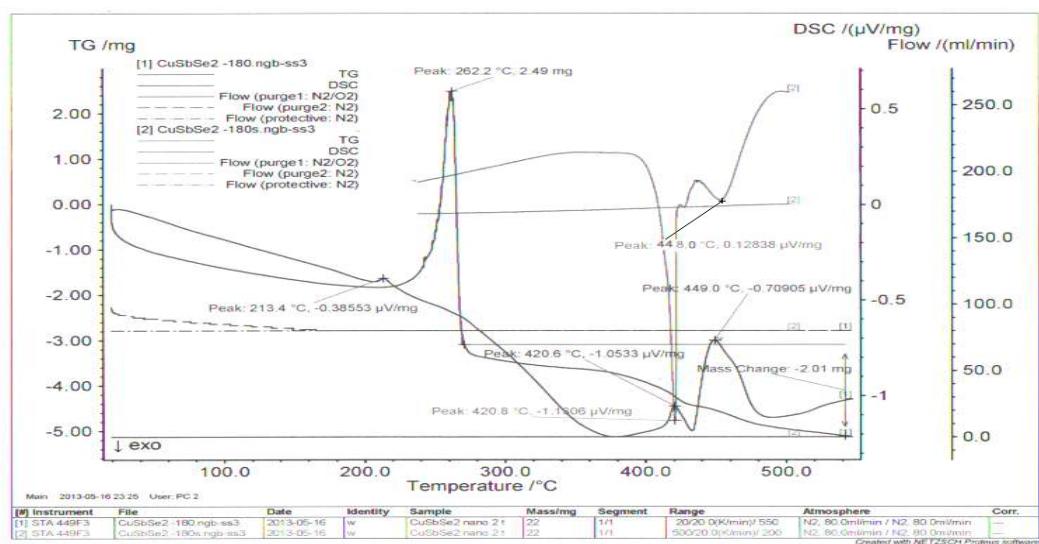


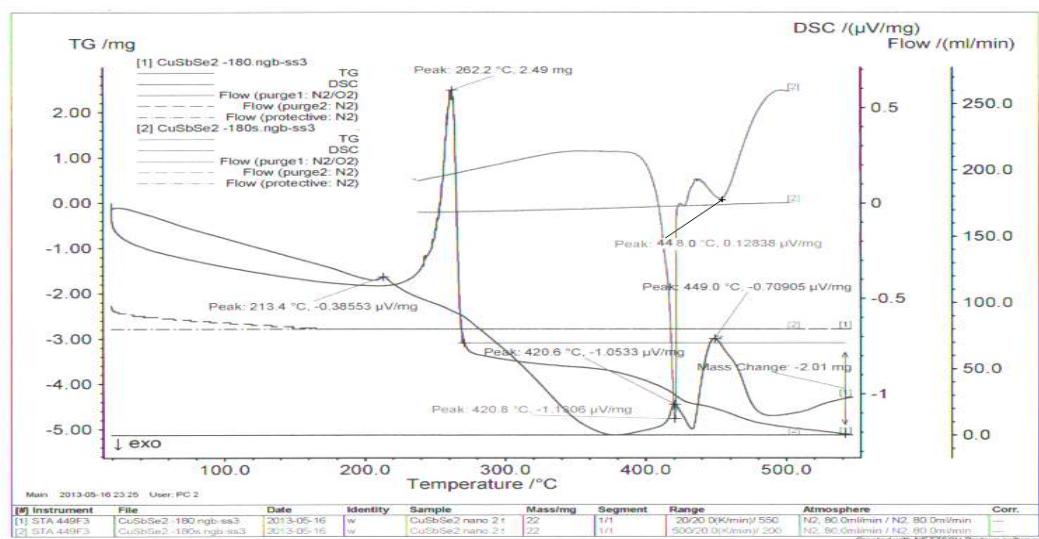
Figure 1. Thermogravimetric and differential and calorimetric analyses of nanoconnection of CuSbSe₂ received at 180⁰C within 10 hours.

Keywords

Antimony copper selenide, solvothermal method, chemical analysis, thermographic analysis, nanorod.

ÖZET

Kaliumantimoniltartrat ile bakır (I) klorür karışımı etilen glikol çözülverek üzerine selenidləşdirici reaktif olarak sodyum selenesulfat çözeltisi ilave edilir. Deney kabı teflon kütvetdə Speedwave four mikrodalga elektrik qızdırıcısında 180^0C de 10 saat süreyle saklanır. Alınan çöküntü szülür, zayıf hidroklorik asit solüsyonu, ultra saf su ve alkoller yıkandıktan sonra $60 - 70^0\text{C}$ de vakumda kurutulur. Çıxım 92 - 94% teşkil etmiştir. Alınan CuSbSe₂-in kimyasal, termoqrafik, morfolojik analizleri yapıldı ve kristaller nanoçubuqdan ibaret olduğu belirlenmiştir.



Şekil 1: 180^0C -de ve 10 saat süreyle alınmış CuSbSe₂ nanobirləşməsinin termoqrafimetrik ve ayırıçı kalorimetrik analizi

Anahtar Kelimeler

Bakır sürme selenid, solvotermal yöntem, kimyasal analiz, termoqrafik analiz, nanoçubuq.

Kaynaklar / References

- [1] Ding Tang, Jia Yang, Fangyang Liu//One-Step Electrodeposition and Annealing of CuSbSe₂ Thin Films. *Electrochemical and Solid-State Letters (ESL)*, 2011, China.
- [2] Tan D, Jan of J, Liu FY//Growth and the characteristic of CuSbSe₂ of the thin films received by a method of electrolytic sedimentation. *Electrochimica*.
- [3] Chongyin Yang, Yaoming Wang, Shaotang Li//CuSbSe₂-assisted sintering of CulnSe₂ at low temperature. *Journal of Materials Science*, 2012, Volume 47, Issue 20, pp 7085-7089.