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Optimization of Ultrasound-Assisted Extraction of Antioxidants from *Apium graveolens L.* Seeds using Response Surface Methodology

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Abstract: In this study, optimum conditions for ultrasound-assisted extraction (UAE) of antioxidants from *Apium graveolens L.* seeds were investigated by Response Surface Methodology (RSM). A Box-Behnken Design (BBD) was used to evaluate the effect of sonication time (5, 10, 15 min), ultrasound power (60, 120, 180 W), and the ratio of extraction solvent in terms of methanol (0, 50, 100%) on antioxidant capacity. The optimal conditions of UAE for the parameters investigated were 11 min of sonication time, ultrasound power of 131 W and 100% methanol as an extraction solvent. Under these conditions, UAE of antioxidants from the seeds achieved a maximum of 95.08% in respect to 1,1-diphenyl-2-picryl hydrazyl (DPPH) radical scavenging activity. Additionally, the high value of the adjusted coefficient of determination (R^{2}_{adj} = 0.9192) and the non-significant difference between experimental and predicted values confirmed the validity of the quadratic polynomial model. Hence, UAE is a suitable, fast, economical and practical technique for the extraction of antioxidants from *Apium graveolens L.* seeds.

Keywords: *Apium graveolens L.* seed, ultrasound-assisted extraction, optimization, Box-Behnken Design, Response Surface Methodology (RSM).

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INTRODUCTION

In recent years, chemical methodologies have focused on the use of less toxic extractant phases and clean energies along with new techniques and materials in order to minimize human and environmental detriments. For instance, microextraction techniques have been developed instead of the more classical sample pretreatment-preconcentration techniques such as liquidliquid extraction and solid-phase extraction. Ultrasound that is one of the clean and efficient energy types has been commonly utilized for a good deal of sample preparation procedures (1). Various sample pretreatment steps such as digestion, dissolution, and extraction can easily be fulfilled thanks to its acoustic cavitation. This force can expedite the heat and mass transfer rate in order to disintegrate foods, sediments, soils or plant cell walls and allow the release of interested extractable substances. In terms of green chemistry, ultrasonic extraction has minimum solvent and energy consumption, effectiveness, and short operation time. It is also an easy-to-use and inexpensive method, so that it is employed in lots of studies in literature due to these many advantages (2-6).

Antioxidants neutralize reactive oxygen species including hydroxyl radicals, superoxide anions, nitric oxide, hydrogen peroxide, and so forth (7). Since they prevent cellular damage, aging, the creation and progress of cancer, and cardiac diseases, much interest has been shown in extracting natural antioxidant compounds from medicinal plants, herbal sources or vegetables. *Apium graveolens L.* which is widely known as celery contains flavonoids (apigenin, luteolin, and kaempferol), polyphenols (p-coumaric acid, caffeic acid, and ferulic acid), terpenoids, alkaloids, tannins, essential oils, and glycosides as major components along with β -carotene, ascorbic acid, folic acid, minerals such as sodium, potassium and magnesium (8). *Apium graveolens L.* seeds have also antioxidant and antibacterial activity as well as diuretic, carminative, sedative, and antirheumatic effects and they have potential for therapeutic efficacy (Figure 1) (9).



Figure 1: An image of Apium graveolens L. seeds.

Researchers have previously investigated antioxidant activity of *Apium graveolens L.* leaf, tuber, and seeds by using different extraction techniques containing Soxhlet (8-11), maceration, ultrasound-assisted, microwave-assisted, and supercritical carbon dioxide extraction (12).

Soxhlet extraction, which is a traditional extraction method is often time consuming and requires large amounts of organic solvents. Aydemir and Becerik (10) used this extraction technique which employed 150 mL of methanol and took for 4 hours in a shaker at room temperature. Maleki and coworkers (12) did not investigate optimum conditions for ultrasound-assisted extraction (UAE) of antioxidant activity from dried *Apium graveolens L.* in their studies. According to our literature survey, there is no study about optimization of UAE of antioxidants from *Apium graveolens L.* seeds. The key factors in UAE are type and ratio of extraction solvent, sonication time, ultrasound power, and so on, which need to be optimized. In this respect, He and coworkers optimized for microwave-assisted extraction conditions of total flavonoids, phenols and antioxidants from *Apium graveolens L.* leaves by experimental design (13).

Multivariate statistic techniques have been employed to optimize significant factors, making possible optimization of more than one factor simultaneously and so they give rise to different advantages including short operation-analysis time, cost-effective, reduced number of experiments, and consumption of solvents and materials needed (14). Response surface methodology (RSM) that can evaluate the effect of the factors and their interactions is a multivariate technique and is used to describe mathematical coherence of the experimental domain studied in the theoretical design through a response function (15). So far, RSM has been increasingly used to determine the optimum conditions of extraction techniques (2, 13, 16, 17), chromatographic (18, 19), and electroanalytical (20, 21) measurements.

This study has presented the use of RSM with Box-Behnken design (BBD) including three variables (sonication time, ratio of extraction solvent and ultrasound power) for the optimization of UAE of antioxidants from *Apium graveolens L.* seeds.

EXPERIMENTAL SECTION

Instrumentation

The antioxidant activity of *Apium graveolens L.* seed extracts was determined using a UV-Vis spectrophotometer (Mini UV 1240, Shimadzu, Tokyo, Japan). UAE process was performed using an Isolab ultrasonic bath (Wetheim, Germany) with the fixed-frequency of 40 kHz and at three power settings of 60, 120 and 180 W. A vortex mixer, model IKA MS3 was obtained from Staufen, Germany. A Hettich centrifuge, model EBA20 from Tuttlingen, Germany was used for centrifuging. Both a domestic grinder and an Eppendorf automatic micro-pipettes with variable capacity were employed and obtained from Hamburg, Germany.

Materials and Chemicals

Apium graveolens L. seeds were purchased from a local herbalist in Istanbul, Turkey and were ground in a domestic grinder daily. 1,1-diphenyl-2-picryl-hydrazyl (DPPH) was purchased from Sigma (Sigma-Aldrich GmbH, Sternheim, Germany). All other solvents were of analytical grade.

Ultrasound-assisted Extraction (UAE)

0.50 g of the powdered seeds of *Apium graveolens L.* was placed in a 100 mL Erlenmeyer flask and extracted with 3.0 mL of the solvent investigated (methanol, ethanol or 50%methanol:ethanol) at different time intervals (5, 10, 15 min) and different power values (60, 120 and 180 W) of ultrasonic bath (40 kHz) at room temperature ($25 \pm 1 \circ C$) with shaking. After the extraction process, the contents were centrifuged at 5000 rpm for 15 min. The supernatant was removed into a 50 mL beaker and evaporated under the fume hood. The residue was transferred into a 25 mL volumetric flask and diluted to its volume with absolute ethanol.

Determination of Antioxidant Capacity

The DPPH radical scavenging activity (DPPH SA, %) is often used to evaluate the antioxidant capacity of compounds. The method of Brand-Williams *et al.* (22) was used for the determination of scavenging activity of DPPH radical. 0.75 mL of extract solution mixed with 1.50 mL of freshly prepared DPPH (0.05 mM) in ethanol. The mixture was shaken vigorously and kept at room temperature for 30 min. Then the absorbance was measured at 517 nm against a blank (without extract) in a UV-Vis spectrophotometer. The scavenging activity of DPPH radical (DPPH SA, %) was calculated using the following formula:

$$DPPH SA (\%) = \left[\frac{A_{control} - A_{sample}}{A_{control}}\right] \times 100$$
(Eq. 1)

Where $A_{control}$ and A_{sample} are the absorbance of the control (DPPH solution without sample) and the absorbance of the test sample, respectively.

Box-Behnken Experimental Design and Optimization by RSM

The software Statgraphics Centurion XV (StatPoint, Inc., USA, 2005) was employed for experimental design, data analysis, and model building. The optimization step of the UAE was performed using a BBD with three levels, three factors incomplete factorial design. While the number of experiments required to investigate three factors at three levels should be 27 (3^3), a BBD uses 15 experimental runs (23). Based on preliminary studies, sonication time (X₁), ultrasound power (X₂) and ratio of extraction solvent (X₃, methanol:ethanol) were selected as factors with ranges of 5-15 min, 60-180 W and 0-100 %, respectively. The response was measured as DPPH scaving activity (%) as mentioned before. BBD is based on a second-order polynomial model, as follows (14):

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$$Y = \beta_0 + \beta_1 X_1 + \beta_2 X_2 + \beta_3 X_3 + \beta_{12} X_1 X_2 + \beta_{13} X_1 X_3 + \beta_{23} X_2 X_3 + \beta_{11} X_1^2 + \beta_{22} X_2^2 + \beta_{33} X_3^2$$
(Eq. 2)

Where Y is the predicted response, β_0 is the constant, β_1 , β_2 and β_3 are linear cofficients, β_{12} , β_{13} and β_{23} are interaction coefficients between the three invistigated factors, and β_{11} , β_{22} and β_{33} are quadratic coefficients. The analysis of variance (ANOVA) with 95% confidence level was carried out to confirm the significance of the estimated regression model.

Sonication time, X1 (min)	Ultrasound power, X ₂ (W) Solvent Ratio, X		
5 (-1)	60 (-1)	0 (-1)	
10 (0)	120 (0)	50 (0)	
15 (+1)	180 (+1)	100 (+1)	

Table 1: Coded and actual values of the factors selected for optimization of UAE.

RESULTS AND DISCUSSION

Modeling of Ultrasound-assisted Extraction (UAE) Process and Statistical Analysis

In order to determine optimum conditions and investigate the effects of independent variables (factors) on UAE of antioxidants from the seeds, a Box-Behnken design was carried out and the experimental results obtained and predicted values were represented in Table 2. After regression analysis was performed, the second-order polynomial model was developed between the response and three factors investigated. The best fitted model equation in terms of coded units was obtained as:

$$Y_{(DPPH SA\%)} = 90.41 + 0.896X_2 + 3.625X_3 - 2.703X_1^2 - 1.454X_2^2.....(3) (Eq. 3)$$

The adequacy and fitness of the model was tested by analysis of variance (ANOVA). The ANOVA results showing the significant factors for optimum UAE are presented in Table 3. The influence of critical factors and the model efficiency were evaluated according to Fisher's statistical analysis (F-value) and p-values obtained. The model F-value was determined as 18.70, which indicates that the quadratic polynomial model is significant. The high determination coefficient (R^2 =0.9711) of the estimated model denotes that 97.11% of the variations in UAE process can be explained by the model. The adjusted R^2 (R^2_{adj}), which is more suitable for comparing models with different numbers of independent variables, is 91.92%.

As seen Figure 2 and Table 3, while sonication time (X_1) is not significant to the extraction procedure in the selected working range, ultrasound power (X_2) and solvent ratio (X_3) parameters were found as statistically significant. Quadratic contributions of sonication time (X_1^2) and ultrasound power (X_2^2) also influence the UAE significantly. Meanwhile, solvent ratio

 (X_3) has a positive and the largest influence on UAE. It means the antioxidant activity increases with the increasing methanol ratio in the extraction solvent.

	v	v	v	DPPH SA (%)		
Run	A1 (min)	×2 (W)	×3 (%)	Experimental ^a	Predicted ^b	
1	5 (-1)	60 (-1)	50 (0)	85.00	84.70	
2	15 (+1)	60 (-1)	50 (0)	86.67	86.02	
3	5 (-1)	180 (+1)	50 (0)	85.56	86.21	
4	15 (+1)	180 (+1)	50 (0)	87.78	88.08	
5	5 (-1)	120 (0)	0 (-1)	85.02	84.33	
6	15 (+1)	120 (0)	0 (-1)	86.00	85.67	
7	5 (-1)	120 (0)	100 (+1)	91.00	91.33	
8	15 (+1)	120 (0)	100 (+1)	92.50	93.17	
9	10 (0)	60 (-1)	0 (-1)	84.00	84.98	
10	10 (0)	180 (+1)	0 (-1)	87.50	87.52	
11	10 (0)	60 (-1)	100 (+1)	93.00	92.98	
12	10 (0)	180 (+1)	100 (+1)	95.00	94.02	
13 ^c	10 (0)	120 (0)	50 (0)	90.23	90.41	
14 ^c	10 (0)	120 (0)	50 (0)	90.00	90.41	
15 ^c	10 (0)	120 (0)	50 (0)	91.00	90.41	

Table 2: Box-Behnken design matrix together with the experimental and predicted responses.

^a Experimental values of response.

^b Predicted values of response by the second-order polynomial model.

^c Central points used to determine the experimental error.

Table 3: ANOVA results of optimization of UAE condition							
Source	Sum of	Degree of	Mean of	F-Value	P-Value	Coefficient	
	Square	Freedom	Square			Estimate	
Model	155.63	9	17.29	18.70	0.0078	90.41	
X1	5.10	1	5.10	5.52	0.0656	0.799	
X2	6.43	1	6.43	6.95	0.0462	0.896	
Х3	105.12	1	105.12	113.68	0.0001	3.625	
X ₁ ²	26.99	1	26.99	29.19	0.0029	-2.703	
X ₁ X ₂	0.07	1	0.07	0.08	0.7864	0.137	
X ₁ X ₃	0.06	1	0.06	0.07	0.8053	0.125	
X ₂ ²	7.80	1	7.80	8.44	0.0336	-1.454	
X ₂ X ₃	0.56	1	0.56	0.61	0.4707	-0.375	
X ₃ ²	3.12	1	3.12	3.37	0.1258	0.919	
Total error	4.62	5	0.92				
Correct Total	160.25	14					
Model Summary			R ²		R ² adi		
Statistics			0.9711		0.9192		



Figure 2. Pareto chart (P=0.05) of standardized effects for BBD.

Additionally, it can be inferred that although sonication time (X_1) and ultrasound power (X_2) do not greatly influence the extraction procedure, better results are obtained for central or medium values of these parameters (Figure 3). Figure 4 also shows a good fitness between experimental and predicted values (this fact can also be observed in Table 2). Main Effects Plot for DPPH SA



Figure 3: Main effect of each factor investigated on DPPH scavenging activity (%).



Figure 4: Correlation of experimental and predicted values for DPPH scavenging activity (%).

RSM Graphs and Optimum Conditions for UAE Process

Three-dimensional (3D) response surface graphs were also plotted for the UAE procedure. The influence of sonication time and ultrasound power on DPPH scavenging activity was illustrated in Fig. 5a, with solvent ratio kept at its level +1 (100% methanol). At the central values of these parameters, maximum responses were obtained for the UAE. Fig. 5b reveals that an increase in the ratio of methanol in the extraction solvent resulted in higher antioxidant activity of the seeds and reached a maximum at 100% (predicted DPPH SA 95.0%). In terms of methanol use, this result comforms to those of the studies conducted by Shanmugapriya and Ushadevi (9), Aydemir and Becerik (10) and Sameh *et al.* (11). Considering green chemistry, methanol is not a good option as an extraction solvent. But this study wants to indicate that the use of ultrasound for

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the extraction of antioxidants from *Apium graveolens L.* seeds is an efficient technique compared to conventional extraction techniques such as Soxhlet extraction and maceration which are required for much longer operation times and larger amounts of organic solvent. On the other hand, Maleki *et al.* found to be 98.71% the antioxidant activity of the tuber of celery by using water:ethanol (50:50) with the help of ultrasound without optimization (12). Sonication time of this study is 45 min in a ultrasonic bath having the power of 80 W at 37 kHz frequency. For our study, the volume of methanol employed for each study is only 3.0 mL and the sonication time is 11 min.

Fig. 5c also demonstrates that using intermediate power of ultrasound and only methanol as an extraction solvent lead to obtain better extraction efficiency.

According to overall results of optimization study, optimum conditions for the UAE of antioxidants from *Apium graveolens L.* seeds were obtained: sonication time: 11 min; ultrasound power: 131 W; and methanol ratio: 100%.

It can be said that the UAE process developed is a fast and efficient method for extracting antioxidants from *Apium graveolens L.* seeds, thereby reducing the energy input. For the purpose of comparison in terms of operation time, extraction solvent type and ratio and so forth, Table 4 shows several applications of UAE of organic compounds from various matrix together with the optimal conditions of this study. These studies show that UAE has been used as an alternative to Soxhlet extraction and maceration.

Table 4: Summary of some reported studies on the UAE of natural compounds from differentmatrix.

Analyte	Matrix	UAE Conditions	Analytical Technique	Ref
Polyphenols	Olive tree leaves	50 kHz, 60 min, Ethanol (50%), 500 mg solid to 10 mL solvent ratio	UV-Vis Spectrophotometer	2
Antioxidants, phenolic compounds, anthocyanins	Red raspberry puree	Water, 20 kHz, 10 min	Photochemiluminescence , UV-Vis Spectrophotometer	3
Catechins and caffeine	Tea samples	10 mL methanol (8/2, v/v) or acetonitrile (1:1, v/v), 10 min	HPLC-Diode Array Dedector (DAD)	4
Fatty acids	Caper seed oil	25 mL of hexane, 40 °C, 45 min, 50 kHz, 550 W	Gas chromatography– MS	5
Polysaccharides	Longan fruit pericap	120 W, 22 min, 60 °C or 241 W, 18 min, 51 °C, in	UV-Vis Spectrophotometer	6
Polyphenols	Orange peel	25 kHz, 150 W, 40 °C, Ethanol: water (4:1, v/v),	HPLC, UV-Vis Spectrophotometer	16
Phenolic compounds, antioxidants, anthocyanins	Grape	10 L, 50% Ethanol, 24 min	UV-Vis Spectrophotometer	24
Antioxidants, flavonoids	Citrus peel	Methanol, 40 °C, 30 min	HPLC-DAD-mass spectrometry (MS)	25
Phenolic and silymarin compounds	Milk thistle seeds	Methanol, different time intervals (30, 60, 90, 120, and 150 min)	UV-Vis Spectrophotometer, HPLC	26
Flavonoids	<i>Adinandra nitida</i> leaves	45 kHz, Ethanol (63.84%), 30 25 min	UV-Vis Spectrophotometer	27
Phenolic and flavonol compounds, antioxidants	Grape pomace	Water as an extraction solvent, 40 kHz, 150 W/L, 25 min	UV-Vis Spectrophotometer	28
Anthocyanins	Red raspberries	1.5 M HCI: Ethanol (85%, v/v%) (15:85, v/v), 3.3 min	HPLC-MS	29
Anthocyanins	Haskap berries	40 kHz, 100 W, 80% Ethanol (contain 0.5% formic acid), 35 °C, 20 min	UV-Vis Spectrophotometer, HPLC	30
Antioxidants	<i>Apium graveolens</i> L. seeds	40 kHz, 131 W, in methanol, 11 min	UV-Vis Spectrophotometer	This work

84.0 85.0 86.0 87.0

88.0

89.0

90.0

91.0

92.0

93,0 94.0

95.0

96.0

-1^{-0.6}

ratio (%)





-0.2

Ultrasound power (W)

0,2

0,6

1

-0.6

DPPH SA (%)

94

92

90

88

86

84 -1

Figure 5: Response surface plots for the antioxidant capacity of the extracts as a function of (a) sonication time to ultrasound power (solvent ratio = methanol 100%); (b) sonication time to solvent ratio (ultrasound power = 131 W); (c) ultrasound power to solvent ratio (sonication time = 11 min).

CONCLUSIONS

The ultrasound-assisted extraction of antioxidants from *Apium graveolens L*. seeds has been developed and demonstrated to be effective, fast, low-cost, and practical. Conditions of the UAE process were successfully optimized by response surface methodology using Box–Behnken design. Furthermore, the effects of main factors and possible interactions between factors were investigated. As expected, the results revealed that antioxidant activity of *Apium graveolens L*. seeds was influenced most significantly by the ratio of extraction solvent in the selected working range. The mathematical model obtained by RSM also has a high determination coefficient (R² = 0.9711, R²_{adj} = 0.9192) and shows a good agreement between the predicted and corresponding experimental results. Consequently, antioxidants from *Apium graveolens L*. seeds can be extracted at a high rate and efficiency by using ultrasound-assisted extraction.

COMPLIANCE WITH ETHICAL STANDARDS

The authors declare that there is not an unethical situation.

CONFLICT OF INTEREST

The authors declare that they have no conflict of interest.

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