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# Investigation of the Effects of Alcohol-Based Fuel Additive on Engine Performance and Exhaust Emissions in a Spark-Ignition Gasoline Engine

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#### Abstract

Although fossil fuel reserves have been decreasing day by day, the high carbon emission of fossil-based vehicles and the fact that these carbon emissions trigger global warming caus researchers to focus on alternative fuels. CO and HC, which form the exhaust emissions of motor vehicles, called unburned fuel, have been, threatened human health. Efforts to reduc  $CO_2$ , which triggers global warming, and  $NO_3$  emissions, which turn into acid rain, are increased asing day by day. It is tried to reduce harmful exhaust emissions by mixing ethanol, metha nol, propanol, butanol and their derivatives fuels as alcohol-based fuels in motor fuels in cer tain proportions. The aim of this study is to investigate the usability of waste fusel oil an plant-based methanol as an alternative fuel in internal combustion engines without the nee for any auxiliary system and investigate the effects of fusel oil and methanol fuels added t gasoline on engine performance and emissions. In the experimental study, as engine perfor mance, it was determined that the engine torque decreased by 11.62% in F20 fuel and 6.45% in M20 fuel compared to G fuel. While an increase in BSFC values was detected in both tes fuels, there was a decrease in EGT values. According to the emission results, an average dec rease of 7.31% in F20 fuel and 13.40% in M20 fuel was detected in CO values compared t G fuel, while an increase in CO<sub>2</sub> and NO<sub>x</sub> values occurred.

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## 1. Introduction

Many researches have been carried out around the world for suitable alternative fuel potentials that can replace fossil-based fuel resources with limited reserves in the future. In addition to meeting the fossil-based fuel characteristics of alternative fuels, pollutant exhaust emission values are expected to be low. Studies have shown that alcohol-based fuels give values close to fossil fuel properties and pollutant exhaust emissions are lower. In addition to the use of pure alcohol-based fuels, research continues within the framework of mixtures applied into fossil fuels at certain rates. Motor vehicle emissions have been a major risk in the last few decades in terms of global warming, noise and environmental pollution. In addition, although it is clearly understood that the exhaust emissions of motor vehicles, which are increasing day by day, threaten human health, it is almost impossible to reduce the number of vehicles due to their importance in daily life. Motor vehicle manufacturers are obliged to produce their vehicles within the scope of emission restrictions within the framework of international agreements. In many state management policies, there are strategic management policies to reduce harmful emission values and reduce the carbon footprints of countries. In this context, emission reduction studies are becoming more important day by day and lead researchers to search for interesting alternatives [1,2]. Fusel oil is the waste liquid left over from the production of ethyl alcohol obtained from molasses formed after sugar beet production. Fusel oil contains many different types of alcohol. [3,4,5]. Fusel oil blended test fuels, which were put in 10%, 20% and 30% gasoline by volume, were tested in a gasoline engine and the highest torque increase was obtained in 30% fusel blended fuel. Specific fuel consumption at all engine speeds increased depending on the amount of fusel oil additive. The highest increase was obtained with F30 fuel as 7.7%. Depending on the amount of fusel oil in the mixtures, nitrogen oxide (NO<sub>x</sub>) emissions decreased while hydrocarbon (HC) and carbon monoxide (CO) emissions have been increased [6]. In the experimental study of Eyidogan et al., on the addition of ethanol and methanol to gasoline, it was determined that ethanol-gasoline and methanol-gasoline mixtures (E5, E10, M5 and M10) brake specific fuel consumption increased compared to



gasoline, and there was a general decrease in CO, HC, CO<sub>2</sub> and NO<sub>x</sub> emissions with the use of alcohol mixtures. Similar studies have been done also by Danaiah et al. [7,8]. Calam et al. compared the engine performance and emission characteristics of mixtures of Ethanol (E25), Methanol (M25), Fusel oil (F25), Naphtha (N25), isopropanol (IP25), Butanol (B25) and nheptane in an HCCI engine. In this study, it was determined that CO and HC values increased due to the amount of water contained in the fusel oil. The amount of water in fusel oil affects combustion and emissions. Purifying the amount of water in the fusel oil with sodium chloride improves the test results. Ethanol-Methanol fuel mixtures have been seen to increase engine performances (volumetric efficiency, torque and braking power) in general, and reduce pollutant emissions (carbon monoxide (CO) and unburned hydrocarbons (UHC)) [9, 10,5]. Methanol fuel is an alcohol-based fuel that can be mixed with gasoline at a lower mixing ratio and provides better performance in spark ignition engines. However, methanol poses problems, especially with regard to its energy content and vapor lock properties. Focusing on this subject, Sharudin et al. investigated engine performance and emission results by mixing 5-15% isobutanol in methanol-gasoline mixtures (M5). It has been seen in an experimental study that M5B15 provides improvement in engine braking power, BTE (brake thermal efficiency) and EGT (exhaust gas temperature) compared to other blended fuels [11]. It has been revealed that fusel oil increases the octane number of the fuel, volumetric efficiency and engine torque in mixtures with gasoline, reduces knocking and nitrogen oxides  $(NO_x)$ , and affects engine performance depending on the amount of water it contains. There are also experimental studies done by mixing fusel oil into diesel fuel [12,13]. Şimşek et al., in their experimental study, performed experiments at constant speed and different ratios for six different fuel mixture ratios (F0, F10, F20, F30, F40 and F50) and compared with the original operating parameters of a gasoline engine. According to the test results, optimum engine performance was obtained with 9.12/1 compression ratio and F30 fuel mixture. It was found that by increasing the compression ratio from 8/1 to 9-12/1 for the F30 fuel mixture, the overall efficiency increased by 6.91% and the specific fuel consumption decreased by 2.35%. In the effects of fusel oil on emissions, it was determined that CO emissions increased by 36.82%, HC emissions by 23.07%, NO<sub>x</sub> emissions by 15.42%, and CO<sub>2</sub> emissions by 13.88% [14]. Ehtasham et al., In their experimental study, they tested 0%, 3%, 6%, 9%, 12% methanol-gasoline mixtures in a four-stroke, single-cylinder SI engine and demonstrated their artificial neural network (ANN) assisted performance and emission analysis. In the study, it was observed that the continuous addition of methanol up to 12% (M12) increased engine performance. However, a decrease was observed in emissions other than NO<sub>x</sub> emissions [15]. In his experimental study, Calam investigated the effects of fusel oil on combustion, performance and exhaust emissions in an HCCI engine at different lambda and inlet temperatures. In the use of prepared test fuels (n-heptane, F20, F40 and F60),

it has been determined that fusel oil not having economic benefit, improves engine performance and emissions in HCCI engines under appropriate conditions. Similarly, Solmaz tested it in an HCCI engine by blending 37.5% fusel oil and 62.5% nheptane (Fusel40) in his experimental study. It was determined that fusel oil had a better effect in reducing CO and HC emissions than increasing the compression ratio [16,17]. In his experimental study, Ozer, by mixing secondary fatty acids with 20% by volume of diesel fuel and adding alcohol (ethanol, methanol, isopropyl, n-butanol and fusel oil) at the same rate, on a single-cylinder, direct-injection diesel engine, combustion, fuel line pressure. investigated its effects on engine performance and exhaust emissions. According to the results of the study, it was determined that all fuel mixtures provided a decrease in CO and smoke opacity emissions and an increase in HC and NO<sub>x</sub> emissions. On the other hand, Fusel oil and high oil waste costs are very low compared to diesel fuel. The use of fusel oil as a fuel also provides an important advantage [18]. Balki et al. obtained M5, M10, M15 and M20 fuels by mixing methanol at 5%, 10%, 15% and 20% by volume, respectively. These test fuels were tested in a single-cylinder gasoline engine at two different loads (10 and 20 Nm) and at different compression ratios (CR), (7:1, 8:1, 9:1). When exhaust emissions are evaluated, it has been found that M20 is more suitable than gasoline when used in a small SI engine at 9:1 CR. [19]. Yaman et al., in their experimental study, n-pentanol was mixed in gasoline at 5%, 10%, 15% and 20% by volume, and tests were carried out on a single-cylinder gasoline engine, at a constant speed of 1600 rpm and variable loads between 1-5 kW, at 1kW intervals. Findings from tests showed that infusion of n-pentanol into gasoline reduced HC, CO, CO<sub>2</sub> and NO<sub>x</sub> emissions compared to basic gasoline, but higher O<sub>2</sub> levels [20]. Örs et al. mixed 10% methanol and 10% ethanol into gasoline and compared it in terms of engine performance and emissions. In the study, the addition of methanol and ethanol showed similar combustion properties. Addition of methanol and ethanol to gasoline increased BSFC' Exhaust gas temperature (EGT), CO<sub>2</sub>, NO<sub>x</sub> and decreased CO and HC emissions. It has been stated that methanol and ethanol can be mixed with gasoline in certain proportions [21]. In the current experimental study by Puricelli et al, the Euro 6d-TEMP GDI passenger car was tested with four different fuels, both in the laboratory and on the road. Prepared test fuels Fuel B (bioethanol and bionaphtha), Fuel C (a high level of renewable ethers ) and Fuel D (methanol and bioethanol) gasoline fuel were compared with Fuel A. An increase in  $NO_x$  amounts (+48.1%) was detected in methanol blended fuel [22]. When the literature studies are reviewed, the interest in alcohol blended fuels is high.

In this study, in parallel with the literature studies, alcoholbased fuel mixtures were tested in a gasoline engine and the results were presented. Different from the literature studies, tests were carried out at different loads and speeds at the same mixing ratios. Test fuels prepared in experimental studies were mixed directly into gasoline by volume 20% methanol and



20% fusel oil without the need for any auxiliary system to the engine, and engine performance and emission values were compared with gasoline, which is a fossil-based fuel. 20% methanol and 20% fusel oil by volume were mixed in gasoline and engine performance and emission values were compared with gasoline, which is a fossil-based fuel. The aim of this study is to investigate the usability of waste fusel oil and plant-based methanol as alternative fuels in internal combustion engines. One the the other hand, it is to reveal the effects on alternative fuel availability, engine performance and emission output efficiencies by reducing harmful exhaust emissions.

## 2. Material and Method

## 2.1 Test Fuel Preparations

In the experimental study, 95 octane unleaded gasoline fuel (G) and fusel oil and pure methanol produced on vegetable basis were used. By mixing 20% Fusel oil (F20) and 20% pure Methanol (M20) by volume into gasoline, engine performance (Torque, Brake Specific Fuel Consumption (BSFC), Exhaust gas temperature (EGT)) and Exhaust emissions, carbon monoxide (CO), Carbon dioxide (CO<sub>2</sub>) and Nitrous oxide (NO<sub>x</sub>) values were measured. Experimental studies were carried out under full load conditions of the engine and the data were instantly recorded on the computer. Table 1 gives the physical and chemical properties of Fusel oil.

Table 1. Fusel oil physical and chemical properties [14,3].

Chemical content	Chemical formula	Molecular density (g/mol)	Density(g/cm <sup>3</sup> ) ASTM D1007	Boiling point (°C) ASTM D1007	Poor point(°C) ASTM D1007	% Volumetric	% Massive
i-amyl alcohol	$C_5H_{12}O$	88.148	0.8104	131.1	-117.2	63.93	61.52
i-butyl alcohol	$C_4H_{10}O$	74.122	0.802	108	-108	16.66	15.87
n-butyl alcohol	$C_4H_{10}O$	74.122	0.8098	117.73	-89.5	0.736	0.708
n-propyl alcohol	C <sub>3</sub> H <sub>8</sub> O	60.09	0.8034	97.1	-126.5	0.738	0.704
Ethanol	C <sub>2</sub> H <sub>6</sub> O	46.07	0.789	78.4	-114	9.58	8.98
Water	H <sub>2</sub> O	18	1	100	0	10.3	12.23

The test fuels were mixed with a magnetic stirrer for 15 minutes in a graduated beaker at the rates of M20 (20% methanol-80% Gasoline) and F20 (20% Fusel oil and 80% Gasoline) by volume, and the mixture became homogeneous. Samples

were taken from this prepared test fuel and its physical and chemical properties were determined within the framework of standard measurement methods. The physical and chemical properties of pure methanol, fusel oil and gasoline are given in Table 2.

Table 2. Physical and chemical properties of Methanol, Fusel oil and
Gasoline[23-25,14].

Properties	Methanol	Fusel oil	Gasoline
Molecular weight	32.04	74,122	91.4
Lower Calorific Value (Mj/kg)	20.10	29.514	42.9
RON/MON	106/96	103.61/106.82	95/85
Density (20°C'de) (kg/m3)	790	852.1	720,0- 775,0
Poor point (°C)	>50	>50	-53
Dynamic viscosity (20 °C)[mPas]	0.57	2,278	1.2

## 2.2 Test Equipments

Test equipments consists of test engine, exhaust emission device and engine dynamometer. The engine dynamometer is connected to the test engine by means of a coupling. The exhaust temperature sensor and the emission device exhaust probe are connected to the test engine exhaust. Dynamometer torque sensor, fuel gauge sensor, exhaust temperature sensor are connected to the dynamometer control module. Figure 1 shows the schematic picture of the experimental study.

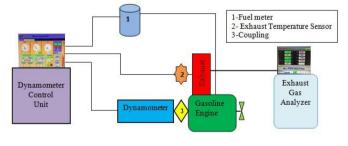


Fig. 1. Experimental study schematic picture

## 2.2.1 Test Engine

In the experimental studies, a single-cylinder, four-stroke atmospheric gasoline engine was used. The motor was combined with the coupling connection to the dynamometer. Periodic maintenance of the test engine was carried out before each measurement. The air filter was cleaned, the engine oil was checked, the fuel hose connections and the engine fixing connections were checked. The test engine was run at idle until it reached the regime temperature (300 °C) and measurements were started after this temperature. The engine throttle was



brought to the full throttle position and the test starting speed was determined as 3500 rpm. Engine performance and emission values were recorded to the computer as instant data at every 500 rpm intervals. The maximum torque speed of the test engine is 2500 rpm, and the maximum power speed is 3400 rpm. The test engine specifications are given in Table 3.

Engine Type	Single cylinder – 4 stroke – Air Cooled	
Torque	32/ 2500 Nm/rpm	
Piston Diameter x Stroke	68 x 54 mm	
Cylinder Volume	196 cc	
Compression Ratio	8,5:1	
Maximum power – HP (kW) @ rpm	17 (12.4) 3400 rpm	

Table 3 Test engine specifications

In the exhaust emission measurements, an emission device integrated with the computer was used. For gasoline vehicles, the device performs CO, CO<sub>2</sub>, HC, O<sub>2</sub>, Lambda and AFR measurements with a non-dispersive infrared system. In Table 4, the technical specifications of the exhaust emission device are given.

Table 4. Properties of the exhaust gas analyzer

Emission	Mesuring range	Measurement Accuracy
СО	0-%10	%0,001
CO <sub>2</sub>	0-%20	%0,001
НС	0-10000	1 ppm
O <sub>2</sub>	0-%25	%0,01
CO Corr	0-%10	%0,001
NO <sub>x</sub> (OPS.)	0-5000	1 ppm
Lambda	0,5-2.00	0.001
AFR	5-30	-
Engine Oil Temperature (OPS.)	0-150 °C	1 °C
RPM (OPS.)	0-9990 rpm	10 rpm.
Opacity	%0-100	%0,1

## 2.2.3 Engine Dynamometer

The test motor has been connected to the dynamometer by means of a coupling. It consists of dynamometer, fuel meter, dynamometer control module and control panel. Starting and stopping the test engine is done on the dynamometer control screen. After the engine has been started, it is operated at idle until it reaches the regime temperature. Engine exhaust gas temperature, engine speed, engine torque, engine power, hourly fuel consumption, specific fuel consumption are instantly seen and recorded on the dynamometer screen. After the engine reaches the regime temperature, full throttle is applied, and the engine is loaded from the dynamometer until the test start speed reaches 3500 rpm. The dynamometer reverse-loads the engine so that the engine speed drops at 25 rpm with each command. When it reaches 3500 rpm, engine performance and exhaust emission values are recorded to the computer as instant data. Measurements were repeated every 500 rpm. Engine dynamometer specifications are given in Table 5.

Maximum braking power	26 kW
Maximum braking torque	83 Nm
Exhaust temperature measuring range	0 - 1000 °C (Thermocouple)
Incremental Encoder	0 - 8000 rpm
Operating voltage	380V, AC,
Ambient temperature	40 °C
Control unit	Engine speed and load control, instan- taneous speed, torque, power, hourly fuel consumption, specific fuel con- sumption, exhaust gas temperature.

#### Table 5. Engine dynamometer specifications

## 3. Experiment Results and Discussion

In this study, 20% fusel oil (F20) and 20% pure methanol (M20) by volume were added to 95 octane gasoline (G). Test fuels were tested in a single-cylinder atmospheric engine under full load conditions, and engine performance and exhaust emission values were compared with gasoline fuel.

## 3.1 Engine performances

Torque changes of test fuels are shown in Figure 2. At the maximum torque speed, G fuel produced 28.82 Nm, F20 fuel produced 26.72 Nm and M20 fuel produced 27.55 Nm. On average, a torque decrease of 11.62% in F20 fuel and 6.45% in M20 fuel was determined. M20 fuel provided less torque drop than F20 fuel. The amount of water in the fusel oil is quite high and this affects the combustion performance [10]. When Table 2 is examined, although the lower calorific value of fusel oil is higher than that of methanol fuel, the torque reduction is higher than that of M20 fuel. The density of M20 fuel is lower than fusel oil. In addition, the viscosity of F20 fuel is higher than M20 fuel. This is thought to improve fuel atomization and combustion efficiency. When Table 1 is examined, there is a small amount of water in fusel oil. The water in the fuel worsens the combustion and lowers the end-of-combustion pressure. The lower performance of F20 fuel compared to M20 fuel can be shown as a reason for this situation. The amount of oxygen in alcohol-based fuels is higher than in fossil fuels. This affects the in-cylinder combustion performance. It can be thought that the excess oxygen in the fuel improves the engine performance by positively affecting the combustion in the cylinder. However, since the heating values, fuel densities and viscosities of M20 and F20 fuels are lower than G fuel, it causes the engine torque values to decrease. This situation has also been ob-



served in similar studies in the literature [5,11,30-32].

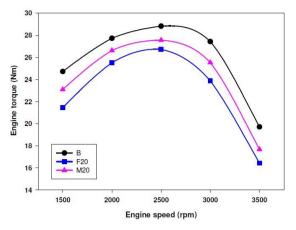


Fig.2. The effect of methanol and fusel oil addition in test fuels on engine torque

The brake specific fuel consumption of the test fuels is shown in Figure 3. Brake specific fuel consumption (BSFC) was measured as 362.08 g/kWh in G fuel, 397.77 g/kWh in F20 fuel and 374.34 g/kWh in M20 fuel at maximum torque speed. Compared to gasoline, BSFC showed an increase in F20 and M20 fuels. This increase was found to be 9.79% in F20 fuel and 4.23% in M20 fuel on average. Heating value differences of fuels affect BSFC results [26,28]. Although the heating value of fusel oil is higher than that of methanol, the BSFC value of F20 fuel was higher than that of M20 fuel. The specific fuel consumption value is proportional to the calorific value of the fuels and their conversion to useful work in the cylinder [18]. The reason why the BSFC value of M20 fuel is lower than that of F20 fuel is due to the differences in the heating value as well as the different density and viscosity values of the fuels. Fusel oil worsens combustion and increases BSFC because it contains water. Uslu and Celik [29] stated that alcohol fuels have lower calorific values and higher stoichiometric fuel/air ratios compared to gasoline, so more fuel will be used to provide the same output power in SI engines, which will increase the BSFC. There are many studies showing that BSFC values increase in alcohol-based fuel usage due to low calorific value [7,12,21,26].

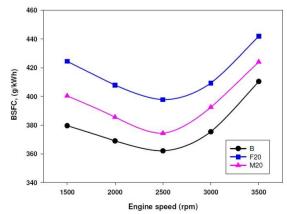


Fig.3. The effect of methanol and fusel oil addition in test fuels on

Figure 4 shows the exhaust gas temperature (EGT) changes of the test fuels. At maximum torque speed, it was measured as 701.88 °C in G fuel, 681.19 °C in F20 fuel and 658.20 °C in M20 fuel. F20 and M20 EGT temperatures showed a decrease compared to G fuel. The highest EGT decrease occurred in M20 fuel. For all test fuels, maximum temperatures were obtained close to maximum power speed. At all engine speeds, on average, EGT decreased by 3.22% in F20 fuel and 5.91% in M20 fuel. The reason for the EGT decreases of F20 and M20 fuels compared to G fuel is that alcohol-based fuels have a higher latent heat of vaporization than G fuel [7]. The high BSFCs of F20 and M20 test fuels were interpreted as the EGT temperatures decreased as a result of the increase in the amount of mixture taken into the cylinder to provide the same power. Sharudin et al., Alcohol-based fuels have higher latent heat of vaporization and lower heating value than Gasoline fuel, which reduces EGT. This situation has also been found in similar literature studies [12,13,16,20,27].

BSFC

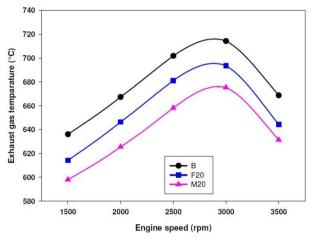


Fig.4. The effect of methanol and fusel oil addition in test fuels on EGT

#### 3.2 Exhaust emissions

CO Carbon monoxide (CO) changes of test fuels are shown in Figure 5. Compared to G fuel, CO emission values decreased in F20 and M20 fuels. At maximum torque speed, CO values were measured as 8.32% in G fuel, 7.78% in F20 fuel and 7.31 in M20 fuel. The highest CO reduction was achieved in M20 fuel. On average, CO values decreased by 7.31% in F20 fuel and 13.40% in M20 fuel at all engine speeds. Excess CO is a product of incomplete combustion. On the contrary, when full combustion is provided in the cylinder, the CO values decrease. The excess of oxygen in the content of alcoholbased fuels improves combustion compared to fossil fuels. This reduces the CO values [4,7,8]. The use of alcohol-based fuels reduces CO emissions. Due to the high oxygen content of alcohol-based fuels, the reason for the decrease in CO is complete combustion, especially at low rpm. The fact that M20 fuel further reduces the amount of CO compared to F20 fuel indicates that M20 fuel properties are better than F20 fuel. Da-408



ta on CO reduction and cause-effect relationships are consistent with previous studies [12-14,18-22]

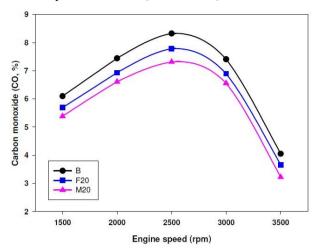


Fig.5. The effect of methanol and fusel oil addition in test fuels on CO

Figure 6 shows the  $CO_2$  changes of the test fuels. The  $CO_2$ values of the test fuels increased compared to G fuel. On aveage, CO<sub>2</sub> values increased at all engine speeds, 10.73% in F20 fuel and 5.90% increase in M20 fuel. M20 fuel produced less CO<sub>2</sub> than F20 fuel. An increase in carbon dioxide emissions is due to the effect of oxygen increases, which contributes to an increase in the rate of complete combustion. Increasing oxygen during combustion will result in more carbon that can be converted to carbon dioxide. Excess oxygen in alcoholbased fuels improves combustion. This affects the formation of CO<sub>2</sub> [12]. Since fusel oil contains oxygen, the hydroxyl radical OH (one of the major oxidizing agents) converts carbon monoxide to carbon dioxide with the presence of sufficient oxygen O<sub>2</sub> [14].. The fact that the amount of oxygen in fusel oil and methanol is higher than gasoline fuel affects combustion and it can be said that this situation causes an increase in CO<sub>2</sub>. Similar studies in the literature also refer to the high oxygen content of alcoholic fuels as the main reason for the increase in CO<sub>2</sub> [18-22, 26-30].

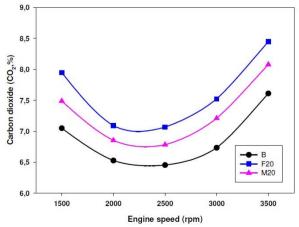


Fig.6. The effect of methanol and fusel oil addition in test fuels on-CO<sub>2</sub>

Figure 7 shows the  $NO_x$  changes of the test fuels. The  $NO_x$ values of the test fuels increased compared to G fuel. F20 fuel produced more NO<sub>x</sub> than M20 fuel. On average, an increase of 12.62% in F20 fuel and 9.67% in M20 fuel was detected at all engine speeds compared to G fuel. NO<sub>x</sub> formation is an emission due to high temperatures resulting from combustion in the cylinder. At temperatures above 1500 °C, NO<sub>x</sub> formation occurs [6,7]. Fusel oil and methanol are alcohol-based fuels. The high amount of oxygen in alcohol-based fuels affects the endof-combustion temperature and this increases the formation of NO<sub>x</sub>. Looking at Figure 7, the reason why M20 fuel produces less NO<sub>x</sub> than F20 fuel may be due to the oxygen differences in the fuel. On the other hand, BSFC values of F20 and M20 fuels were higher than G fuel. The high BSFC causes an increase in NO<sub>x</sub> as a result of more fuel-air mixture being taken into the cylinder to provide the power corresponding to the unit volume and therefore the increase in the in-cylinder combustion end temperatures. Similar studies and similar results triggering NO<sub>x</sub> increase were obtained by other researchers [1,12,19,21,32].

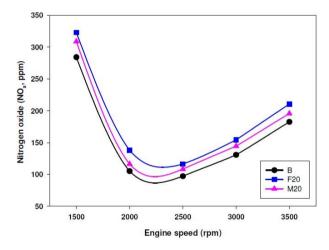


Fig.7. The effect of methanol and fusel oil addition in test fuels on NO<sub>x</sub>

## 4. Conclusions

In this study, engine performance and exhaust emission values were measured by adding 20% fusel oil and 20% methanol into gasoline, which is a fossil-based fuel. As a result of the experimental study;

- Compared to G fuel, torque decreased in F20 and M20 test fuels.
- BSFC increased 9.79% in F20 fuel and 4.23% in M20 fuel compared to G fuel at all engine speeds,
- The EGT temperatures of the test fuels decreased on average by 3.22% in F20 fuel and 5.91% in M20 fuel.
- It was determined that CO values decreased by 7.31% in F20 fuel and 13.40% in M20 fuel,
- CO<sub>2</sub> values increased by 10.73% in F20 fuel and 5.90% in M20 fuel,
- Compared to NO<sub>x</sub>G fuel, it increased 12.62% in F20 fuel and 9.67% in M20 fuel.



- M20 and F20 ratios can be used by mixing directly into gasoline without the need for any additional system to the engine system.
- Studies can be expanded with different mixing ratios and different fuels.

## Nomenclature

M20	:Test fuel consisting of a mixture of 20%
	Methanol-80% gasoline by volume
F20	:Test fuel consisting of a mixture of 20%
	Fusel oil and 80% gasoline by volume
BSFC	:Brake Specific Fuel Consumption
CO	:Carbon monoxide
$CO_2$	:Carbon dioxide
NO <sub>x</sub>	:Nitrogen oxide
G	:Gasoline
EGT	:Exhaust Gas Temperature
RON	:Research Octane Number
MON	:Engine Octane Number

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