

# **Experimental Study on Gasification of Banknote Waste: Effects of Torrefaction Pre-Treatment and Co-Gasification on Producer Gas Composition**

Hakan KAVUSTU<sup>[1](https://orcid.org/0000-0002-3492-2754)</sup> Emir Hüseyin SIMSEK<sup>1[\\*](https://orcid.org/0000-0001-7945-8222)</sup>

*<sup>1</sup> Ankara University, Department of Chemical Engineering, Ankara, Türkiye*



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## **1. INTRODUCTION**

Energy production is one of the most important economic factors that provides development and increases efficiency. In the context of energy sustainability, the importance of renewable energy sources is increasing due to (i) rapidly growing demand for energy, (ii) the fact that most of this demand is met by fossil fuels, and (iii) increasing concerns about global warming and environmental pollution caused by the use of fossil fuels (Jahromi et al., 2021; Zhang et al., 2024). The total energy supply increased by 15% between 2000 and 2021, and oil (29.5%), coal (27.2%), natural gas (23.6%), biofuels and waste (9.5%) were the sources with the largest share in the total energy supply in 2021 (IEA, 2021). Modern bioenergy sources, which provide more than half of the renewable energy supply, can exist in solid, liquid and gaseous forms. According to the 802

International Energy Agency (IEA) Energy Policy Scenario, the modern solid bioenergy supply will reach 57 Exajoules (EJ) in 2050 (IEA, 2023).

Biomass, which is a carbon-neutral and renewable energy source, has an important role in sustainable energy policies (George et al., 2019). Biomass can be classified according to the different raw material types such as agricultural, forest, municipal, industrial wastes and wood (Hussain et al., 2023). Biomass differs in terms of their properties such as molecular morphology and thermal stability, and the type and structure of biomass directly affect the product quality and yield to be obtained as a result of thermochemical conversion. Lignocellulosic biomass is basically composed of cellulose, hemicellulose and lignin (Chan et al., 2019).

In principle, banknotes are printed on cotton and polymer (polypropylene) based materials (Luján-Ornelas et al., 2018). The vast majority of banknotes produced worldwide are cotton-based banknotes with a high cotton content (DLR, 2022). Cotton-based banknote production generally includes (i) the production of banknote paper containing cotton, security thread, foil and (ii) the printing on banknote paper with inks. The Euro banknote consists of an average 0.815 g of cotton, 0.082 g of ink, 0.010 g of security thread, and 0.049 g of foil (Hanegraaf et al., 2019).

Banknotes in circulation are routinely sorted by central banks. At the end of banknote processing, worn or damaged banknotes which are not fit for circulation in terms of quality standards are considered as end-oflife banknotes (BoI, 2023). End-of-life cotton banknotes can be classified as lignocellulosic biomass due to their high cotton content (Yousef et al., 2020; Kavuştu & Şimşek, 2023). Banknote demand is basically determined by (i) the increase in the amount of banknotes in circulation, (ii) the issuance of new series of banknotes, and (iii) the renewal of end-of-life banknotes (DLR, 2023). It is estimated that there are approximately 500 billion banknotes in circulation worldwide and more than 150 billion banknotes are produced each year (G+D, 2024). The amount of banknote production worldwide tends to increase by 2-3% per year (K&B, 2022). These production data mean that (i) the potential amount of end-of-life banknote waste is approximately 500,000 tonnes assuming an average banknote weight of 1 g, (ii) nearly 150,000 tonnes of potential waste will be generated annually, and (iii) this amount will continue to increase by 2-3% each year. End-of-life banknotes are mostly disposed of by landfill and incineration (RDK, 2022). The European Central Bank (ECB) has increased the use of incineration and recycling with the ban of the disposal of the waste banknotes by landfill and combustion in order to reduce their environmental impact (ECB, 2023)

In gasification, biomass is converted into a gas mixture [\(synthesis gas/](https://www.sciencedirect.com/topics/earth-and-planetary-sciences/synthesis-gas)syngas/producer gas), which can be used for the production of liquid fuels, chemicals and energy, with the gasification agent at the gasifier temperature of 600-1400 $^{\circ}$ C (Ng et al., 2013). Air, oxygen, steam and CO<sub>2</sub> are generally used as gasification agents. The high nitrogen content in the gasification agent reduces the calorific value of the syngas, while the use of pure oxygen is costly. In addition to providing a higher gasification rate compared to the use of  $CO<sub>2</sub>$ with gasification by feeding steam, a product with high calorific value and low tar content can be obtained (Xie et al., 2021). The gasification process basically consists of the following stages: (i) drying, where the moisture in the biomass is removed, (ii) pyrolysis, where volatile components are vaporized, (iii) oxidation (combustion) of combustible materials with oxygen  $(O_2)$ , and (iv) reduction (char gasification) occurring as a result of the contact of coal formed by pyrolysis with carbon dioxide  $(CO<sub>2</sub>)$  and water (H<sub>2</sub>O) (Baruah & Baruah, 2014).

In gasification, syngas is produced, which contains combustible gases such as hydrogen  $(H<sub>2</sub>)$ , carbon monoxide (CO), methane (CH<sub>4</sub>) and non-combustible gases such as nitrogen (N<sub>2</sub>) and carbon dioxide (CO<sub>2</sub>). Operating conditions, the types of feedstock and gasifier affect the composition of the syngas (Das et al., 2020). Transition metal catalysts contribute to the reduction of tar formation and the increase of the quality of the syngas (Meng et al., 2023). Banknotes can contain many of the elements that can be used as catalysts, such as zeolites that can be synthesized at different Si/Al ratios (Yousef et al., 2020; Kavuştu & Şimşek, 2023). Bubbling fluidized-bed gasifier has advantages such as (i) ease of operation, (ii) good mixing and high-quality gas-solid contact, (iii) increased mass, heat transfer and chemical conversion efficiency, and (iv) ability to operate under different operating conditions (Parrillo et al., 2021). Kinetic studies for the devolatilisation of lignocellulosic biomass contribute to understanding biomass decomposition behaviour. Non-isothermal methods such as Flynn-Wall-Ozawa (FWO) and Kissinger-Akahira-Sunose (KAS) are widely used in estimating kinetic parameters and extent of conversion (Mohammed et al., 2022).

With the torrefaction pre-treatment, which is pyrolysis that occurs at relatively low temperatures, the fuel quality of biomass and the syngas yield can be increased in addition to facilitating handling, storage and transportation of biomass (Cabuk et al., 2020). In the study of Sadaka (2013), as a result of torrefaction of cotton gin waste, the moisture and volatile solid content decreased, while the bulk density and calorific value increased, and generally more carbon conversion and gasification efficiency were obtained compared to raw waste by gasification. With co-gasification, where two or more feedstocks are mixed and gasified, the mass transfer, gasification rate, reactivity and efficiency can be increased and the amount of pollutant gases can be reduced (Hanchate et al., 2021).

Polymer banknotes are printed on substrates made of biaxially-oriented polypropylene film (BOPP) (BoC, 2011). Using only plastic as feedstock in gasification has some disadvantages such as (i) having a very high calorific value, making gasifier temperature control difficult, (ii) its sticky structure making operation difficult, and (iii) increasing tar formation. In order to overcome these disadvantages, the product quality and yield can be increased by co-gasification of plastic and biomass in a bubbling fluidized-bed gasifier (Parrillo et al., 2023). One of the most suitable plastic types for co-gasification with biomass is polypropylene (PP) (Block et al., 2018). Co-gasification of biomass and plastic has advantages such as accelerating plastic degradation with the catalytic effect of the biomass char, and reducing char formation due to the increase in syngas yield (Ajorloo et al., 2024).

There are studies examining the end-of-life cotton-rich banknotes by (i) biochemical processes such as methane production by "anaerobic digestion" (Mancilla-Leytón et al., 2024), glucose production by "fermentation" (Sheikh et al., 2013; Yousef et al., 2021) and (ii) thermochemical processes such as bio-oil, bio-gas and char formation by "pyrolysis" (Yousef et al., 2020), producer gas production by "gasification" (Kavuştu & Şimşek, 2023).

In the present study, Differential Scanning Calorimetry (DSC) and Fourier Transform Infrared Spectroscopy (FTIR) analysis were performed within the scope of characterization tests of the end-of-life cotton-rich samples. Thermal decomposition kinetics were examined by Flynn-Wall-Ozawa (FWO) and Kissinger-Akahira-Sunose (KAS) methods. The effects of (i) torrefaction of cotton-based banknote sample and (ii) cogasification of cotton and polymer-based banknote samples on the producer gas content as maximum mole fractions were investigated within the scope of experiments on gasification in a fluidized-bed reactor. The main novelty of this study is to investigate the effects of torrefaction pre-treatment and co-gasification on producer gas composition with gasification of end-of-life banknotes.

# **2. MATERIAL AND METHOD**

## **2.1. Materials**

In this study, the following samples were used: (i) 5 Turkish Lira (TL) banknote samples, which were put into circulation in 2009 and were printed on a substrate made of cotton fibers and (ii) 1 Romanian Leu (RON) banknote samples, which entered into circulation in 2005 and were printed on a substrate made of polypropylene resin. The physical properties of banknote samples are given in Table 1.

<b>Sample</b>	<b>Physical Properties</b>						
	Thickness (µm)	Grammage $(g/m^2)$	<b>Particle Density</b> $(g/cm^3)$				
5 TL (Kavuştu & Şimşek, 2023)	116	98.5	0.85				
1 RON	96	90.8	0.95				

*Table 1. Physical properties of 5 TL and 1 RON samples*

#### **2.2. Preparation of Samples**

The samples were sieved from a mesh after turning into small particles. 5 TL and 1 RON samples with a mean particle size of 890 μm were prepared. Additionally, the 5 TL sample with a mean size of 890 μm was torrefied with inert gas  $(N_2: 99.9\%)$  at 250°C for 10 min in the gasifier. The prepared samples are demonstrated in Figure 1.



*Figure 1. Prepared banknote samples: a) raw and b) torrefied cotton-based; c) polymer-based*

#### **2.3. Characterization Techniques**

Cotton-based raw sample shown in Figure 1a was used in characterization tests. Differential Scanning Calorimetry (DSC) analysis was performed using a TA Instruments SDT 650 Simultaneous DSC/TGA, at six different heating rates between 5-30°C/min and a temperature range of 25-1000°C in nitrogen. Fourier Transformed Infrared Spectroscopy (FTIR) analysis was performed using a Perkin Elmer Pyris STA 6000 Simultaneous TGA/FTIR at six different heating rates between 5-30°C/min and a temperature range of 30- 950°C in nitrogen.

## **2.4. Activation Energy Determination**

The thermal analysis data used to investigate the thermal decomposition kinetics were obtained from our previous study (Kavustu & Simsek, 2023) in which thermogravimetric analysis was performed on 5TL banknote sample at six different heating rates in the range of 5-30°C/min. The formula used to calculate the activation energy with FWO and KAS methods is given in Table 2 (Yousef et al., 2020).

<b>Method</b>	<b>Formula</b>	X		<b>Slope</b>
Flynn-Wall-Ozawa (FWO)	$\left(\frac{\text{lnA} \text{Ea}}{\text{Rf}(v)}\right) - 5.335 - \frac{1.051 \text{Ea}}{\text{RT}}$ $\ln \beta =$	1/T	$ln(\beta)$	$-1.0516$ Ea/R
Kissinger-Akahira-Sunose (KAS)	$=\ln\left(\frac{AR}{\text{Eaf(y)}}\right)-\frac{\text{-}\text{Ea}}{RT}$ $\ln\left(\frac{\beta}{T^2}\right)$		$ln(\beta/T^2)$	$-Ea/R$

*Table 2. FWO and KAS methods*

### **2.5. Gasification Experiments**

An experimental setup consisting of a gas preparation unit, a fluidized-bed reactor and a gas analyzer was used (Figure 2). In the gas preparation unit consisting of PID controlled flow meters (MKS), the mixture of steam and air was prepared at the desired rates by adjusting the bubble column temperature. The mixture of steam-air was fed to the fluidized-bed reactor at the minimum fluidization velocity in order to prevent the elutriation of sample particles in the reactor. 3 g of sample was fed to the furnace-reactor combined system consisting of a gas distributor, a ceramic furnace and an atmospheric pressure cylindrical fluidized-bed reactor. Experiments were performed for 600 s of operation time. After the undesired components such as sulfur oxides  $(SO_x)$  and nitrogen oxides  $(NO_x)$  in the gas formed in the reactor were trapped in the conditioning unit, the CO,  $CO_2$ , CH<sub>4</sub>, H<sub>2</sub> and  $O_2$  mole fraction (%  $v_i$ ) values were determined with the gas analyzer (GASBOARD-3100P).

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*Figure 2. The experimental set-up*

## **2.6. Gasification Reactions**

Gasification reactions that may affect the producer gas content obtained from gasification experiments were investigated. Water-gas, water-gas shift, steam reforming and hydrogasification reactions occur in the presence of steam (Kumar et al., 2009). With the Boudouard reaction, which includes the reaction of C with CO2, the syngas with a high CO content can be produced. The need for a gas purification process is reduced by converting of CO2, which has a greenhouse effect, into CO (Huang et al., 2021). Main gasification reactions are shown in Table 3 (Block et al., 2018).

<b>Gasification Reactions</b>	$\Delta H^0$ <sub>298K</sub> (kJ/mol)	
<b>Complete Oxidation</b>	$C+0.5O_2 \Leftrightarrow CO$	$-111$
<b>Partial Oxidation</b>	$C + 0.5O_2 \Leftrightarrow CO_2$	$-283$
<b>Water-gas Reaction</b>	$C + H2O \Leftrightarrow CO + H2$	$+131$
<b>Water-gas Shift Reaction</b>	$CO + H2O \Leftrightarrow CO2 + H2$	$-42$
<b>Steam Reforming Reaction</b>	$CH4 + H2O \Leftrightarrow CO + 3H2$	$+206$
<b>Hydrogasification Reaction</b>	$C+2H_2 \Leftrightarrow CH_4$	$-75$
$C+CO_2 \Leftrightarrow 2CO$ <b>Boudouard Reaction</b>		$+173$

*Table 3. Main gasification reactions*

## **3. RESULTS AND DISCUSSION**

The cotton-based banknote sample has a similar structure to cotton gin waste and cotton fibers in terms of characterization test results (Table 4).

<b>Sample</b>	<b>Proximate</b> Analysis $(wt. \% )$		<b>Calorific</b> <b>Value</b> (MJ/kg)	<b>Elemental Analysis</b> $(wt. \% )$					<b>Lignocellulosic Analysis</b> $(wt. \% )$		
	Volatile Solid Matter	Ash	Higher Calorific Value	$\mathcal{C}$	H	N	S	$\Omega$	Cellulose	Hemicellulose	Lignin
<b>Cotton-based Banknote (5 TL)</b> (Kavuştu & Şimşek, 2023)	81.12	4.83	17.42	38.95	6.70	0.65	$\overline{\phantom{a}}$	53.70	71.82	2.83	17.64
<b>Cotton Gin Waste</b> (Raveendran et al., 1995)	88.00	5.40	17.48	42.70	6.00	0.10	$\overline{\phantom{a}}$	49.50	77.80	16.00	
<b>Cotton Fiber</b> (Corradini et al., 2009)									$88.3 \pm 0.3$	$8.0 \pm 0.3$	$4.8 \pm 0.5$

*Table 4. Comparison of characterization analysis of cotton and cotton-based banknote sample*

The higher calorific value of the cotton-based banknote sample is higher than that of oil palm empty fruit bunch (17.02 MJ/kg), beef cattle manure (15.93 MJ/kg), rice straw (16.28 MJ/kg) and rice hulls (16.14 MJ/kg) (Widjaya et al., 2018). The effect of lignin content of biomass on energy content is greater than cellulose and hemicellulose (Elbersen et al., 2017). The lignin content in the cotton-based banknote sample is higher than that in the wheat straw (17 %), rice straw (12 %) and poultry litter (11 %) (Ren et al., 2019).

## **3.1. DSC Analysis Results**

The DSC curves obtained at six different heating rates in the range of 5-30°C/min are shown in Figure 3. Endothermic peaks observed in the range of 93.51-179.91°C at different heating rates indicated physical changes and the removal of moisture from the fibers. Exothermic peaks observed in the range of 357.04- 398.11°C represented radical and char formation reactions of small molecules such as CO, CO<sup>2</sup> and acetaldehyde, which were converted into intermediate products with the increase in temperature (Smith et al., 2024). Afterwards, the endothermic values obtained indicated the ongoing decomposition, which was more prominent at the heating rate of  $10^{\circ}$ C/min. In the study of Smith et al. (2024), where DSC analysis was performed on cotton, an endothermic peak was observed in the range of 120-150°C and an exothermic peak at 372°C, similar to the cotton-based banknote sample.



*Figure 3. DSC curves of cotton-based banknote sample*

## **3.2. FTIR Analysis Results**

The FTIR spectrum obtained at six different heating rates in the range of 5-30°C/min are shown in Figure 4. The bands from 3500-3700 cm<sup>-1</sup> represented the O–H or N–H stretching of carbohydrates, proteins and adsorbed water. The peaks in the 2800-3200 cm<sup>-1</sup> presented the C-H stretching in cellulose, hemicellulose and hydrocarbons. The peaks in the  $1300-1800$  cm<sup>-1</sup> indicated the C=O stretching modes of carbonyl groups originating from esters, carboxylic acids, ketones and aldehydes. The C-O stretching in the bands of 1000- 1200 cm<sup>-1</sup> indicated the presence of ethers, alcohols and phenols (Portella et al., 2016; He et al., 2022; Pang et al., 2024). With FTIR analysis of cotton-based banknote sample, similar peaks were obtained with the cotton organs and tissues in the study of He et al. (2022) and cotton fibers in the study of Portella et al. (2016).



*Figure 4. FTIR spectrum of cotton-based banknote sample at heating rates (*°*C/min) of a) 5, b) 10, c) 15, d) 20, e) 25, and f) 30*

## **3.3. Activation Energy Results**

The plots obtained for different extent of conversions using the KAS and FWO methods are shown in Figure 5, and the calculated activation energies are given in Table 5. The regression lines and calculated activation energies obtained from the KAS and FWO methods were compatible with each other. The activation energies of the cotton-based banknote sample were calculated in the range of 134-171 kJ/mol, and these values continued to increase until the x=0.6 conversion and then started to decrease. In the study of Yousef et al. (2020), the activation energies of the end-of-life Egyptian Pound (EGP) cotton-rich banknote were calculated in the range of 157-276 kJ/mol by FWO and KAS methods, and these values continued to increase until the x=0.7 conversion and then started to decrease. Multi-step reactions occurred due to the ink, hemicellulose, lignin and other materials contained in the sample, making it difficult to examine thermal degradation (Yousef et al., 2020). The difference in activation energies of TL and EGP banknotes is due to the difference in their chemical composition and structure. In the study of Corradini et al. (2009), the average activation energy for cotton fibers was calculated as 150.4±3.6 kJ/mol by FWO method. This value was nearly the same as the value calculated with FWO as 154.25 kJ/mol for the cotton-based banknote sample.



*Figure 5. a) KAS and b) FWO plots of cotton-based banknote sample*

Ea (kJ/mol)	<b>Extent of Conversion (x)</b>									
	0.1	0.2	0.3	0.4	0.5	0.6	0.7	0.8	Mean	
Flynn-Wall-Ozawa (FWO)	136.638	152.946   162.841   167.563						167.689   171.419   161.734   113.168   154.250		
<b>Kissinger-Akahira-Sunose (KAS)</b>   134.011   150.767   160.970   165.793					165.800		169.612   159.269		107.308   151.691	

*Table 5. Activation energies for cotton-based banknote sample*

#### **3.4. Effect of Torrefaction Pre-Treatment on Experimental Mole Fractions**

Raw and torrefied (250°C for 10 min) cotton-based banknote samples were used as feedstocks. Gasification experiments were carried out at  $800^{\circ}$ C with an inlet H<sub>2</sub>O/O<sub>2</sub> ratio of 25. Mole fraction-time curves obtained with gasification are given in Figure 6a and 6b. Considering the maximum mole fractions achieved, with torrefaction pre-treatment,  $H_2$  mole fraction increased from 4.44% to 4.64%; CO mole fraction increased from 9.32% to 10.71% and CH<sub>4</sub> mole fraction increased from 3.47% to 3.65%, while  $CO<sub>2</sub>$  mole fraction decreased from 6.28% to 5.48%. The decrease in  $CO<sub>2</sub>$  mole fraction showed the effects of Boudouard reaction, the increase in CH<sup>4</sup> mole fraction showed the effects of hydrogasification reaction, and the increase in mole fractions of  $H_2$  and CO showed the effects of water-gas and steam reforming reactions. The increase

in CO mole fraction and the decrease in  $CO<sub>2</sub>$  mole fraction also showed that the effect of water-gas shift reaction was limited. The experimental results were consistent with the studies of Singh and Yadav (2021) and Sarker et al. (2022) in which the syngas yield increased as a result of gasification with torrefied biomass.

#### **3.5. Effect of Co-Gasification Pre-Treatment on Experimental Mole Fractions**

Samples prepared by mixing cotton and polymer-based banknote samples with the ratios of 50:50 and 75:25 (wt.%) were used as feedstocks. Gasification experiments were carried out at 800 $^{\circ}$ C with an inlet H<sub>2</sub>O/O<sub>2</sub> ratio of 25. Mole fraction-time curves obtained with gasification are given in Figure 6c and 6d. Considering the maximum mole fractions achieved, with the co-gasification of cotton and polymer-based samples at 75:25 ratio, H<sup>2</sup> mole fraction decreased from 4.44% to 3.71%; CO mole fraction decreased from 9.32% to 8.02% and  $CO<sub>2</sub>$  mole fraction decreased from 6.28% to 4.77%, while CH<sub>4</sub> mole fraction increased from 3.47% to 4.69%, compared to the raw cotton-based sample. With the co-gasification of cotton and polymerbased samples at 50:50 ratio compared to 75:25 ratio, H<sup>2</sup> mole fraction decreased from 3.71% to 3.68%; CO mole fraction decreased from 8.02% to  $6.68\%$  and  $CO<sub>2</sub>$  mole fraction from decreased 4.77% to 4.74%, while CH<sup>4</sup> mole fraction increased from 4.69% to 5.65%. Producer gas production was generally achieved efficiently and the decreasing maximum mole fractions indicated that the conversion efficiency of the gasification reactions and therefore the energy content of the gas decreased with the increasing ratio of the polymer-based sample in the feedstock. The increase in only the CH<sup>4</sup> mole fraction indicated that the hydrogasification reaction was effective and the effect of the steam reforming reaction was reduced. The conversion of feedstock primarily to tar with low char yield with the gasification of cellulose-rich biomass with plastic may have limited gas efficiency (Ajorloo et al., 2024). On the other hand, the difference in materials and inks contained in cotton and polymer-based samples may have also caused this result.



*Figure 6. Mole fraction-time curves for a) raw, b) torrefied cotton-based banknote samples; c) 50:50, d) 75:25 cotton:polymer-based banknote samples*

## **4. CONCLUSION**

The results of DSC and FTIR analysis performed on the cotton-rich banknote sample and the calculated mean activation energy by FWO method showed that the cotton-based banknote sample had characteristic structures similar to those of cotton and thus end-of-life cotton banknotes could be used as a biomass source. With the non-isothermal kinetic methods of Flynn-Wall-Ozawa (FWO) and Kissinger-Akahira-Sunose (KAS), the activation energies were calculated as 134-171 kJ/mol at different conversions and the activation energy started to decrease after the  $x=0.6$  conversion. This situation showed that the multi-step reactions occurring due to the complex structure of the sample had made it difficult to study the thermal decomposition.

In the gasification experiments performed at 800 $^{\circ}$ C with an inlet H<sub>2</sub>O/O<sub>2</sub> ratio of 25, the effects of torrefaction pre-treatment and co-gasification on the producer gas content were investigated with mole fraction- time curves. With the gasification torrefied (250°C for 10 min) cotton-based banknote sample, the  $H_2$ , CO, CH<sub>4</sub> mole fractions and product yield increased, while the CO<sub>2</sub> mole fraction decreased. With the increase in the polymer-based sample ratio in the feedstock, the gas product yield and  $H_2$ , CO, CO<sub>2</sub> mole fractions decreased, while the CH<sup>4</sup> mole fraction increased. In the gasification of torrefied cotton-based banknote sample, the water-gas, water-gas shift, steam reforming and hydrogasification reactions, which took place in the steam environment, were effective; while in the co-gasification of the cotton and polymerbased banknote samples, the hydrogasification reaction was effective.

In conclusion, the results of the study showed that (i) end-of-life cotton banknotes can be effectively used as a biomass source in thermochemical processes, (ii) [torrefaction](https://tureng.com/tr/turkce-ingilizce/torrefaction) can facilitate the storage and transportation of dispersed waste banknotes, and thus improved fuel properties can increase producer gas efficiency, (iii) producer gas can be efficiently obtained by co-gasification of end-of-life cotton and polymer-based banknotes considering the diversity of these wastes.

## **AUTHOR CONTRIBUTIONS**

Methodology, H.K. and E.H.Ş.; laboratory work, H.K.; manuscript-original draft, H.K.; manuscript-review and editing, H.K. and E.H.Ş. All authors have read and legally accepted the final version of the article published in the journal.

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#### **CONFLICT OF INTEREST**

The authors declare no conflict of interest.

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