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*Research Article*

# **A Water Absorption Chamber Design for Cable Capacitance Measurement Tests**

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*Abstract:* XLPO cables used in ships should be tested for water absorption. Test standards require the cables be submerged in a water absorption chamber and kept at 75 <sup>0</sup>C for 14 days. Once every 7 days, the cables within the chamber are tested by applying high voltage, and their capacitance, dissipation factor, stability factor, and relative Permittivity are measured. In this study, a water absorption chamber used in the experiments is designed using fiberglass material and controlled with an ENDA ET 4420 temperature controller. It is experimentally shown that the water absorption chamber performs well.

*Keywords: Power Cables, Water Absorption Test, Cable Tests, Thermal Design, Heater Design.*

## **Kablo Kapasitans Ölçüm Testleri için Su Emme Odası Tasarımı**

*Öz.* Gemilerde kullanılan XLPO kablolar su emme açısından test edilmelidir. Test standartları, kabloların su emme odasına batırılıp 75 <sup>0</sup>C'de 14 gün saklanmasını gerektirmektedir. Her 7 günde bir, oda içerisindeki kablolar yüksek gerilim uygulanarak test edilerek kapasite, kayıp faktörü, stabilite faktörü ve bağıl geçirgenlik ölçümleri yapılır. Bu çalışmada deneylerde kullanılan bir su emme odası fiberglas malzeme kullanılarak tasarlanmış ve bir ENDA ET 4420 sıcaklık kontrolörü kullanarak kontrol edilmiştir. Deneysel olarak su emme odasının iyi başarımla çalıştığı gösterilmiştir.

*Anahtar Kelimeler: Güç Kabloları, Su Emme Testi, Kablo Testleri, Termal Tasarım, Isıtıcı Tasarımı*

#### *.***1. Introduction**

Cables, which are an important part of electrical systems, should be tested according to standards [1-4]. Marine cables are an important cable type, and their market share is everincreasing [5, 6]. Marine cables also age due to water trees occurring within [7, 8]. Marine cables are tested according to IEEE 1580-2021, NEMA WC 53, and NEMA WC 57 to make sure that they operate without failure [9-11]. The Accelerated Water Absorption test is defined in the IEEE 1580:2021 standard [9]. The electrical and mechanical tests are performed on the insulation material of the cable after removing its sheath according to Table 13 given in IEEE 1580-2021 standard. All the values measured are used to calculate the dielectric constant, capacitance increase, and stability factor and the calculated cable parameters are compared to the standard values given in Table 13 of the IEEE 1580-2021 standard [10]. After the comparison, it can be decided whether the test is successful or not. The note at the bottom of Table 13 refers to Section 6.14 of the standard NEMA WC 57 for the test procedure [11]. Information about the cables to be tested is given in the NEMA WC 57 standard, and, therein, it is stated that the test can be done in two different methods which are the electrical and gravimetric methods. The electrical method is defined in the IEEE 1580-2021 standard for testing. NEMA WC 57 standard tests and calculations direct us to the NEMA

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**ENGINEERING AND**<br>APPLIED SCIENCES

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WC 53 standard (Section 2.6 of NEMA WC 53) [10]. The tests specified in the IEEE 1580-2021, Table 13 Accelerated Water Absorption section of the NEMA WC 53 standard are explained and how the calculations can be made is stated. A low-voltage XLPO cable is tested using an accelerated water absorption test [12].

A water chamber should be used during the testing of marine cables according to IEEE 1580-2021. Such a water chamber must be designed to keep the water temperature at 75  $^{\circ}$ C [9, 12]. A water heater design needs a good experience [12]. The electrical heaters can be controlled using several different techniques [13-15]. In [16-20], a PLC is used for this purpose. A PID controller is preferred in [18, 19, 21]. A microcontroller is used to design an intelligent bath water heater [22]. The temperature of a water heater must be sensed with an accurate temperature sensor [14, 23]. The temperature of electrical heaters can be monitored with a transmitter [24]. The effect of several parameters on a heater is studied to increase its efficiency [25]. Such a water chamber must be designed using good heat insulators and good controllers [26]. Choosing a bad material may result in a failed operation [27]. In this study, a water test absorption chamber is designed to be used in the tests. Fiberglass is used to make the chamber due to its low thermal conductivity [28]. The temperature controller shown to perform well in the polymerization process is used to control its water temperature [29, 30]. The heat transfer model of the chamber is also made. The performance of the chamber is tested experimentally.

This paper is ordered as follows. In the second section, the water absorption bath mechanical system is introduced. In the third section, the water absorption bath electrical system is briefly described. In the fourth section, the heat transfer model of the system is given. In the fifth section, the simulation results are given. In the sixth section, the experimental results are given. The study is finished with the conclusion section.

#### **2. Water Absorption Bath Mechanical System**

A sketch of the water absorption test chamber can be seen in Figure 2. It is made of a box, a lid, and an internal grid. All the parts of the box are fiberglass which is chosen due to its low thermal conductivity and thermal endurance [28]. It can stand temperatures more than  $100<sup>0</sup>C$ . The lid has also two holes for guiding the cable ends in and out. The grid is used to prevent direct contact of the heater resistance to the power cable and its damage. The box has two holes. The heater resistance is placed in one of them and the thermocouple in the other. The controller circuit is placed onto the front side of the box. A photograph of the designed device is given in Figure 2.







*Figure 2. A Photograph of the device made.*

#### **3. Water Absorption Bath Electrical Circuit**

The schematic of the Water Absorption Bath electric circuit is shown in Figure 3. It operates at 220 Volt and dissipates power up to 2000 Watt. The electrical system of the device consists of a water heater resistance, a solid-state relay ISISO ISS-100- 3, a 25 A W-automat fuse, a 0/1 rotary power switch with a signal LED, a Fe-Const J Tipi thermocouple, and an ENDA ET 4420 temperature controller [29]. A 220 V 2000 W water heater resistance is used to heat water within the chamber. The components of the circuit are shown in Figure 4.

### **4. On the Control of the System**

A PID controller is used to control the water absorption chamber as shown in Figure 5. The PID controller parameters used in the design are given in Table 1.



*Figure 5. PID temperature control of the water absorption chamber.*





The reference water temperature is set to  $75 \,^0\text{C}$  degrees with ENDA ET 4420. The controller uses a hysteresis controller. When the power switch is turned on, its green LED starts illuminating and indicates that the device is energized. ENDA ET 4420 temperature controller measures the temperature using a Fe-Const J type thermocouple with a 25 cm long measurement probe with a diameter of 8 mm placed within the chamber in contact with the water. If the measured temperature is under the upper-temperature reference, the ENDA ET 4420



*Figure 3. Electrical Circuit of the Device.*



*Figure 4. The electrical components: a) a 25 A W-automat fuse (Schneider IK60N C 25A), b) a 0/1 rotary power switch with a signal LED, c) ENDA ET 4420 temperature controller, d) a Fe-Const J Tipi Thermocouple with a 25 cm long measurement probe with a diameter of 8 mm, e) the solid-state relay (SSR) ISISO ISS-100-3, and f) a 220V-2000W 18 Ohm heater resistance.* 

The accuracy of a Fe-Const J Tipi thermocouple can vary depending on several factors, including the quality of the thermocouple, the measurement conditions, and the calibration. Generally, the accuracy of a Type J thermocouple is around  $\pm 2.2$ °C or  $\pm 0.75$ % of the reading, whichever is greater. Such accuracy is good enough for the test purposes.

turns on and turns off the solid-state relay (SSR) ISISO ISS-100-3, which connects the heater resistance to the utility. The heater resistance starts heating the water and the water temperature increases. If the measured temperature reaches the upper-temperature reference, the ENDA ET 4420 turns off the solid-state relay (SSR) ISISO ISS-100-3 and the heater resistance is disconnected from the utility. The water temperature starts decreasing. When it falls to the lowertemperature reference, the ENDA ET 4420 turns on the solidstate relay (SSR) ISISO ISS-100-3 and, therefore, the heater resistance again. This operation cycle continues, and the average water temperature is kept at the set temperature of 75  ${}^{0}C.$ 

The instantaneous temperature error fed to the input of the PID controller is calculated as:

$$
e(t) = T(t) - T_{ref} \tag{1}
$$

where  $T(t)$  is the instantaneous measured water temperature and  $T_{ref}$  is the reference temperature.

The controller output  $PID_{out}$  can be expressed as:

$$
PID_{out} = K_P e(t) + K_I \int_0^t e(t)dt + K_D \frac{de(t)}{dt}
$$
 (2)

Where  $K_P$ ,  $K_I$ , and  $K_D$  is the proportional, the integral, and the derivative constants of the PID controller, respectively.

The controller output  $PID_{out}$ , is equal to the duty ratio of the heater power:

$$
D = PID_{out} \tag{3}
$$

where  $D$  is the duty ratio.

When the heater resistance is fed by the utility, the power consumed by the heater resistance is given as

$$
P_{heater} = R_{AC} I_{rms}^2 \tag{4}
$$

or

$$
P_{heater} = V_{rms}^2 / R_{AC}(T) \tag{5}
$$

where  $I_{rms}$  is the rms current of the heater resistance,  $V_{rms}$  is the rms voltage of the utility, and  $R_{AC}(T)$  is the temperature dependent heater resistance.

The average power of the heater is given as

$$
p = DP = R_{AC}I_{rms}^2 = DV_{rms}^2/R_{AC}(T)
$$
 (6)

#### **5. Heat Transfer Model of the System**

In this section, a simplified thermal circuit model of the circuit is provided. The geometry of the chamber in the study consists of four vertical walls and two horizontal walls as shown in Figure 6. It is assumed that the heat transfer between the water chamber with a certain amount of water inside and a certain amount of air above the water surface and the external environment is realized by heat conduction in the solid material, while the heat transfer between the solid surfaces and the fluid interfaces is realized by the effect of natural convection mechanisms, and the radiation heat transfer is neglected since the temperature differences are not high in the system.

Equations 7-11 contain the steps of the method used in the heat convection coefficient calculations used in the model, whose value range is given in Table 2. Heat transfer relations in natural convection are based on experimental studies except for some simple cases. The simple empirical relation for the mean Nusselt number  $\overline{Nu}_L$  in natural convection is as follows:

$$
\overline{Nu}_L = \frac{h}{k} = C \cdot Ra_L^n \tag{7}
$$

where *h* is the heat convection coefficient  $(W/m<sup>2</sup>K)$ , *L* is the length of the geometry  $(m)$ ,  $k$  is the heat conductivity (W/m∙K), *C* is a constant coefficient, *Ra<sup>L</sup>* is the Rayleigh number which is the product of Grashoff  $(Gr_L)$  and Prandtl *(Pr)* numbers, and *n* is a constant exponent. They are given as:

$$
Ra_L = Gr_L \cdot Pr = \frac{g \cdot \beta \cdot (T_s - T_\infty) \cdot L^3}{\nu \cdot \alpha} \tag{8}
$$

$$
Gr_L \equiv \frac{g \cdot \beta \cdot (T_s - T_\infty) \cdot L^3}{v^2} \tag{9}
$$

where *g* is the acceleration of gravity  $(m/s^2)$ ,  $\beta$  is the coefficient of volumetric expansion (1/<sup>o</sup>K) and can be expressed as *β=*1*/T* for ideal gases,  $T_s$  is the surface temperature ( $\degree$ C),  $T_\infty$  is the fluid temperature far enough from the surface  $({}^{\circ}C)$ , *v* is the kinematic viscosity of the fluid  $(m^2/s)$  and  $\alpha$  is the thermal diffusivity of the fluid  $(m<sup>2</sup>/s)$ .



*Figure 6.* Schematic view of the geometry of the container.

The Nusselt number for the vertical walls can be expressed by the following relation:

$$
\overline{Nu}_L = \left\{ 0.825 + \frac{0.387 \cdot Ra_L^{1/6}}{\left[1 + (0.492/Pr)^{9/16}\right]^{8/27}} \right\}^2 \tag{10}
$$

For the horizontal top cover or horizontal bottom base, the appropriate one of the following relations is used to determine the Nusselt number:

$$
\overline{Nu}_L = 0.54 \cdot Ra_L^{1/4} \quad (10^4 \le Ra_L \le 10^7)
$$
  

$$
\overline{Nu}_L = 0.15 \cdot Ra_L^{1/3} \quad (10^7 \le Ra_L \le 10^{11})
$$
  

$$
\overline{Nu}_L = 0.27 \cdot Ra_L^{1/4} \quad (10^5 \le Ra_L \le 10^{10})
$$
  
(11)

After the  $\overline{Nu}_L$  number was calculated with the help of the relations given above, the heat convection coefficients (*h*) on the surfaces were calculated using Eq. (7). The range of heat

convection coefficient values calculated by this method is given in Table 2.

In this study, the thermal resistance model is used, and this concept is based on the analogy between electric current and heat dissipation. The relationship between electrical resistance and transmitted electric current is analogues to the relationship between thermal resistance and transmitted heat power. Resistance is defined as the ratio of a potential difference to current while thermal resistance is defined as the ratio of the temperature difference to the heat power transmitted. Equivalent thermal circuits can also be used for complex systems such as mixed walls. When the layers of such walls are composed of different materials, they can contain many series and parallel thermal resistances. The schematic of the thermal resistance circuit of the device in the steady-state for the geometry in the study is shown in Figure7.a. The thermal circuit consists of the side thermal resistances and the thermal capacitances of the fiberglass box, water and the still air, Combining the series and parallel thermal resistances, their equivalent circuit shown in Figure 7.b is obtained.

By adding a thermal capacitance to the equivalent thermal resistor in parallel to model thermal dynamics, the transient thermal circuit model shown in Figure 7.c is obtained. For the device, the side thermal resistance components of the box, the bottom thermal resistance of the box, and the top thermal resistance of the box, are, respectively, given as

$$
R_{con-inner} = \frac{1}{h_{inner} A_{side}}
$$
 (12)

$$
R_{THwall} = \frac{d_{wall}}{k_{fg} A_{side}} \tag{13}
$$

$$
R_{con-outer} = \frac{1}{h_{inner} A_{side}}\tag{14}
$$

and

$$
R_{THdown} = R_{THup} = \frac{d}{k_{fg}A_{down}}
$$
 (15)

where  $d_{wall}$  is the thickness of the glass fiber,  $h_{outer}$  is the wall-to-air heat convection constant,  $h_{inner}$  is the water-towall heat convection constant, and  $k_{fg}$  is the thermal conductivity of the glass fiber,  $A_{side}$  is the side (wall) area which is the vertical surface area  $(m^2)$  where heat transfer takes place, and  $A_{down}$  is the base or ceiling area.

Since they are four side thermal resistances, and one cover thermal resistance, and one base thermal resistance, their equivalent thermal resistance of the device is given as

$$
R_{TH} = (R_{con}/4) / (R_{THdown}/2)
$$
 (16)

$$
R_{TH} = \frac{R_{con} R_{THdown}}{8\left(\frac{R_{con}}{4} + \frac{R_{THdown}}{2}\right)}\tag{17}
$$

where  $R_{con} = R_{con-inner} + R_{THwall} + R_{con-outer}$ ,

The heat power transferred to the environment in the steadystate can be expressed using the thermal resistance method as follows:

$$
p = \frac{\Delta T}{R_{TH}} = \frac{(T_w - T_a)}{R_{TH}}\tag{18}
$$

where  $T_w$  is the water temperature and assumed to be the measured water temperature  $T(t)$ .



*Figure 7. a) The steady-state thermal circuit of the chamber, b) the steady-state equivalent thermal circuit of the chamber, and c) the transient thermal circuit model of the chamber.*

Heat convection coefficient values were calculated for the outer, inner, top, and bottom surfaces of the container side walls. The equivalent thermal resistance is calculated as 0.0243 W/K with the parameters given in Table 2.





The thermal capacitance of water is found as

 $C = mc$  (18)

where m is the mass of water and *c* is the specific heat of water.

The mass of water within the chamber is approximately 100 kg, The thermal capacitance of the water (*C*) or the device ignoring the mass of other sections is calculated to be 418740 J/°C. In the transient, the heat transfer model of the device can be modeled as

$$
p = DV_{rms}^2/R_{AC}(T) = mc\frac{d(T_W - T_A)}{dt} + \frac{T_W - T_A}{R_{eq}} \tag{19}
$$

where  $T_A$  the ambient temperature,  $T_W$  is the water temperature,  $m$  is the water mass, and  $c$  is the specific heat of water.

In the steady-state, since the temperature of the device is not varying with time and the water temperature is  $75 \text{ °C}$ , the power consumption of the device can be modeled as

$$
p = DR_{AC}(T_W)I_{rms}^2 = DV_{rms}^2/R_{AC}(T_W) = \frac{T_W - T_A}{R_{eq}}\tag{20}
$$

where  $I_{rms}$  is the rms current of the heater resistance and  $R_{AC}(T)$  is the temperature dependent heater resistance.

#### **6. Simulation Results**

The system simulations are performed in Matlab program. The system is simulated under heating conditions when the full electrical power 2000 W is applied to the device, and the water temperature during the heating process is shown in Figure 8.a. The simulated water temperature is almost a ramp in this case. The system is simulated under natural cooling conditions, and

the water temperature during the cooling process is shown in Figure 8.b. The simulated temperature falls exponentially when simulated for 6 hours.



*Figure 8. Simulated temperature vs. time during a) the natural cooling of the device and b) the heating process.* 

#### **7. Experimental Results**

The water absorption test on a marine power cable is made to ascertain the quantity of water absorbed by the insulation when it comes into contact with water. To be able to use the device in such a test, the device's performance must be examined. In this study, experiments are done to see the performance of the water absorption chamber designed. Before starting the test, the device is filled with tap water. The water chamber is placed in the test room for protection purposes during high voltage experiments. As a result, there is neither forced nor natural cooling of the device during the test. In the test, a 4,5 meter long 0.6 /1 kV Volt rated marine cable after removing the outer TPO sheath and armoring is placed within the chamber as shown in Figure 9 and the device is turned on. Its temperature is read from the controller display shown in Figure 10 and recorded with respect to time. Two thermal experiments were conducted on the device. The first experiment was performed during the heating of the device, and the other during its natural cooling. In the first experiment, the device was turned on at room temperature, and the water temperature was read every few minutes until it reached 75 °C. The heating duration or the setting time of the device can be assumed to be almost 3 hours. After this time, its resistor is turned on and off to compensate for the ambient power loss. The time and temperature data were recorded and plotted. In the second experiment, the water temperature was at 75 °C, and the device

was in the thermal steady-state when it was turned off. As the device cooled, the temperature and time data were also recorded. The graphs showing the temperature as a function of time from these experiments are shown in Figure 11.



*Figure 9. A photograph of the water absorption test from the laboratory* 



*Figure 10. The appearance of the device monitor during the experiment in the steady-state.*



*Figure 11. Temperature vs. time during a) the heating process and b) the natural cooling of the device.*

The experiment shows that the device performs well. When the utility voltage is 220 Volt, the water temperature reaches from 18  $^0C$  to 75  $^0C$ , i.e., to the thermal steady-state in almost 3 hours. When the device is turned off, the water temperature falls to the ambient temperature in almost 5,5 hours as seen in Figure 11.b. The specific heat capacitance of the model is found to be 4.026e+05 J/°C which gives an acceptable error of just -3.99%. The error of the thermal resistance is found to be much higher than the calculated value. The decay of the temperature is not exponential enough. Therefore, the thermal dynamics are more complex than the behavior of the simplified model. This means the thermal model of the device should be modified. Heat transfer equation can be solved in Cartesian coordinates with proper boundary conditions using finite difference or finite element method to improve the heat transfer model of the circuit. Also, an experiment with multiple temperature sensors can be done to measure the thermal conductivity of the insulator material to check its accuracy. However, the device operates well performing its duty during the tests without any problem.

## **8. Conclusion**

In this study, a water absorption chamber is designed using fiberglass and it is controlled with a cheap industrial controller by sensing its temperature with a thermocouple. It is tested using a power cable inside. It has been found that it operates well and it is usable to make the tests described in IEEE 1580- 2021, NEMA WC 53, and NEMA WC 57 standards.

As future work, we suggest using a fuzzy logic controller to reduce the power consumption of the device. Such a system could benefit from user experience and consider ambient temperature and its derivative as additional inputs. This approach can decrease the setting time and provide tighter temperature error control [31]. The system can also be improved by incorporating IoT for monitoring [32]. IoT can make tracking the water temperature easier and can also be used to stop the experiment if something goes wrong with the operation. The proposed device can be easily modified for larger-scale industrial applications. For instance, using a higher power heater can reduce the heating duration. A larger version of this device can be made for medium voltage cables. The heater power should also be increased accordingly in this case. Additionally, the device is already suitable for mass production due to its simplicity and robustness. Such a chamber can also be combined with a high-voltage generator for better control and monitoring of the experiments.

As another future work, the thermal model of the system can be improved with the methods suggested in the experimental results section.

## **Author Contribution**

Formal analysis – Ercan Ulutay (EU), Reşat Mutlu (RM), Uğur Akyol (UA), Metin Yurtsever (MY); Investigation – RM, UA, MY; Experiments – Avşin Öztaş (AÖ), EU; Processing – MY, EU, RM, AÖ; Literature review – RM, AÖ; Writing – MY, EU, RM, UA, AÖ; Review and editing – RM, UA, MY, AÖ, EU.

## **Declaration of Competing Interest**

The authors declared no conflicts of interest with respect to the research, authorship, and/or publication of this article.

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