

An Investigation of Structural, Optical and Pyroelectrical Properties of LiTaO₃

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Highlights

- Structural, optical, and pyroelectric properties of bulk LiTaO3 crystals is investigated
- The effect of sample size and thickness on pyroelectrical properties of LiTaO3 is reported,
- The thinner LiTaO₃ generated larger pyroelectric current density.
- The bandgap of LiTaO3 is determined as being direct and falls into deep UV region.

Article Info

Abstract

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Keywords

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1. INTRODUCTION

In this study, structural, optical, and pyroelectric properties of Z-cut single crystal LiTaO3 bulk materials with thicknesses of 27 µm and 250 µm are analyzed. XRD results show characteristic diffraction peaks of Z-cut LiTaO3 at (012), (006), and (202), along with a Ta2O5 peak due to Lideficiency. The strong (006) peak confirms a high c-orientation, indicating pyroelectric potential. Raman spectroscopy confirms agreement with known vibration modes of bulk LiTaO3. Band gap values for the 27 µm and 250 µm samples are determined as 4.44 eV and 4.65 eV, respectively, with both showing a direct band gap. Temperature changes from 30 °C to 180 °C were applied at rates of 50 °C, 100 °C, and 150 °C. As temperatures rose, negative pyroelectric currents were observed; with cooling, currents shifted positive. The 250 µm thick, 24 mm² LiTaO₃ wafer produced about 4 nA at 50 °C rate, rising to 12-13 nA at 150 °C. With larger surface areas yielding higher currents, measurements on three wafers at a 50 °C change showed the highest-area sample producing ~7.5 nA, while the smallest yielded ~0.5 nA. The mean pyroelectric current density was higher in 27 μ m (180 μ A/m²) than in 250 μ m (125 μ A/m²), and the pyroelectric coefficient increased with decreasing thickness, measured at 33.43 µC/m²·K (27 µm) and 23.22 µC/m²·K (250 µm). These results suggest the potential of LiTaO3 crystals in IR detectors and self-powered deep UV detector applications due to their wide band gap.

Due to rapid advancements in infrared (IR) sensing technology, pyroelectric materials and detectors have garnered significant attention in recent years by virtue of their simple and cost-effective production processes and unique characteristics [1, 2]. Pyroelectric detectors have been extensively studied for a wide range of potential applications, including environmental sensing, security communication, fire prediction, and military services. However, they are frequently used in practical applications such as wavelength selectivity, gas detection, motion sensing, and fire detection [3, 4]. Pyroelectric detectors are constructed from materials in which the orientation of their dipoles' changes with temperature. Pyroelectric materials also exhibit ferroelectric and piezoelectric properties. When such substances are placed between metal electrodes, they form a temperature-sensitive capacitor. If a temperature change occurs, the polarization of the ferroelectric material changes, leading to a time-varying surface charge, and thus the generation of current. In the absence of temperature variation, no current is generated.

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Important pyroelectric materials commonly cited are Lithium Tantalate (LiTaO₃), Lead Zirconate Titanate (PZT), Barium Titanate (BTO), and Triglycine Sulfate (TGS). Among these materials, LiTaO₃, as can also be seen in Table 1, exhibits a high Curie temperature and a low dielectric constant. It also has a high melting point (1650°C) and is insoluble in water [5]. Studies on thin films LiTaO₃ is scarce, as it is typically used in single crystal form and grown by Czochralski method in bulk. Its excellent piezoelectric, pyroelectric, and ferroelectric properties, including a large spontaneous polarization, moderate dielectric constant, low dielectric loss, and a high pyroelectric coefficient, make it a suitable material for pyroelectric IR detectors [1, 6–8]. Recently, potential of pyroelectric materials in self-powered photon detector applications has been also demonstrated [9, 10]. Therefore, being a wide bandgap material, LiTaO₃ can be also a candidate for self-powered detectors to be used in the UV region.

	Curie temperature (°C)	Pyroelectric coefficient (µC/m ² K)	Dielectric constant (ε ₀)	Reference
LiTaO ₃	620	180	47	[11–13]
BaTiO ₃	120	800	2350	[14]
PbZr _{0.52} Ti _{0.48} O ₃	350	400	500	[15]
TGS	49	250	3323	[16]

Table 1. Theoretical properties of typical bulk pyroelectric mate	erials
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Pyroelectric materials are non-centrosymmetric materials, falling within the category of piezoelectric materials. Materials with centrosymmetric possess a single polar axis, which is an axis parallel to the spontaneous polarization of the crystal without any external influence; this spontaneous polarization is found in non-centrosymmetric crystals. In pyroelectric materials, the change in net dipole moment of the material with temperature determines its pyroelectricity. The pyroelectric effect involves the movement of positive and negative charges towards opposite ends of the material when exposed to an electromagnetic wave, as illustrated in Figure 1. Continual exposure to the electromagnetic wave leads to the build-up of static electricity. This property can be harnessed to generate electrical current from various devices [17, 18].

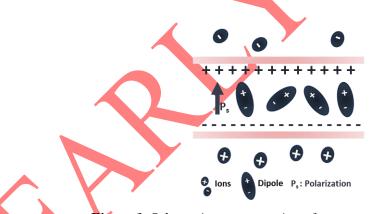


Figure 1. Schematic representation of spontaneous polarization

Pyroelectric sensing requires specific material characteristics. The surface of the material should be smooth, and its dielectric loss factor should be low. The dielectric constant is frequency-dependent, which can lead to noise issues; therefore, it should also be low. In pyroelectric materials, as well as in all dielectric materials, it is essential for the material to remain below its Curie temperature to maintain its properties. As the material approaches the Curie temperature, its pyroelectricity decreases and becomes zero at the Curie point. Beyond this temperature, the material loses its characteristics [18].

Figure 2 illustrates the principle behind the generation of current in pyroelectric materials. As shown, pyroelectric current varies with temperature changes. When the temperature increases, the pyroelectric current also increases, and in the case of constant temperature, the pyroelectric current decreases until it reaches zero. This is because maintaining a constant temperature will result in no change, and pyroelectric detectors operate based on temperature fluctuations. Similarly, as the temperature decreases, the direction

of the pyroelectric current changes, increasing in the negative direction. Therefore, changes in pyroelectric current occur with each temperature variation and change in direction [17].

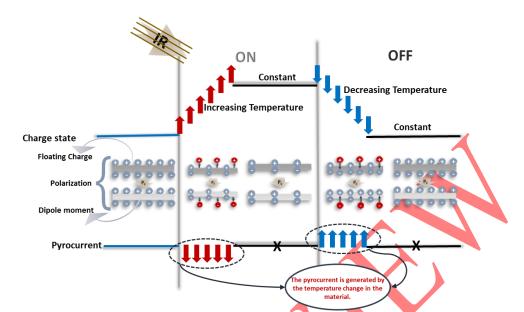


Figure 2. Schematic representation of occurrence of pyrocurrent and polarization in the pyroelectric material

The dependence of the pyroelectric coefficient and spontaneous polarization on temperature change is given in Equation (1)

$$\Delta \mathbf{P}_{\mathbf{s}} = \pi dT$$

The pyroelectric coefficient of the material (p) is a π vector that is equal to the rate of change of spontaneous polarization (P_s) with temperature change (dT) As shown in Figure 2. *Schematic representation of occurrence of pyrocurrent and polarization in the pyroelectric material* and Equation (1), the pyroelectric effect manifests itself as the flow of current in an external circuit connected to the material when the temperature of the material changes. Imagine a thin piece of pyroelectric material with bottom and top electrodes, such that the material has a π component perpendicular to the electrode surfaces and associated with an area A_s. While not all, but many detectors are known to be constructed in a way that the direction of π is perpendicular to the electrode plane, that is, $p = |\pi|$. Therefore, it will be assumed in the subsequent sections that the situation is as such, and the term "pyroelectric coefficient" will be attributed to "p" [17]. The electric current occurs due to the alteration in charge over time, where electric charges flow within a closed circuit, and is defined as [19, 20]

$$i_{p} = pA_{s} \frac{dT}{dt}.$$
(2)

Here, p is the pyroelectric coefficient, A_s is the electrode overlapping area of the pyroelectric material, and $\frac{dT}{dt}$ is the temperature gradient. It has been reported that pyroelectric properties of pyroelectric materials tend to increase with decreasing sample thickness, therefore a great effort is spent to make the bulk material thinner [21, 22]. So far, most of the studies of bulk LiTaO₃ is focused on its pyroelectric properties [11–13, 17, 18], without considering its optical properties.

In this study, structural, optical and pyroelectrical characterization of Z-cut LiTaO₃ bulk samples grown by Czochralski method and mechanically thinned to 27 μ m and 250 μ m are presented. LiTaO₃ wafers with thicknesses of 27 μ m and 250 μ m were cut into different sizes and coated with Pt/Ti bottom and Au/Ti top electrodes to investigate the influence of sample size on pyroelectric properties of LiTaO₃.

2. MATERIAL METHOD

There are different methods for pyroelectric current measurements of LiTaO₃ single crystal wafers in the literature. The most common of these methods is to perform temperature-dependent current measurements by placing the pyroelectric material-based chip on a plate and heating or cooling it [7]. On the other hand, pyroelectric materials can be heated and cooled in a temperature-controlled chamber and the currents they create at different temperature change rates can be examined. With this technique [19], materials are heated or cooled to the desired temperature with a constant rate of change, while pyroelectric current is measured in response to a constant temperature change over time by operating with a sawtooth signal. In this study, measurements were made by heating and cooling LiTaO₃ samples between 30 °C and 180 °C with a temperature change rate of 50 °C, 100 °C and 150 °C using Radiant Technologies Precision LC II ferroelectric parameter analyzer equipped with Linkam T96-S cryostat. It is shown that as the thickness of LiTaO₃ decreases, pyroelectric properties improve [23], therefore two samples with thicknesses of 27 µm and 250 µm are investigated in this study. Since the pyroelectric current depends on the area of the pyroelectric material in Equation (2), 250 μ m thick LiTaO₃ wafer is cut to have 24 mm² and 4 mm² areas and 27 µm thick LiTaO₃ wafer is cut to obtain a 50 mm² area, The whole surfaces of LiTaO₃ samples were coated with Au/Cr as the front electrode and Pt/Ti as the back electrode with NANOVAK NVEB-600 ebeam evaporator. Before investigation of pyroelectric characteristic, of LiTaO₃, structural and optical properties of LiTaO₃ were examined using Raman scattering, X-Ray-Diffraction (XRD) and transmittance measurements. Raman scattering analyses was performed using a free-space custom set-up. A 532 nm laser was used excitation source, and the signal was collected using a thermoelectric cooled CCD [24]. XRD measurements were carried out using Rigaku Smartlab X-ray diffraction (XRD) spectrometer at an angular 2θ from 20° to 50° in the increment of 0.02°/minute. Finally, the optical properties of LiTaO₃ single crystal wafers were investigated using the Ocean Insight DH 2000 mini spectrometer equipped with a Deuterium & Tungsten Halogen light source.

3. RESULTS

3.1. Structural Characterization

In Figure 3, glancing-angle XRD spectra of LiTaO₃ samples with thicknesses of 250 μ m and 27 μ m are shown. The characteristic peaks of bulk stoichiometric Z-cut LiTaO₃ are clearly observed in the spectra, which are indexed according to the XRD card number JCPDS #01-020-0631 [25], which reveals that growth of LiTaO₃ is (006) plane- preferential, however observation of diffraction peaks from other planes and Ta₂O₅ shows that the crystal quality of the samples is not good enough. The (006) diffraction peak is responsible for the pyroelectricity of LiTaO₅ and observation of the strong (006) peak proves that bulk samples are highly c-oriented and can exhibit pyroelectric characteristics. The peak positions remain approximately the same for both samples. The XRD results reveal the characteristic diffraction peaks of Ta₂O₅ from (012), (006) and (202) crystal planes together with a characteristic diffraction peak of Ta₂O₅ from (012) plane. The observation of Ta₂O₅- related diffraction peak can be attributed to the Li-deficient growth of the crystal. The peak at 42,44° corresponding to the (202) plane of LiTaO₃ is only observed for 250 μ m thick samples.

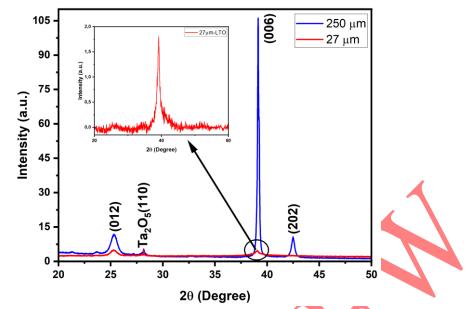


Figure 3. XRD graphs of 250 μ m and 27 μ m thick LiTaO₃ single crystal wafers

3.2. Analysis of Raman Scattering

The Raman scattering spectra of single crystal LiTaO₃ wafers with thicknesses of 250 μ m and 27 μ m are presented in Figure 4, covering the range of 230 to 1160 cm⁻¹. The characteristic Raman active modes of LiTaO₃ are observed in the spectra [26].The peak positions are approximately the same for each sample. For the 27 μ m thickness LiTaO₃ sample, the peaks are observed at 256 cm⁻¹ (3E TO), 314 cm⁻¹ (4E TO), 355 cm⁻¹ (2A₁ LO), 381 cm⁻¹ (5E LO), 405 cm⁻¹ (3A1 LO), 462 cm⁻¹ (7E TO), 591 cm⁻¹ (8E TO), 661 cm⁻¹ (9E TO), and 864 cm⁻¹ (4A₁ LO) [27, 28]. Here, TO represents transverse optic polar oscillations, while LO represents longitudinal optic polar oscillations. The peak at 749 cm⁻¹ is attributed to the fundamental doubly degenerate E(TO) and E(LO) symmetry-type vibrations perpendicular to the polar axis of LiTaO₃ [29]. The peaks of the 250 μ m thick LiTaO₃ sample are observed at the same wavenumber values with higher peak intensities, as expected. However, the LO peak at 381 cm⁻¹ is not observed.

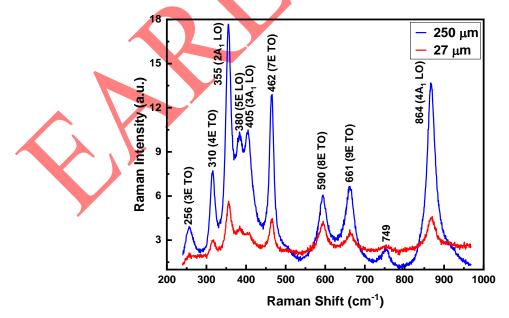


Figure 4. Raman graphs of 250 µm and 27 µm thick LiTaO₃ single crystal wafers

3.3. Optical Characterization

Transmittance measurements were conducted in the wavelength range of 250-1100 nm. As can be seen from Figure 4, the transmittance (\geq 300 nm) of 27 µm thick LiTaO₃ is over 70% and is 3 times higher than the 250 µm thick sample. The bandgap value was determined from Tauc's equation. If the intensity of the incoming photons onto a film of thickness *t* is I₀ and the wavelength of incoming photons is λ , the intensity of the transmitted light is defined as,

(3)

(4)

$$I_t = I_0 e^{-\alpha t}.$$

where α is absorption coefficient and the band gap is calculated using Tauc's equation given as [30],

$$\alpha h v = A (h v - E_g)^{\gamma}$$
.

Using the obtained transmittance results, the band gap values of LiTaO₃ wafer were determined by Tauc plot analysis. The wavelenght dependence of absorption coefficient as well as Tauc's plot is presented in Figure 5. The best fit to the experimental data is obtained for $\gamma = 2$, which indicates LiTaO₃ is a direct band gap material. The band gap values of 250 µm and 27 µm thick LiTaO₃ single crystal wafers calculated with the Tauc plot are 4.69 eV and 4.44 eV, respectively. There is no reported bandgap value of bulk LiTaO₃, however the obtained band gap values are higher than the reported ones for the LiTaO₃ thin films [31].

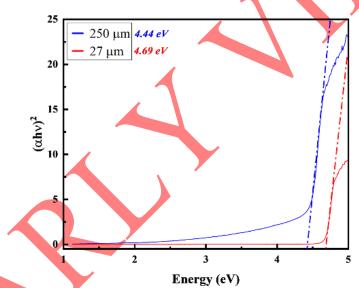


Figure 5. Tauc's plot of transmittance measurements and the extracted band gap of 250 μ m and 27 μ m thickness LiTaO₃ single crystals

3.4. Pyroelectric Characterization

In Figure 6, generated pyroelectric current in LiTaO₃ samples at different temperature change rates are shown. While a pyroelectric current of approximately 3 nA was generated in a 250 μ m thick 24 mm² LiTaO₃ sample with a temperature change rate of 50 °C, the generated pyroelectric current at a temperature change rate of 150 °C increases to approximately 11 nA. As the temperature change rate increases to three-fold from 50 °C to 150 °C (Figure 6b), the pyroelectric current also increases linearly at the same rate as expected in Equation (2).

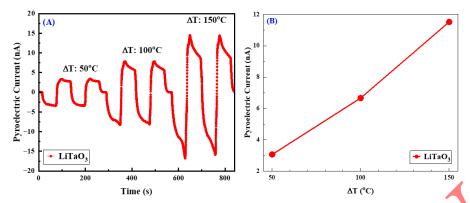


Figure 6. (a) Pyroelectric current graphs of LiTaO₃ samples with 250 μ m thickness and 24 mm² area at different temperature change rates, (b) mean pyroelectric current at different temperature changes

Since the generated pyroelectric current depends on the area and the thickness of pyroelectric material, pyroelectric characterization was carried out on three different LiTaO₃ samples, 250 μ m thick with 24 mm² and 4 mm² area and 27 μ m thick with 50 mm² area, at a temperature change rate of 50 °C. Figure 7 shows the pyroelectric current measurement results of LiTaO₃ samples of different sizes. The highest pyroelectric current value is observed for 27 μ m thick sample.

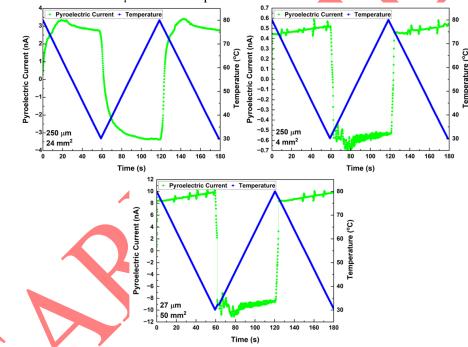


Figure 7. Pyroelectric current graphs of LiTaO₃ samples with different sizes at a 50 °C temperature change rate

Based on the data presented in Figure 7, Pyroelectric current density was calculated for the samples with a thickness of 250 μ m (area 4 mm²) and 27 μ m (area 50 mm²) as seen in Figure 8a and 8b, respectively. A slightly higher pyroelectric current density is obtained for 27 μ m thick sample. Using pyroelectric current density in Equation (2), the mean values of pyroelectric coefficient of the 27 μ m and 250 μ m thick samples are determined as 33.43 $\mu C/m^2 \cdot K$ and 23.22 $\mu C/m^2 \cdot K$, respectively. During the heating-cooling cycle, pyroelectric current tends to increase not constant therefore the mean value of the pyroelectric current is taken to calculate pyroelectric coefficient. The pyroelectric coefficient tends to be higher for the thinner bulk LiTaO₃, which agrees with the literature [32]. The obtained values of pyroelectric coefficient of the samples is found to be about one-sixth lower than the reported theoretical value of bulk LiTaO₃ given in Table 1, which can be related to insufficient single crystal quality as discussed in XRD results. Our results reveals that the bulk 27 μ m thick LiTaO₃ sample is more promising to be used in thermal detector applications.

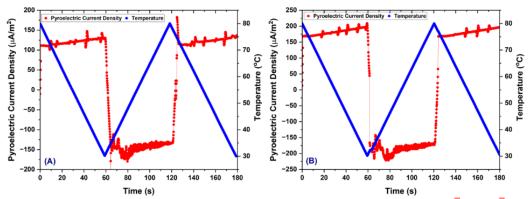


Figure 8. Pyroelectric current density graphs of (a) 250 μm and (b) 27 μm thick LiTaO₃ samples at a 50 °C temperature change rate

4. CONCLUSION

The structural, optical, and pyroelectrical properties of bulk single crystal Z-cut LiTaO₃ samples with different thicknesses (250 μ m and 27 μ m) have been investigated. XRD results has confirmed that bulk LiTaO₃ is c-oriented with a strong (006) peak. The observation of other XRD peaks is an indication of por crsyal quality. Raman scattering results have exhibited the characteristic of bulk LiTaO₃ crystal. As for thinner LiTaO₃ sample, intensity of both Raman scattering and XRD peaks has been found to be lower. The band gap of the samples obtained as 4.69eV (27 μ m) and 4.44 eV (270 μ m) indicates LiTaO₃ has potential to be used in self-powered deep-UV detectors thanks to its wide band gap. The mean value of pyroelectric current density of approximately 180 μ A/m² measured from 27 μ m thick LiTaO₃ sample can be promising to use the sample as an active material of a pyroelectric detector.

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CONFLICTS OF INTEREST

No conflict of interest was declared by the authors.

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