



Study on pH in water and potassium chloride for Bulgarian soils

Alexander N. Sadovski *

Bulgarian Science Center of the IEAS, Sofia, Bulgaria

Article Info

Received : 26.01.2018

Accepted : 26.09.2018

Abstract

Soil pH is commonly measured in water pH(H₂O) or pH(KCl). The relationship between pH(H₂O) and pH(KCl) across all Bulgarian soils were investigated and results examining the effect of soil type on the relationship were presented. Several functions were used to estimate dependence between the two measures. For all soils and depths, a linear regression accounted for 95.32% of the variation, which predicts pH(KCl) very well. From the analysis of data follows that they were differentiated into three clusters.

Keywords: Soil pH, soil database, cluster analysis.

© 2019 Federation of Eurasian Soil Science Societies. All rights reserved

Introduction

Study on pH measured by different methods are in progress in different countries such as: Romania (Gavriloaei, 2012), Australia (Ahem et al. 1995; Minasyan et al., 2011), Poland (Kabała et al., 2016). Unfortunately, there are several measures of soil reaction used worldwide. The most common extracts are distilled water (H₂O), 1 mol L⁻¹ KCl (KCl), and 0.01 mol L⁻¹ CaCl₂ (CaCl₂). Different measurement methods lead, however, to incompatibility of data from various countries and disturb data integration in the international soil databases.

In this article the relationship between pH(H₂O) and pH(KCl) across all Bulgarian soils were investigated and results examining the effect of soil type on this relationship were presented. For all soils and depths, several regression equations were calculated, which predicts pH(KCl) in dependence of pH(H₂O). The study was intended to help scientists and practitioners in using both methods of pH when dealing with problems of liming of acidic soils.

Material and Methods

Since 1956 the data from the large-scale soil survey have been used to compile soil maps of Bulgarian geographical regions at different scale. Thematic maps of the whole of Bulgaria have been prepared also to facilitate the soil agro-ecological partition at a scale of 1:600,000 (Yolevski et al., 1980), land evaluation for crop production at 1:1,000,000 scale (Kabakchiev et al., 1985). Until that time the so-called agro-ecological grouping of soils was adopted for the needs of agriculture. In Table 1 the total areas and arable areas are presented.

The materials of the study are the values of pH(H₂O) and pH(KCl) given in Reference database for soils in Bulgaria (Teoharov et al., 2009). This valuable source contains 306 data from different soils, namely: Chernozems (64), Gray Forest soils (33), Pseudopodzolic Forest soils (54), Cinnamonic Forest soils (33), Zheltozem soils (30), Leached Smolnitsa (15), Brown Forest soils (11), Mountainous Meadow soils (3), Alluvial Meadow soils (21), Peat-gley soils (11), Rankers (21), Regosols (9), Rendzinas (2), and Technogenic soils (17).

* Corresponding author.

Bulgarian Science Center of the IEAS, Sofia, Bulgaria

Tel.: +3592 9815586

e-ISSN: 2147-4249

E-mail address: bsc.ieas@yahoo.com

DOI: [10.18393/ejss.477560](https://doi.org/10.18393/ejss.477560)

Table 1. Agro-ecological groups of the total and arable area (Yolevski, 1986).

| Code | Agro-ecological groups | Total area | | Arable area | |
|------|------------------------------------------------|------------|--------|-------------|--------|
| | | 1000 ha | (%) | 1000 ha | (%) |
| 01 | Calcareous and typical chernozems | 1047.25 | 9.435 | 444.04 | 16.332 |
| 02 | Leached chernozems | 967.29 | 8.714 | 410.13 | 13.965 |
| 03 | Podzolic chernozems and dark gray forest soils | 538.62 | 4.852 | 228.37 | 5.598 |
| 04 | Gray forest soils | 1158.31 | 10.435 | 491.12 | 11.657 |
| 05 | Light gray pseudopodzolic forest soils | 577.49 | 5.203 | 244.86 | 3.742 |
| 06 | Smolnitsa soils | 659.67 | 5.943 | 279.70 | 8.698 |
| 07 | Cinnamonic forest soils | 2037.87 | 18.359 | 864.06 | 14.617 |
| 08 | Cinnamonic podzolic forest soils | 928.42 | 8.364 | 393.65 | 4.374 |
| 09 | Brown forest soils | 1910.16 | 17.209 | 809.91 | 2.909 |
| 10 | Meadow soils | 999.50 | 9.005 | 423.79 | 16.653 |
| 11 | Rendzinas (humus-carbonate soils) | 275.42 | 2.481 | 116.78 | 1.455 |
| | Sum | 11100.00 | 100.00 | 4706.40 | 100.00 |

Preliminary check of the data shows that three points were erroneous (Chernozem profile No. 32/3/C1k and C2k; Gray Forest profile No. 8 B2t) and were excluded from the analysis. The Technogenic soils also have been excluded. So we had 304 pairs for analysis. First step was performing descriptive statistical analysis. Results are given in Table 2 and 3.

Table 2. Statistical characteristics of different soils.

| Soil | Variable | Valid N | Mean | Median | Std.Dev. | Minimum | Maximum | Skewness | Kurtosis |
|----------------------|------------------|---------|--------|--------|----------|---------|---------|----------|----------|
| Chernozems | H ₂ O | 62 | 7.4403 | 7.75 | 0.7221 | 6.00 | 8.30 | -0.686 | -0.945 |
| | KCl | 62 | 6.6226 | 7.10 | 0.9510 | 4.60 | 7.80 | -0.725 | -0.897 |
| Gray Forest | H ₂ O | 32 | 6.2531 | 5.50 | 1.2324 | 4.30 | 8.20 | 0.541 | -1.307 |
| | KCl | 32 | 5.2922 | 4.70 | 1.2422 | 3.40 | 7.40 | 0.524 | -1.333 |
| Pseudopozolic Forest | H ₂ O | 54 | 5.3796 | 4.85 | 1.1690 | 4.00 | 8.40 | 1.209 | 0.269 |
| | KCl | 54 | 4.3287 | 3.85 | 1.2783 | 2.90 | 7.20 | 1.174 | 0.044 |
| Cinnamonic Forest | H ₂ O | 32 | 7.0500 | 7.70 | 1.1188 | 4.70 | 8.20 | -1.049 | -0.249 |
| | KCl | 32 | 6.1016 | 6.80 | 1.2652 | 3.40 | 7.40 | -1.048 | -0.219 |
| Zheltozem | H ₂ O | 30 | 4.9300 | 4.90 | 0.2818 | 4.40 | 5.90 | 1.322 | 4.224 |
| | KCl | 30 | 3.8017 | 3.75 | 0.3865 | 3.20 | 4.95 | 1.489 | 2.905 |
| Leached Smolnitsa | H ₂ O | 15 | 6.9867 | 7.10 | 0.7130 | 6.10 | 8.20 | 0.332 | -1.095 |
| | KCl | 15 | 5.9933 | 5.90 | 0.7363 | 5.10 | 7.20 | 0.377 | -1.329 |
| Brown Forest | H ₂ O | 11 | 5.4864 | 5.40 | 1.1845 | 4.00 | 7.20 | 0.074 | -1.640 |
| | KCl | 11 | 4.4055 | 4.40 | 1.0340 | 3.10 | 5.90 | 0.067 | -1.667 |
| Alluvial Meadow | H ₂ O | 21 | 7.6810 | 8.10 | 0.6853 | 5.80 | 8.30 | -1.236 | 1.135 |
| | KCl | 21 | 7.0714 | 7.50 | 0.8113 | 5.40 | 7.80 | -1.104 | -0.246 |
| Peat-gley | H ₂ O | 11 | 7.5727 | 7.70 | 0.4052 | 6.80 | 8.00 | -0.839 | -0.493 |
| | KCl | 11 | 7.0727 | 7.10 | 0.2687 | 6.50 | 7.40 | -0.818 | 0.757 |
| Rankers | H ₂ O | 21 | 4.9643 | 4.95 | 0.5730 | 4.10 | 6.20 | 0.035 | -0.418 |
| | KCl | 21 | 4.2333 | 4.20 | 0.6187 | 3.10 | 5.50 | 0.217 | -0.500 |
| Regosols | H ₂ O | 9 | 7.9333 | 8.00 | 0.5745 | 6.70 | 8.60 | -1.275 | 1.873 |
| | KCl | 9 | 6.8556 | 6.90 | 0.6002 | 5.60 | 7.50 | -1.139 | 1.301 |

Table 3. Descriptive statistics from combined analysis of pH(H₂O) and pH(KCl).

| pH | N | Mean | Median | Std.Dev. | Min | Max | Skewness | Kurtosis |
|------------------|-----|--------|--------|----------|------|------|----------|----------|
| H ₂ O | 304 | 6.4158 | 6.45 | 1.3879 | 4.00 | 8.60 | -0.095 | -1.519 |
| KCl | 304 | 5.4976 | 5.50 | 1.5262 | 2.90 | 7.80 | -0.047 | -1.528 |

There is another collection of data with the properties of soils in Bulgaria (Ninov et al., 1975). Unfortunately, it is too limited and not all pH analyzes by both methods are included simultaneously for different soils.

Next step was performing regression analysis to describe the link between pH(H₂O) and pH(KCl). The aim was to find the most appropriate function, which accurately describes the relationship between the values of both pH analyzes. It was also interesting to investigate the distribution of pH and its differentiation across the different soil groups. For this purpose cluster analysis was applied.

We are looking for a regression of the type

$$y = f(x),$$

where $y = \text{pH}(\text{KCl})$, $x = \text{pH}(\text{H}_2\text{O})$ and f is a selected regression model.

We consider the following types of equations:

- (a) $y = a + bx$, Linear function
 (b) $y = a + bx + cx^2$, Quadratic function
 (c) $y = \frac{1}{a + bx + cx^2}$, Reciprocal Quadratic function
 (d) $y = ab^x x^c$, Hoerl function

Selected function have no more than three parameters to be estimated. The principle of Ocam is followed: "Of two competing theories, the simpler explanation of an entity is to be preferred" (Duignan, 2017). If you have a few hypotheses that could explain an observation, it is usually best to start with the simplest one.

Results and Discussion

The United States Department of Agriculture, formerly Soil Conservation Service (Soil Survey Staff, 1993) classifies soil pH in water ranges as follows in Table 4. FAO classification applicable in Bulgaria is given in (Gyurov and Artinova, 2015).

Table 4. pH classification by USA and FAO (applicable in Bulgaria).

| USA accepted classification | | FAO accepted classification | |
|-----------------------------|----------|-----------------------------|------------------------|
| Denomination | pH range | pH(H ₂ O) | Reaction |
| Ultra acid | < 3.5 | < 3.0 | Extremely acid |
| Extremely acid | 3.5–4.4 | 3.0–4.0 | Very strongly acid |
| Very strongly acid | 4.5–5.0 | 4.1–5.0 | Strongly acid |
| Strongly acid | 5.1–5.5 | 5.1–6.0 | Moderately acid |
| Moderately acid | 5.6–6.0 | 6.1–6.9 | Slightly acid |
| Slightly acid | 6.1–6.5 | 7.0 | Neutral |
| Neutral | 6.6–7.3 | 7.1–7.5 | Very slightly alkaline |
| Slightly alkaline | 7.4–7.8 | 7.6–8.1 | Slightly alkaline |
| Moderately alkaline | 7.9–8.4 | 8.2–8.6 | Moderately alkaline |
| Strongly alkaline | 8.5–9.0 | 8.7–8.9 | Alkaline |
| Very strongly alkaline | > 9.0 | 9.0–10.0 | Strongly alkaline |
| | | 10.1–11.0 | Very strongly alkaline |

First, for each of the soils all models (a), (b), (c) and (d) are calculated. Results of corresponding correlation coefficients R^2 are presented in Table 5. Because of their small numbers of data Mountainous Meadow soils and Rendzinas are excluded from separate consideration, but they are included in the combined analysis.

Table 5. Values of correlation coefficients R^2 of models for different soils.

| Soils | N | Models | | | |
|-----------------------------|----|--------|--------|--------|--------|
| | | (a) | (b) | (c) | (d) |
| Chernozems | 62 | 0.9078 | 0.9174 | 0.9206 | 0.9183 |
| Gray Forest soils | 32 | 0.9291 | 0.9291 | 0.9299 | 0.9292 |
| Pseudopodzolic Forest soils | 54 | 0.9623 | 0.9636 | 0.9660 | 0.9634 |
| Cinnamonic Forest soils | 33 | 0.9786 | 0.9786 | 0.9791 | 0.9786 |
| Zheltozem soils | 30 | 0.9152 | 0.9161 | 0.9176 | 0.9161 |
| Leached Smolnitsa | 15 | 0.9060 | 0.9080 | 0.9082 | 0.9081 |
| Brown Forest soils | 11 | 0.9884 | 0.9884 | 0.9893 | 0.9884 |
| Alluvial Meadow soils | 21 | 0.9298 | 0.9328 | 0.9328 | 0.9323 |
| Peat-gley soils | 11 | 0.8008 | 0.8597 | 0.8600 | 0.8596 |
| Rankers | 21 | 0.8438 | 0.8439 | 0.8424 | 0.8438 |
| Regosols | 9 | 0.9945 | 0.9943 | 0.9949 | 0.9948 |

Types of models: (a) Linear function, (b) Quadratic function, (c) Reciprocal Quadratic function, (d) Hoerl function.

It is evident that the highest values of R^2 are for the model (c) Reciprocal Quadratic function in almost all soils. But let's not rush to the conclusions about the most suitable function before looking at combined data and applying Ocam's principle.

In the analysis of all 304 value pairs, the following regressions are obtained,

where $y = \text{pH}(\text{KCl})$ and $x = \text{pH}(\text{H}_2\text{O})$:

| | | | | | |
|-----|--------------------------------|----------------|----------------|-----------------|----------------|
| (a) | $y = a + bx,$ | $a = -1.39116$ | $b = 1.07371$ | | $R^2 = 0.9533$ |
| (b) | $y = a + bx + cx^2,$ | $a = -0.38044$ | $b = 0.73943$ | $c = 0.02632$ | $R^2 = 0.9766$ |
| (c) | $y = \frac{1}{a + bx + cx^2},$ | $a = 0.72916$ | $b = -0.13037$ | $c = 0.0006526$ | $R^2 = 0.9776$ |
| (d) | $y = ab^x x^c,$ | $a = 0.49694$ | $b = 0.99356$ | $c = 1.31154$ | $R^2 = 0.9767$ |

If you follow the Ocam's principle, it is natural to accept the linear regression, which has only two coefficients:

$$\text{pH}(\text{KCl}) = -1.39116 + 1.07371 \times \text{pH}(\text{H}_2\text{O}) \text{ with } R^2 = 0.9533.$$

The R-Squared statistic indicates that the model as fitted explains 95.33% of the variability in $\text{pH}(\text{KCl})$. The adjusted R-squared statistic, which is more suitable for comparing models with different numbers of independent variables, is 95.32%. The standard error of the estimate shows the standard deviation of the residuals to be 0.33029. This value can be used to construct prediction limits for new observations.

If available data for pH are analyzed with KCl , it can be used the reverse analysis, where $y = \text{pH}(\text{H}_2\text{O})$, $x = \text{pH}(\text{KCl})$ and gives the equation:

$$\text{pH}(\text{H}_2\text{O}) = 1.53466 + 0.88787 \times \text{pH}(\text{KCl}) \text{ with } R^2 = 0.9533.$$

It should not be forgotten that the analyzes of both methods - in water and potassium chloride, produce results with a certain error. It is logical to use the orthogonal regression method (Total least squares), which is appropriate in such case. That gives the equation:

$$\text{pH}(\text{KCl}) = -0.076319 + 0.86878 \times \text{pH}(\text{H}_2\text{O}) \text{ with } R^2 = 0.9536.$$

It should be noted that this method gives the same average value of the dataset, but with a smaller standard error equal to 1.2057. Figure 1 shows the straight lines of linear and orthogonal regression.

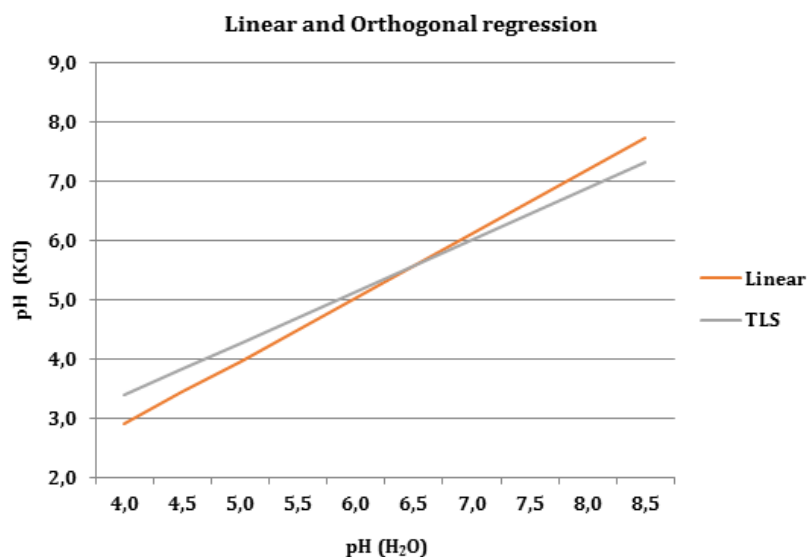


Figure 1. Both regression lines.

It is interesting to see the histograms of the distributions of the two variables.

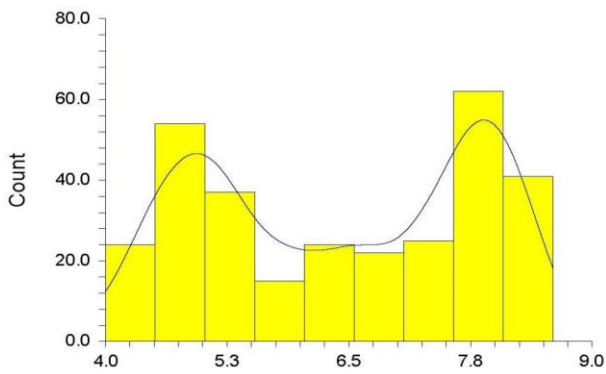


Figure 2. Histogram of pH(H₂O)

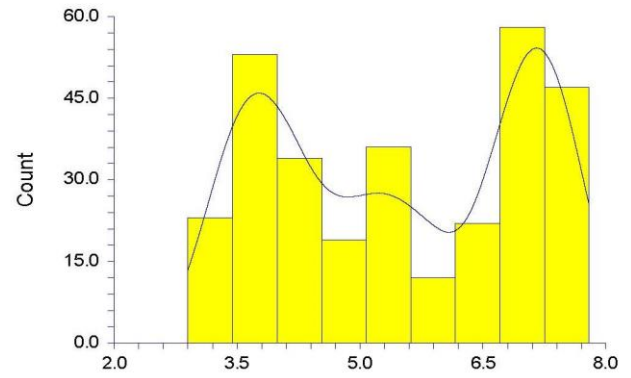


Figure 3. Histogram of pH(KCl)

Obviously, it differs significantly from Normal (Gaussian) distribution and is a mixture of two or more different distributions. This justifies the use of cluster analysis to identify the groups that determine the differences in analyses. As a result of this analysis, three clusters were obtained, the statistical characteristics of which are given in Table 6.

Table 6. Statistical characteristics of three Clusters.

| | pH | N | Mean | Median | St. Dev. | Min | Max | Skewness | Kurtosis |
|-----------|------------------|-----|--------|--------|----------|------|------|----------|----------|
| Cluster 1 | H ₂ O | 123 | 7.8728 | 7.90 | 0.3343 | 6.80 | 8.60 | -0.470 | 0.208 |
| | KCl | 123 | 7.1305 | 7.20 | 0.3612 | 6.30 | 7.80 | -0.311 | -0.447 |
| Cluster 2 | H ₂ O | 62 | 6.4565 | 6.40 | 0.4344 | 5.40 | 7.20 | -0.095 | -0.741 |
| | KCl | 62 | 5.4387 | 5.45 | 0.4112 | 4.60 | 6.30 | 0.100 | -0.264 |
| Cluster 3 | H ₂ O | 119 | 4.8887 | 4.90 | 0.4090 | 4.00 | 5.60 | -0.171 | -0.620 |
| | KCl | 119 | 3.8404 | 3.80 | 0.4563 | 2.90 | 4.90 | 0.294 | -0.562 |

Closer examination of the data shows the following:

- Members of Cluster 1 are predominantly data from Alluvial Meadow soils, Chernozems and Gray Forest soils. Hypothesis for Normal distribution can't be confirmed.
- Members of Cluster 2 are data from Chernozems, Brown Forest soils and Leached Smolnitsa. Hypothesis for Normal distribution can't be rejected.
- Members of Cluster 3 are data from Pseudopodzolic Forest soils, Rankers and Zheltozem soils. Hypothesis for Normal distribution can't be rejected.

Figure 4 shows a distinct differentiation between the three clusters.

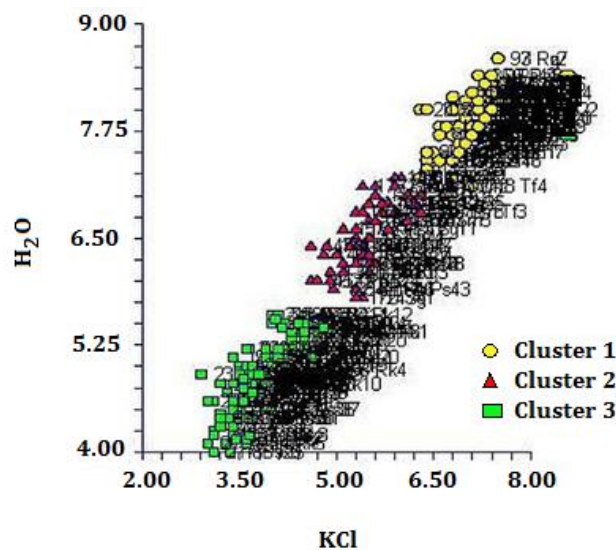


Figure 4. Cluster analysis

The presence of objects from the same soil group as members of different clusters can be explained by the large soil diversity in Bulgaria and some inaccuracies in the identification of soil profiles.

The simple linear model equation (a) seems to perform very well and is almost accurate as the nonlinear models equations (b, c, d) across all the datasets. As a result, values of pH in KCl can be predicted as a function of pH in water and vice versa. From a purely statistical point of view, it is advisable to use orthogonal linear regression, which takes into account the fact that in both methods the results are obtained with a certain error.

Conclusion

The analysis of 304 pairs results from soil samples is the basis for reliable conclusions. The results of this study allow soil reaction data obtained from different methods - with distilled water (H₂O) and 1 mol L⁻¹ KCl (KCl), to be converted and integrated into national and international soil databases. The difference between pH in water and pH in KCl is that the first refers to the acidity of the soil solution, while the pH in KCl refers to the acidity of the soil solution plus the reserve acidity in the colloids and therefore it is always more acid than pH in water. Regression between pH(H₂O) and pH(KCl) is important because it gives the possibility for soil scientists to directly compare own values with the data already existing in literature from other country.

Monitoring pH changes over time is an important management tool. By comparing past and present soil tests, it is possible to see if the soil acidity is increasing over time and, if it is, to alter management methods to prevent this trend from continuing. Analysis of the results from the soil survey in Bulgaria shows that almost half of the soil resources are vulnerable to anthropogenic acidification. Special attention must be paid to genetically acid soils under cultivation. Their additional acid loading has to be controlled to avoid anthropogenic soil degradation.

References

- Ahem, C.R., Baker, D.E., Aitken, R.L., 1995. Models for relating pH measurements in water and calcium chloride for a wide range of pH, soil types and depths. *Plant and Soil* 171(1): 47-52.
- Duignan, B., 2017. Occam's razor. Encyclopaedia Britannica. Available at [access date: 26.03.2018]: <https://www.britannica.com/topic/Occams-razor>
- Gavriloaei, T., 2012. The influence of electrolyte solutions on soil pH measurements. *Revista de Chimie* 63(4): 396-400.
- Gyurov G., Artinova, N., 2015. Soil Science. Intelxpert-94 Ltd Ood Publishing House, Plovdiv, Bulgaria. 258 p. [in Bulgarian].
- Kabakchiev I., Georgiev, B., Bojinova, P., 1985. Map of the land evaluation for crop production. N. Poushkarov Institute of Soil Science and Agroecology, Sofia, Bulgaria. [in Bulgarian].
- Kabała, C., Muszytyfaga, E., Gałka, B., Łabuńska, D., Mańczyńska, P., 2016. Conversion of soil pH 1:2.5 KCl and 1:2.5 H₂O to 1:5 H₂O: Conclusions for soil management, environmental monitoring, and international soil databases. *Polish Journal of Environmental Studies* 25(2): 647-653.
- Minasny B., McBratney, A.B., Brough, D.M., Jacquier, D., 2011. Models relating soil pH measurements in water and calcium chloride that incorporate electrolyte concentration. *European Journal of Soil Science* 62(5): 728-732.
- Ninov N., Yolevski, M., Kukularov, P., 1975. Ecological characterization of experimental fields. N. Poushkarov Institute of Soil Science and Agroecology, Sofia, Bulgaria. 193p. [in Bulgarian].
- Soil Survey Staff, 1993. Soil survey manual. Chapter 3 Examination and Description of Soil Profiles. C. Ditzler, K. Scheffe, H.C. Monger (Eds.). United States Department of Agriculture (USDA) Handbook 18. Government Printing Office, Washington, D.C. USA. Available at [access date: 26.03.2018]: https://www.nrcs.usda.gov/wps/portal/nrcs/detail/soils/ref/?cid=nrcs142p2_054253
- Spurway, C.H., 1941. Soil reaction (pH) preferences of plants. Michigan State University Agricultural Experiment Station. Special Bulletin Vol. 306. 36p.
- Teoharov M., Sv. Popandova, R. Kancheva, T. Atanasova, V. Tsołova, M. Banov, P. Ivanov, E. Filcheva, R. Ilieva. 2009. Reference database for soils in Bulgaria. N. Poushkarov Institute of Soil Science and Agroecology, Poni Publishing House, Sofia, Bulgaria. 416p. [in Bulgarian].
- Yolevski, M., 1986. Study, maintenance and enhancement of soil fertility, restoration of damaged land. Conservation and Effective Use of Soil Resources in Bulgaria, Discussion, June 27, 1985, 18-24. NRC, BPD, Sofia, Bulgaria. [in Bulgarian].
- Yolevski, M., A. Hadzhiyanakiev, Y. Georgieva, I. Kabakchiev. 1980. Map of the soil agro-ecological regions in Bulgaria at a scale of 1:600,000. KIPP Cartography, Sofia, Bulgaria. [in Bulgarian].