

# COMMUNICATIONS

DE LA FACULTÉ DES SCIENCES  
DE L'UNIVERSITÉ D'ANKARA

Série B Chimie

---

TOME 28

ANNÉE 1982

---

**Synthesis of some o.o'-Dihydroxy Schiff Bases**

By

Birgül ERK and Neelâ GÜNDÜZ

8

Faculté des Sciences de l'Université d'Ankara  
Ankara, Turquie

# Communications de la Faculté des Sciences de l'Université d'Ankara

Comité de Redaction de la Série B

E. Alper, Ş. Gümüş, T. Gündüz, C. Tüzün Y. Sarıkaya,

Secrétaire de Publication

Ö. Çakar

---

La Revue "Communications de la Faculté des Sciences de l'Université d'Ankara" est un organe de publication englobant toutes les disciplines scientifiques représentées à la Faculté des Sciences de l'Université d'Ankara.

La Revue, jusqu'à 1975 à l'exception des tomes I, II, III était composée de trois séries

Série A: Mathématiques, Physique et Astronomie,

Série B: Chimie,

Série C: Sciences Naturelles.

A partir de 1975 la Revue comprend sept séries:

Série A<sub>1</sub>: Mathématiques,

Série A<sub>2</sub>: Physique,

Série A<sub>3</sub>: Astronomie,

Série B: Chimie,

Série C<sub>1</sub>: Géologie,

Série C<sub>2</sub>: Botanique,

Série C<sub>3</sub>: Zoologie.

En principe, la Revue est réservée aux mémoires originaux des membres de la Faculté des Sciences de l'Université d'Ankara. Elle accepte cependant, dans la mesure de la place disponible les communications des auteurs étrangers. Les langues Allemande, Anglaise et Française seront acceptées indifféremment. Tout article doit être accompagnés d'un résumé.

Les articles soumis pour publications doivent être remis en trois exemplaires dactylographiés et ne doit pas dépasser 25 pages des Communications, les dessins et figures portés sur les feuilles séparées devant pouvoir être reproduits sans modifications.

Les auteurs reçoivent 25 extraits sans couverture.

l'Adresse : Dergi Yayın Sekreteri,  
Ankara Üniversitesi,  
Fen Fakültesi,  
Beşevler-Ankara

# Synthesis of some *o,o'*-Dihydroxy Schiff Bases

by

Birgül ERK and Neclâ GÜNDÜZ

(Department of Chemistry, Faculty of Science, Ankara University)

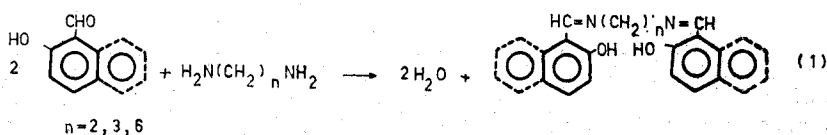
## ABSTRACT

Some Schiff bases have been prepared by using *o*-hydroxy benzaldehyde and  $\beta$ -hydroxy- $\alpha$ -naphthaldehyde with some diamines in medium of methyl alcohols. Their properties have been examined.

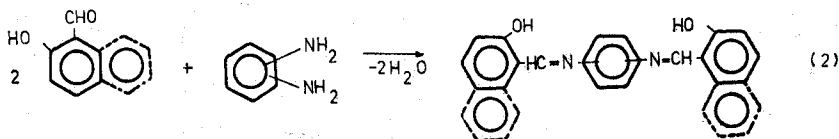
## INTRODUCTION

*o,o'*-dihydroxy schiff bases give very strong 5 or 6 membered chelate rings complexes with transition element cations (1, 2). In this work some *o,o'*-dihydroxy schiff bases like polymethylene dialdimine, polyethyleneamine dialdimine, 4,4'-(dialdimine), diphenyl methane and *p*-*N,N*-dimethyaminophenyl aldimine, have been prepared and their properties have been examined by spectrophotometrically.

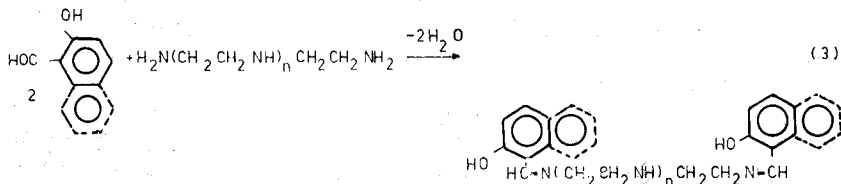
Polymethylenedialdimines were obtained by condensation of polymethylene dialdimines with salicylaldehyde and  $\beta$ -hydroxy- $\alpha$ -naphthaldehyde in medium of methyl alcohols (Equation-1).



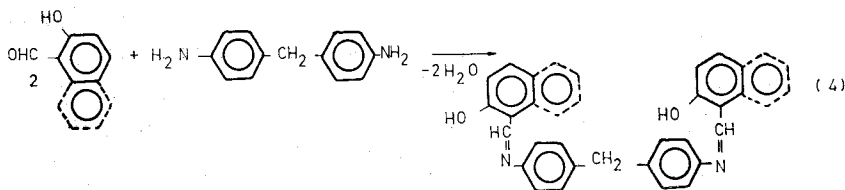
Phenylene dialdimines were obtained by condensation of phenylene diamine dihydrochlorides with salicylaldehyde and  $\beta$ -hydroxy- $\alpha$ -naphthaldehyde in medium of pyridine (Equation-2).



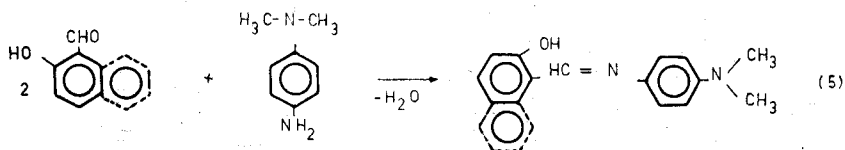
Polyethyleneamine dialdimines were obtained by condensation of polyethylene amines with salicylaldehyde  $\beta$ -hydroxy  $\alpha$ -naphthaldehyde in medium of methylalcohol. Only salicylaldimines are liquid at room temperature and were purified by chromatographic method (Equation-3).



The compound 4,4'-(dialdimine) diphenylmethane were obtained by condensation of 4,4'-diamino diphenylmethane with salicylaldehyde and  $\beta$ -hydroxy  $\alpha$ -naphthaldehyde in medium of methylalcohol (Equation-4).



*p*-N,N-dimethylaminophenylaldimines were obtained by condensation of N,N-dimethyl-*p*-phenylene diamine dihydrochloride with salicylaldehyde and  $\beta$ -hydroxy  $\alpha$ -naphthaldehyde in medium of methylalcohol (Equation-5).



One of the striking properties of o-hydroxy schiff bases is to give intramolecular hydrogen bondings (like O - H... N). Because of this hydrogen bonding in their I.R. Spectra a large broadening is observed in region of O - H vibrations, covering  $3600-2800\text{ cm}^{-1}$  (3). This seems to be in great agreement with work done by Freedman (4).

Studies on the schiff bases also gave a sharp absorption band in the region of  $1640-1633\text{ cm}^{-1}$  (5, 6) which is due to  $C = N$  double bond.

In their UV spectra all schiff bases showed absorption band round about  $330-320\text{ nm}$  due to  $C = N$  double bond.

### EXPERIMENTAL

#### Propane-1,3-disalicylaldimine:

24.42 g (0.2 M) salicylaldehyde dissolved in 50 cc methyl alcohol is placed in a 250 cc balloon. 7.41 g (0.1 M) 1,3-diamino propane is added dropwise on to the solution which is stirred and cooled down to  $-10^\circ\text{C}$ . After little time yellow crystals formed. They are filtered off and collected on a buchner funnel and recrystallized from petroleum ether. Yield 75% m.p  $59^\circ\text{C}$ .

#### Propane-1,3-di ( $\beta$ -hydroxy) - $\alpha$ - naphthaldimine:

34.4 g (0.2 M)  $\beta$ -hydroxy - $\alpha$ - naphthaldehyde dissolved in 175 cc methylalcohol by heating is placed in a 250 cc balloon. 7.41 g (0.1 M) 1,3-diamino propane is added dropwise on to the solution which is stirred and cooled down to  $-10^\circ\text{C}$ . Yellow crystals are filtered off and collected on a buchner funnel and recrystallized from chloroform. Yield 67% m.p  $212^\circ\text{C}$ .

#### Hexane-1,6-disalicylaldimine:

24.42 g (0.2 M) salicylaldehyde dissolved 50 cc methyl alcohol is put in a 250 cc balloon. 11.62 g (0.1 M) 1,6-diaminohexane dissolved in 50 cc methyl alcohol is added dropwise on to the solution which is stirred and cooled down to  $-10^\circ\text{C}$ . Yellow crystals are filtered off and collected on a buchner funnel and recrystallized from hot methyl alcohol. Yield 73% m.p  $72-73^\circ\text{C}$ .

#### Hexane-1,6-di ( $\beta$ -hydroxy) - $\alpha$ -naphthaldimine:

34.4 g (0.2 M)  $\beta$ -hydroxy - $\alpha$ - naphthaldehyde dissolved in 150 cc methylalcohol by heating is placed in a 250 cc balloon. 11.62 g (0.1 M)

1,6-diamine hexane dissolved in 50 cc methylalcohol is added dropwise on to the solution which is stirred and cooled down to  $-10^{\circ}\text{C}$ . After some time, the yellow crystals are filtered off and collected on a buchner funnel and recrystallized from hot ethylacetate. Yield 65% m.p  $173^{\circ}\text{C}$ .

**o-phenylene-disalicylaldimine:**

24.42 g (0.2 M) salicylaldehyde placed in a 250 cc baloon, is dissolved in 50 cc methylalcohol. 10.8 g (0.1 M) o-phenylene diamine dissolved in 50 cc methylalcohol is added dropwise on to the solution which is stirred and cooled down to  $-10^{\circ}\text{C}$ . Orange crystals are filtered off and collected on a buchner funnel and washed with mixture of diethylether and petroleum ether (1:1) and recrystallized from ethylalcohol. Yield 60% m.p  $156^{\circ}\text{C}$ .

**o-phenylene di ( $\beta$ -hydroxy)  $-\alpha$ -naphthaldimine:**

34.4 g (0.2 M)  $\beta$ -hydroxy  $-\alpha$ - naphthaldehyde placed in a 250 cc baloon is dissolved in 50 cc methylalcohol by heating. 10.8 g (0.1 M) o-phenylene diamine dissolved in 50 cc methylalcohol is added dropwise on to the solution which is stirred when the resulting red coloured mixture is evaporated by an evaporator orange crystals are formed. They are filtered off and collected on a buchner funnel and recrystallized from ethylacetate. Yield 65%. Decomp:  $195-198^{\circ}\text{C}$ .

**m-phenylene disalicylaldimine:**

24.42 g (0.2 M) salicylaldehyde placed in a 250 cc baloon is dissolved in 50 cc methyl alcohol. 18.10 g (0.1 M) m-phenylene diamine dihydrochloride dissolved in mixture of 150 cc pyridine and methyl alcohol is added dropwise on to the solution which is stirred and cooled down to  $-10^{\circ}\text{C}$ . The resulting purple coloured mixture is added drop by drop on to 50 cc distilled water and stirred. Yellow crystals are obtained. They are filtered off and collected on a buchner funnel and washed with 50 cc water and recrystallized from 100 cc methyl alcohol. Yield 63% m.p  $109.5^{\circ}\text{C}$ .

**m-phenylene di ( $\beta$ -hydroxy)  $-\alpha$ - naphthaldimine:**

34.4 g (0.2 M)  $\beta$ -hydroxy  $-\alpha$ - naphthaldehyde placed in a 250 cc baloon is dissolved 175 cc methyl alcohol by heating. 18.10 g (0.1 M) m-phenylene diamine dihydrochloride dissolved in hot mixture of 50 cc methyl alcohol and pyridine (1:1) is added on to the above solution. The resulting red coloured solution is cooled down to  $-10^{\circ}\text{C}$ . After some

time, yellow crystals are obtained. They are filtered off and collected on a buchner funnel and washed with the mixture of petroleum ether and diethylether (1:1). Yield 60%. Decomp at 222–225 °C.

**p-phenylene disalicylaldimine:**

24.42 g (0.2 M) salicylaldehyde placed in a 250 cc baloon is dissolved in 50 cc methyl alcohol. 18.10 g (0.1 M) p-phenylene diamine dihydrochloride dissolved in hot mixture of 50 cc pyridine and methyl alcohol (1:1) is added dropwise on the above solution. The resulting red coloured solution is cooled down to -10 °C. After some time oranges crystals are obtained. They are filtered off and collected on a buchner funnel and washed through petroleum ether and recrystallized from hot chloroform. Yield 73% m.p 210–11 °C.

**p-phenylene di (β-hydroxy) -α- naphthaldimine:**

34.4 g (0.2) β-hydroxy -α- naphthaldehyde placed in a 250 cc baloon is dissolved in 175 cc ethyl alcohol by heating. 18.10 g (0.1 M) p-phenylene diamine dihydrochloride dissolved in hot mixture of 250 cc pyridine and ethy alcohol, is added dropwise on to the above solution is cooled down -10 °C. After some time red crystals come down. They are filtered off and collected on a buchner funnel washed through benzene and dried. Yield 65%. Decomp at 298 °C.

**N,N'-bis (salicylidene) diethylenetriamine:**

24.42 g (0.2M) salicylaldehyde placed in a 250 cc baloon is dissolved 50 cc methylalcohol. 10.3 g (0.1M) diethylenetriamine dissolved in 50 cc methylalcohol is added dropwise on to above solution which is stirred and cooled. Resulting red coloured solution is evaporated by an evaporater and 50 cc benzene is added. It is left over night and yellowish-orangeviscos liquid is seperated by chromatographically using a column of 3 × 10 silicium dioxide and eluated by 250 cc ethylalcohol. Then the eluate is evaporated. Yield 62–63%.

**N,N'-bis [(β-hydroxy) -α- naphthalydene] diethylenetriamine:**

34.4 g (0.2 M) β-hydroxy -α- naphthaldehyde placed in a 250 cc baloon is dissolved in hot 100 °C methylalcohol. 10.3 g (0.1 M) diethylene triamine dissolved in 50 cc methylalcohol is added dropwise on to the above solution which is stirred and cooled down to -10 °C. Yellow crystals form. They are filtered off and collected on a buchner funnel and

washed with the mixture of chloroform and petroleum ether (1:9) and recrystallized from 100 cc methylalcohol. Yield 70–71%. m.p. 151°C.

N.N<sup>2</sup>-bis (salicylidene) triethylene tetramine:

24.42 g (0.2 M) salicylaldehyde placed in a 250 cc baloon is dissolved 50 cc methylalcohol. 14.6 g (0.1 M) triethylene tetramine dissolved in 50 cc methylalcohol is added dropwise on to the above solution which is stirred and cooled down to -10°C. Yellow crystals form. They are filtered off and collected on a buchner funnel and recrystallized from methylalcohol. Yield 81–82%. m.p. 102–103°C.

N.N<sup>2</sup>-bis (β-hydroxy) -α-naphthaldehyde triethylenetetramine:

34.4 g (0.2 M) β-hydroxy -α- naphthaldehyde placed in a 250 cc baloon is dissolved in 150 cc hot methylalcohol. 14.6 g (0.1 M) triethylene tetramine dissolved in 50 cc methylalcohol is added dropwise on to the above solution which is stirred and cooled down to -10°C. After the resulting dark yellow mixture is evaporated and waited 3–4 hours yellow crystals form. They are filtered off and collected on a kuchner funnel and washed with the mixture of diethyl ether and methylalcohol (1:1) and recrystallized from 100 cc methylalcohol. Yield 78%. m.p. 192°C.

N.N<sup>2</sup>-bis (salicylidene) tetraethylenepentamine:

24.42 g (0.2 M) salicylaldehyde placed in a 250 cc baloon is dissolved in 50 cc methylalcohol. 18.9 g (0.1 M) tetraethylenepentamine dissolved in 50 cc methylalcohol is added dropwise on to the above solution which is stirred and cooled down to -10°C. After the resulting red coloured solution is evaporated by an evaporater, 50 cc benzene is added. It is left over night and dark red viscos liquid is seperated by chromatographically using a column of 3×10 silicium dioxide and eluated by 250 cc ethylalcohol. Then the eluate is evaporated. Yield 69%.

N.N<sup>2</sup>-bis (salicylidene) pentaethylenehexamine:

24.42 g (0.2 M) salicylaldehyde placed in a 250 cc baloon is dissolved 50 cc methylalcohol. 22.6 g (0.1 M) pentaethylenehexamine dissolved in 50 cc methylalcohol is added dropwise on to the above solution which is stirred and cooled down to -10°C. The resulting red colour solution is evaporated by an evaporator and 50 cc benzene is added. It is left over night and dark red viscos liquid seperated by chromatographically



using a column of silicium dioxide and eluated by 250 cc ethylalcohol. Then the eluate is evaporated. Yield 65-66%

4.4'-(disalicylaldimine) diphenylmethane:

24.42 g (0.2 M) salicylaldehyde placed in a 250 cc baloon is dissolved in 50 cc methylalcohol. 19.8 g (0.1 M) 4.4-diaminodiphenyl methane dissolved in 100 cc dimethylformamide is added dropwise on to the above solution which is stirred and cooled down to  $-10^{\circ}\text{C}$ . Yellow crystall form. After some time they are filtered off a buchner funnel and washed with 50 cc methylalcohol. Yield 72-73%. m.p. 212-213 $^{\circ}\text{C}$ .

4.4'- di ( $\beta$ -hydroxy)  $-\alpha$ -naphaldymine diphenylmethane:

34.4 g(0.2 M)  $\beta$ -hydroxy  $-\alpha$ - naphthaldehyde placed in a 250 cc baloon is dissolved in 175 cc hot methylalcohol. 19.8 g (0.1 M) 4.4'-diamine diphenylmethane dissolved in 50 cc methylalcohol is added dropwise on to the above solution. Resulting dark yellow mixture is slowly poured in to 100 cc distilled water. Yellow crystall form. After some time they are filtered off a buchner funnel and washed with mixture of distilled water and methylalcohol (1:1) and recrystallized from 150 cc chloroform. Yield 68%. m.p. 198-200 $^{\circ}\text{C}$ .

p-N.N-dimethylaminophenyl salicylaldimine:

12.21 g(0.1M) salicylaldehyde placed in a 250 cc baloon is dissolved in 50 cc methylalcohol. 17.25 g (0.1 M) N.N-dimethyl-p-phenylene diamine hydrochloride dissolved hot mixture of 50 cc pyridine and methyl alcohol (1:1) is added dropwise on to the above solution which is stirred and cooled down to  $-10^{\circ}\text{C}$ . Orange crystall form. They are filtered off a buchner funnel and recrystallized from methylalcohol. Yield 83-84%. m.p. 133-35 $^{\circ}\text{C}$ .

p-N.N-dimethylamino phenyl ( $\beta$ -hydroxy)  $-\alpha$ - naphthaldimine:

17.2 g (0.1 M)  $\beta$ -hydroxy  $-\alpha$ - naphthaldehyde placed in a 250 cc baloon is dissolved in 50 cc methylalcohol. 17.25 g (0.1 M) N.N-dimethyl-p-phenylenediamine dissolved in hot mixture of 50 cc pyridine and methylalcohol is added dropwise on to the above solution which is stirred and cooled down to  $-10^{\circ}\text{C}$ . Red crystall form. They are filtered off a buchner funnel and recrystallized from methylalcohol. Yield 80%. m.p. 163-64 $^{\circ}\text{C}$ .

Yields, m.ps., and analytical data for the products are recorded in the table.

TABLE I.

COMPOUND	Yield%	M.p. °C	Found %			Formula	Required %		
			C	H	N		C	H	N
Propane-1,3-disalicylaldimine	75	59	68.99	5.82	10.10	$C_{17}H_{18}N_2O_2$	72.34	6.38	9.93
Propane-1,3-di ( $\beta$ -hydroxy)- $\alpha$ -naphthalidime	67	212	78.87	5.25	7.10	$C_{23}H_{22}N_2O_2$	78.87	5.76	7.33
Hexane-1,6-disalicylaldimine	73	72-73	74.31	7.37	9.08	$C_{20}H_{24}N_2O_2$	74.07	7.41	8.64
Hexane-1,6-di( $\beta$ -hydroxy)- $\alpha$ -naphthalidime	65	173	75.10	6.40	6.82	$C_{28}H_{28}N_2O_2$	79.20	6.60	6.60
o-phenylene-disalicylaldimine	60	156	75.72	4.81	8.99	$C_{20}H_{18}N_2O_2$	75.95	5.06	8.86
o-phenylene-di( $\beta$ -hydroxy)- $\alpha$ -naphthalidime	65	195-98	80.47	4.71	7.33	$C_{28}H_{28}N_2O_2$	80.77	4.80	6.73
m-phenylene-disalicylaldimine	63	109.5	75.41	5.01	8.01	$C_{20}H_{16}N_2O_2$	75.95	5.06	8.86
m-phenylene-di( $\beta$ -hydroxy)- $\alpha$ -naphthalidime	60	222-25*	80.20	4.90	5.26	$C_{28}H_{28}N_2O_2$	80.77	4.80	6.73
p-phenylene-disalicylaldimine	73	210-11	76.05	5.28	9.20	$C_{20}H_{16}N_2O_2$	75.95	5.06	8.86
p-phenylene-di( $\beta$ -hydroxy)- $\alpha$ -naphthalidime	65	298*	81.16	4.32	9.96	$C_{28}H_{28}N_2O_2$	80.70	4.80	6.73

N,N'-bis(salicylidene)diethylene triamine	62-63	-	67.47	6.33	13.43	$C_{18}H_{21}N_3O_2$	69.45	6.75	13.51
N,N'-bis[( $\beta$ -hydroxy- $\alpha$ -naphthalidene] diethylenetriamine	70-71	151	76.06	6.87	9.59	$C_{26}H_{23}N_3O_2$	75.91	6.08	10.22
N,N'-bis(salicylidene)triethylene tetramine	81-82	102-3	67.70	6.88	14.42	$C_{20}H_{26}N_4O_2$	67.80	7.34	15.82
N,N'-bis[( $\beta$ -hydroxy- $\alpha$ -naphthalidene]triethylenetetramine	78	192	73.33	6.94	12.42	$C_{28}H_{30}N_4O_2$	74.00	6.60	12.33
N,N'-bis(salicylidene)tetra-ethylenepentamine	69	-	67.14	7.50	16.88	$C_{22}H_{21}N_5O_2$	66.50	7.81	17.63
N,N'-bis(salicylidene) pentaethylenhexamine	65-66	-	65.47	7.95	18.17	$C_{24}H_{26}N_6O_2$	65.49	8.18	19.10
4,4'-(disalicylidimine)diphenylmethane	72-73	212-13	79.65	5.20	7.04	$C_{17}H_{17}N_2O_2$	79.80	5.42	6.90
4,4'-di( $\beta$ -hydroxy- $\alpha$ -naphthalidimine) diphenylmethane	68	198-200	82.75	5.84	6.11	$C_{33}H_{26}N_2O_2$	83.00	5.14	5.53
p-N,N-dimethylamino-phenylsalicylidimine	83-84	133-35	72.18	6.99	11.74	$C_{15}H_{16}N_2O$	75.00	6.67	11.67
p-N,N-dimethylaminophenyl( $\beta$ -hydroxy- $\alpha$ -naphthalidimine) decomposition*	80	163-64	78.59	6.25	9.73	$C_{19}H_{18}N_2O$	78.62	6.21	9.66

## REFERENCES

- 1- Pfeiffer, P., Hesse, Th., Pfitzner, H., Scholl, W., and Thielert, H., J. Prakt. Chem., 149, 217 (1937).
- 2- Dikmen, C., and Gündüz, T., Chem. Ber. 2637-2641 (1956).
- 3- L.J. Bellamy, "The Infrared Spectra of Complex Molecules", Methuen, London, 1960, Chapter 22.
- 4- Freedman, Harolt. H., J. Am. Chem. Soc., 83, 2900 (1961).
- 5- Ueno, K., and Martel, A.E., *ibid.*, 59,998 (1955).
- 6- Ueno, K., and Martel, A.E., *ibid.*, 60,1270 (1956).

## ÖZET

*o*-hidroksi benzaldehit ve  $\beta$ -hidroksi - $\alpha$ - naftaldehitin metanollü ortamda bazı diaminler ile verdiği Schiff bazları hazırlandı ve özellikleri incelendi.