

Effect of EDTA on the removal Cd from artificially contaminated water by *Tradescantia fluminensis*

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Abstract

Heavy metal polluted water is a great threat to sustainable marine, agriculture and environment. Plants can have the natural ability to absorb essential metals in its tissues from water solution, and this ability of plants can be advantage to extract trace metals from the polluted water. In this study, Cd(II) accumulation and distribution in the *Tradescantia fluminensis* were investigated, with a focus on the role of ethylenediaminetetracetic acid (EDTA) as a chelating agent in hydroponics. The results showed that in plant, EDTA, addition to media produced the highest major increase in Cd(II) accumulation but significantly decreased biomass of the plant. Also the Enrichment Coefficient (EC) and Translocation Factor (TF) values were calculated to evaluate the removal efficiency of the EDTA. Based on higher BCF ($9.95 \pm 0.54 - 12.06 \pm 0.40$) and TF ($1.03 \pm 0.11 - 1.27 \pm 0.10$) values, the high Cd(II) accumulation in the stems and leaves indicated that *Tradescantia fluminensis* has the potential of hyperaccumulation under EDTA enhanced treatments in removal for Cd(II) contaminated water.

Keywords: *Tradescantia fluminensis*, cadmium, EDTA, water.

***Tradescantia fluminensis* Bitkisi Kullanılarak Sulardan Cd Uzaklaştırılmasına EDTA'nın Etkisi**

Öz

Ağır metaller ile kirlenmiş sular sürdürülebilir tarım ve çevre için büyük bir tehdittir. Bitkiler, beslenmeleri için ihtiyacı olan elementleri sulardan bünyelerine alma yeteneğine sahiptir ve bitkilerin bu yeteneği, kirlenmiş sulardan ağır metalleri uzaklaştırmak için kullanılabilir. Bu çalışmada, *Tradescantia fluminensis* bitkisinin Cd(II) biriktirme ve bünyesindeki dağılımı incelenmiş ve etilen diamintetrasetik asit (EDTA) 'nın şelatlaştırıcı madde olarak bunlara etkisi araştırılmıştır. Sonuçta, EDTA'nın ortama ilave edilmesinin Cd(II) birikiminde yüksek artışa neden olduğu ancak bitkinin biyokütlesini önemli ölçüde azalttığı görülmüştür. Ayrıca, EDTA'nın giderim etkinliğini değerlendirmek için Biriktirme Katsayısı (EC) ve Taşıma Faktörü (TF) değerleri hesaplanmıştır. Yüksek BCF ($9.95 \pm 0.54 - 12.06 \pm 0.40$) ve TF ($1.03 \pm 0.11 - 1.27 \pm 0.10$) değerlerine bakılarak, gövde ve yapraklardaki yüksek Cd birikimi, *Tradescantia fluminensis* bitkisinin Cd(II) ile kirlenmiş sular için yardımcı kimyasal EDTA kullanıldığında hiperakümülatör bitki potansiyeline sahip olduğunu sonucuna varılmıştır.

Anahtar Kelimeler: *Tradescantia fluminensis*, Kadmiyum, EDTA, su.

1. Introduction

Rapid increases in world population and industrialisation have seriously contributed to highly toxic heavy metal pollution in ecosystems. Due to the shortage of freshwater for irrigation, municipal and industrial sewage water is being used for irrigation in various countries including Turkey (Anwar et al., 2016).

Heavy metals are well known risk elements for the environment, and cadmium (Cd) is the one of the most toxic of the heavy metals. Once in the water, Cd is highly persistent and maybe come available or uptake by fishes and plants and it is a threat to food security (H. W. Liu, Wang, Ma, Wang, & Shi, 2016). Phytoremediation refers to cleaning pollutants from contaminated waters by plants, and it has attracted much attention since it is an environmentally friendly and relatively cheap technique (Cay, 2016; Sidhu, Singh, Batish, & Kohli, 2017). Each plant species has different levels of tolerance toward different contaminants. The plants have evolved different mechanisms to maintain physiological collections of essential metals and to minimize exposure to nonessential heavy metals and to minimize the damage caused in plants by heavy metals (Bouchama, Rouabhi, & Djebar, 2016; Cay, Uyanik, & Engin, 2016). Excess Cd can interfere with numerous biochemical and physiological processes in plants by forming unspecific complex compounds in the cells. There is a need for the development of inexpensive and environmentally friendly new methods to extract heavy metals, thus improving water quality. Thanks to its cost effectiveness and environmentally friendly nature, phytoextraction (the use of green plants for remediation) of heavy metal contaminated water has become increasingly popular currently.

Spiderwort (*Tradescantia fluminensis*) is a monocotyledonous plant with soft fleshy leaves and a fleshy body belonging to the Commelinaceae family, native to southwestern Brazil and Northern Argentina. To date, spiderwort is also presented in Portugal, Italy, New Zealand, Russia, Japan, and south – eastern regions of Australia and USA, growing in damp and shaded places not like the bulk areas of forests but such as banks, parks, gardens, streamsides, and forest edges (R. Y. Liu et al., 2016). The aim of the present study was, therefore, to explore the Cd uptake capacity phytoremediation potential of *Tradescantia fluminensis* and effect of EDTA for Cd removal from wastewater.

2. Materials and methods

2.1. Experiment Design

Tradescantia fluminensis seedling was used for this study. When the seedlings reached 10 cm in length and reached about 15 cm in length with mature roots, fidelities of similar shape and size were selected and washed three times with tap water to remove adhering particles. These plants were then used for experiments and separated into the beaker containing 100 ml of the nutrient solution for 2 days preculture. The feed medium was a modified half-strength Hoagland nutrient solution (pH 5.5) made with the following salts (J. Y. Liu & Schnoor, 2008). 1 mmol L⁻¹ KNO₃, 5 mmol L⁻¹ Ca(NO₃)·4H₂O, 5 mmol L⁻¹ KH₂PO₄, 2 mmol L⁻¹ MgSO₄·7H₂O, 46.26 μmol L⁻¹ H₃BO₃, 9.15 μmol L⁻¹ ZnSO₄·7H₂O, 0.77 μmol L⁻¹ MnCl₂·4H₂O, 0.32 μmol L⁻¹ H₃MoO₄·H₂O, 0.12 μmol L⁻¹ CuSO₄·5H₂O, 20.01 μmol L⁻¹ FeSO₄·7H₂O, 20.03 μmol L⁻¹ EDTA-2Na· 2H₂O). Three rebuilds for each treatment with two *Tradescantia fluminensis* seedling in each beaker. The seven treatments were set as follows: Cd concentrations (CdCl₂·2H₂O (Merck, Germany) (0, 25, 50, and 100 mg kg⁻¹) and EDTA concentrations (EDTA (Merck, Germany) (0.5, 1.0 and 1.5 mmolL⁻¹) (See table 1). The plants were harvested after being grown for 21 days. Plant samples were carefully washed, dried, separated and weighed into roots, stems and leaves.

Table 1. Concentrations of Cd and EDTA

Treatment	Cd (mg kg ⁻¹)	EDTA (mmol kg ⁻¹)	Treatment	Cd (mg kg ⁻¹)	EDTA (mmol kg ⁻¹)
C0	-	-	EC0	25	-
Cd1	10	-	ECd1	25	0.5
Cd2	25	-	ECd2	25	1.0
Cd3	50	-	ECd3	25	1.5
Cd4	100	-			

E: EDTA, C0: Control

2.2. Measurement methods

After 21 days for hydroponics plants were harvested and divided into roots, stems and leaves. The separated pieces were then used to identify the accumulated Cd. Harvested plants were first washed with tap water and than washed with distilled water. The clean parts were dried in an oven at 70–80°C until constant weight was achieved. The dried plant samples were finely pulverized and their dry weights were measured. For the preparation of plant samples for analysis, microwave wet digestion technique was used. The aqueous phase was separated from the rest of the plant by centrifugation (Engin, Uyanik, Cay, & Icbudak, 2010). The concentration of heavy metals in plant

tissues was determined by a Bruker (820-MS) Inductive Coupled Plasma Mass Spectrometer (ICP-MS).

2.3.Data Analysis

In addition to the cumulative concentration, BCF (Bioconcentration Factor) and TF (Translocation Factor) were used to assess the Cd accumulation capacity of *Tradescantia fluminensis*. BCF is the ratio of metal concentration to soil or water in the plant collection site (root, stem or leaf) (Zayed, Lytle, Qian, & Terry, 1998). TF shows the ability to transport shoots defined as the ratio of the root of a plant to the concentration of metal in plant roots at plant roots (Mattina, Lannucci-Berger, Musante, & White, 2003).

$BCF = \text{Dry weight (dw) Cd concentration in plant tissue} / \text{The initial concentration of Cd in the medium}$

$TF = \text{Cd concentration in stem or leaf (dw)} / \text{Cd concentration in root (dw)}$ (Chanu & Gupta, 2016).

The metal concentrations of water and plants were reported as mg kg^{-1} dry weight, and each result was the vehicle of three replicates. Anova was implemented with SPSS®16.0 statistical package Microsoft Excel® 2010 was used to draw the charts.

3. Results and discussion

Cd addition level of 10 and 50 mg kg^{-1} can be tolerated by *Tradescantia fluminensis* as there is a marginal decrease in the plant dry matter and no inhibitory symptoms appeared at this Cd addition level. As shown in Table 2, greater Cd addition into the water led to a significant decrease in plant dry matter. A similar trend was observed in plants treated with EDTA.

Table 2. Effects of Cd and EDTA on *Tradescantia fluminensis* dry weight

Treatment	Dry weight (g)	Treatment	Dry weight (g)
C0	1.67±0.37	ECd0	1.89±0.31
Cd1	1.36±0.36	ECd1	1.22±0.28
Cd2	1.89±0.31	ECd2	1.16±0.27
Cd3	1.23±0.30	ECd3	1.08±0.25
Cd4	1.05±0.33		

E: EDTA, C0: Control (Values within the same column and followed by the same letter are not different at $P < 0.05$ by an ANOVA-protected LSD test)

Cd accumulation in plant increased significantly with increasing Cd concentration. The greatest Cd accumulation was observed when 100 mg kg⁻¹ Cd was applied (See Fig. 1).

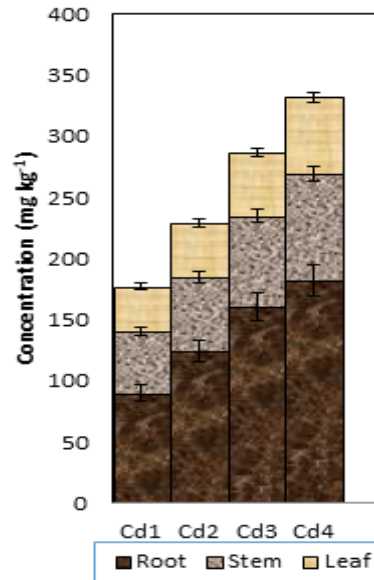


Figure 1. Concentration of Cd in root, stem and leaf of *Tradescantia fluminensis*.

It was found that plants supplemented with EDTA showed significantly greater uptake of Cd as compared to control plants without EDTA (See Fig. 2). This is because EDTA has the potential to mobilize Cd in the water and Cd-EDTA also absorbed by plants. Cd accumulation in roots was affected by EDTA so that the greatest accumulation was obtained when EDTA was applied. EDTA lead to a greater uptake of metals into the plant root and through the cell membrane.

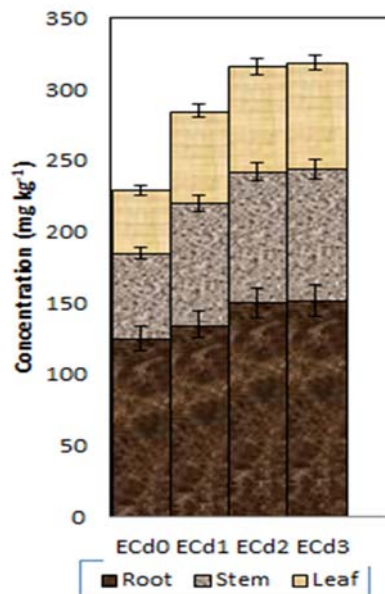


Figure 2. Comparison of Cd accumulation in each part of *Tradescantia fluminensis* with EDTA

The Cd Bioconcentration Factor (BCF) and Translocation Factor (TF) of *Tradescantia fluminensis* was calculated by dividing the lead content in the various parts of the corn by the total Cd concentration in the Cd (See Table 3). The BCF values of Cd in the plant were lower if the water Cd concentrations were higher. The translocation factor (TF) values shown in Table 3 denote the ratio of the Cd concentration in the plant parts to that in the roots. The TF of the plant was the highest (1.27) when EDTA was added. TF values less than 1.0 indicate that a plant can tolerate Cd content, whereas TF values greater than 1.0 indicate that a plant is a Cd hyperaccumulator.

Table 3. Bioconcentration factors and Translocation Factors under different treatments

Treatment	BCF	TF	Treatment	BCF	TF
Cd1	11.17±1.47	0.87±0.15	ECd1	9.95±0.54	1.03±0.11
Cd2	9.83±0.68	0.80±0.12	ECd2	11.70±0.61	1.15±0.14
Cd3	7.85±0.42	0.69±0.09	ECd3	12.06±0.40	1.27±0.10
Cd4	3.43±0.36	0.52±0.10			

E: EDTA (Values within the same column and followed by the same letter are not different at $P < 0.05$ by an ANOVA-protected LSD test)

4. Conclusion

This study demonstrates the ability to store excess Cd in water and to store excess Cd in the stem and lower stem at high Cd exposure. The plant could well tolerate up to 100 mg kg^{-1} Cd concentration levels, no visual symptom was observed in *Tradescantia fluminensis*. The plants with high efficiency for Cd accumulation may be exploited in extracting metals from polluted water individually or in combination with chelating agent as EDTA based treatment system involving interdisciplinary approach.

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