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Received: 31.07.2022 Accepted: 08.12.2022 Research Article *Effects of flavonoids on SARS–CoV–2 main protease (6W63): A molecular docking study* 

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Abstract: Public health is still under attack by a worldwide pandemic caused by a coronavirus which is known to cause mainly respiratory and enteric disease in humans. Currently, still limited knowledge exists on the exact action mechanism and biology of SARS–CoV–2 although there are several effective vaccines and antiviral treatment. Besides, there is a considerable amount of 3D protein structures for SARS–CoV–2, related to its main protease resolved by X–ray diffraction. Here, we used molecular docking strategy to predict possible inhibitory activities of flavonoids on SARS–CoV–2 Mpro enzyme. For this, 800 flavonoids were retrieved from the ZINC database. Results suggested that avicularin was the lead flavonoid which docked to Mpro with the best binding energy. However, most of flavonoids showed H–bond interactions with Hie–41 and Cys–145 catalytic dyad, which were important residues for the catalytic activity of SARS–CoV–2 Mpro. Strong hydrogen bonding (2.36 Å) with S $\gamma$  atom of Cys145 residue was observed. This might suggest an initial formation of covalent bonding. Findings showed that selected flavonoids could be promising inhibitors of this enzyme and have the potential for future therapeutic drugs against COVID–19 after immediate experimental validation and clinical approvals.

*Keywords:* COVID–19; SARS-CoV-2; Flavonoids; Molecular docking; catalytic dyad.

### 1. Introduction

Coronaviruses (CoVs) are known to cause mainly respiratory and enteric diseases in humans and animals [1]. They are mainly divided into four genera, alpha, beta, gamma and delta–CoV [2]. As of July 10, 2022, this virus which has already spread to almost all countries with 555,030,991 confirmed total cases and 6,350,601 global deaths [3]. Currently, limited knowledge exists on the exact action mechanism and biology of SARS–CoV–2 ("Severe Acute Respiratory Syndrome Corona Virus–2" as seventh member of beta group [4]) and there is a limited number of effective vaccines and antiviral treatment against it. Regarding the efforts to discover a more effective vaccine, there are numerous achievements globally which produced vaccines since November 2020, Pfizer Inc [5], Moderna [6] and the University of Oxford (in collaboration with AstraZeneca) [7,8] announced positive results from provisional analyses of their Phase III vaccine trials. However, according to Coalition for Epidemic Preparedness Innovations (CEPI) most of the platforms of vaccine candidates in clinical trials are focused on the coronavirus spike protein and its variants as the primary antigen of COVID–19 infection [9]. Based on the data from COVID19 Vaccine Tracker website, there are 38 approved vaccines [10]. A research reported that 76

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total vaccine candidates dominantly exist from protein subunit among other molecular platforms such as non-replicating viral (31) vector, RNAbased (31), replicating viral vector (21), DNAbased (19), inactivated virus (14), virus-like particle (13), and live attenuated virus (4) platforms [11]. On the other hand, there is a considerable amount of 3D protein structures for SARS-CoV-2, generally related to its main protease structure resolved by mostly X-ray diffraction crystallography, available in Protein Data Bank (RCSB PDB). Although, exact action mechanism/(s) of COVID-19 is still a mystery itself, it was reported that it has the same cell-entry receptor, ACE2 (Angiotensin-Converting Enzyme 2), for infection as SARS-CoV [12,13]. Like main protease and ACE2 structures, 3C-like Proteinase of COVID-19 was also successfully expressed [14,15]. Moreover, the number of infections is globally still rising [16] and immediate increase in the publication of over 10,000 peer-reviewed papers showed the importance and urgent need for discovery and development of effective and preventive therapeutic drugs/protocols and vaccines.

Traditional herbal medicines are known to be used and it was recently reported that these medicines have been used in China from the beginning of the outbreak and they were seen to recover of 90% of the 214 patients treated [17,18]. Some promising results were also published by Xu et al. in Zhejiang Province-China Chinese [19]. traditional medicines, Shu Feng Jie Du and Lianhuaqingwen, were recommended because of their efficiency against previous influenza A (H1N1) or SARS-CoV-1 [20]. According to a recent investigation, researchers from the Zhongnan Hospital of Wuhan University recommended traditional Chinese medicines in the guidelines for the treatment and prevention of COVID-19 to treat the disease by using different herbal mixtures according to the disease-stage [21]. Flavonoids are a large class of occurring phenolic naturally compounds distributed in the plant kingdom. Almost more than 4,000 varieties of flavonoids have been previously identified. These broad-spectrum compounds act as potent antioxidants, and display antiallergic, antiinflammatory, antimutagenic, antihemorrhagic, antineoplastic, and hepatoprotective activities [22,23]. They also exhibit biological activities including anticancer, antibacterial, antifungal, and antiviral activities. Among flavonoids, apigenin, luteolin, and quercetin were shown to possess antiviral activities both in-vivo and in-vitro [24,25]. Several natural compounds including baicalin, scutellarin, hesperetin, nicotianamine and glycyrrhizin were predicted to have capacity for binding ACE2 receptor with potential anti-2019nCoV effects [25]. Moreover, quercetin, daidzein, puerarin, epigallocatechin, epigallocatechin gallate, gallocatechin gallate and kaempferol were reported to inhibit the proteolytic activity of SARS-CoV 3CLpro [26,27]. Among flavonoids, quercetin, quercetin derivatives, catechin, epicatechin, epicatechin gallate and epigallocatechin gallate were reported to inhibit Lpro SARS-3C expressed in Escherichia coli [28-30]. Usage of flavonoids that exist in many herbal plants can be evaluated as an alternative approach to chemically synthesized therapeutic drugs against many viruses including CoVs. In addition, chemical structures and vibrational spectroscopic properties of flavonoids such as baicalein, naringenin, and selected amino-, chloro-, and bromo-flavones were previously investigated in detail [31-36] and they have also been investigated against a wide range of DNA and RNA viruses [37]. For example, apigenin is active against picornavirus (RNA virus), inhibiting protein synthesis by suppressing internal ribosome entry site (IRES) viral activity [38]. Epigallocatechin-3-gallate (EGCG), active polyphenolic catechin that accounts for approximately 59% of the total catechins from the leaves of the green tea (Camellia sinensis (L.), Kuntze) interferes with the replication cycle of DNA viruses, such as hepatitis B virus, herpes simplex, and adenovirus [39].

It is quite useful and efficient to apply computeraided drug design and molecular docking techniques to quickly identify promising candidates against diseases, especially after detailed 3Dstructures of various key proteins of that disease are resolved experimentally. Although there are efforts appeared currently, there is still a lack of docking studies performed with especially flavonoids on different main proteases of SARS–CoV–2. The importance of Mpro in the life cycle of SARS– CoV–2 identifies the Mpro as an attractive target for antiviral drug design. SARS–CoV–2 Mpro (6LU7) was docked with fourteen flavonoid

compounds as well as four already existing drugs and the binding energies were determined [40]. We used 6W63 for our non-covalent docking, as its natural ligand (X77) is non-covalent whereas 6LU7 ligands are covalent. Taking the advantage of a recently deposited crystal structure of SARS-CoV-2 Mpro (6W63) in complex with its natural inhibitor [41], we used virtual screening approach and found that three FDA approved drugs can be used against inhibition of Mpro of SARS-CoV-2 [42]. Virtual screening research already was shown be able to replace time-consuming efforts in determination of new targets for the existing drug compounds as demonstrated in earlier articles including SARS-CoV-2 [43-63]. In addition, some recent investigations on flavonoids showed and discussed the high potential of flavonoids docked to MPro of SARS-CoV-2 [64-70]. In addition, our previous virtual screening study [42] showed that two FDA approved drugs, fenoterol and dobutamine, can be considered as promising inhibitors for SARS-CoV-2 Mpro. Moreover, regarding 113 FDA approved drugs in clinical trials for COVID-19 treatment, it has been identified activated signaling pathways associated with the infection caused SARS-CoV-2 in human lung epithelial cells through integrative analysis. Then, the activated gene ontologies (GOs) and top lead super GOs were identified [71] and fenoterol was among these GOs. Their findings appear to support our conclusions on fenoterol and dobutamine [42]. The objectives of this study were: i) to determine important active site amino acids by structure-based sequence alignment of SARS-CoV-2 Mpro ii) to identify potential non-covalent Mpro inhibitors by screening protease-inhibitor-like compounds. particularly flavonoids as candidates, which are available in the ZINC database by molecular docking studies iii) to validate the stable binding of the lead compounds with SARS-CoV-2Mpro (6W63) and iv) to calculate binding affinities (kcal/mol) for each lead compound.

#### 2. Computational Method

Molecular docking procedures are used for the purpose of selecting the hits that exhibit chemical, structural, and electronic characteristics. The information of the target protein can be derived from in silico technique or experimental data. In order to predict the possible inhibitory activities of flavonoids, we performed docking studies on SARS–CoV–2 Mpro enzyme using Schrödinger 2019-4 software, with Maestro 12.2 and the Glide 8.5 module to predict the binding energies [39,72,73].

#### 2.1. Protein preparation

X-ray crystallographic structure of SARS–CoV–2 Mpro was retrieved from Protein Data Bank (6W63) and prepared for docking process. This enzyme have 306 residues and resolution of the structure is 2.10Å [74]. In order to prepare the enzyme, we used the protein preparation wizard module. During preparation hydrogen atoms were added and bond orders were assigned with zero order bonds to disulfide bonds and metals as well. Water molecules were removed within 3 Å of het groups. OPLS-2005 force field for minimization and pH =7.0  $\pm$  2.0 were chosen to minimization step.

### 2.2. Ligand preparation

Approximately 800 flavonoids were retrieved from the ZINC [75] database to perform the molecular docking studies and prepared by using Schrödinger, LigPrep module [76]. Flavonoids were prepared and 3D structures were generated by adding hydrogen atoms and removing salt. The bond angles and bond orders were assigned after ligand minimization step. In order to keep the ligands in the right protonation state in biological conditions, Epik option was used. LigPrep can generate the expected ionized forms at significant concentrations corresponding to the pH 7.0±2.0, generate variations, perform verification, and optimize structures. It generates a maximum of 32 stereochemical structures per ligand.

#### 2.3. Grid Generation

The active site of SARS–CoV–2 Mpro was defined for generating the grid in Maestro. The grid box was limited to the size of 10 Å in –20.46, 18.11, and – 26.91 directions at the active site. First, docking procedure was validated by extracting the nature ligand, X77 from the binding site and re-docking it to SARS–CoV–2 Mpro by using the Glide SP (standard precision glide docking) module [77]. Glide generates conformations internally and passes these through a series of filters. Glide successfully reproduced the experimental binding conformations of X77 in SARS–CoV–2 MPro with an acceptable root-mean-square deviation (RMSD) value of 0.68 Å. Then, flavonoids were screened by using the same grid with Glide XP (Extra precision glide docking) module [73]. According to the docking scores 16 hit flavonoids were selected.

#### 2.4. DFT computation

In addition, geometric structure of the top hit compound, avicularin, was optimized by Gaussian 09 software by using DFT method with B3LYP functional by 6-311++G(d,p) basis set. Vibrational wavenumbers were scaled with 0.967 and 0.955 for the wavenumbers below and above 1800 cm $\Box$ 1, respectively.

#### 3. Results and discussion

In this study, we used molecular docking strategies to predict the possible inhibitory activities of some flavonoids on SARS-CoV-2 Mpro enzyme. For this purpose, approximately 800 flavonoids were retrieved from the ZINC database. According to docking results, we selected 16 hit flavonoids and the docking scores and the interactions were shown in Table 1 together with the data for natural ligand, X77. Structures of these top 16 flavonoids were given in Figure 1. Binding modes of the selected flavonoids on SARS-CoV-2 Mpro (6W63) and their interactions with the surrounding residues in two-dimension representation were given in Figure 2. Compound 469 (Avicularin) and 471 showed the best two docking scores as -11.799 and -10.789, respectively. As presented in Table 1, avicularin showed H-bonds with Gly143, Hie163, and water mediated H-bonds with Hie41 and Glu166 whereas the compound 471 showed H-bonds with Gly143, Cys145, Hie163, water mediated H-bond with Glu166 and  $\pi$ - $\pi$  stacking with Hie41 (Figure 2). Although avicularin has the best docking score, we presented compound 471 in Figure 3 because it is most interacting compound with the its environment. It is interesting to note that both compounds (avicularin and 471) have the same flavonoid skeleton with the four (two in B ring and two in A ring) identically located OH groups in their molecular structure but act differently for the inhibition of 6W63. Furan rings are above and below the skeleton for 471 and avicularin, respectively. OH groups are consecutively located in both compounds. For 471, OH groups are in upup-down fashion whereas OH groups of avicularin in down-up-down fashion. The interaction type and the number of the interactions with the surrounding residues is strongly affected by these orientations of these groups. Binding mode of compound 471 with SARS-CoV-2 Mpro (6W63) together with the catalytic center was shown in Figure 3. Most of the selected flavonoids showed H-bond interactions with Hie41 and/or Cys145 [78-80], which were important residues for the catalytic activity of SARS-CoV-2 Mpro with the better docking scores than the natural ligand X77. Avicularin was recently shown to have docking capabilities with different interaction sites of various main proteases such as, 6W4B, 6VYB, 6LVN, 6M0J, and 6LU7 and it showed no toxicity and undesired effects like mutagenicity, tumorigenicity, irritating, reproductive effects [65].

Distances between the active site residues of Mpro on interaction with the selected flavonoids were presented in Table 2. From our findings, it is evident that the distances between Cys145 and the investigated compounds are found to be between 2.30 (compound 471) and 4.0 Å (compounds 425 and 424). Regarding Hie41, these data fell into the region between 2.57 (compound 454) and 5.66 Å (compound 468). Aside from Cys-Hie residues, Gly143 revealed closer distances when compared to Cys-Hie residues. For example, we observed strong hydrogen bonding with Gly143 (with 1.65 Å distance) for compound 435. The structure of the 6W63 complex at 2.10 Å resolution [41] is well enough for us to distinguish the hydrogen bonds given in Table 2. In addition, although the thiol group of Cys145 interacts to the compound 471 with a bond of 2.30 Å, it was confirmed that the main chains of Gly143, Hie163, and Glu166 also interact with each inhibitor mostly. In addition, it is interesting to note that we could observe the Pistacking formation only between imidazole ring of Hie41 and the ligand 420. This ligand was also seen to be interaction via Sy atom of the Cys145. According to the results of Yoshino et al. (2020), hydrogen bond donor pharmacophore sphere is located near Hie41. Their results suggested that any hydrogen bond acceptor functional group could have the potential to contact with Hie41. As depicted in Figure 5, our results are in agreement with their work [81] and we could identify both strong hydrogen bonding (2.36 Å) with Sy atom of

the Cys145 and Pi-stacking formation between the imidazole ring of Hie41 and phenyl ring of compound 420. Moreover, the water bridges constructed among the residues and the ligands also contribute to Mpro and inhibitor complex structure to stabilize the structure, functional groups of ligands.



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Table 1. Docking results of	the 16 hit flavonoids.			
Compound	ZINC ID	Docking Score (kcal/mol)	Glide Score (kcal/mol)	Interactions
<b>469</b> (Avicularin)	ZINC28540146	-11.799	-11.828	Thr25 <sup>d</sup> , Thr26 <sup>d</sup> , Leu27 <sup>c</sup> , <b>Hie41</b> (water mediated) <sup>d</sup> , Phe140 <sup>c</sup> , Leu141 <sup>c</sup> , Asn142 <sup>d</sup> , Gly143 <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , <b>Hie163<sup>d</sup></b> , His164 <sup>d</sup> , Met165 <sup>c</sup> , Glu166 (water mediated) <sup>a</sup> , Pro168 <sup>c</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Thr190 <sup>d</sup> , Gln192 <sup>d</sup> , H <sub>2</sub> O
471	ZINC33833455	-10.789	-10.817	Thr25 <sup>d</sup> , Thr26 <sup>d</sup> , Leu27 <sup>c</sup> , Hie41 <sup>d,e</sup> , Met49 <sup>c</sup> , Phe140 <sup>c</sup> , Leu141 <sup>c</sup> , Asn142 <sup>d</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> ,

				<b>Cys145</b> °, <b>Hie163</b> <sup>d</sup> , His164 <sup>d</sup> , Met165°, <b>Glu166</b> (water mediated) <sup>a</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Thr190 <sup>d</sup> , Gln192 <sup>d</sup> , <b>H</b> <sub>2</sub> <b>O</b>
425	ZINC4416344	-10.467	-10.496	Hie41 (water mediated) <sup>d</sup> , Cys44 <sup>c</sup> , Met49 <sup>c</sup> , Tyr54 <sup>c</sup> , Phe140 <sup>c</sup> , Leu141(2) <sup>c</sup> , Asn142 <sup>d</sup> , Gly143 <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , Hie163 <sup>d</sup> , His164 <sup>d</sup> , Met165 <sup>c</sup> , Glu166 (water mediated) <sup>a</sup> , Pro168 <sup>c</sup> , Hie172 <sup>d</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Thr190 <sup>d</sup> , Gln192 <sup>d</sup>
424	ZINC4416342	-10.167	-10.195	Thr25 <sup>d</sup> , Hie41 <sup>d</sup> , Cys44 <sup>c</sup> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Tyr118 <sup>c</sup> , Phe140 <sup>c</sup> , Leu141(2) <sup>c</sup> , Asn142 <sup>d</sup> , Gly143 <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , Hie163 <sup>d</sup> , His164 <sup>d</sup> , Met165 <sup>c</sup> , Glu166 <sup>a</sup> , Pro168 <sup>c</sup> , Hie172 <sup>d</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup>
472	ZINC33833712	-9.968	-9.997	Hie41 <sup>d</sup> , <b>Cys44</b> <sup>c</sup> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Leu141 <sup>c</sup> , Asn142 <sup>d</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , Met165 <sup>c</sup> , <b>Glu166</b> (water mediated) <sup>a</sup> , Leu167 <sup>c</sup> , Pro168 <sup>c</sup> , <b>Asp187</b> <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Gln192 <sup>d</sup>
<b>435</b> (3-O-Methylquercetin)	ZINC5998596	-9.837	-9.871	Hie41 <sup>d</sup> , <b>Cys44</b> <sup>c</sup> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Leu141 <sup>c</sup> , Asn142 <sup>d</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , Met165 <sup>c</sup> , <b>Glu166</b> <sup>a</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , <b>H<sub>2</sub>O</b>
466	ZINC15657718	-9.652	-9.681	Hie41 <sup>d</sup> , Cys44 <sup>c</sup> , Met49 <sup>c</sup> , Phe140 <sup>c</sup> , <b>Leu141(2)</b> <sup>c</sup> , Asn142 <sup>d</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , <b>Hie163</b> <sup>d</sup> , His164 <sup>d</sup> , Met165 <sup>c</sup> , Glu166 <sup>a</sup> , Pro168 <sup>c</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup>
422	ZINC4416338	-9.627	-9.655	Thr25 <sup>d</sup> , <b>Hie41 (water mediated)</b> <sup>d</sup> , Cys44 <sup>c</sup> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Phe140 <sup>c</sup> , <b>Leu141</b> <sup>c</sup> , Asn142 <sup>d</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , <b>Hie163</b> <sup>d</sup> , Met165 <sup>c</sup> , Glu166 <sup>a</sup> , Pro168 <sup>c</sup> , Hie172 <sup>d</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup>
478	ZINC61948742	-9.547	-9.575	Hie41 <sup>d</sup> , Cys44 <sup>c</sup> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Leu141 <sup>c</sup> , Asn142 <sup>d</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , Met165 <sup>c</sup> , <b>Glu166 (water mediated)</b> <sup>a</sup> , Pro168 <sup>c</sup> , <b>Asp187</b> <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Thr190 <sup>d</sup> , Gln192 <sup>d</sup>
454	ZINC9147119	-9.502	-9.531	Hie41 <sup>d</sup> , <b>Cys44</b> <sup>c</sup> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Phe140 <sup>c</sup> , <b>Leu141</b> <sup>c</sup> , Asn142 <sup>d</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , <b>Hie163</b> <sup>d</sup> , His164 <sup>d</sup> , Met165 <sup>c</sup> , <b>Glu166</b> <sup>a</sup> , Pro168 <sup>c</sup> , Phe181 <sup>c</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Thr190 <sup>d</sup> , Gln192 <sup>d</sup>
495	ZINC104891686	-9.345	-9.374	<b>Hie41 (water mediated)</b> <sup>d</sup> , Cys44 <sup>c</sup> , Met49 <sup>c</sup> , Tyr118 <sup>c</sup> , Phe140 <sup>c</sup> , <b>Leu141(2)</b> <sup>c</sup> , Asn142 <sup>d</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , <b>Hie163</b> <sup>d</sup> , His164 <sup>d</sup> , Met165 <sup>c</sup> , Glu166 <sup>a</sup> , Pro168 <sup>c</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup>
462	ZINC14684606	-9.310	-9.339	Thr25 <sup>d</sup> , Thr26 <sup>d</sup> , Leu27 <sup>c</sup> , Hie41 <sup>d</sup> , <b>Cys44<sup>c</sup></b> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Asn142 <sup>d</sup> , Gly143 <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , Hie163 <sup>d</sup> , Met165 <sup>c</sup> , <b>Glu166</b> <sup>a</sup> , Pro168 <sup>c</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , <b>H</b> <sub>2</sub> <b>O</b>
420	ZINC4349611	-9.205	-9.234	Thr25 <sup>d</sup> , Hie41 <sup>d,e</sup> , <b>Cys44</b> <sup>c</sup> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Leu141 <sup>c</sup> , Asn142 <sup>d</sup> , Gly143 <sup>f</sup> , Ser144 <sup>d</sup> , <b>Cys145</b> <sup>c</sup> , His164 <sup>d</sup> , Met165 <sup>c</sup> , <b>Glu166</b> <sup>a</sup> , Leu167 <sup>c</sup> , Pro168 <sup>c</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Thr190 <sup>d</sup>
468	ZINC25763686	-9.172	-9.201	Thr25 <sup>d</sup> , Hie41 <sup>d</sup> , <b>Cys44</b> <sup>c</sup> , Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , <b>Asn142</b> <sup>d</sup> , Gly143 <sup>f</sup> , <b>Cys145</b> <sup>c</sup> , Met165 <sup>c</sup> , <b>Glu166</b> <sup>a</sup> , Pro168 <sup>c</sup> , <b>Asp187</b> <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Thr190 <sup>d</sup>
474	ZINC43465464	-9.150	-9.178	Hie41 <sup>d</sup> , Cys44 <sup>c</sup> , Met49 <sup>c</sup> , Tyr54 <sup>c</sup> , Phe140 <sup>c</sup> , Leu141 <sup>c</sup> , Asn142 <sup>d</sup> , Gly143 <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , Hie163 <sup>d</sup> , His164 <sup>d</sup> , Met165 <sup>c</sup> , Glu166 <sup>a</sup> , Pro168 <sup>c</sup> ,

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				Hie172 <sup>d</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup> , Thr190 <sup>d</sup> ,
				Gln192 <sup>d</sup>
				<b>Hie41 (water mediated)</b> <sup>d</sup> , Cys44 <sup>c</sup> , Met49 <sup>c</sup> ,
				Tyr54 <sup>c</sup> , Phe140 <sup>c</sup> , Leu141 <sup>c</sup> , Asn142 <sup>d</sup> , Gly143 <sup>f</sup> ,
463	ZINC14684625	-9.062	-9.091	Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , <b>Hie163</b> <sup>d</sup> , His164 <sup>d</sup> , Met165 <sup>c</sup> ,
				Glu166 <sup>a</sup> , Leu167 <sup>c</sup> , Pro168 <sup>c</sup> , Hie172 <sup>d</sup> , Asp187 <sup>a</sup> ,
				Arg188 <sup>b</sup> , Gln189 <sup>d</sup>
				Thr25 <sup>d</sup> , Thr26 <sup>d</sup> , Leu27 <sup>c</sup> , Hie41 <sup>d,e</sup> , Cys44 <sup>c</sup> ,
V77				Met49 <sup>c</sup> , Pro52 <sup>c</sup> , Tyr54 <sup>c</sup> , Phe140 <sup>c</sup> , Leu141 <sup>c</sup> ,
$\Delta / /$		-8.938	-8.938	Asn142 <sup>d,w</sup> , <b>Gly143</b> <sup>f</sup> , Ser144 <sup>d</sup> , Cys145 <sup>c</sup> , <b>Hie163</b> <sup>d</sup> ,
(natural ligand)				His164 <sup>d</sup> , Met165 <sup>c</sup> , <b>Glu166</b> <sup>a</sup> , Leu167 <sup>c</sup> , Pro168 <sup>c</sup> ,
				Hie172 <sup>d</sup> , Asp187 <sup>a</sup> , Arg188 <sup>b</sup> , Gln189 <sup>d</sup>

**Bold:** H-bond, <sup>a</sup>negative charge (orange), <sup>b</sup>positive charge (cyan), <sup>c</sup>Hydrophobic (green), <sup>d</sup>Polar (turquoise), <sup>e</sup>π-π stacking, <sup>f</sup>glycine. Colors in this table refer to the colors given in Figure 2.



Figure 2 Binding modes of the selected lead flavonoids on SARS–CoV–2  $M^{pro}$  (6W63) and their interactions with the surrounding residues.



**Figure 3 a)** Binding mode of compound **471** with SARS–CoV–2 M<sup>pro</sup> (6W63) together with the catalytic center shown as magnified view. The residues that participate in ligand are shown as stick models. Ligand is shown as ball-and-stick model. Strands represent the SARS–CoV–2 M<sup>pro</sup> polypeptide with the side chain of Cys145 protruding. **b)** Ligand **471** in the binding pocket.

**Table 2.** Distances (Å) between the active site residues (Hie41, Gly143, Cys145) of  $M^{pro}$  on interaction with the selected flavonoids.

Compound	Hie41	Gly143	Cys145
420	5.41 <sup>e</sup>	2.24	2.36
422	3.69 (water mediated)	2.15	4.00
424	4.12	2.14	4.00
425	4.12 (water mediated)	2.14	4.00
435	4.78	1.65	3.90
454	2.57	1.97	3.27
462	5.01	2.01	3.55
463	3.32 (water mediated)	2.12	3.98
466	4.04	2.04	3.98
468	5.66	4.23	2.36
469	3.41 (water mediated)	1.93	2.41
471	5.43 <sup>e</sup>	2.43	2.30
472	4.82	1.91	3.43
474	2.67	1.94	3.35
478	4.87	1.93	2.41
495	3.33 (water mediated)	2.10	3.98

**Bold:** H-bond, e:  $\pi$ - $\pi$  stacking (ring to ring distance was measured).

A recent investigation on the identification of key interactions between SARS–CoV–2 and various inhibitor drug candidates focused on the structures of three proteins, 6LU7, 2A5I, and 2OP9 took the advantage of molecular dynamics simulations and they suggested that the probability of the interaction of the amino acid residues for drug candidate indinavir was higher for Gly143 (54%) than their corresponding probabilities of natural ligands (30 and 36% for 2A51, and 2OP9) [81]. On the other hand, it is known that there are differences between protein structures, as 6W63 in our study, and binding profiles and interactions are quite different than each other. Consequently, this would lead difference interactions of drug candidates and their

surrounding residues. We found that Gly143 is an important residue since our flavonoids are mostly in close contact with this residue.

The Mpro helps in replication and transcription of novel coronavirus and SARS–CoV–2 Mpro has a Cys-His catalytic dyad, and the substrate-binding site is in a cleft between domain I and domain II. These features are similar to previously reported Mpro from other coronaviruses [78–80]. It is widely accepted that increased inhibitor potency is related to the covalent bond formed between the active site residue and the designed compound. It was previously reported that the S $\gamma$  atom of Cys145 forms a covalent bond with the substrate, confirming that the Michael addition, has occurred

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[82]. In contrast to that work, we were not able to detect any signal of Michael addition mechanism for none of our lead flavonoids, but this does not mean that such mechanism would not occur in the case of flavonoids. It is obvious that there are rapidly increasing numbers of MPro structures for this disease, and this would need more systematic research including molecular dynamics and experimental studies. However, our docking investigation revealed that almost all the selected flavonoids showed interactions with Hie41 and/or Cys145, with the better docking scores than the natural ligand X77. Even though numerous works emphasize Cys-His catalytic dyad residues Cys145 and Hie41, still there are a limited number of studies on Cys-His-Gly catalytic triad even this is a more complex mechanism compared to catalytic dyad. Furthermore, very rare Cys-His-Gly type of catalytic triad examples were previously be reported [83-85]. Even though the short distance (1.65 Å) between the Gly-143 and compound 435 might indicate a very strong hydrogen bond, this can be also attributed to an initial formation of a covalent bond. Of course, we are aware that it is still difficult to suggest or comment on this quickly, because occurring mechanisms of both catalytic dyads and triads are still not clear. Furthermore, regarding non-covalent inhibition aspects of compounds on main proteases, including flavonoids, appeared in recent publications demonstrated that it is rarely possible to design/discover non-covalent inhibitors, such as ML188 for SARS-CoV-1 Mpro which was shown to exhibit inhibition effects experimentally [86]. In that study, authors characterized the complex of SARS-CoV-2 Mpro with the non-covalent inhibitor ML188 and showed that ML188 has enhanced binding potency to SARS-CoV-2 compared to SARS-CoV-1. In addition, a recent X-Ray screening investigation identified active site and allosteric inhibitors of SRAS-CoV-2 Mpro. Researchers successfully found two allosteric binding sites from the yield of 37 compounds that bind to Mpro. A noncovalent binder MUT056399 (4-(4-ethyl-5-fluoro-2-hydroxyphenoxy)-3-

fluorobenzamide) was also shown to block the active site of Mpro [87]. Even these rare compounds were encountered related to noncovalent bonding in terms of their inhibition effects experimentally, we must admit that experimental noncovalent binding profiles and inhibition effects of flavonoids particularly are still not crystal clear.



Figure 4 Compound 435 with its surrounding environment (Cys145-Hie41-Gly143).

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**Figure 5** Stick diagram depicting the geometry and atomic interactions of the oxygen atom of compound **420** in H-bond linkage between the  $S_{\gamma}$  atom of Cys145. Light blue color denotes the  $\pi$ - $\pi$  stacking between the phenyl ring of compound **420** and the Hie41.

Quantum chemical computations were performed on the top hit compound, avicularin, and optimized geometric structure was shown in Figure 6 together with the optimized parameters and vibrational wavenumbers (Table 3 and 4, respectively). Computed and scaled IR spectrum of avicularin was also given in Figure 7.



Figure 6. Optimized geometric structure of avicularin.

Optimized geometric structure of avicularin revealed that the twisted dihedral angle between the plane where A and C rings locate and the ring B was computed to be  $\Box 29.94^{\circ}$  which confirmed previously reported dihedral angle of several flavonoid analogs [35,88–91] that vary between ~18 to 21° with the exception of the group attached to C-ring's via its hydrogen atom (H12 in Figure 6). Due to the difference of the basic flavone structure (A-C-B structure), this value of dihedral angle would still be acceptable and this can be explained by weak interaction of the group between the B-ring of avicularin, thus this bending would take place.

C-H stretching, in-plane and out of plane CH bending modes were previously reported to be observed between 3100-3000, 1100-1500, and 800-1000 cm-1, respectively [92,93]. Our findings for avicularin were computed between 3079-2946 cm-1, 1104-1446 cm-1 799-905 cm-1 respectively and there is a good agreement with Varsanyi and Mohan's work [92,93]. O-H stretching vibrations were computed between 3675 and 3624 cm-1 whereas the peak that was observed at 1655 cm-1 was assigned to C=O stretching vibration and this assignment is in line with Heneczkowski and co-workers' previous work [94] where this band was assigned between 1649 and 1652 cm-1 for selected flavonoids in that study.

C=C stretching vibration were determined between 1606 and 1563 cm-1 while the bands computed between 1488 and 1480 cm-1 were attributed to CC stretching vibration of avicularin. In addition, CO stretching vibration was assigned between 1067 and 993 cm-1. Out-of-plane bending vibrations of avicularin were also computed below 904 cm-1 where these vibrations were previously emphasized to be observed in the same region [95].



Figure 7. Computed (scaled) IR spectrum of avicularin.

	Т	able 3. Optimi	zed geomet	ric structure of av	vicularin*.
R(1,2)	1.363	A(2,1,6)	121.67	D(6,1,2,3)	2.18
R(1,6)	1.370	A(1,2,3)	121.59	D(6,1,2,7)	-178.51
R(2,3)	1.405	A(1,2,7)	114.70	D(2,1,6,5)	0.96
R(2,7)	1.393	A(3,2,7)	123.70	D(2,1,6,26)	-179.78
R(3,4)	1.477	A(2,3,4)	119.45	D(1,2,3,4)	-1.84
R(3,10)	1.419	A(2,3,10)	116.40	D(1,2,3,10)	179.10
R(4,5)	1.478	A(4,3,10)	124.13	D(7,2,3,4)	178.92
R(4,11)	1.222	A(3,4,5)	114.11	D(7,2,3,10)	-0.14
R(5,6)	1.357	A(3,4,11)	124.85	D(1,2,7,8)	-179.40
R(5,12)	1.372	A(5,4,11)	121.04	D(1,2,7,32)	0.28
R(6,26)	1.474	A(4,5,6)	122.49	D(3,2,7,8)	-0.11
R(7,8)	1.385	A(4,5,12)	115.83	D(3,2,7,32)	179.57
R(7,32)	1.081	A(6,5,12)	121.57	D(2,3,4,5)	-1.28
R(8,9)	1.400	A(1,6,5)	120.51	D(2,3,4,11)	178.49
R(8,31)	1.364	A(1,6,26)	111.37	D(10,3,4,5)	177.70
R(9,10)	1.392	A(5,6,26)	128.11	D(10,3,4,11)	-2.53
R(9,33)	1.087	A(2,7,8)	118.07	D(2,3,10,9)	0.32
R(10,30)	1.352	A(2,7,32)	120.70	D(2,3,10,30)	-179.80
R(12,13)	1.442	A(8,7,32)	121.23	D(4,3,10,9)	-178.69
R(13,14)	1.536	A(7,8,9)	120.72	D(4,3,10,30)	1.19
R(13,19)	1.402	A(7,8,31)	117.40	D(3,4,5,6)	4.39
R(13,34)	1.092	A(9,8,31)	121.89	D(3,4,5,12)	-179.38
R(14,15)	1.422	A(8,9,10)	120.37	D(11,4,5,6)	-175.39
R(14,16)	1.533	A(8,9,33)	120.05	D(11,4,5,12)	0.84
R(14,35)	1.094	A(10,9,33)	119.58	D(4,5,6,1)	-4.37
R(15,36)	0.962	A(3,10,9)	120.73	D(4,5,6,26)	176.50
R(16,17)	1.422	A(3,10,30)	118.46	D(12,5,6,1)	179.62
R(16,18)	1.539	A(9,10,30)	120.81	D(12,5,6,26)	0.49
R(16,37)	1.099	A(5,12,13)	116.62	D(4,5,12,13)	98.27
R(17,38)	0.962	A(12,13,14)	106.53	D(6,5,12,13)	-85.47

R(18,19)	1.443	A(12,13,19)	112.82	D(1,6,26,25)	-29.94
R(18,20)	1.513	A(12,13,34)	108.98	D(1,6,26,27)	148.45
R(18,39)	1.095	A(14,13,19)	107.61	D(5,6,26,25)	149.25
R(20,21)	1.422	A(14,13,34)	113.11	D(5,6,26,27)	-32.35
R(20,40)	1.102	A(19,13,34)	107.88	D(2,7,8,9)	0.19
R(20,41)	1.098	A(13,14,15)	109.82	D(2,7,8,31)	-179.92
R(21,42)	0.961	A(13,14,16)	103.51	D(32,7,8,9)	-179.49
R(22,23)	1.403	A(13,14,35)	108.81	D(32,7,8,31)	0.40
R(22,27)	1.387	A(15,14,16)	114.40	D(7,8,9,10)	-0.02
R(22,28)	1.362	A(15,14,35)	111.15	D(7,8,9,33)	179.83
R(23,24)	1.389	A(16,14,35)	108.79	D(31,8,9,10)	-179.90
R(23,29)	1.375	A(14,15,36)	108.30	D(31,8,9,33)	-0.06
R(24,25)	1.392	A(14,16,17)	109.48	D(7,8,31,49)	-179.76
R(24,43)	1.086	A(14,16,18)	102.94	D(9,8,31,49)	0.13
R(25,26)	1.402	A(14,16,37)	108.00	D(8,9,10,3)	-0.24
R(25,44)	1.081	A(17,16,18)	115.82	D(8,9,10,30)	179.87
R(26,27)	1.404	A(17,16,37)	110.66	D(33,9,10,3)	179.91
R(27,45)	1.081	A(18,16,37)	109.44	D(33,9,10,30)	0.02
R(28,46)	0.966	A(16,17,38)	109.49	D(3,10,30,48)	-177.92
R(29,47)	0.962	A(16,18,19)	102.35	D(9,10,30,48)	1.96
R(30,48)	0.964	A(16,18,20)	113.41	D(5,12,13,14)	-163.92
R(31,49)	0.963	A(16,18,39)	111.11	D(5,12,13,19)	-46.07
		A(19,18,20)	109.34	D(5,12,13,34)	73.73
		A(19,18,39)	109.83	D(12,13,14,15)	-114.95
		A(20,18,39)	110.49	D(12,13,14,16)	122.49
		A(13,19,18)	108.88	D(12,13,14,35)	6.91
		A(18,20,21)	109.79	D(19,13,14,15)	123.81
		A(18,20,40)	108.23	D(19,13,14,16)	1.25
		A(18,20,41)	108.62	D(19,13,14,35)	-114.33
		A(21,20,40)	110.42	D(34,13,14,15)	4.76
		A(21,20,41)	111.36	D(34,13,14,16)	-117.80
		A(40,20,41)	108.33	D(34,13,14,35)	126.62
		A(20,21,42)	108.79	D(12,13,19,18)	-94.08
		A(23,22,27)	119.79	D(14,13,19,18)	23.13
		A(23,22,28)	120.62	D(34,13,19,18)	145.48
		A(27,22,28)	119.59	D(13,14,15,36)	167.12
		A(22,23,24)	120.17	D(16,14,15,36)	-77.02
		A(22,23,29)	115.32	D(35,14,15,36)	46.66
		A(24,23,29)	124.51	D(13,14,16,17)	-146.62
		A(23,24,25)	120.06	D(13,14,16,18)	-22.88
		A(23, 24, 43)	119.81	D(13,14,16,37)	92.82
		A(25,24,43)	120.13	D(15,14,16,17)	93.91
		A(24,25,26)	120.32	D(15,14,16,18)	-142.35
		A(24,25,44)	119.70	D(15,14,16,37)	-26.65
		A(26,25,44)	119.98	D(35,14,16,17)	-31.03

A(6,26,25)	119.60	D(35,14,16,18)	92.71
A(6,26,27)	121.19	D(35,14,16,37)	-151.59
A(25,26,27)	119.19	D(14,16,17,38)	-172.90
A(22,27,26)	120.46	D(18,16,17,38)	71.31
A(22,27,45)	118.58	D(37,16,17,38)	-53.97
A(26,27,45)	120.96	D(14,16,18,19)	36.37
A(22,28,46)	108.58	D(14,16,18,20)	154.03
A(23,29,47)	110.65	D(14,16,18,39)	-80.81
A(10,30,48)	109.38	D(17,16,18,19)	155.80
A(8,31,49)	110.10	D(17,16,18,20)	-86.54
		D(17,16,18,39)	38.62
		D(37,16,18,19)	-78.29
		D(37,16,18,20)	39.37
		D(37,16,18,39)	164.53
		D(16,18,19,13)	-37.56
		D(20,18,19,13)	-158.08
		D(39.18.19.13)	80.53
		D(16.18.20.21)	171.33
		D(16,18,20,40)	50.73
		D(16.18.20.41)	-66.68
		D(19.18.20.21)	-75.16
		D(19,18,20,40)	164.24
		D(19,18,20,41)	46.83
		D(39.18.20.21)	45.83
		D(39,18,20,40)	-74.77
		D(39.18.20.41)	167.82
		D(18,20,21,42)	169.43
		D(40,20,21,42)	-71.31
		D(41.20.21.42)	49.09
		D(27,22,23,24)	0.54
		D(27,22,23,29)	-179.73
		D(28,22,23,24)	-179.01
		D(28,22,23,29)	0.72
		D(23,22,27,26)	-0.39
		D(23.22.27.45)	179.97
		D(28.22.27.26)	179.17
		D(28,22,27,45)	-0.47
		D(23.22.28.46)	-0.89
		D(27 22, 28, 46)	179 56
		D(22,23,24,25)	0.01
		D(22.23.24.43)	179.92
		D(29,23,24,25)	-179.69
		D(29,23,24,23)	0.22
		D(22,23,29,47)	179.89
		D(24.23.29.47)	-0.40
		-(-,,,-/)	0.10

			D(23 24 25 26)	0.71
			D(23,24,23,20)	-0.71
			D(23,24,25,44)	179.48
			D(43,24,25,26)	179.37
			D(43,24,25,44)	-0.43
			D(24,25,26,6)	179.29
			D(24,25,26,27)	0.86
			D(44,25,26,6)	-0.91
			D(44,25,26,27)	-179.33
			D(6,26,27,22)	-178.71
			D(6,26,27,45)	0.92
			D(25,26,27,22)	-0.31
			D(25,26,27,45)	179.32
*R: Bond lengt	h (Å), A: Bond	l angle (°), D	: Dihedral angle (	(°).

Table 4.	<b>Table 4.</b> Computed vibrational wavenumbers (cm <sup>-1</sup> ) of avicularin.					
Mode	Wavenumbers	Wavenumbers	Assignments (%)			
number	(computed)	(scaled)				
141	3848,73	3675.54	<b>v</b> OH (80.8)			
140	3848,29	3675.12	<b>v</b> OH (81.6)			
139	3841,40	3668.54	<b>v</b> OH (82.3)			
138	3839,47	3666.70	<b>v</b> OH (82.6)			
137	3836,57	3663.93	vOH (86)			
136	3820,21	3648.30	<b>v</b> OH (87.7)			
135	3795,73	3624.93	<b>v</b> OH (90.7)			
134	3224,17	3079.08	<b>v</b> CH (73.6)			
133	3219,35	3074.48	<b>v</b> CH (74.2)			
132	3211,67	3067.14	<b>v</b> CH (76.4)			
131	3155,68	3013.67	<b>v</b> CH (75.7)			
130	3141,99	3000.60	<b>v</b> CH (74.6)			
129	3085,30	2946.46	vCH (49.8)			
128	3063,00	2925.16	<b>v</b> <sub>s</sub> CH (46.9)			
127	3056,96	2919.40	<b>v</b> <sub>as</sub> CH (44.6)			
126	3012,28	2876.73	<b>v</b> <sub>as</sub> CH (56.7)			
125	2995,74	2860.93	<b>v</b> <sub>s</sub> CH (42.9)			
124	2951,38	2818.56	<b>v</b> <sub>s</sub> CH (65.6)			
123	1711,54	1655.06	vC=O (14.4)			
122	1661,32	1606.50	$\nu$ C=C (13.6) + $\nu$ CC (13.2)			
121	1654,13	1599.54	$\nu$ C=C (11.3) + $\nu$ CC (12.4)			
120	1653,11	1598.56	<b>v</b> C=C (15.4) + <b>v</b> CC (11.9)			
119	1632,78	1578.90	vC=C (16.2) + vCC (11.6)			
118	1616,61	1563.26	$\nu$ C=C (15.6) + $\nu$ CC (13.2)			
117	1539,26	1488.46	$\nu CC (12.1) + \nu C = C (8.2)$			
116	1530,58	1480.07	$\nu CC(14) + \nu C = C(8)$			

115	1495,87	1446.51	δHCH(12.6)
114	1494,38	1445.07	$\nu$ C=C (8.9) + $\nu$ CC (8.5)
113	1475,97	1427.26	$\nu$ C=C (8.4) + $\nu$ CC (7.9)
112	1466,74	1418.34	δCCH(17.8)
111	1451,00	1403.11	δOCH(8.1)
110	1417,82	1371.03	δOCH(8)
109	1408,07	1361.61	vC=C (3.9)
108	1395,16	1349.12	$\delta OCH(9.3) + \delta OCH(8.5)$
107	1383,73	1338.07	δOCH(9.8)
106	1376,68	1331.25	$\nu$ CC (12.7) + $\nu$ C=C (9.2)
105	1375,52	1330.13	$\nu$ CC (15.5) + $\nu$ C=C (12.6)
104	1355,25	1310.53	<i>τ</i> HCCH (10.1)
103	1347,00	1302.55	τHCCH (15.3)
102	1338,42	1294.25	τHCCH (9.2)
101	1333,55	1289.54	δCCH(8.9)
100	1320,31	1276.74	δCCH(12.5)
99	1313,03	1269.70	δCCH(8.3)
98	1289,15	1246.61	δC=CH(12)
97	1287,20	1244.72	$\tau$ HCCH (14.4) + $\delta$ CCH(13.6)
96	1274,58	1232.52	$\delta OCH(15.8) + \tau OCCH(13)$
95	1269,73	1227.83	δC=CH(13.4)
94	1235,88	1195.10	δCOH(13.7)
93	1234,70	1193.95	δCOH(12.7)
92	1230,24	1189.65	$\delta \text{COH}(12.3) + \delta \text{CCH}(11.7)$
91	1217,71	1177.52	8CCH(12.3)
90	1216,19	1176.06	8CCH(11.3)
89	1212,36	1172.35	$\frac{\delta \text{COH}(16.1) + \delta \text{CCH}(13.4)}{\delta \text{COH}(13.4)}$
88	1204,48	1164.73	$\frac{\delta C = CH(11.8) + \delta COH(9.4)}{\delta COH(9.4)}$
8/	1192,11	1152.77	$\frac{\partial CCH(8.6)}{\partial C}$
86	11/3,95	1135.20	$\delta C = CH(18.6) + \delta COH(20.6)$
85	11/1,59	1132.93	0CCH(14.9)
04	1142,32	1104.82	0C - CH(10.7)
0.5	1122,02	1083.37	VCC (0.8)
82	1120,02	1083.03	8C-CU(5.6)
80	1114,02	1077.84	3C - CH(5.0)
00	1104,43	1007.90	
79	1092,83	1036.//	
/8	1081,33	1045.64	0CCH(7.4)
	10//,01	1041.47	
/6	1009,95	1034.64	
/5	1042,83	1008.42	<b>v</b> CO (9.6)
74	1027,14	993.24	<b>v</b> CO (9.1)
73	1017,13	983.56	VCC (8.3)
72	997,96	965.03	vCC (8.8)
71	988,05	955.44	vCC (8.1)

70	957,15	925.56	<b>v</b> OC (5.8)
69	935,48	904.61	<i>τ</i> HC=CH (15.3)
68	916,36	886.12	τCC=CH (16)
67	906,95	877.02	$\tau CC = CH(6.2)$
66	857,54	829.24	тнсон (10.6)
65	847,26	819.30	$\delta CC = C(6.3)$
64	826,78	799.50	$\tau HC=CO$ (13.4) + $\tau OCCH$
	,		$(13.1) + \tau HC = CC (10.2) +$
			$\tau C = CCH (10.1)$
63	815,39	788.48	τOCCH (5.1)
62	807,12	780.48	τOCCH (6.3)
61	793,63	767.44	vC=C (8.1)
60	779,71	753.98	<i>т</i> HC=CO (11.1) + <i>т</i> ОССН
			(10.7)
59	743,81	719.27	$\tau OC = CC (3.9)$
58	729,57	705.49	$\tau CC = CC (5.7)$
57	725,32	701.38	τCOCO (2.3)
56	710,99	687.53	$\tau CC = CC (2.6)$
55	694,20	671.29	$\tau C = CCH (3.7)$
54	674,91	652.64	$\tau C = CC = C(8)$
53	643,66	622.42	<i>τ</i> HC=CC (3.8)
52	631,48	610.64	<i>τ</i> HC=CC (4.2)
51	621,09	600.59	<i>τ</i> HC=CC (2.7)
50	616,09	595.76	<i>τ</i> HC=CC (5.2)
49	601,05	581.21	δCC=C (8.4)
48	594,43	574.82	δC=CO (2.8)
47	589,51	570.05	δCCH (2.7)
46	571,80	552.93	$\tau CC = CC (4.4)$
45	565,54	546.88	δC=CO (4.2)
44	560,57	542.07	τНСОН (5)
43	530,93	513.41	δC=CO (7.3)
42	504,55	487.90	τCCOH (2.8)
41	479,59	463.76	δCCO (5.7)
40	462,89	447.61	δССО (2.8)
39	455,09	440.07	τC=COH (9.4)
38	434,91	420.56	τНСОН (3.2)
37	408,53	395.05	$\tau$ CCOH (31.4) + $\tau$ C=COH
			(29)
36	403,70	390.38	τCCOH (3.2)
35	396,32	383.24	τCCOH (5.3)
34	365,96	353.89	<i>τ</i> CCOH (7.1)
33	362,59	350.62	$\tau C = COH (11) + \tau C = COH$
			(10.2)
32	355,64	343.90	<i>τ</i> CCOH (12.9)
31	333,21	322.21	τCCOH (4.3)
30	326,78	316.00	<i>τ</i> CCOH (14.5)

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29	323,09	312.43	$\tau \text{COC}=\text{C}(2.6)$
28	310,56	300.32	δC=CO (10.1)
27	304,25	294.21	$\tau$ CCOH (19.5) + $\tau$ C=COH
			(18.1)
26	293,67	283.98	$\tau$ CCOH (16.7) + $\tau$ C=COH
			(7.9)
25	273,81	264.77	$\tau$ CCOH (16.9) + $\tau$ HCOH
			(9.1)
24	264,17	255.46	$\tau$ CCOH (27.7) + $\tau$ HCOH
			(14)
23	257,84	249.33	$\tau CCOH(8) + \tau C = COH(6.4)$
22	245,29	237.20	$\tau$ CCOH (29.4) + $\tau$ HCOH
			(14.7)
21	235,91	228.13	$\tau$ CCOH (24.3) + $\tau$ HCOH
			(11.8)
20	231,14	223.51	$\tau$ CCOH (16.4) + $\tau$ HCOH
			(7.8)
19	218,90	211.68	$\tau$ CCOH (12.7) + $\tau$ C=COH
			(9.3)
18	212,57	205.56	$\tau$ HCOH (19.8) + $\tau$ CCOH
			(9.8)
17	209,94	203.02	$\tau CCOH (18) + \tau C=COH$
			(12.2) +
			<i>τ</i> HCOH (10.4)
16	204,35	197.60	$\tau$ CCOH (15) + $\tau$ C=COH (13)
15	197,52	191.00	$\tau$ CCOH (10.9) + $\tau$ HCOH (8)
14	182,40	176.39	$\tau$ HCOH (21.6) + $\tau$ CCOH
			(11.2)
13	174,78	169.01	$\tau$ HCOH (17.2) + $\tau$ CCOH (9)
12	137,62	133.08	$\tau CC = CO(6.2)$
11	117,26	113.39	<i>τ</i> HCOH (4.7)
10	102,47	99.09	$\tau$ CCCH (10.1) + $\tau$ OCCH
			(9.5)
9	90,51	87.52	$\tau C = CC = O(3.2)$
8	74,50	72.04	<i>τ</i> CCOC (7.6)
7	66,29	64.10	<i>τ</i> CCOC (3.8)
6	44,67	43.20	τΟССО (7.9)
5	40,18	38.85	τΟССО (7.7)
4	33,73	32.61	$\tau$ COCCa< mq (7.1)
3	28,53	27.59	$\tau C = CCC (5.5) + \tau C = CC = C$
			(5.4)
2	21,43	20.73	$\tau C = CCC (4.3)$
1	15,11	14.62	τΟССО (6.6)

#### 4. Conclusions

Our research has highlighted the importance of the application of docking strategies based on selected flavonoids we retrieved from the ZINC database. We have presented how these flavonoids were oriented within the active site of the SARS–CoV–2 MPro. Our results suggested that a strong hydrogen bonding (1.65 Å) formed between the Gly–143 and

we suggest that this can be attributed to an initial formation of a covalent bond. Most of our lead flavonoids showed H–bond interactions with Hie–41 and/or Cys–145, which were important residues for the catalytic activity of SARS–CoV–2 Mpro. At this point, we are aware that it is still not easy to make any direct conclusion because the nature and action mechanisms of both catalytic dyads and triads are still not clear. It should be noted that we did not use covalent bonding docking strategies, but our results can still be regarded as satisfactory.

Since molecular docking studies are almost the gold standard for screening new drugs and their targets, any type of candidate drug and determination of its interactions with SARS-CoV-2 Mpro is of importance. Based on our findings, the selected flavonoids could be promising inhibitors of SARS-CoV-2 MPro and they also need to be rapidly confirmed by experimental and clinical studies. We also suggest that the outcomes of this study can be led to design new and more potent drugs against fatal COVID-19 disease. It must be also considered that these findings solely might not be generalized to any other representative MPros of SARS-CoV-2 MPro. Taken together, our work clearly has some limitations, and the most important one is the lack of clinical applications that would confirm our results. Finally, our findings would seem to suggest that we have provided further evidence that flavonoids are potential inhibitors of SARS-CoV-2 MPro. Since the whole picture is still not complete, we believe that our results provide a stimulus for further research on docking of flavonoids to more Mpros and add to a growing body of literature on the way of understanding of mechanism of SARS-CoV-2.

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