

The Turkish Journal of Occupational / Environmental Medicine and Safety

Vol:1, Issue Supplement 1 Web: <u>http://www.turjoem.com</u> ISSN : 2149-4711 Poster Presentation

P29. PHOTOCATALYTIC REMOVAL OF TOXIC CR(VI) WITH PHTHALOCYANINE IMMOBILIZED TIO $_2$

Melek KOÇ, Cansu ALBAY, İlknur ALTIN, İsmail DEĞİRMENCİOĞLU, Rıza BAYRAK, Münevver SÖKMEN Department of Chemistry, ScienceFaculty, Karadeniz Technical University, 61080, Trabzon, TÜRKİYE

Chromium is one of the metal pollutants which is used in different industrial fields. Because of its toxicity and mobility hexavalent chromium Cr(VI) is trouble matter and remove it from the waste waters is quitediffucult. Cr(VI) exposure via inhalation cause carcinogenic effect on the living mechanism. As Cr(VI), a potential carcinogen present in the environment, represents a significant safety concern, it is currently the subject of an EPA health risk assessment. Therefore, sensitive and selective removal of this specie is highly desired. Removal of Cr(VI) in waste waters have been improved physical, chemical, biological procedures and so forth. In recent years, photocatalytic reduction as advanced chemical techniques is a suitable way for effective removal of Cr(VI). This study have been focused on photo reduction of Cr(VI) with dye sensitized TiO₂ under UV-vis light. In this study, phthalocyanine (Pc) compounds are used as dye sensitizer. Pc molecules are also known to have reductive properties and removing it from the aqueous solution after treatment is difficult so they are stabilized on to TiO₂ support surface. For this, 1%(w/w) Pc (two different chemical structures) was immobilized on TiO2. Pc/TiO2 derivatives (2g/L) were used for photocatalytic reduction of 10 ppm Cr(VI) as model pollutant. In this process, Cr(VI) solution was exposured with UV source (λ =365 nm) for 150 minutes. After a certain illumination period remaining unreduced Cr(VI) concentration was measured by spectrophotometric method. Control experiments were carried out with neat Pc and the results were compared with Pc-TiO₂.