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## A new approach for prevention the oxidations and mutations: Zinc borate

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### ABSTRACT

Zinc borate is a white crystalline powder substance with variable composition (34% B<sub>2</sub>O<sub>2</sub>, 45% ZnO and 20% H<sub>2</sub>O). It is used as a fungus and mildew inhibitor, to fire proof textiles, and for other uses. Although there are limited studies about the biologic properties of this compound. The present study was aimed to investigate the antioxidant and antimutagenic activity of zinc borate and also to figure out its cytotoxic effect. Antioxidant activity of zinc borate was determined with βcarotene linoleic acid and DPPH radical scavenging assays. The mutagenicity and antimutagenic activity was determined with AMES/ Salmonella microsomal test systems using Salmonella typhimurium TA98 and TA100 strains. The cytotoxic effect of zinc borate was evaluated by 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay using mouse fibroblast L929 cell line. When compared to the control, % inhibition values of 1 mg/mL zinc borate for DPPH radical scavenging and  $\beta$ - carotene linoleic acid tests were calculated as 13.50% and 30%, respectively. At the tested concentrations, zinc borate did not exhibit any mutagenic effect. Zinc borate exhibited low antimutagenic activity at 0.2 and 0.04 mg/plate concentrations on S. typhimurium TA98 strain while it exhibited no antimutagenic activity on S. typhimurium TA100 strain. In vitro toxicity against L929 cell line with IC  $_{_{50}}$  values of zinc borate were found to be 40.70  $\mu\text{g/mL}$  and 32.93  $\mu\text{g/}$ mL for 24h and 72h, respectively. In spite of the low DPPH free radical scavenging activity, zinc borate has found to have moderate total antioxidant activity besides low antimutagenic potential. Zinc borate has a potential to use for medical purpose with its antioxidative and antimutagenic properties.

#### 1. Introduction

Zinc is an essential micronutrient for the human body [1,2]. It is important for human health and diseases as it plays a significant role in the immune system function, central nervous system, growth and development, bone metabolism and wound healing [3]. Zn is also a cofactor for many metalloenzymes necessary for cell membrane, cell repair, growth and proliferation. Deficiency in zinc can cause health complications, excessive intake can also lead to negative side effects. Pathological manifestations of zinc deficiency include the impaired immune function, growth retardation, formation of skin lesions and impaired wound healing [4].

Boron is a required element for plants, and recent studies on the biological importance of boron to various hormonal, nutritional, metabolic and physiological processes shown that is also probably necessary for animals and humans [5-8]. Boron plays a crucial role in embryogenesis, psychomotor skills, immune functions, hormone and lipid metabolism [9], vitamin D3 deficiency [10], wound healing [11], and healthy bone/ teeth growth and maintenance [12].

Zinc borate is an inorganic compound in the form of white crystalline powder. Important properties of zinc borate contain comparatively low water solubility and a high dehydration temperature. The second feature is that it allows processing on a wide variety of polymer systems [13].

There are some variants of zinc borates, differing by the zinc/boron ratio and the water content. Many of which can find significant industrial use [13]. For example, it is largely used as a polymer additive. It is also used as flame retardant and preservative in plastic, rubbers, cellulose fibres, paper and textiles. As a polymer additive, it is a fire retardant synergist, charpromoting agent, antidrip agent, smoke and last radiation suppressor, and serves as a modifier of electrical and optical properties [14,15].

Today's most significant commercial zinc borate was produced 40 years ago and currently the annual production in the world exceeds 10 000 metric tons [13]. Zinc borate as  $2ZnO.3B_2O_3.3,5H_2O$  (hexaboron dizinc undecaoxide) was developed by the field experts at Eti Maden. It can be used all over life because it can be combined with safer, long lasting, durable and economical products in our living space [14].

The living cells are often exposed to potentially damaging free radical species, which origin may be both extra- and intracellular. The cellular targets of reactive oxygen species (ROS) include DNA, lipids and proteins; and they are highly mutagenic/genotoxic and damage cellular macromolecules [16]. Oxidative stress caused by ROS may be the primary cause of various degenerative diseases (e.g., cardiovascular, cancer and neurodegenerative) [17,18]. Antioxidants react with ROS to remove radicals and produce less aggressive chemical species [19]. It is also known that antimutagens play a significant role in reducing the damage caused by oxidants [20].

The present study was aimed to determine the antioxidant and antimutagenic activities of zinc borate and also to figure out its cytotoxic effect. Antioxidant activity of zinc borate was determined with DPPH radical scavenging activity and  $\beta$ - carotene linoleic acid assays. The mutagenicity and antimutagenic activity was determined with AMES/*Salmonella* microsomal test systems using *Salmonella typhimurium* strains (TA98 and TA100). The cytotoxic effect of zinc borate was evaluated by MTT assay using mouse fibroblast L929 cell line.

## 2. Materials and methods

## 2.1. Antioxidant activities

The antioxidant activity of zinc borate (Sigma Aldrich)( $B_2O_6Zn_3$ , MA: 313.758 g/mol) using the  $\beta$ -carotene/linoleic acid assay was assayed by the method of Rauter et al. [21]. In brief, 1 mL of b-carotene (0.5 mg/mL in chloroform) was mixed with 25  $\mu$ L of linoleic acid and 200 mg of Tween 60. The mixture was shaken and

evaporated to remove chloroform. Then, 100 mL of oxygenated distilled water was added to the mixture and agitated. From this emulsion, 2.5 mL transferred into different test tubes containing 0.5 mL of the zinc borate. The initial absorbance of samples was measured after 1 min of vortexing at 470 nm. Samples were incubated for 60 min at 50°C, and the second absorbance was measured at 470 nm after 1 min of vortexing. The measurement was carried out at 30 min intervals. Deionised water and ascorbic acid were used as a blank and positive controls, respectively. Antioxidant activity of zinc borate was calculated using the following formulation:

 $\ln (Abs) = \ln (Abs_0) + R \times t, \tag{1}$ 

where R, the bleaching rate

AA (%) = [(
$$R_{\text{blank}} - R_{\text{sample}}$$
)/ $R_{\text{blank}}$ ] × 100, (2)

where  $R_{\rm blank}$  and  $R_{\rm sample}$  are the oxidation rates of negative control and sample containing the zinc borate, respectively.

DPPH radical scavenging activity of the zinc borate was detected by the method of Ebrahimabadi et al. [22] with minor modifications [23]. One mL DPPH (0.2 mM in 95% ethanol) was mixed with 1 mL of the zinc borate (1 mg/ml) of various concentrations in 95% ethanol. After vortexing, the tubes were left in the dark for 30 min at room temperature, after which the absorbance was measured against a blank at 517 nm. Deionised water and ascorbic acid were used as a blank and positive controls, respectively.

DPPH radical scavenging activity was calculated as follows:

Inhibition (%)= [(OD<sub>blank</sub>-OD<sub>sample</sub>)/OD<sub>blank</sub>]x100 (3)

OD<sub>blank</sub>: Absorbance of the blank

OD<sub>sample</sub>: Absorbance of the sample

Each experimental group was repeated 3 times.

#### 2.2. Mutagenicity and antimutagenic activity

Test bacterial strains were examined primarily for their genetic integrity and spontaneous mutation rates. Then cytotoxic doses of the zinc borate were detected according to Mortelmans and Zeiger [24].

The mutagenicity of zinc borate at the subcytotoxic doses on the test bacteria, was made using the plate/ incorporation method [25]. The positive controls used 4-nitro-o-phenylenediamine (4-NPD) for the *S. ty-phimurium* TA98 strain and sodium azide (NaN<sub>3</sub>) for the *S. typhimurium* TA100 strain. The plates were incubated for 48 h at 37°C and His<sup>+</sup> revertant colonies were counted. The antimutagenicity against 4-NPD and NaN<sub>3</sub> was assayed by incubating without and with zinc borate at the subcytotoxic doses on the test bacteria (200 and 40  $\mu$ g/plate for TA 98, 500, 100 and 20  $\mu$ g/plate for TA 100), using *S. typhimurium* TA98 or TA100. Ethanol/ water (1:1, v/v) was used to determine spontaneous reversion. The plates were incubated for 72 h at 37°C and the His<sup>+</sup> revertants were counted. For the antimutagenicity assays, the % inhibition was calculated as follows:

% inhibition= [1 - (T - S)/(M - S)]x100 (4)

T is the number of revertant colonies in the presence of mutagen and the zinc borate, M is the number of revertant colonies induced by the mutagen and, S is the number of spontaneous revertants. To monitor the mutagenicity and antimutagenicity, all doses were studied in three replicates and four parallels.

## 2.3. Cytotoxicity

Cytotoxicity evaluation of the zinc borate on L929 fibroblast cells were screened using MTT colorimetric assay and doneby Anadolu University Plant, Drug and Scientific Research Center. Briefly, cells was cultured in Dulbecco's Modified Eagle's Medium (DMEM)-high glucose, supplemented with 10% heat-inactivated Fetal Calf Serum (FCS), L-glutamine (2 mM), antibiotic-antimycotic solution (10.000-unit penicillin, 10 mg streptomycin and 25 µg amphotericin B per mL). Cells were maintained at 37°C, in a humidified atmosphere of 5% CO<sub>2</sub>. After being detached from culture flasks with 0.05% trypsin solution, cells were passaged to 96-well microtiter tissue culture plates (10.000 cells per well), and incubated at 37°C, in a humidified atmosphere of 5% CO<sub>2</sub> for 24 h. After then 76.19, 19.05, 4.76, 1.19 and 0.3 µg/mL of zinc borate were added to the wells and incubated for 24 and 72 h at the same conditions, after which 20 µl of MTT (5 mg/mL, prepared in phosphate-buffered saline) was added to each well and incubated for additional 3 h. The medium containing MTT was then poured off and 100  $\mu$ L of DMSO was used to solubilize the formed formazan crystals in each well. Plates were put in an orbital shaker for 15 minutes at orbital shaker and the absorbance was measured at 540 nm using microplate reader (Thermo Scientific Multiskan FC, Thermo Fischer, Vantaa, Finland). Each experimental group was repeated 3 times with six replicates.

The % cell inhibition was determined using the following formula and the graph between % cell inhibition and concentration were plotted, from which  $IC_{_{50}}$  was calculated:

(%) =  $[100 \times (\text{Sample}_{abs})/(\text{Control}_{abs})]$  (5)

### 3. Results and discussion

The antioxidative potential of zinc borate at tested concentration are shown at Table 1. Zinc borate exhibit low DPPH free radical scavenging activity at 1 mg/mL concentration while at the same concentration it exhibited moderate total antioxidant activity, respectively.

It is known that ROS play a bi-directional act in biological systems, since they can be beneficial or detrimental to living systems [26]. The detrimental impacts of ROS are offset by the activity of antioxidant enzymes as well as non-enzymatic antioxidants [27]. Although the cell's antioxidant defense system opposes oxidative damage, it has been suggested that oxidative damage accumulates throughout the life cycle and radical damage to DNA, proteins and lipids play a key role in age-dependent diseases development such as arthritis, neurodegenerative disorders, cancer, arteriosclerosis and other conditions [28].

Boron was found to have potential effects against genotoxicity and lipid peroxidation by increasing antioxidant defence mechanism [29] and it reduces oxidative stress caused by malathion and acetylcholinesterase inhibition [30]. Boron was also reported to inhibit the molybdenum containing proteins, proliferating cell nuclear antigen index and improve oxidative stress in hepatocellular carcinoma [31]. Prasad [32] reported that zinc supplementation to elderly individuals reduced the incidence of infections, plasma oxidative stress markers and inflammatory cytokine formation, and reported that plasma zinc levels increased.

According to the mutagenicity assay, zinc borate didn't exhibit any mutagenic efficiency at the tested concentrations, including the  $IC_{50}$  doses on the L929 fibroblast cells, made with *S. typhimurium* TA98 and TA100 (data not shown).

The effects on 4-NPD and  $NaN_3$  mutagens were used to determine the antimutagenic property of zinc borate. The results were estimated using the standard

Commis	Concentration —	% inhibition		
Sample		DPPH <sup>a</sup>	β-carotene-linoleic acid <sup>6</sup>	
Zinc borate	1 mg	13.5±0.034	30±0.024	
Ascorbic acid	10 mg	83±0.062	66.4±0.054	
	1 mg	72±0.076	48±0.044	

Table 1. The antioxidant activity results of zinc borate

		Number of revertant colonies			
Test items	Concentration (µg/plate)	TA98		TA100	
		Mean ± S.D.	Inhibition %	Mean ± S.D.	Inhibition %
Negative control		21.66±3.05 <sup>c</sup>		199.5±16.2 <sup>b</sup>	
Positive control					
4-NPD <sup>a</sup>	3.0	636±49.75			
NaN <sub>3</sub> <sup>b</sup>	8.0			1686.66±172.4	
	500	NT		1444±108.29	14.38
Zinc borate	200	505.66±46.5	20.49	NT	
	100	NT		1470±155.56	12.84
	40	499.5±72.8	21.46	NT	
	20	NT		1513.33±90.18	10.27

Table 2. The antimutagenicity results of the zinc borate for S. typhimurium TA98 and TA100 strains.

<sup>a</sup>4-NPD: positive control for *S. typhimurium* TA98, <sup>b</sup>NaN<sub>3</sub>: positive control for *S. typhimurium* TA100, <sup>c</sup>Values expressed are means ± S.D. of three replications. Regression analysis for mutagenicity inhibition (%) was performed using Microsoft Excel. NT-not tested

plate incorporation method and summarized in Table 2. The zinc borate showed low antimutagenic effects against 4-NPD at 0.2 and 0.04 mg/plate concentrations. Whereas, zinc borate didnt have any antimutagenic efficiency against NaN<sub>3</sub>, as all values were below the limit of 20%.

Considering that mutations are an early stage of carcinogenesis, short-term tests such as *Salmonellal* reversion testing can be said to be very important in the early identification of mutagens/carcinogens as well as antimutagens/anticarcinogens [33]. Therefore, prior identification of substances that can induce mutations has happen an considerable process in safety evaluation. Chemicals that can cause/induce mutations can potentially harm the reproductive tract that leads to mutations and reproductive problems in future generations. Mutagenic chemicals can also induce cancer, and this concern has led most of the mutagenic testing programs [24]. The *Salmonella* mutagenicity test is conceived to define chemically induced mutagenesis [34]. In this study, Ames *Salmonella*/microsome mutagenicity test was used to define the mutagenic and antimutagenic potential of zinc borate. The current research focuses on the development of efforts to diminish the risk of cancer through the use of chemoprophylactic and chemopreventive compounds, as cancer prevalence rates rise in the worldwide [35].

In addition, the effect of zinc borate on cell viability of L929 fibroblast cells are given at Figure 1. According to the MTT analysis results applied at 24 h and 72 h, the IC<sub>50</sub> values of zinc borate were found to be 40.70  $\mu$ g/mL and 32.93  $\mu$ g/mL, respectively (Table 3). According to the data obtained from MTT test, there was no significant difference between zinc borate treated groups and non-zinc borate treated groups (p<0.05).

It is known that boron is significant for animal cell development and replication, but the basic mechanism remains unclear [36]. Studies in the literature show that boron compounds have no mutagenic or carcinogenic activity on cell and living organisms. These



Figure 1. The effect of zinc borate on cell viability of L929 fibroblast cells (n=6).

Table 3. The IC<sub>50</sub> values of zinc borate on L929 fibroblast cells (n= 6).

Material	24 h	R <sup>2</sup>	72 h	R <sup>2</sup>
Zinc borate (µg/ml)	40.70 0.98		32.93	0.96
Zinc borate (µM)	129.7	0.90	104.9	0.90

studies have been conducted with well-known boron compounds such as borax, boric acid or sodium tetraborate [37]. Studies have shown that boric acid is not genotoxic in human blood cultures and supports antioxidant enzyme activities [38]. In a study done by the same researchers, the hepatoprotective effects of boric acid were investigated [39]. As the effects on malignancies have recently emerged, attention to boron compounds is increasing.

It is possible to find studies done using different boron compounds and different cell groups in the literature. In a study, sodium tetraborate and boric acid at 1.000  $\mu$ M concentration applied to healthy human lymphocytes by MTT methods caused a 25% and 20% reduction in mitochondrial activity, respectively. It was found that the same concentration of sodium tetraborate and boric acid resulted in a 20% and 50% reduction in mitochondrial activity of human leukemia cells, respectively. Especially, the cell viability of human leukemia cells was significantly decreased at a concentration of 100  $\mu$ M for an incubation period of 48 h [40]. Similarly, in our study, the cell viability of L929 fibroblast cells was 50% reduced at 129.7  $\mu$ M concentrations after 24 h incubation.

In another study the 400 mg/L concentrations of some synthesized borenium compounds were non-cytotoxic on cultured human blood cells [41]. Similarly, in a study reveals that both synthesized hexagonal boron nitrides and boric acid are not cytotoxic at lower than 22 µg/mL boron containing concentrations on brain neuronal mHippo E-14 cells [42]. Administration of <0.5 mmol/L boron significantly promoted the viability of Sertoli cells (P<0.01); however, the exposure to high dose (> 10 mmol/L) of boron exhibited significant adverse effects [43]. Deshayes et al. [44] observed cytotoxic effects on B16F10 murine melanoma cells of phenyl boronic acid nanoparticles (IC<sub>50</sub>:186 µM). In our study, the IC<sub>50</sub> values of zinc borate on L929 fibroblast was determined as 129.7 µM after 24 h incubation and 104.9  $\mu$ M after 72 h. The IC<sub>50</sub> values of zinc borate is lower than the boronic acid nanoparticles which given above study. However, there is no article that considers the evaluation of zinc borate on fibroblast cells.

### 4. Conclusions

The study was performed to evaluate the antioxidant and antimutagenic activity, and the mutagenic and cytotoxic potential of zinc borate. Zinc borate found to have moderate total antioxidant activity. At the tested concentrations, zinc borate didn't exhibit any mutagenic effect, including the  $IC_{50}$  values on L929 fibroblast cell lines. Zinc borate exhibited low antimutagenic effect on *S. typhimurium* TA98 strain which was induced with 4-NPD whereas not founded on TA100 strain. *In vitro* toxicity against L929 cell line with  $IC_{50}$  values of zinc borate were found to be 40.70 µg/mL and 32.93 µg/mL for 24h and 72h, respectively. As a result of this study, zinc borate which was found to have no genotoxic effect and cytotoxicity at the tested concentrations, it has a potential for medical applications with its antioxidative and antimutagenic properties.

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