



COMPARISON OF CHEMICAL AND ANTIOXIDANT PROPERTIES OF MENTHA PIPERITA L., SALVIA OFFICINALIS L., ROSMARINUS OFFICINALIS L. AND LAVANDULA ANGUSTIFOLIA L.

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Abstract: 4 different plant species belonging to the same family mint (*Mentha piperita*), sage (*Salvia officinalis*), rosemary (*Rosmarinus officinalis*), lavender (*Lavandula angustifolia*) were harvested from the Afyon region. Total antioxidant and phenolic contents were analyzed by DPPH and Folin-Ciocalteu method. The effect of extra virgin olive oils oxidation stability was tested by means of rancimat method. According to DPPH analysis results, *M. piperita* is the highest antioxidant capacity and the radical scavenging activity is 95.31%. The radical scavenging activities of *S. officinalis*, *R. officinalis* and *L. angustifolia* are respectively 91.83%, 54.28%, and 18.85%. The total phenolic content of *M. piperita*, *S. officinalis*, *R. officinalis* and *L. angustifolia* plants as gallic acid were respectively; 0.32, 0.28; 0.26 and 0.18 mg/100ml. According to the results of the study, the order of the plants did not change in terms of phenolic and antioxidant content, *M. piperita* has the highest values and *L. angustifolia* has the lowest. It was also found that the plants are effective on the oxidation stability of extra virgin olive oil and prevent oxidation in the order of large to small *R. officinalis*, *M. piperita*, *S. officinalis* and *L. angustifolia*.

Keywords: *Mentha piperita* L, *Salvia officinalis* L, *Rosmarinus officinalis* L, *Lavandula angustifolia* L, Phenolic content, Antioxidant

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1. Introduction

Mentha piperita is one of the most important, most common and aromatic herbs. It belongs to the Lamiaceae family (Tsai et al., 2013). The *M. piperita* has antibacterial, antifungal and antioxidant properties, and its essential oil include menthol (36.02%), menthone (24.56%), menthyl acetate (8.95%), and menthofuran (6.88%) (Singh et al., 2015; Ilboudo et al., 2016; Desam et al., 2019) Some essential components and phenolic compounds of *M. piperita* leaves are caffeic acid, rosmarinic acid, cinnamic acid, eryositrin, luteolin-7-O-glucoside, and represent about 20% weight of dry matter (Dorman et al., 2009; Farnad et al., 2014). *S. officinalis* is a plant in the Lamiaceae family. Grows in the Middle East and Mediterranean regions. In folk medicine, *S. officinalis* has been used for the treatment of various disorders such as seizures, ulcers, gout, rheumatism, inflammation, dizziness, tremor, stroke, diarrhea and hyperglycemia. The essential components of essential oil are borneol, camphor, caryophyllene, cineol, sieve, humulene, leden, pinene and thujone. The alcoholic and aqueous extracts of *S. officinalis* are rich flavonoids, especially rosmarinic acid and luteolin-7-glucoside. In addition, phenolic acids such as caffeic acid and 3-Caffeoylquinic acid were found

in the methanolic extract of *S. officinalis*. Various flavonoids such as chlorogenic acid, ellagic acid, epicatechin, epigallocatechin gallate, quercetin, rosmarinic acid, routine and luteolin-7-glucoside, as well as volatile components such as borneol, cineole, camphor and thujone have been described (Ghorbani and Esmaeilzadeh, 2017). *Rosmarinus officinalis* is a widely consumed aromatic plant of the Lamiaceae family. Fresh and dried leaves are often used in traditional Mediterranean cuisine and folk medicine. *R. officinalis* is known to have antioxidant, antibacterial and antifungal, anti-cancer, anti-inflammatory effects. Therefore, it has a wide range of industrial applications such as food and food packaging, pharmaceutical, perfumery and cosmetic industries (Ribeiro-Santos et al., 2015). *Lavandula angustifolia* is a perennial plant of the Lamiaceae family. Its origin is Southern European and Mediterranean and commercially grown in many countries. It has antifungal antibacterial and antioxidant properties (Blažeković et al., 2018). *L. angustifolia* is also a very popular aromatic plant and is widely used in perfumes, cosmetics and medicines (Tang et al., 2017). Borneol (22.4%), epi- α -muurolol (13.4%), α -bisabolol (13.1%), precocene I (13.0%), and eucalyptol (7.9%) are the major essential



oil constituents (Mantovani et al., 2013). Lavandunat, lavandufurandiol, lavandufluoren lavandupyrones A and B and five known compounds 4- (1-hydroxy-1-methylethyl) benzoic acid, methyl 3- (3,4-dihydroxyphenyl) propanoate, 3,4, α -trihydroxyl-ethyl phenylpropionate, rosmarinic acid and isosalvianolic acid are primary phenolic compounds (Yadikar et al., 2018).

Nowadays, since the safety of synthetic antioxidants is questioned in edible oils, the addition of natural antioxidants is recommended. This is one of the aims of our study and to determine whether the plants in question will be used for this purpose. In addition, the antioxidant and phenolic content of these plants is compared with each other. There are studies on these plants in the literature and there is no comparison under the same experimental conditions. This is one of the factors that make up the originality of our study. The antioxidant test of these plants in edible oils by Rancimat method is another of our original values.

2. Material and Methods

2.1. Plant Extraction

Plants harvested from Afyonkarahisar Medicinal and Aromatic Plants Center in 2018. All analyses were done at the same year. The plants were dried at 37°C for 72 hours and then prepared for analysis. 0.1 g plant samples were kept in 20 ml of methyl alcohol for 2 hours and the solution was filtered through ordinary filter paper. The resulting solutions were used for DPPH and total phenolic analysis. Total antioxidant and total phenolic analyzes were performed in three replicates and the average of the experimental results were given.

2.2. Detection of Free Radical Scavenging Activity

Total antioxidant test DPPH method was used with some modifications of antioxidant determination of *M. piperita*, *S. officinalis*, *R. officinalis* and *L. angustifolia* (Blois, 1958; Brand-Williams et al., 1995). Accordingly, 1 ml of plant extracts were mixed with 1 ml of 0.002 g / 100 ml DPPH (in methanol). The resulting solution incubated in dark for 30 minute and the absorbance of each sample mixture was measured at 517 nm. 1 ml of DPPH solution with 1 ml of methyl alcohol was mixed and read at 517 nm at 0 minutes and recorded as a control. Methyl alcohol solution was evaluated as a blank sample. The percentage of Radical scavenging activity was calculated using this formula % Radical scavenging activity = (Absorbance control-Absorbance sample) x100 / Absorbance control equation.

2.3. Total Phenolic Content Analysis

The total phenol content in *M. piperita*, *S. officinalis*, *R. officinalis* and *L. angustifolia* was calculated according to the Folin-Ciocalteu method (Singleton et al., 1965). Five different concentrations of 0.0312-0.125-0.25-0.5-1mg / ml of gallic acid were prepared with 99.9% methanol. 20 μ l of this solution was taken and 680 μ l of distilled water, 400 μ l of 0.5 N folin reagent (in water), 400 μ l of 10% Na₂CO₃ (in water) were added and absorbance at 760nm wavelength were read. In addition, 700 μ l distilled water,

400 μ l 0.5 N folin reagent, 400 μ l 10% Na₂CO₃ was mixed as the blank sample and the absorbance values from the blank sample was reset for samples and standard samples. From the data, graph of absorption versus concentration was obtained which is regression coefficient 0.95. 20 μ l of the extracted plant solution mixed with 680 μ l of distilled water, 400 μ l of 0.5 N folin reagent, 400 μ l of 10% Na₂CO₃ and the absorbance values were measured at 760 nm after 30 minutes. In the obtained equation, the absorbance values replaced by y and the x values were calculated. When the x multiply by the dilution factor phenolic equivalents in terms of gallic acid was calculated.

2.4. Rancimat Analysis

Rancimat analyzes were performed according to the standard rancimat method. In this method, the temperature is 120 ° C, the sample amount is 3 g, the air flow is 20 L / h, and the water amount is 60 mL. 0.055 μ s ultra-pure water was used in the experiments. Extra virgin olive oil produced in 2018 was used as oil. Since there is no standard for these plants for use as an antioxidant in edible oils, they were added to olive oil in 2% ratio and kept for 24 hours. Rancimat analysis results were compared in its pure form. The experiments were performed in two replicates and the average of the test results were given.

3. Results

3.1. % Radical Scavenging Activity

Data obtained as a result of DPPH experiments were given in Table 1. When the radical scavenging activity values of the plants are examined, it was observed that *M. piperita* was the highest with 95.31% radical scavenging activity and it was followed by *S. officinalis* with 91.83%. In terms of antioxidant content, the lowest *L. angustifolia* was found to be 18.85% under the same conditions. The radical scavenging activity of *R. officinalis* was tested as 54.28%.

3.2. Phenolic Content

The absorption concentration graph of the gallic acid standard formed according to the folin-ciocalteu method was given in Figure 1. The phenolic equivalents of the plants in terms of gallic acid were given in Table 2.

Table 1. % Radical scavenging activities of plant materials

Plant materials name	% Radical scavenging activity
<i>M. piperita</i>	95.31
<i>S. officinalis</i>	91.83
<i>R. officinalis</i>	54.28
<i>L. angustifolia</i>	18.85

Table 2. Phenolic content of plants in terms of gallic acid (mg / 100ml)

Plant materials name	The phenolic equivalents of the plants in terms of gallic acid mg/100ml
<i>M. piperita</i>	0.32
<i>S. officinalis</i>	0.28
<i>R. officinalis</i>	0.26
<i>L. angustifolia</i>	0.18

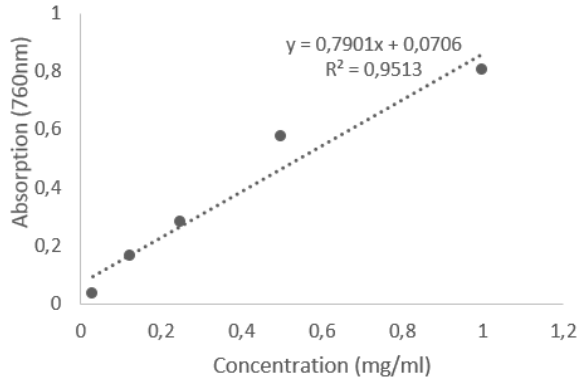


Figure 1. Absorption- concentration graph of five different concentrations of gallic acid methanol solution at 760 nm.

3.3. Rancimat Results

In Table 3, the induction periods obtained as a result of rancimat analysis were given.

From the data, adding these plants to virgin olive oils increased oxidation stability of virgin olive oil. While virgin olives induction period was 5.68, *L. angustifolia* added virgin olive oil was 5.90, *R. officinalis* added virgin olive oil was 6.63, *S. officinalis* added virgin olive oil was 6.07, and *M. piperita* added virgin olive oil was 6.43.

Table 3. Induction periods of olive oil and olive oil with 2% plant additive obtained by rancimat device

Plant materials name	Average
<i>M. piperita</i>	6.43±0.07
<i>S. officinalis</i>	6.07±0.01
<i>R. officinalis</i>	6.63±0.14
<i>L. angustifolia</i>	5.90±0.00
Virgin Olive Oil	5.68±0.16

4. Discussion

In this study, these four different plants were evaluated comparatively in terms of % radical scavenging activity, phenolic analysis, and effects of olive oil on oxidation stability. *M. piperita*'s % radical scavenging activity was 95.31%, *S. officinalis*'s 91.83%, *R. officinalis*'s 54.28%, and *L. angustifolia*'s 18.85%. *M. piperita* is the highest with 0.32 mg/100ml in terms of gallic acid. The *M. piperita* plant was followed by *S. officinalis* 0.28 *R. officinalis* 0.26 and *L. angustifolia* 0.18 mg/100ml, respectively. The total phenolic content of the *S.*

officinalis leaves was calculated as 0.324g 100g⁻¹ in terms of caffeic acid (Türk Baydir et al., 2021). In a scientific study, it was observed that *S. officinalis* was higher in terms of antioxidant and total phenolic content compared to *R. officinalis* (Armatu et al., 2010). According to the results of another study, *S. officinalis* is rich in myricetin and p-coumaric acid, while *R. officinalis* is rich in components such as rutin catechin quercetin (Bianchin et al., 2020). Extra virgin olive oil deteriorates at 120 °C in about 5.68 hours, 5.90 hours when 2% lavender was added, 6.07 hours when sage was added in, 6.63 hours when rosemary was added, 6.43 hours when mint was added. As a result, it was concluded that these plants can be used as anti-oxidation or retardant of extra virgin olive oil oxidation. The effect of *L. angustifolia* on oxidation stability of olive oil was the lowest as in all analyzes and the order of other plants had changed. In a study, it was determined with the help of the rancimat device that sage prevented oxidation in sunflower oil more than rosemary. This situation contradicts with the results of our study and we attribute this situation to regional factors (Upadhyay and Mishra, 2014). The toxicity of these plants was not emphasized in this study. Considering this situation, it is recommended to establish a standard and use it as a natural antioxidant in extra virgin olive oil. In addition, the effects of other cooking oils on oxidation stability can also be studied.

Author Contributions

All tasks have been done by the single author. The author reviewed and approved the manuscript.

Conflict of Interest

The author declared that there is no conflict of interest.

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