



## Synthesis and Characterization of Novel Selenium N,S-Heterocyclic Carbene Compounds

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### Yeni Selenyum N,S-heterosiklik Karben Bileşiklerinin Sentezi ve Karakterizasyonu

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#### Öz

Benzotiyazol ve ilgili benzil halojenürler kullanılarak iki yeni benzotiyazolium tuzu N-(4-metoksibenzil)benzotiyazolium klorür (1a) ve N-(4-tert-bütilbenzil)benzotiyazolium bromür (1b) hazırlandı. Selenoüre N,S-heterosiklik bazlı türevler (2a ve 2b), tuzların KOBu<sup>t</sup> bazı varlığında selenyum ile reaksiyonu sonucu elde edildi. Sentezlenen bileşiklerin karakterizasyonu için <sup>1</sup>H NMR, <sup>13</sup>C NMR, FT-IR ve elementel analiz kullanıldı. Ayrıca, selenyum katılma ürünlerinin molekül ağırlıkları LC-MS/MS analizi ile doğrulandı.

#### Abstract

Two novel benzothiazolium salts, N-(4-methoxybenzyl)-benzothiazolium chloride (1a) and N-(4-tert-butylbenzyl)-benzothiazolium bromide (1b), were prepared by using benzothiazole and related benzyl halides. Selenourea N,S-heterocyclic-based derivatives (2a and 2b) were obtained by the reaction of the salts with selenium in the presence of KOBu<sup>t</sup> base. <sup>1</sup>H NMR, <sup>13</sup>C NMR, FT-IR and elemental analysis were used for characterization of the synthesized compounds. Also, molecular weights of the selenium adducts were validated by LC-MS/MS analysis.

**Anahtar Kelimeler:** Benzotiyazolium; N,S-Heterosiklik Karben; Selenyum; Selenoüre

**Keywords:** Benzothiazolium; N,S-Heterocyclic Carbene; Selenium; Selenourea

#### 1. Introduction

Strongly  $\sigma$ -basic N-heterocyclic carbenes (NHCs) and their stable organometallic compounds are of great importance in organometallic chemistry. NHCs have numerous applications, especially in homogeneous catalysis. While studies of NHCs mainly concern N,N-heterocyclic carbenes (Zhao et al. 2021, Huang et al. 2021, Demir Atli 2021), studies of N,S-heterocyclic carbenes (NSHCs) are less common. In an earlier study on NSHCs, vitamin B1 was used as a catalyst for the condensation reactions of benzoin (Breslow 1958). The first NSHC, a thiazol-2-ylidene compound, was isolated in 1997 (Arduengo et al. 1997). The use of electron-rich olefins can be mentioned as the first applied method when listing the synthesis methods of NSHC complexes. The first nucleophilic compounds were synthesized by this method (Cardin et al. 1973). Another one is transmetallation of lithiated thiazoles (Fraser et al. 1974). Nowadays, direct or indirect metallation methods have been used to prepare many transition metal complexes (Huynh et al. 2006, Han et al. 2009, Yen et al. 2009, Purohit et al. 2016). NSHC ligands have become an alternative in catalysis due to the presence of more

electropositive and softer sulphur compared to nitrogen. For this reason, the use of metal complexes as effective catalysts has been the subject of much research. The reactions catalysed by NSHC complexes are as follows: Multicomponent Click reaction catalyzed by copper(I) complexes (Purohit et al. 2014), Ullmann coupling catalyzed by nickel(II) complexes (Ding et al. 2009), hydroalkoxylation of allenes catalyzed by gold(I) complexes (Dince et al. 2017), transfer hydrogenation of carbonyls catalyzed by ruthenium(II) complexes (Ding and Hor 2010, Oruc et al. 2016, Karakaş et al. 2022), Heck and Suzuki coupling catalyzed by palladium(II) complexes (Calo et al. 2000, Yen et al. 2006, Yen et al. 2007, Yen et al. 2008). In addition, gold(I) NSHC compounds derived from peptides with excellent cytotoxicity have been prepared (Gutiérrez et al. 2014).

Selenium is one of the most important trace elements in human biology and health. The main source of supply is food. Selenium, which is toxic in high amounts, is essential for the body. It is essential for the cellular metabolism of seleno-proteins such as thioredoxin reductase, glutathione peroxidase, etc., which are involved in maintaining antioxidant functions (Sun et al.

2017, Spengler et al. 2019, Papp et al. 2007). Selenium adducts are potent anti-carcinogens (Banerjee and Koketsu 2017, Rafique et al. 2015, Liang et al. 2016, Cierpiat et al. 2016).

In this work, two new benzothiazolium salts and their selenium adducts were synthesized and characterized.

## 2. Experimental

### 2.1. General Remarks

Benzothiazole was used as freshly distilled. THF was distilled over sodium/benzophenone. Other solvents and chemicals were purchased commercially and used without any pre-treatment. Elemental analysis was carried out on a CHNS-932 (LECO) elemental analyser. A BRUKER ULTRASHIELD PLUS BIOSPIN GmbH Nuclear Magnetic Resonance Spectrometer, Agilent LC-MS/MS device and Perkin Elmer Spectrophotometer were used for NMR, LC-MS/MS and FT-IR measurements. The measurements of the melting point (mp) were carried out with an Electrothermal 9100 instrument.

### 2.2. Synthesis of N-(4-methoxybenzyl)benzothiazolium chloride (1a)

The mixture of benzothiazole (0.48 g, 3.58 mmol) and 4-methoxybenzyl chloride (0.5 mL, 3.58 mmol) was stirred at 80 °C for 24h. The solidified mixture was dissolved in a small amount of methanol and the solids were precipitated with the addition of Et<sub>2</sub>O. After filtration, the product was abundantly washed with Et<sub>2</sub>O. Yield: 1.03 g, 98%. mp: 134-135 °C. FT-IR (KBr)  $\nu_{\text{CN}}$ : 1423 cm<sup>-1</sup>. <sup>1</sup>H NMR (400 MHz, dms<sub>o</sub>-d<sub>6</sub>):  $\delta$  = 3.68 (s, 3H, OCH<sub>3</sub>), 6.09 (s, 2H, NCH<sub>2</sub>), 6.92 (d, *J* = 8.8 Hz, 2H, Ar-H), 7.52 (d, *J* = 8.8 Hz, 2H, Ar-H), 7.80 (dt, *J* = 27.5, 7.1 Hz, 2H, Ar-H), 8.37 (d, *J* = 8.4 Hz, 1H, Ar-H), 8.53 (d, *J* = 8.0 Hz, 1H, Ar-H), 10.99 (s, 1H, NCHS) ppm. <sup>13</sup>C NMR (100 MHz, dms<sub>o</sub>-d<sub>6</sub>):  $\delta$  = 55.19, 55.64 (NCH<sub>2</sub>, OCH<sub>3</sub>), 114.85, 117.96, 125.06, 125.94, 128.82, 129.94, 130.36, 132.35, 140.41, 160.13 (Ar-C), 165.45 (NCS) ppm. Anal. Calcd. for C<sub>15</sub>H<sub>14</sub>NSOCl (%): C, 65.33; H, 5.13; N, 5.08; S, 11.61. Found: C, 65.21; H, 5.10; N, 5.01; S, 11.53.

### 2.3. Synthesis of N-(4-tert-butylbenzyl)benzothiazolium bromide (1b)

Benzothiazole (0.32 g, 2.39 mmol) ve 4-tert-butylbenzyl bromide (0.46 mL, 2.39 mmol) in DMF (2 mL) were stirred at 80 °C for 24h. After cooling to room temperature, Et<sub>2</sub>O was added to the orange solution for precipitation. Filtration and washing with Et<sub>2</sub>O processes were performed. Yield: 0.75 g, 87%. mp: 192-193 °C. FT-IR (KBr)  $\nu_{\text{CN}}$ : 1430 cm<sup>-1</sup>. <sup>1</sup>H NMR (400 MHz, dms<sub>o</sub>-d<sub>6</sub>):  $\delta$  = 1.21 (s,

9H, Bu<sup>t</sup>), 6.09 (s, 2H, NCH<sub>2</sub>), 7.38 – 7.48 (m, 4H, Ar-H), 7.78 – 7.90 (m, 2H, Ar-H), 8.35 (d, *J* = 9.1 Hz, 1H, Ar-H), 8.53 (d, *J* = 8.0 Hz, 1H, Ar-H), 10.79 (s, 1H, NCHS) ppm. <sup>13</sup>C NMR (100 MHz, dms<sub>o</sub>-d<sub>6</sub>):  $\delta$  = 31.40, 34.84 (Bu<sup>t</sup>), 55.30 (NCH<sub>2</sub>), 117.90, 125.93, 126.35, 128.67, 128.93, 130.10, 130.35, 132.34, 140.50, 152.83 (Ar-C), 165.91 (NCS) ppm. Anal. Calcd. for C<sub>18</sub>H<sub>20</sub>NSBr (%): C, 59.67; H, 5.58; N, 3.87; S, 8.83. Found: C, 58.89; H, 5.83; N, 3.75; S, 8.84.

### 2.4. General procedure for the synthesis of NSHC-based selenoureas

The mixture of benzothiazolium salt (1 mmol), KOBu<sup>t</sup> (2 mmol) and Se (2.5 mmol) was stirred in dry THF (10 mL) for 24h at room temperature under argon atmosphere. The post-reaction processes were carried out in air. THF was removed. CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was added and the suspension was filtrated through celite. After removing of CH<sub>2</sub>Cl<sub>2</sub>, the residue was purified by column chromatography on silica gel.

#### 2.4.1. N-(4-methoxybenzyl)benzothiazole-2-selenone (2a)

For the column chromatography, CH<sub>2</sub>Cl<sub>2</sub> was used as eluent. Then, the obtained product was recrystallized CH<sub>2</sub>Cl<sub>2</sub> / n-hexane (1/4) at room temperature. Yield: 40%. mp: 157-158 °C. FT-IR (KBr)  $\nu_{\text{CN}}$ : 1366 cm<sup>-1</sup>,  $\nu_{\text{C-Se}}$ : 963 cm<sup>-1</sup>. <sup>1</sup>H NMR (400 MHz, dms<sub>o</sub>-d<sub>6</sub>):  $\delta$  = 3.71 (s, 3H, OCH<sub>3</sub>), 5.81 (s, 2H, NCH<sub>2</sub>), 6.88 – 6.93 (m, 2H, Ar-H), 7.33 – 7.38 (m, 2H, Ar-H), 7.39 – 7.47 (m, 2H, Ar-H), 7.57 (dd, *J* = 7.5, 1.1 Hz, 1H, Ar-H), 7.86 (dd, *J* = 7.7, 1.2 Hz, 1H, Ar-H) ppm. <sup>13</sup>C NMR (100 MHz, dms<sub>o</sub>-d<sub>6</sub>):  $\delta$  = 50.21, 55.09 (NCH<sub>2</sub>, OCH<sub>3</sub>), 114.15, 114.35, 121.94, 125.38, 126.39, 127.43, 128.72, 129.55, 142.01, 158.87 (Ar-C), 184.80 (C=Se) ppm. Anal. Calcd. for C<sub>15</sub>H<sub>13</sub>NOS<sub>2</sub> (%): C, 53.86; H, 3.93; N, 4.19; S, 9.57. Found: C, 53.99; H, 3.78; N, 4.16; S, 9.57. LC-MS/MS (CH<sub>2</sub>Cl<sub>2</sub>): [M + H + Na]<sup>+</sup> at m/z 358.30, [4-methoxybenzyl]<sup>+</sup> at 121.24.

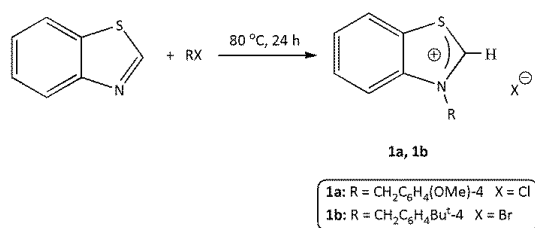
#### 2.4.2. N-(4-tert-butylbenzyl)benzothiazole-2-selenone (2b)

For the column chromatography, CH<sub>2</sub>Cl<sub>2</sub> / n-hexane (4/1) mixture was used as eluent. Yield: 36%. mp: 203-204 °C. FT-IR (KBr)  $\nu_{\text{CN}}$ : 1368 cm<sup>-1</sup>,  $\nu_{\text{C-Se}}$ : 950 cm<sup>-1</sup>. <sup>1</sup>H NMR (400 MHz, dms<sub>o</sub>-d<sub>6</sub>):  $\delta$  = 1.23 (s, 9H, Bu<sup>t</sup>), 5.84 (s, 2H, NCH<sub>2</sub>), 7.28 (d, *J* = 8.5 Hz, 2H, Ar-H), 7.34 – 7.38 (m, 2H, Ar-H), 7.39 – 7.47 (m, 2H, Ar-H), 7.55 (d, *J* = 7.0 Hz, 1H, Ar-H), 7.88 (dd, *J* = 7.4, 1.7 Hz, 1H, Ar-H) ppm. <sup>13</sup>C NMR (100 MHz, dms<sub>o</sub>-d<sub>6</sub>):  $\delta$  = 31.02, 34.23 (Bu<sup>t</sup>), 50.39 (NCH<sub>2</sub>), 114.34, 121.97, 125.40, 125.51, 126.88, 127.52, 129.53, 131.47, 142.10, 150.24 (Ar-C), 184.96 (C=Se) ppm. Anal. Calcd. for C<sub>18</sub>H<sub>19</sub>NSS<sub>2</sub> (%): C, 59.99; H, 5.32; N, 3.89; S,

8.88. Found: C, 59.70; H, 5.09; N, 3.88; S, 8.80. LC-MS/MS ( $\text{CH}_2\text{Cl}_2$ ):  $[\text{M} + \text{H} + \text{Na}]^+$  at  $m/z$  384.30, [4-tert-butylbenzyl] $^+$  at 147.30.

### 3. Results and Discussion

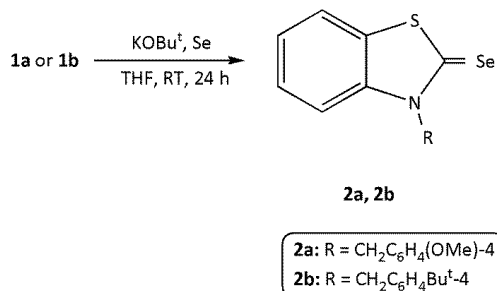
The synthetic routes are presented in Scheme 1 and Scheme 2. Benzothiazolium salts **1a** and **1b** were prepared according to the literature (Oruc et al. 2016, Karakaş et al. 2022, Ding et al. 2009, Gutierrez et al. 2014, Yen et al. 2007). The reaction of benzothiazole with 4-tert-butylbenzyl bromide in DMF gave **1b** in 87% yield. When the same reaction was repeated with 4-methoxybenzyl chloride, the product **1a** was obtained in very low yield. Therefore, this reaction was carried out in solvent-free condition rather than in DMF and it was observed that the product was formed in very high yield. The salts formed in high yields are air stable. They are soluble in methanol, water, dimethylformamide and insoluble diethyl ether and n-hexane.



**Scheme 1.** Synthesis of **1a** and **1b**

The selenium adducts of the salts (**2a** and **2b**) were obtained by the reactions with excess selenium.  $\text{KOBU}^t$  was used as base for deprotonation in these reactions. After column chromatography, pure products were obtained, albeit with slightly lower yields. They are soluble in most of the organic solvents.

Since the experimental elemental analysis results of all the compounds synthesized are in agreement with the theoretical ones, this is an indication that the target compounds have been obtained.



**Scheme 2.** Synthesis of **2a** and **2b**

NMR spectra are shown in Figures 1 to 4. Analysis of the  $^1\text{H}$  NMR spectra of **1a** and **1b** reveals that the acidic C2 proton is present at 10.99 and 10.79 ppm, respectively. These downfield singlet signals are characteristic of such NSHC precursors (Karakas et al. 2022). In the  $^1\text{H}$  NMR spectrum of **1a**, the signals at 3.68 and 6.09 ppm were observed for the methoxy group and for the methylene protons bonded to the nitrogen atom. In the case of compound **1b**, the signals associated with the tert-butyl group and the  $\text{NCH}_2$  protons are located at 1.21 ppm and 5.84 ppm, respectively. The downfield signals seen in the  $^1\text{H}$  NMR spectra of the salts are absent in the  $^1\text{H}$  NMR spectra of the selenoureas (Kamal et al. 2019). This is an indication that the salts are undergoing deprotonation and the formation of a carbene-selenium bond. In the  $^{13}\text{C}$  NMR spectra of the salts and of the selenourea compounds, the carbon atoms number 2 resonate at about 166 ppm and at about 185 ppm, respectively. The formation of selenium compounds is indicated by this difference in values. These  $^{13}\text{C}$  NMR chemical shift values are consistent with the literature (Demir Atli 2020, Oruc et al. 2016, Časar et al. 2002, Yaqoob et al. 2020).

IR peaks concerning the stretching vibrations of  $-\text{CN}$  groups for the salts are present at 1423 and 1430  $\text{cm}^{-1}$ . These frequency values decrease to 1366 and 1368  $\text{cm}^{-1}$  for the selenoureas. Considering the literature data, the peaks at 963  $\text{cm}^{-1}$  and 950  $\text{cm}^{-1}$  in the spectra of **2a** and **2b** are thought to be the stretching vibration frequencies related to the  $\text{C}=\text{Se}$  bond (Devillanova et al. 1980). The proposed structures for the selenourea compounds **2a** and **2b** were also supported by LC-MS/MS spectra (Figures 5 and 6). In both spectra, signals were observed that were related to the  $[\text{M}+\text{H}+\text{Na}]^+$  and to the R group attached to the benzothiazole ring.

### 4. Conclusions

In this study, N-(4-methoxybenzyl)benzothiazolium chloride (**1a**) and N-(4-tert-butylbenzyl)benzothiazolium bromide (**1b**) salts were prepared. These compounds were used for the synthesis of the selenium adducts, N-(4-methoxybenzyl)benzothiazole-2-selenone (**2a**) and N-(4-tert-butylbenzyl)benzothiazole-2-selenone (**2b**). The structures of the prepared compounds were corroborated by NMR spectroscopy and elemental analysis. Also, formation of the selenium adducts were supported by LC-MS/MS analysis. Synthesis of new complexes derived from these adducts and their catalytic and biological studies proceed.

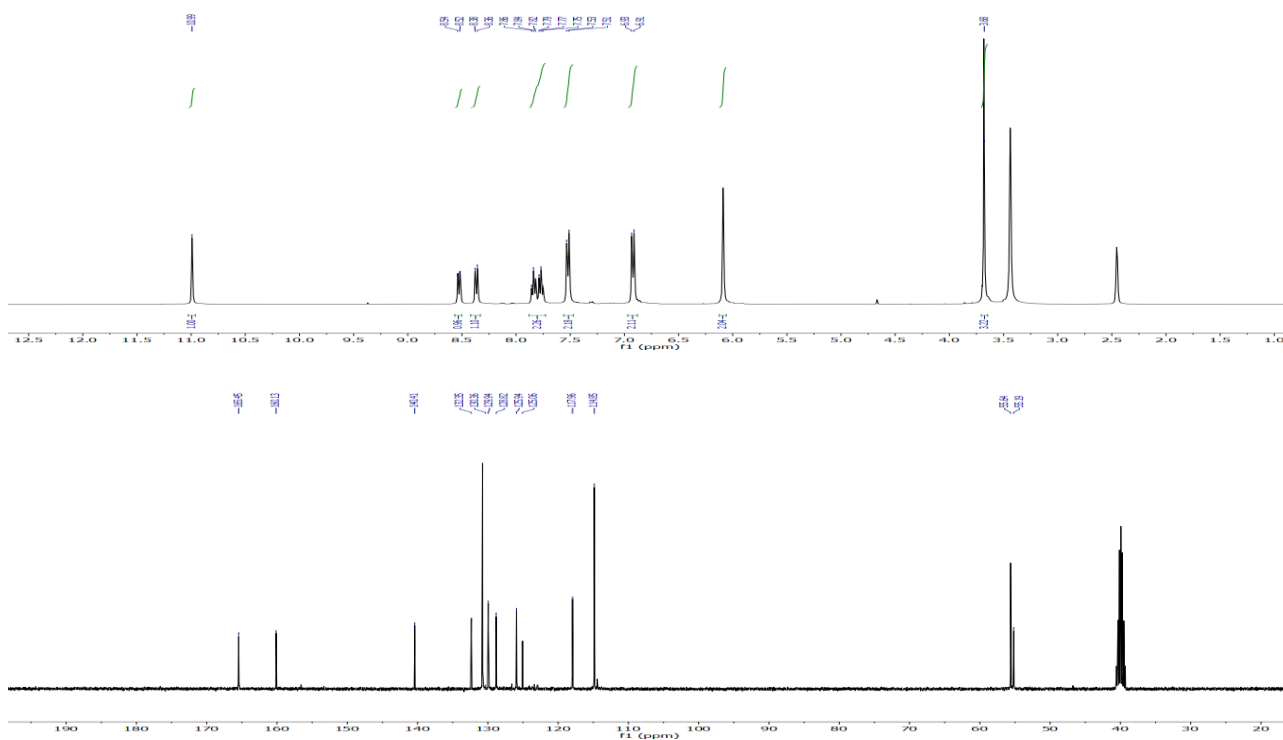


Figure 1. <sup>1</sup>H and <sup>13</sup>C NMR spectra of 1a

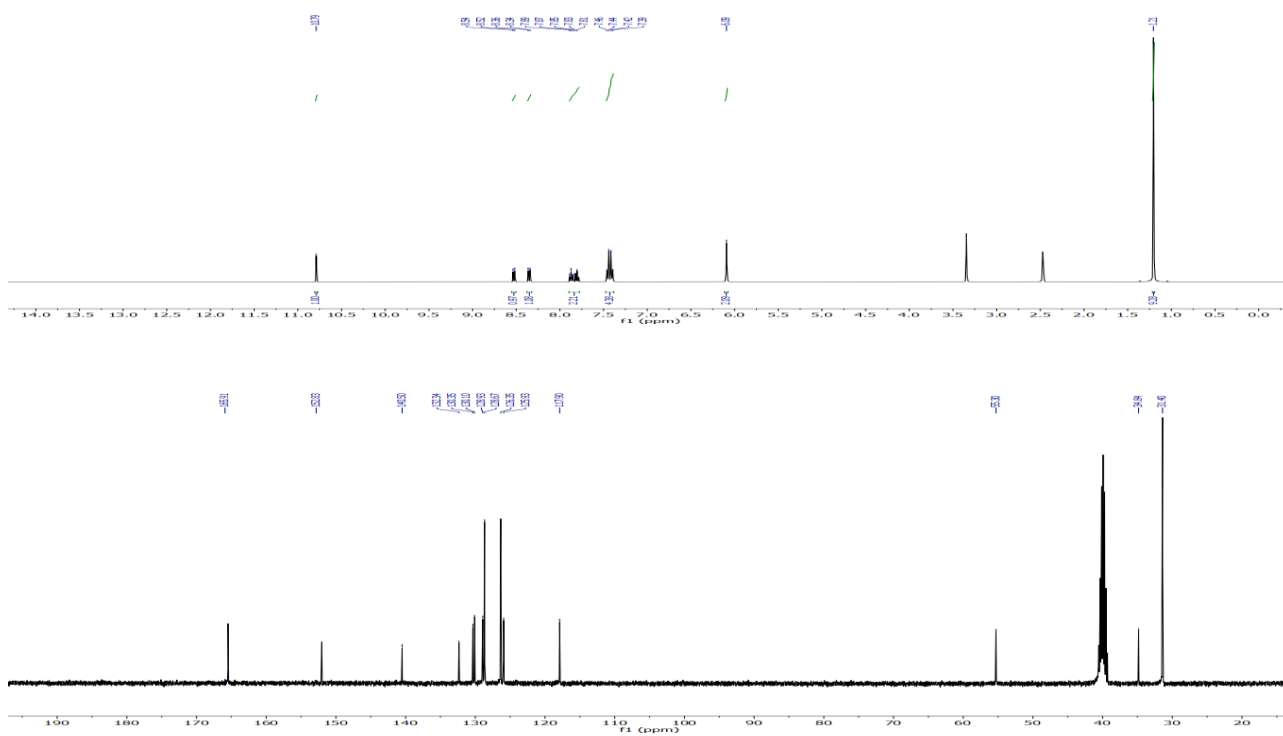


Figure 2. <sup>1</sup>H and <sup>13</sup>C NMR spectra of 1b

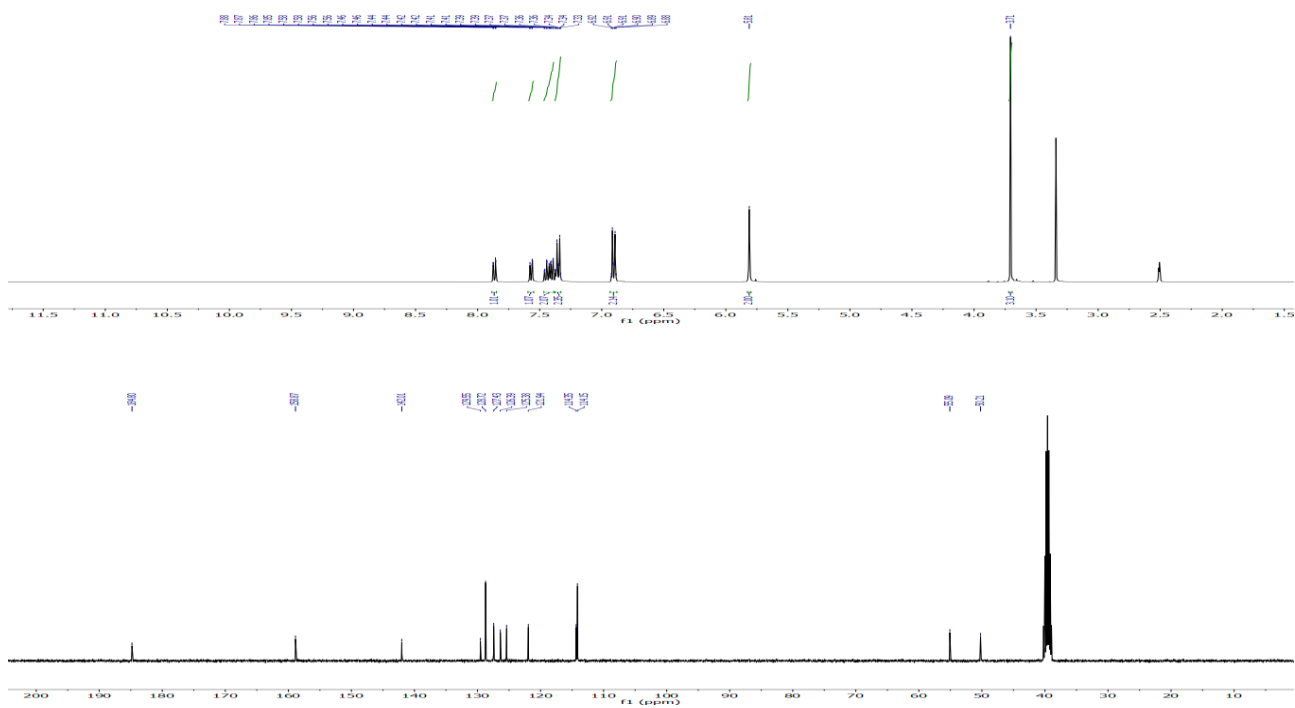


Figure 3.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of 2a

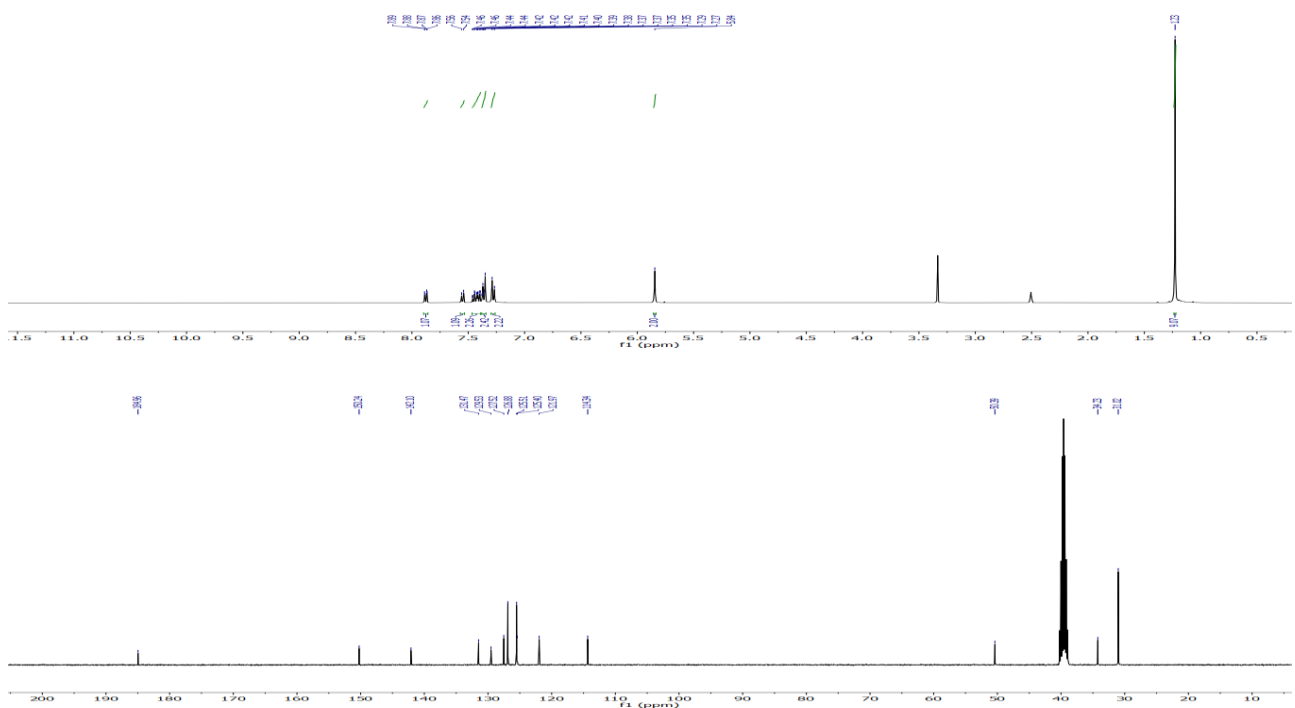


Figure 4.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of 2b

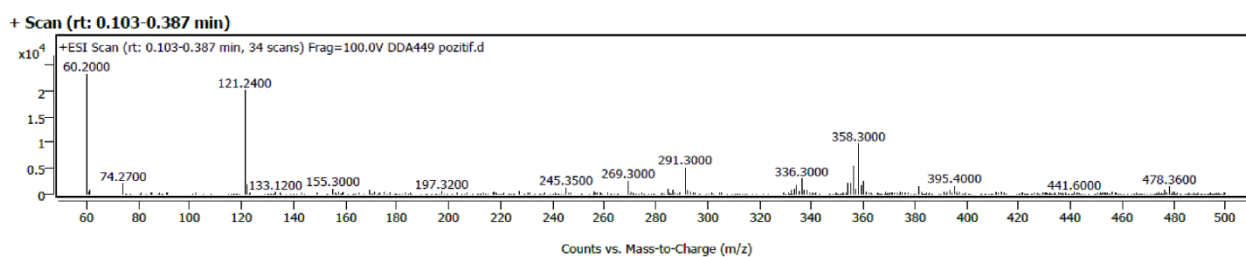
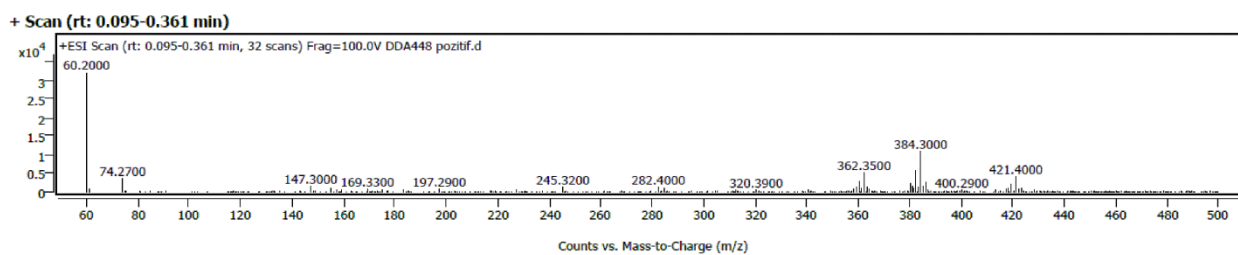


Figure 5. LC-MS/MS spectrum of 2a

Figure 6. LC-MS/MS spectrum of **2b****Declaration of Ethical Standards**

All ethical rules regarding authorship, citation, reporting and publishing original research were followed.

**Declaration of Competing Interest**

The author has no conflicts of interest to declare regarding the content of this article.

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