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Original Research Article

Performance, emission, energetic and exergetic analyses of a compression ignition engine fuelled with sunflower oil methyl esters

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1. Introduction

Nowadays, biodiesel is the most used alternative fuel instead of petroleum-based fuels. It is produced from renewable biomass, such as waste oils, vegetable oils or animal fats. The main advantages of biodiesel usage over conventional fossil-based fuels are being nontoxic, biodegradable, and environmentally beneficial [1]. However, with these significant advantages, biodiesel usage also causes SOB negative consequences. Biodiesel fuels generally have about 10–12% less energy content in comparison with fossil diesel on the mass basis. High price, injector coking, higher viscosity, engine compatibility, and engine wear are the other major disadvantages of biodiesel usage as fuel [2]. Many studies have been conducted in order to determine the characteristics of performance and emission of the compression ignition engines fueled with various biodiesel [3-7].

The efficiency and the performance of internal combustion engines (ICE) can be enhanced by controlling the combustion, proper thermal insulation and recuperation of the engine. From this point of view, the efficiencies of ICEs operating with alternative fuels, in the means of energy and exergy is very crucial [8]. There are SOB studies which used just the first law of thermodynamics to determine performance parameters of the engine. However, to evaluate the energy resource utilization, using just the

first law of thermodynamics is not enough. The second law of the thermodynamics is essential to be able to understand the complete system thermodynamics and to determine magnitudes, locations, and causes of energy waste in a thermal system [9]. There are SOB studies which contain exergy analyses of the ICEs. [10- 12].

Figure 1. Biodiesel Production Processes

In this study, characteristics of the test engine which is a diesel engine fueled with SOB and conventional diesel fuel in the means of performance and emission were determined and compared. After this step, energy and exergy analysis were conducted by using the extracted data from the experiments to be able to define energetic and exergetic performance of the test engine.

2. Material and Method

2.1. Biodiesel Production and Specifications

Experiments were carried out in Department of Automotive Engineering Laboratories in Çukurova University, Adana, Turkey.

The transesterification of SOB biodiesel was illustrated in Figure 1.

The analysis of test fuels was conducted at the Çukurova University Automotive Engineering Laboratories. Density, kinematic viscosity, flash point, cetane number, pour point, heating value characteristic was tested according to standards. Fuel properties of diesel and biodiesel fuels were given in Table 1. Also European petrodiesel standard EN 590 and European biodiesel standard EN 14214 was given in Table 1_{\ldots}

2.2. Engine Test Rig

Engine performance tests were performed on a commercial four cylinder, four-stroke, naturally aspirated, water-cooled direct injection diesel engine with a 89 kW maximum power at 3200 rpm engine speed and has 295 Nm maximum torque at 1800 rpm engine speed. Technical specifications of engine were given in Table 2. Before the tests, the engine was run for 10 minutes with diesel fuel to attain the operation temperature. Test fuels were examined from 1000 to 3000 rpm with a range of 200 rpm at full load condition.

Torque and power output were quantified with the aid of hydraulic dynamometer. Schematic representation of experimental setup was showed in Figure 2. The speed sensor is used to determine the engine speed. Torque of dynamometer is quantified by the help of S type load cell. Dynamometer control panel and computer was utilized for recording all the data that was produced by the system.

Figure 2. Schematic representation of experimental setup

TESTO 350 XL gas analyzer is utilized to quantify exhaust emissions. Emission data was recorded by means of computer program. Measurement accuracy of the gas analyzer is ± 10 ppm for CO, 1% for CO2 and ± 1 ppm for NOx. Measurement capacity of the device is 0- 10000 ppm for CO, 0-50% for CO2 emission and 0-3000 ppm for NOx.

3. Results and Discussion 3.1. Brake Power and Torque

The alteration of the brake power and torque values of the test fuels at different engine speeds are illustrated in the following figures, figures 3 and 4. It is clear that both the values of the brake power and torque values of the test engine were reduced by using biodiesel. There was nearly 4.8% decrement in brake power and 5.4% decrement in torque values when the engine was operated with SOB instead of the diesel fuel. Lower engine power can be explained by lower energy content and poor atomization because of higher density and viscosity values of SOB.

3.2. Emissions

CO emission results of the experiments at different engine speed for fuels was compared in Figure 5. Incomplete combustion is the main reason for CO emission in engine. With the usage of biodiesel, CO emission was decreased by 5.7% with respect to diesel fuel. Extra oxygen content of biodiesel provides extra oxidation of CO . $CO₂$ emission was increased by using SOB with respect to diesel fuel as anticipated. When biodiesel was used, oxygen content of the combustion chamber increased. Therefore, this extra oxidation effect of

biodiesel caused to increase $CO₂$ emission. With the biodiesel usage, $CO₂$ emission was increased by 5.4 % with respect to diesel fuel. It was shown in Figure 6. Temperature is the most important parameter to form oxides of nitrogen (NO_x) . Figure 7 showed that biodiesel increased the content of nitric oxide emission as compared to standard diesel. The content of oxygen of biodiesel enhances the combustion. Enhanced combustion means higher end-combustion temperatures. Therefore, when biodiesel was used, 4.8 % increment of NO_x emission was obtained.

3.3. Energetic and Exergetic Performance

Energy is inevitable part of life and industrialization. Therefore, usage of energy sources effectively and efficiently is very important issue. For welfare and composure of human, there is need for energy usage on every sector [13].

To be able to perform energy and exergy analyses, general form of chemical reaction was considered as:

$$
aC_xH_yO_z + b(O_2 + 3.76N_2) \rightarrow cO_2 + dCO + eCO_2 + fNO
$$

+ $gNO_2 + hN_2 + jH_2O$

By applying conservation of mass principle and using emission data procured from tests, all coefficients were found.

The assumptions made in the energy and exergy analyses are described below:

 \checkmark Whole engine is considered as control volume and it operates as a steady state open system,

 \checkmark The exhaust gas mixtures and the combustion air (79% N_2 , 21% O_2) are ideal,

 \checkmark The potential and kinetic energy effects of combustion air, fuel stream and exhaust gases are ignored,

 \checkmark The reference environment is, P_o=1 atm and $T_0 = 25$ °C [14].

The balances of mass and for the control volume, steady state open system can be expressed as:

$$
\sum_{\substack{0 \text{odd} \\ Q-W}}^{\mathsf{d}} = \sum_{\substack{m_{out} \\ m_{out}}}^{\mathsf{d}} h_{out} - \sum_{\substack{m_{in} \\ m_{in}}}^{\mathsf{d}} h_{in}
$$

Energy input rate (*Efuel*) can be found as:

$$
E_{\text{fuel}} = m_{\text{fuel}} H_{\text{u}}
$$

Energy efficiency $(1st$ law efficiency) is computed by dividing the work rate to the energy input rate of fuel;

$$
\eta_{\iota} = \frac{\overset{\circ}{W}}{E_{\text{fuel}}}
$$

Energy input rate is the energy supplied to the engine that contains work rate and heat losses and computed by multiplying mass flow rate with lower heating value of fuel. Given energy to engine is directly in proportion with lower calorific value of fuel. Therefore, being heating value of diesel fuel is high caused to have higher energy input rate of this fuel. It is clearly seen from the Figure 8 that, diesel fuel supplied 1.4% more energy to control volume according to SOB. Figure 9 shows energetic efficiencies of engine by using diesel and SOB at 1800 rpm engine speed by comparison. Usage of diesel fuel causes more efficient conversion as indicated in the figure. 1st law efficiency of the SOB is 1.05% lower than diesel fuel. Cetane number of diesel fuel and SOB are 56 and 54, respectively. Since cetane number can be seen as an indicator of the combustion speed of diesel fuel; thanks to high cetane number, combustion occurs highly efficiently.

The exergy balances for the control volume, steady state open system can be written as:

$$
\sum \stackrel{\square}{m}_{in} \varepsilon_{in} = \sum \stackrel{\square}{m}_{out} \varepsilon_{out} + \stackrel{\square}{Ex}_{heat} + \stackrel{\square}{Ex}_{work} + \stackrel{\square}{Ex}_{dest}
$$

Input exergy rate contains only chemical exergy of the fuel. Thermomechanical exergy of fuel and total exergy of air can be neglected. Molecular formulas of sunflower oil biodiesel and diesel fuel were considered as $C_{20}H_{38}O_2$ [15] and $C_{14}H_{25}$ [16], respectively.

$$
Ex_{in} = m_{fuel} \varepsilon_{fuel}
$$

$$
\varepsilon_{\text{fuel}} = H_u \varphi
$$

\n
$$
\varphi = 1.041 + 0.1728 \frac{h}{c} + 0.0432 \frac{o}{c} + 0.2169 \frac{\alpha}{c} (1 - 2.0628 \frac{h}{c})
$$

where h, c, o, α represents mass fractions of hydrogen, carbon, oxygen, sulfur contents.

Exergy efficiency $(2nd$ law efficiency) can be found by dividing the exergy work rate to the energy input rate of the fuel.

Figure 10 demonstrates fuel exergy at a specific engine speed by comparison. Same tendency with energy input rates of fuels was attained as anticipated because the term specific flow exergy in input exergy rate includes lower calorific value of fuel. SOB has 0.7% lower input exergy rate value than diesel fuel. Figure 11 shows exergetic efficiencies of engine by using diesel and SOB at 1800 rpm engine speed by comparison. Exergy efficiency of the SOB is 1.73% lower than diesel fuel. SOB has extra oxygen content. That means obtaining higher end-combustion temperature in combustion chamber. Higher end-combustion temperature triggers the destruction of exergy. There is a reverse tendency between exergetic efficiency and exergy destruction. That's way biodiesel showed lower exergetic performance.

4. Conclusions

In this study, besides performance and emission characteristics, $1st$ law and $2nd$ law efficiencies of a compression ignition engine operated with diesel and SOB were achieved.

The main conclusions of study were presented as:

• The power and torque output of engine was decreased by 4.8% and 5.4%, respectively with use of biodiesel.

• CO emission values improved (5.7%) whereas NO_x $(4.8%)$ and CO₂ $(5.4%)$ values raised with biodiesel usage against diesel fuel.

 Diesel fuel indicated better performance than SOB in terms of the $1st$ law and $2nd$ law efficiencies.

• Energy and exergy efficiency values were reduced by 1.05% and 1.73% by use of SOB according to diesel fuel.

Destruction of exergy caused by irreversible processes (mainly by combustion) is the most important factor of system inefficiency.

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