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## Synthesis and Characterization of Cellulose-Based Bio-Polymer Aerogel Isolated from Waste of Blueberry Tree (*Vaccinium myrtillus*)

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**Abstract:** Cellulose aerogel (CA) with a highly porous structure is environmentally friendly, thermally stable, and flame retardant. These properties in material worlds have attracted large interest as a potentially industrial material. In this paper, cellulose aerogel with flame retardant feature was produced from pruned branches and bushes of blueberry wastes (PBW). Firstly, cellulose raw material was obtained from these wastes and then, cellulose aerogel was prepared via freeze-drying, followed by cellulose hydrogel production. The work showed that a three dimensional network aerogel structure was prepared from NaOH/Urea as scaffold solution. The present cellulose aerogel has excellent flame retardant property, which can extinguish within 140 s. By the way, it was inferred that thermal stability performance of cellulose aerogel could be an efficient potential thermal insulating material. Besides, this process is sustainable, easily available at low cost, and suitable for industrial applications.

**Keywords:** Aerogel; Cellulose; Bio-polymer; Blueberry; Hydrogel.

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## INTRODUCTION

Today, composite materials take an important place in the industrial area due to their variety and advantages (1-3). Some materials are inadequate to meet some physical and/or chemical properties required in the fields of application. When these materials are supported by another type of materials in various amounts, composite structures providing the required properties are obtained, which is a method commonly used nowadays. In recent years, people's interest in polymeric composite originating from cellulose obtained from vegetable wastes has gradually increased because of its gripping properties (4, 5). An important reason why cellulose products are preferred is that it provides composite materials with some advantages such as high strength and thermal insulation (6). Being organic substances, cellulosic waste materials also bio-degrade in a short while and they do not have any disadvantages except their unaesthetic effects on the environment. Utilizing cellulosic waste materials and recycling them as new useful composite materials play a crucial role in contributing to the national economy. In addition, it decreases energy costs and consequently provides added value in regions where there are limited energy sources. Recently, aerogels from cellulosic wastes have great interest, because they have their own perfect feature (7). These kind of aerogels, which are among the lightest solid and a new third-generation materials, are one of the finest insulation materials available (8, 9). Traditional aerogels are mainly made of silica, which is not environmentally-friendly. In contrast, cellulose is a major component of wood, the most abundant, low cost naturally occurring polymer of glucose and attracts a lot of interest for producing novel aerogel materials. In addition to thin films, sponges and fibers, functional materials with high added value, such as cellulose nanocrystals, nanofibers, hydrogels and aerogels can be made from cellulose (10-13). Cellulose based aerogels have several unique features such as super absorbent and heat insulating. In this regard, when the variety of sources is taken into consideration, it is aimed to use PBW found abundantly in the Eastern Black Sea region, especially in Rize, as an economic value in Turkey. With this study, a new product which is biopolymer, eco-credential, biodegradable, low-cost, and flame-retardant was achieved using cellulose as a raw material obtained from wastes of the pruned blueberry tree, which is one of the important herbal products in the mentioned region. In addition to the structural and thermal characteristics, flame retardation, and surface properties of CA biopolymer materials obtained from the study were determined with the help of the scanning electron microscope (SEM), Fourier transform infrared spectroscopy (FT-IR), thermal analysis (TG-DTG/DTA), X-ray diffraction (XRD), Brunauer-Emmett-Teller (BET) surface area measurement techniques (pore size, volume, type, and surface areas), and flame retardant measurements.

## EXPERIMENTAL SECTION

### Materials

All chemicals (urea, sodium hydroxide, and ethanol) were of analytical grade and purchased from Merck. All the solutions were prepared with deionized (DI) water. The isolation of cellulose fibers from PBW was performed by a way of an environmental friendly process for cellulose extraction and bleaching .

### Cellulose production from PBW

The PBW was washed with DI water for a few times and filtrated using a cloth strainer to remove impurities such as water soluble sand, soil *etc.* Alkaline treatment was carried out in NaOH solution of 4 wt% at 80°C for three times. Alkaline treated fibers were subjected to NaClO<sub>2</sub> or H<sub>2</sub>O<sub>2</sub> bleaching of 1.7 wt% in acetic acid buffer at 80°C for 4 times. The resulting fibers were strained and rinsed with DI water and cloth strainer until it reached neutrality. Afterwards, the PBW sawdust powder was oven-dried at 105°C.

### Acid Treatment

Bleached PBW product was hydrolyzed in pre-heated H<sub>2</sub>SO<sub>4</sub> (64 wt%) (14). The high concentration of acid was removed through centrifugation at 10.000 rpm for 10 min and repeated until the solution was turbid. The resulting cellulose suspension was filtered using glass filter or cellulose membrane in DI water until the suspension reached a pH of 5.

### Production of Cellulose Aerogels

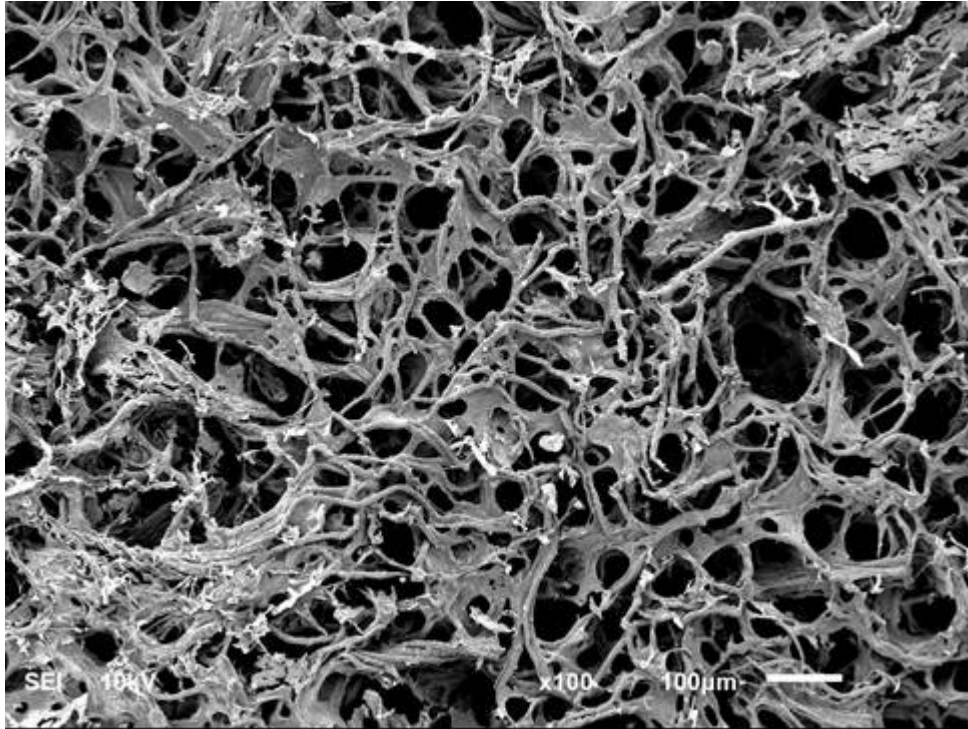
Cellulose fibers (2 wt %) were dispersed into a sodium hydroxide/urea solution (1.9 wt %/10 wt %) by sonicating for 6 min. Then the solution was placed in a freezer for more than 24 h for gelation to take place. After the solution had been frozen, it was then thawed at room temperature, followed by immersion into ethanol (99 vol %) for coagulation (3 h). The specimen's thickness was controlled around 1 cm with a diameter of 3.8 cm using a beaker as a mold. After coagulation, solvent exchange was carried out by immersing the gel in DI water for 2 days. Freeze-drying was carried out for the sample for 2 days at -98 °C with a freeze-dryer after pre-freezing the sample at -18 °C for 12 h (15).

## RESULTS and DISCUSSIONS

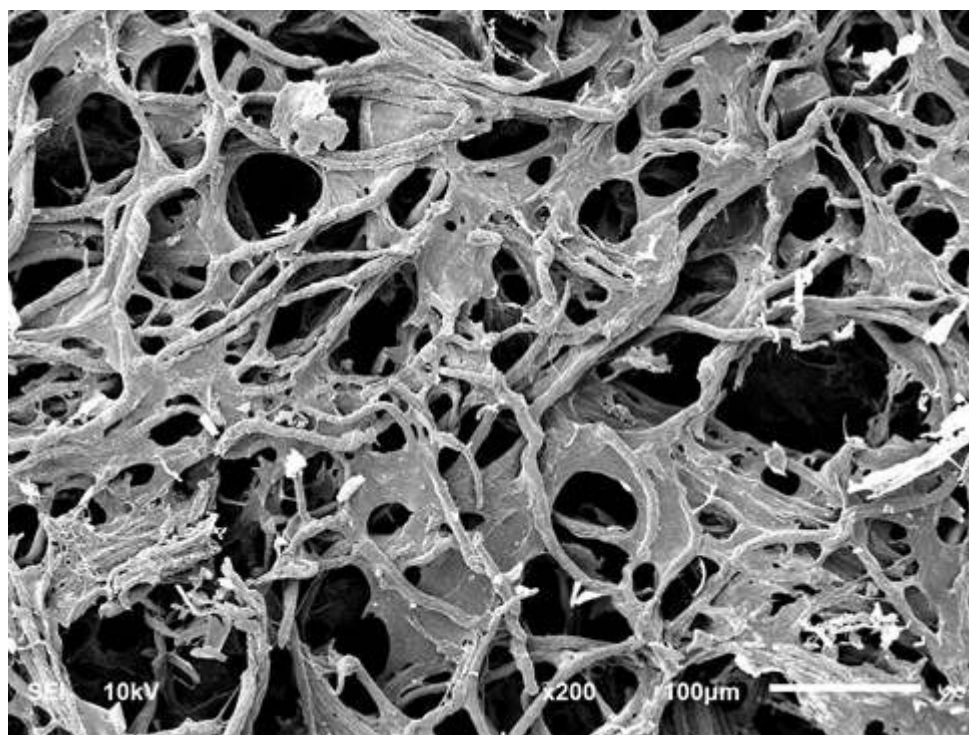
### SEM Analysis

The structural and morphological characteristics of CA were displayed by SEM and are shown in Figures 1-3. It is clear that the CA has a largely homogeneous porous structure consisting of microfibrillar networks (Figure 3). The pore structure results from the sublimation of frozen

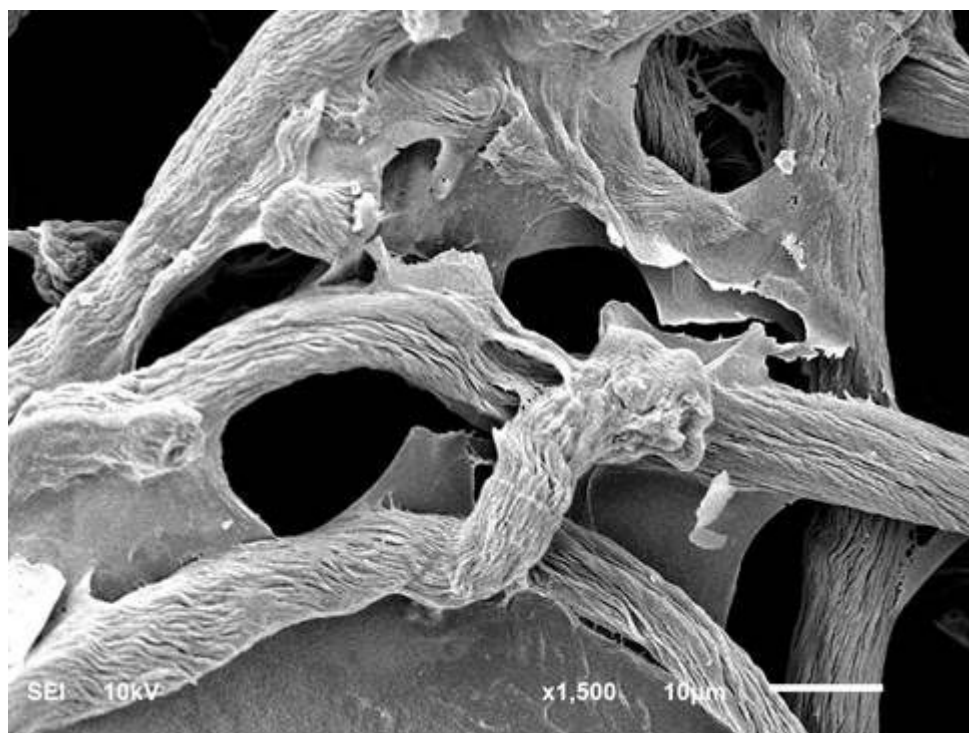
water during the freeze-drying process (16). This property is coherent with those of nitrogen adsorption data, and seems to be reasoned by crystal formation during the relatively slow freezing process, which would have caused to press out fibrils to form the film-like structure (17). According to the aforesaid findings, the CA with a homogeneous microporous structure can be fabricated via *in situ* templated synthesis of sample in the cellulose gel framework (16).



**Figure 1.** SEM image x100 of CA



**Figure 2.** SEM image x200 of CA.

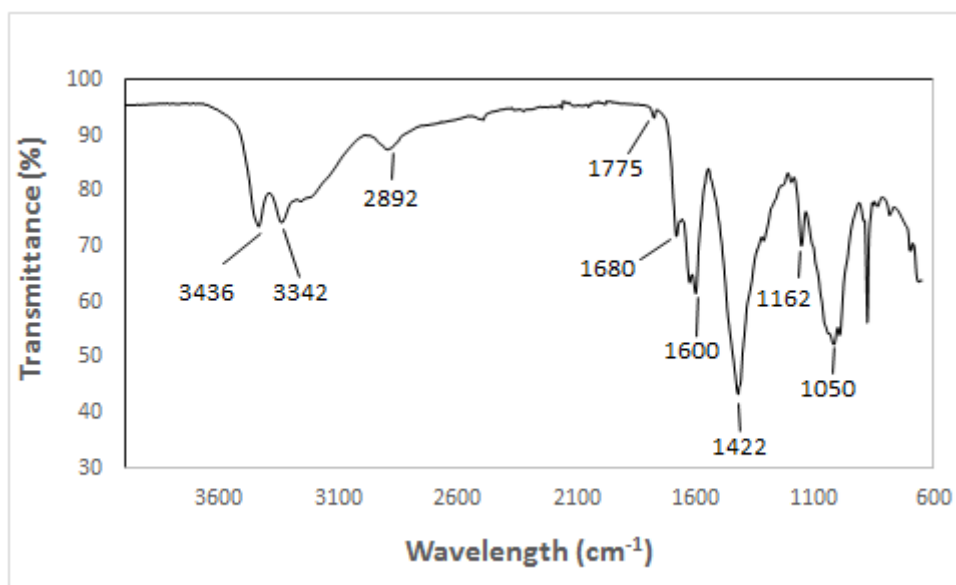


**Figure 3.** SEM image x1500 of CA.

### **FT-IR Analysis**

The FT-IR signal pertaining to the characteristic functional groups are shown in Figure 4. It can be observed that the several components of the CA are most likely consisted of alkene, esters, aromatics, ketone and alcohol, with different oxygen-containing functional groups observed,

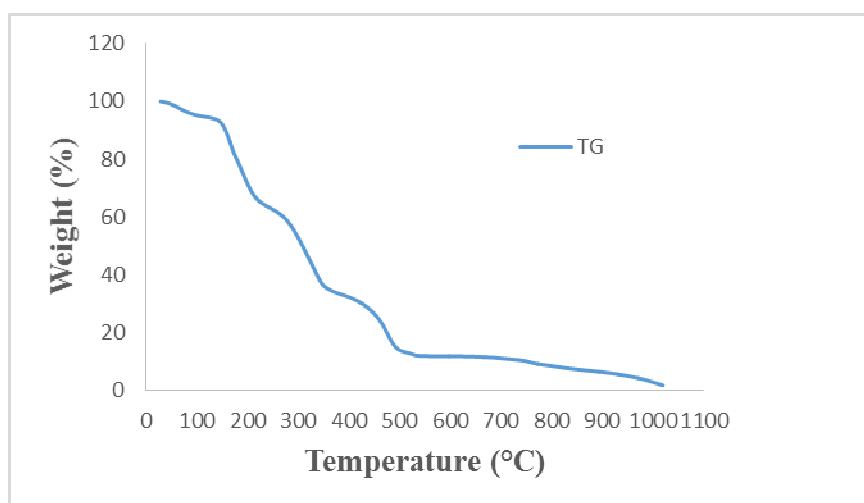
e.g., OH ( $3450\text{--}3300\text{ cm}^{-1}$ ), C=O ( $1775\text{--}1680\text{ cm}^{-1}$ ), C-O-C ( $1162\text{ cm}^{-1}$ ), and C-O-(H) ( $1050\text{ cm}^{-1}$ ) (18).



**Figure 4.** FT-IR Spectrum of CA.

### Thermal Analysis

Considering industrial applications of heat insulating materials, the thermal behavior of the CA was investigated by TGA analysis. As demonstrated in Figure 5, a gradual weight loss occurs up to about 500 degrees. The weight loss at a temperature range of  $0^{\circ}\text{C}$ - $200^{\circ}\text{C}$  can be attributed the loss of moisture in both surface and inner moisture evaporation in porous structure where removal of water is more difficult relatively (18). The stage of weight loss around  $300\text{--}500^{\circ}\text{C}$  can be ascribed to the depolymerization of cellulose with generation of  $\text{CO}_2$  and volatile hydrocarbons (16). Finally, it appeared that the CA has high thermal stability (weight loss only 60 % at about  $350^{\circ}\text{C}$ ).



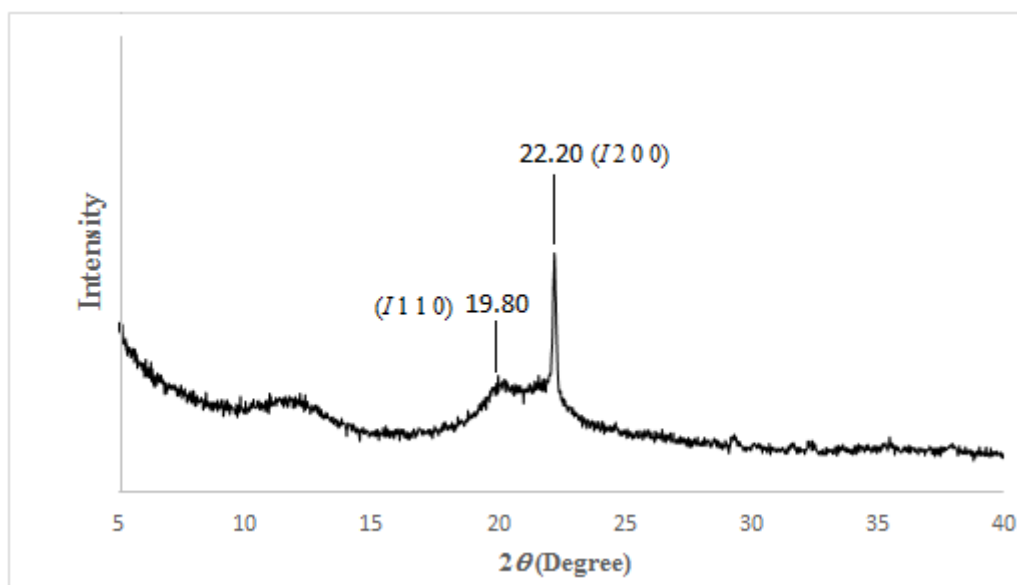
**Figure 5.** Thermal Stability of CA.

### XRD Analysis

XRD spectrum for CA was performed on a Rigaku DMAX-3C automated diffractometer using Ni filtered CuK-beta radiation (40 kV and 30 mA). Diffractograms were recorded from 5° to 40° at a scan rate of 3°/min. Figure 6 displayed a crystal structure of cellulose with typical diffraction peaks for the plane (200) at  $2\theta = 22.20^\circ$  and (110) at  $2\theta = 19.8^\circ$ . Crystallinity index (CrI) was calculated from the intensity of the 200 peak ( $I_{200}$ ,  $2\theta = 22.2^\circ$ ) and the intensity minimum between 200 and 110 ( $I_{am}$ ,  $2\theta = 19.8^\circ$ ) peaks using the empirical equation (8):

$$CrI = \{(I_{200} - I_{am}) / I_{200}\} \times 100 = 76.52 \% \quad (\text{Eq. 1})$$

It was observed that the cellulose aerogel had a higher crystallinity (76.52%) than that of the raw amorphous woody material, probably because of the removal of hemicellulose and lignin. This result also reveals that amorphous and crystalline cellulose regions are considerably divided.



**Figure 6.** XRD Spectrum of CA.

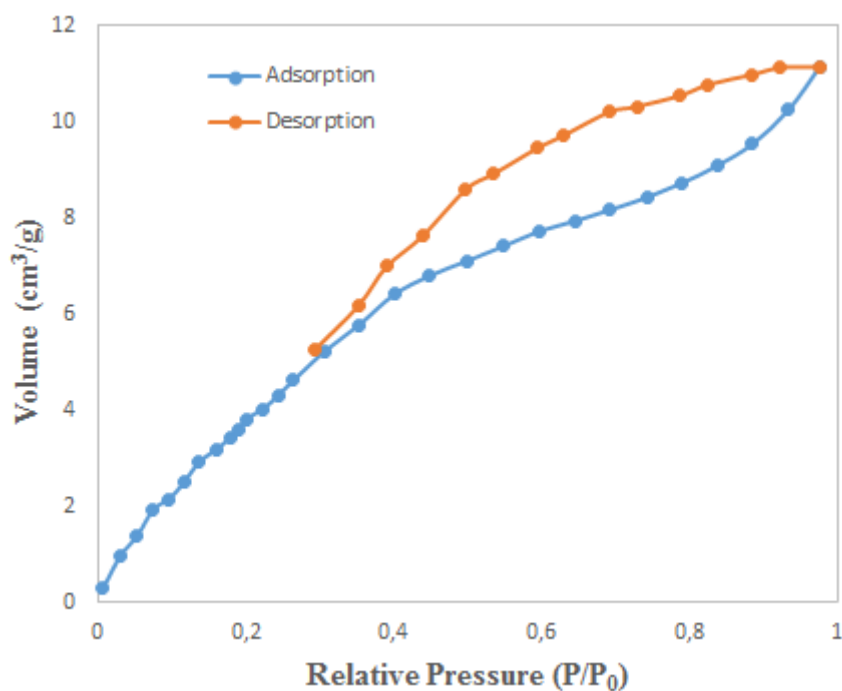
### BET Specific Surface Area

The pore features of the CA involving the total pore volume, average pore diameter, and specific surface area (from N<sub>2</sub> adsorption measurements) are shown in Table 1. The results showed the CA with the total pore volume and average pore diameter as 0.25 cm<sup>3</sup>/g and 4.30 nm, respectively. The nitrogen adsorption at 77 K on CA resembles the type II isotherms (according to IUPAC classification), which are typical for physical adsorption on microporous and non-porous or macroporous solids (Figure 7). The characteristic features of these isotherms are the reversible part at relatively lower pressures and the hysteresis loop at relatively higher pressures (19). To examine the porous structure of the aerogel, nitrogen

adsorption and desorption isotherm techniques were performed. Figure 7 exhibits the adsorption and desorption isotherms, which show a hysteresis typical of a porous system. According to BET analysis, a specific surface area of  $42.25 \text{ m}^2 \text{ g}^{-1}$  is recorded for the CA.

**Table 1.** Specific Surface Area and Porosity.

Sample	Specific surface area ( $\text{m}^2\text{g}^{-1}$ )	Average pore diameter (nm)	Total pore volume ( $\text{cm}^3\text{g}^{-1}$ )
CA	42.25	4.30	0.25



**Figure 7.** Nitrogen Adsorption-Desorption Isotherm.

### Flame Retardant Characteristic of CA

Due to the investigation, the flame retardant performance of the CA produced in this work, the aerogel was emblazed and the burning velocity was calculated. At this stage, it was observed that the CA was ignited slowly, and burned only the part of 25 percent burned after 33 s of combustion (see the video in Supporting Information). However, the neat cellulose aerogel was quickly ignited (about 10 s) and burned completely after 10 s of combustion in the literature (16). The photo in Figure 8 demonstrates the remains of the CA after 34. The combustion velocity of the aerogel decreases from  $4$  to  $3.71 \text{ mm s}^{-1}$  (Figure 8). Besides, when the CA with the length of 40 mm was ignited, flame propagation apparently slowed, and then the flame gradually became smaller and finally extinguished within about 140 s. The results indicate that the CA is a highly an efficient flame retardant.





**Figure 8.** Flame Retardant Study.

## CONCLUSIONS

The cellulose aerogel with a highly porous structure was produced from pruned branches and bushes of blueberries via *in situ* converting cellulose hydrogel scaffold to aerogel. This work is an efficient, environmentally friendly, and low cost process and the bio-polymer material is an effective flame retardant, porous, and potential super absorbent (next studies). It was concluded that the prepared cellulose aerogel has extremely porous networks and perfect flame retardant performance. Besides, the cellulose aerogel has good crystallinity feature and thermal stability. This environmentally friendly and low cost work will offer insight into the new opportunities for the use of inexpensive waste-derived cellulose to fabricate several materials. It has a major potential application in the future of “green” materials chemistry and it may be appropriate for industrial scale production.

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### Türkçe Öz ve Anahtar Kelimeler

## Yabanmersini Ağacı Atıklarından (*Vaccinium myrtillus*) Selüloz Esaslı Biyo-Polimer Aerojelinin Sentezi ve Karakterizasyonu

Mehmet KAYA

**Öz:** Selüloz aerojeli (CA) çok gözenekli bir yapıya sahiptir ve çevreyle dosttur, ısı olarak kararlıdır ve alev geciktirici özelliği vardır. Malzeme dünyasında bu özellikler çok ilgi çekmekte olup potansiyel olarak endüstriyel bir malzeme çıkmasına da yol açabilir. Bu çalışmada, alev geciktirici özelliğine sahip selüloz aerojeli, yabanmersini ağacı atıklarından budanmış dal ve çalılardan üretilmiştir. Öncelikle selüloz ham maddesi elde edilmiş ve daha sonra selüloz hidrojel üretilmiştir. Ardından dondurarak kurutma ile selüloz aerojel oluşturulmuştur. Çalışmada, yapı iskelesi çözültüsü olarak NaOH/Üreden oluşan üç boyutlu ağ sistemine sahip aerojelin üretilmesi anlatılmıştır. Elde edilen selüloz aerojelin mükemmel alev geciktirici özelliği bulunmaktadır ve 140 saniye içinde sönüm sağlanmaktadır. Bunun dışında, selüloz aerojelin ısı kararlılık özelliğinin mükemmel bir ısı yalıtkan olabileceği düşünülmektedir. Son olarak, süreç sürdürülebilir özelliğe sahiptir, düşük maliyetle kolaylıkla elde edilebilir ve endüstriyel uygulamalar için uygundur.

**Anahtar kelimeler:** Aerojel; Selüloz; Biyo-polimer; Yaban mersini; Hidrojel.

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