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Estimation and Optimization of the Radiant Field in Flat Plate Heterogeneous Photoreactors with the P1-approximation of the Radiative Transfer Equation (RTE)

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Abstract: In this work, the P1-approximation of the radiative transfer equation (RTE) was used for the description and optimization of the radiant field in a flat plate photoreactor under solar radiation with three commercial brands of titanium dioxide photocatalysts. The boundary layer of photon absorption ([δ_{abs}]), the average volumetric rate of photon absorption ($_{V\!RPA}$), and a new apparent optical thickness ($\zeta_{\it app1}$) were used as design parameters for optimization. A simple mathematical expression for the calculation of $\delta_{
m abs}$ also called the best reactor thickness was formulated. For the three catalysts, varying the reactor height ($_L$) resulted in a decrease in the local volumetric rate of photon absorption ($_{LVRPA}$) from the top side to the bottom of the reactor for any value of the catalyst loading (C_{cat}). It was also observed that when C_{cat} increases the $_{V\!RP\!A}$ increases exponentially until a fixed value where it remains almost constant. With $_L$ = 1 cm, the optimum C_{cat} (C_{cat-op}) was 0.2 g/L in 0.85 cm of thickness, 0.3 g/L in 0.82 cm of thickness, and 0.4 g/L in 0.89 cm of thickness for the photocatalysts Catalyst D P-25, Catalyst A, and Catalyst H, respectively. The optimum apparent optical thickness ($\zeta_{app1,op}$) was 4.87, 4.62, and 3.7 for the photocatalysts Catalyst D P-25, Catalyst A, and Catalyst H, respectively. These results are in good agreement with the literature. Results found in this work give predictions on radiation absorption in flat plate photocatalytic reactors with different heights.

Keywords: Flat plate reactor, Boundary layer, *LVRPA*, Titanium dioxide, P1-approximation.

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1. INTRODUCTION

Water pollution is one of the major global issues, primarily affecting developing countries. Several approaches have been put out to address this situation. For many decades now, heterogeneous photocatalysis has been considered a potential oxidation method for disinfecting and decontaminating water (He et al., 2021; Li et al., 2022; Nair & Jagadeesh Babu, 2017). It is an advanced oxidation technology based on the photo-excitation of photocatalysts with solar or radiation. artificial The activation of а semiconductor with an energy equal to or greater than its band gap energy generates electron-hole pairs, which in contact with charge carriers (H_2O) , OH⁻, O₂, etc) produce radical oxidative species such as hydroxyl radicals. These radicals are transitory compounds that attack contaminants present in the fluid phase through oxidative or reductive reaction pathways (Li Puma et al., 2020). Research on solar photocatalysis as a clean technology for producing hydrogen and solar fuels, as well as a sustainable alternative to treating industrial wastewater and removing dyes, organic pollutants, pesticides, and emerging contaminants, has increased dramatically as a result of the ongoing concern

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about water remediation (Rizzo et al., 2019; Vaya & Surolia, 2020). Because mathematical modeling is crucial to the design, assessment, optimization, and efficiency estimation of photoreactors at various sizes, it has become more and more relevant in this field (Wang et al., 2021). The mathematical modeling of photocatalysis processes is composed of a sequence of sub-models including the modeling of the radiant field (Ochoa-Gutiérrez et al., 2018). This sub-model includes the solar emission model and the absorption-scattering model for the quantification of the LVRPA, one of the key parameters of the intrinsic kinetic equation (Colina-Márquez et al., 2015). The LVRPA depends on the geometry, radiation source, catalvst loading, and type of photocatalyst, in some cases on the pollutant if this presents absorption of radiant energy. For its determination, one should solve the RTE which remains a challenging task due to its integrodifferential form (Illi et al., 2019). Eq. (1) traduces the steady-state and non-temperaturedependence of the RTE, which describes the different phenomena that occur on the light when it traverses a medium, such as absorption, inscattering, and out-scattering as represented in Eq. (1).

$$\frac{dI_{\lambda}(S,\Omega)}{ds} = -KI_{\lambda}(S,\Omega) - \sigma_{\lambda}I_{\lambda}(S,\Omega) + \frac{\sigma_{\lambda}}{4\pi} \int_{\Omega=4\pi} P(\Omega \to \Omega)I_{\lambda}(S,\Omega') d\Omega' (1)$$

where I_{λ} is the photon irradiance (W/m²), K_{λ} the absorption coefficient (m²/kg), σ_{λ} the scattering coefficient (m²/kg), $P(\Omega' \rightarrow \Omega)$ ^{the} scattering phase function, λ the wavelength (m), S the spatial coordinate (m) and Ω the directional solid angle (Steradian) (Fujii et al., 2022; Ghafoori et al., 2020; Howell et al., 2021).

The RTE gives an account of how light is dispersed or absorbed within a specific medium. Numerous numerical techniques, such as the discrete ordinate methods (DOM) and the Monte Carlo model, were employed to solve the RTE; nevertheless, these methods are time-consuming and need significant processing resources (Acosta-Herazo et al., 2020; Moreno-SanSegundo et al., 2020; Peralta Muniz Moreira & Li Puma, 2021). Alternative analytic methods such as the n-flux or the Pn approximation models (n is a strictly positive integer) have been used to solve the RTE, to describe and quantify satisfactorily the radiant field in various types of photocatalytic reactors (Arancibia-Bulnes et al., 2009; Cuevas et al., 2007). The P1-approximation is the lowest order of the spherical harmonics method (also known as the Pn-approximation), i.e., keeps only the first two terms of the Pn-approximation (Akdemir et al., 2022; Christenson et al., 2018; Harel et al., 2021). It is more versatile than two and four-flux models because it lends itself more easily to different geometries (Arancibia-Bulnes et al., 2009). It has been used to solve the RTE in one dimension in flat plate photoreactors and two dimensions in cylindrical photoreactors (Arancibia-Bulnes et al., 2009; Cuevas & Arancibia-Bulnes, s. f.). The P1 approximation

saves computational time and effort by reducing the mathematical complexity of the RTE to an analytical equation (Arancibia-Bulnes et al., 2009; Cuevas et al., 2007; Cuevas & Arancibia-Bulnes, s. f.). Flat plate reactors can be scaled up and can be used with solar radiation, so they are very attractive and also provide an excellent configuration for efficient activation of the photocatalyst TiO_2 (Li Puma, 2005). Their modeling requires a complex analysis of the radiation field inside the photoreactor (Cassano et al., 1995).

In this paper, the P1 approximation was used to model radiant fields and to find the optimum operational conditions in flat plate heterogeneous photoreactors under solar radiation. Three commercial brands of titanium dioxide photocatalysts (Catalyst D, Catalyst A, and Catalyst H) were used and some comparisons were made between them. Design parameters such as the VRPA, the best reactor thickness

 ${\delta}_{
m abs}$, and a new apparent optical thickness ${\zeta}_{app1}$ were used. ζ_{app1} was formulated with the P1 approximation approach. Kinetic models of pollution degradation and hydrogen production could benefit from the information provided. In kinetic models of pollution degradation and hydrogen production, determining the local volumetric rate of photon absorption might help to improve understanding of the relevant processes. It makes it possible to calculate reaction rates, energy transfer, and overall precisely, which efficiency improves environmental results and leads to optimal systems.

2. MATERIALS AND METHODS

2. 1. Radiant field

The radiant field was estimated by solving the RTE in one dimension in rectangular coordinates, and then the LVRPA was deduced. Eq. (2) is the governing equation of the P1 approximation or the so-called Helmholtz's equation which will be solved later in one dimension with suitable boundary conditions.

$$\Delta G_{\lambda} = k_{d,\lambda}^2 G_{\lambda} \tag{2}$$

where,

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$$k_{d,\lambda} = \beta_{\lambda} \sqrt{3(1 - \omega_{\lambda})(1 - g_{\lambda} \frac{\omega_{\lambda}}{3})}$$
(3)

where g_{λ} is the asymmetry Henyey-Greenstein phase function factor corresponding to the monochromatic radiation of wavelength λ and which is proper to each catalyst, β_{λ} is the monochromatic extinction coefficient being the sum of κ_{λ} and σ_{λ} which are the monochromatic absorption and scattering coefficients respectively defined as,

$$\beta_{\lambda} = \kappa_{\lambda} + \sigma_{\lambda}$$
 (4)

All of these coefficients depend on catalyst particle concentration in a linear fashion as,

$$\beta_{\lambda} = \beta_{\lambda}^{\star} C_{cat} \kappa_{\lambda} = \kappa_{\lambda}^{\star} C_{cat} \sigma_{\lambda} = \sigma_{\lambda}^{\star} C_{cat}$$
(5)

where C_{cat} is the catalyst concentration, $\beta_{\lambda'}^{\star} \kappa_{\lambda}^{\star}$ and σ_{λ}^{\star} are specific coefficients, independent of this concentration (for values in the usual ranges for photocatalysis).

 $\omega_{\lambda} \text{is the scattering albedo which gives the probability that a photon is scattered when colliding with a particle defined as,$

$$\omega_{\lambda} = \frac{\sigma_{\lambda}}{\beta_{\lambda}} \tag{6}$$

Instead of using the monochromatic parameters, one should use their average values in a defined wavelength interval $[\lambda_{min}, \lambda_{max}]$ using the following equation,

$$\Gamma^{*} = \frac{\int_{\lambda_{min}}^{\lambda_{max}} \Gamma_{\lambda} I(\lambda) d\lambda}{\int_{\lambda_{min}}^{\lambda_{max}} I(\lambda) d\lambda}$$
(7)

where Γ_{λ} can be one of the parameters κ_{λ} , $\sigma_{\lambda'}\beta_{\lambda}$ or g_{λ} and λ_{min} =280 nm, λ_{max} =395 nm given by the interception of solar emission spectrum and titanium dioxide absorption spectrum; $I(\lambda)$ is the spectrum of solar emission power (Tourasse & Dumortier, 2014). G_{λ} is the local radiation which corresponds to the integration of the irradiance I_{λ} all over the solid angle $\underline{\Omega}$. After finding the expression of $G_{\lambda'}$ one can easily deduce the local volumetric rate of photon absorption (LVRPA) from Eq. (8); this parameter is very important for the determination of the intrinsic kinetic rate of the photocatalytic reaction in a given reaction system.

$$LVRPA = k_{\lambda}G_{\lambda}$$
 (8)

In this paper, the concept of the boundary layer of photon absorption introduced yet in the literature (Otálvaro-Marín et al., 2014) and which allows the determination of the best thickness has been discussed with the P1 approximation approach. Figure 1 represents a flat plate reactor with thickness L, located perpendicularly to the

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source, where I_0 represents the constant incident radiation intensity. The region where there is a gradient of energy absorption has been called the boundary layer of photon absorption, and its thickness δ_{abs} which can be understood as the reactor thickness measured from the irradiated surface where 99% of total energy is absorbed. Then δ_{abs} is defined as $_X$ -value which satisfies the following equation:

$$\int_{0}^{\delta_{abs}} LVRPA(x) dx = 0.99 \int_{0}^{L} LVRPA(x) dx$$
⁽⁹⁾

This definition of boundary layer thickness for transport phenomena and absorption of radiant energy is analogous to the definition of thickness of hydrodynamic, thermal, and concentration boundary layer on a flat plate (Incropera, s. f.; Otálvaro-Marín et al., 2014).



Figure 1. Collimated incident radiation on a slab reactor.

Oversizing the reactor thickness could lead to the formation of a dark sublayer with very little photon absorption. As a result, while choosing the ideal thickness for flat plate reactors, the boundary layer thickness becomes a design parameter (Otálvaro-Marín et al., 2014). It can be used as a geometrical parameter that is equivalent to the optimal reactor thickness, depending on the type and loading of the catalyst. It is worth mentioning that the new apparent optical thickness was also formulated here with the P1 approximation approach and was used for a design purpose.

2. 2. Solution for the RTE in One Dimension Using the P1 Approximation Approach

Figure 1 shows a schematic representation of a flat plate reactor with thickness \underline{L} , located

perpendicularly to the source.

The P1 approximation in one dimension in cartesian coordinates becomes:

$$\frac{\partial G_{\lambda}}{\partial x} = k_{d,\lambda}^2 G_{\lambda}$$
(10)

The general solution of Eq. 10 is:

$$G_{\lambda} = A e^{k_{d,\lambda}x} + B e^{-k_{d,\lambda}x}$$
(11)

where ${}_{A}$ and ${}_{B}$ are real constants to determine later.

2. 3. Boundary Conditions

where ,

For the determination of the constants $_A$ and $_B$ Marshak's boundary conditions (Marshak, 1947) are considered in Eqs. (1-2), where $_{I_0}$ represents the photon flux reaching perpendicularly the top side of the reactor and with the assumption that the photon flux does not reach the reactor bottom and that the inner reflectance of reactor glass is negligible.

$$G_{\lambda}(0) - \frac{2}{3\xi_{\lambda}} \frac{dG_{\lambda}(x)}{dx}_{x=0} = 4I_{0}$$
(12)

$$G_{\lambda}(L) + \frac{2}{3\xi_{\lambda}} \frac{d G_{\lambda}(x)}{d x} = 0$$
(13)

 $\xi_{\lambda} = \beta_{\lambda} (1 - g_{\lambda} \frac{\omega}{3}) \tag{14}$

That leads to the following system of equations:

$$\left(1 - \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)A + \left(1 + \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)B = 4I_0 \tag{15}$$

$$(1+\frac{2k_{d,\lambda}}{3\xi_{\lambda}})e^{k_{d,\lambda}L}A+(1-\frac{2k_{d,\lambda}}{3\xi_{\lambda}})e^{-k_{d,\lambda}L}B=0$$
 (16)

Which yields

$$A = \frac{4\left(1 - \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)e^{-k_{d,\lambda}L}I_{0}}{\left(1 - \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)^{2}e^{-k_{d,\lambda}L} - \left(1 + \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)^{2}e^{k_{d,\lambda}L}}$$
(17)

$$B = \frac{-4\left(1 + \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)e^{k_{d,\lambda}L}I_{0}}{\left(1 - \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)^{2}e^{-k_{d,\lambda}L} - \left(1 + \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)^{2}e^{k_{d,\lambda}L}}$$
(18)

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Finally, one finds,

$$G_{\lambda}(x) = \frac{4\left(1 - \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)e^{-k_{d,\lambda}(L-x)} - 4\left(1 + \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)e^{k_{d,\lambda}(L-x)}}{\left(1 - \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)^{2}e^{-k_{d,\lambda}L} - \left(1 + \frac{2k_{d,\lambda}}{3\xi_{\lambda}}\right)e^{k_{d,\lambda}L}}I_{0}$$
(19)

The local volumetric rate of photon absorption will be then,

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$$LVRPA(x) = \frac{4\kappa_{\lambda}((1-\frac{2k_{d,\lambda}}{3\xi_{\lambda}})e^{-k_{d,\lambda}(L-x)} - (1+\frac{2k_{d,\lambda}}{3\xi_{\lambda}})e^{k_{d,\lambda}(L-x)})}{(1-\frac{2k_{d,\lambda}}{3\xi_{\lambda}})^{2}e^{-k_{d,\lambda}L} - (1+\frac{2k_{d,\lambda}}{3\xi_{\lambda}})^{2}e^{k_{d,\lambda}L}}I_{0}$$
(20)

2. 4. Parameters of the P1 Approximation

2. 4. Parameters of the P1 Approximation $\omega_{\lambda,mod'} \lambda_{mod'}$ and ζ_{app1} are defined here as the P1 approximation parameters, their equivalent are $\omega_{corr'} \lambda_{corr'}$ and ζ_{app} for the six-flux model (SFM) respectively, ζ_{app1} is a design parameter as ζ_{app} is for the SFM (mod stands for modified). These parameters are defined as follower: These parameters are defined as follows:

$$\omega_{\lambda,mod} = \sqrt{\omega_{\lambda} \left(1 + \frac{g_{\lambda}}{3} - \frac{g_{\lambda}}{3}\omega_{\lambda}\right)} \quad (21)$$

$$\lambda_{mod} = \frac{\lambda_0}{\sqrt{3(1 - \omega_{\lambda, mod}^2)}}$$
(22)

where λ_{0} is the mean free path of photon defined as,

$$\lambda_0 = \frac{1}{\beta_\lambda} \tag{23}$$

 ζ_{app1} , is the apparent optimal thickness defined as

$$\zeta_{app1} = \zeta \sqrt{3(1 - \omega_{\lambda,mod}^2)}$$
(24)

where $\boldsymbol{\zeta}$ is the optical thickness, for a flat plate reactor expressed as,

$$\zeta = \frac{L}{\lambda_0} \tag{25}$$

$$k_{d,\lambda} = \frac{1}{\lambda_{mod}}$$
(26)

$$LVRPA(x) = \frac{4\kappa_{\lambda}((1-\frac{2}{3\xi_{\lambda}})e^{\frac{-L-x}{\lambda_{mod}}} - (1+\frac{2}{3\xi_{\lambda}})e^{\frac{L-x}{\lambda_{mod}}})}{(1-\frac{2}{3\xi_{\lambda}})e^{\frac{-L}{\lambda_{mod}}} - (1+\frac{2}{3\xi_{\lambda}})e^{\frac{L}{\lambda_{mod}}}I_{0}}$$
(27)

The unique solution for Eq. (9) is given by:

where,

$$\delta_{abs} = \lambda_{mod} \ln\left(\frac{\Sigma + \sqrt{\Sigma^2 + 4AB}}{2A}\right)$$
(28)
$$\Sigma = 0.99 \left(A e^{\frac{L}{\lambda_{mod}}} - B e^{\frac{-L}{\lambda_{mod}}} - A + B\right) + A - B$$
⁽²⁹⁾

where $_A$ and $_B$ are constants defined in Eqs. (17-18) respectively.

2. 5. Volumetric Rate of Photon Absorption, VRPA·

VRPA is defined as an average value of VRPA in the entire reactor volume. For a flat plate reactor, when the incident radiation intensity is constant along the reactor wall, the is expressed as:

$$VRPA = \frac{1}{L} \int_{0}^{L} LVRPA(x, C_{cat}) dx$$
(30)

VRPA is a design parameter formulated and validated to determine the optimum catalyst loadings in photocatalytic processes (Brandi et al., 1996; Otálvaro-Marín et al., 2014).

3. RESULTS AND DISCUSSION

The simulation of the P1 approximation in a flat plate photoreactor was performed under solar radiation with three different brands of commercial titanium dioxide with their optical characteristics (albedo, modified albedo, scattering and absorption coefficient, asymmetry factor) reported in the literature (Acosta-Herazo et al., 2016; Otálvaro-Marín et al., 2014) and in Table 1. The reactor thickness L was considered to vary.

Table 1. Average optical properties of commercial photocatalysts based on titanium dioxide under solar radiation (Acosta-Herazo et al., 2016; Otálvaro-Marín et al., 2014), modified albedo, and asymmetry factor $_{a}$.

Catalyst	$\sigma^* \times 10^{-4} (c m^2 g^{-1})$	$\kappa^* \times 10^{-3} \\ (c m^2 g^{-1})$	$\beta^* \times 10^{-4} (c m^2 g^{-1})$	ω	ω_{mod}	g
Catalyst A	3.73	2.43	3.98	0.94	0.97	0.53
Catalyst D	5.42	2.87	5.71	0.95	0.99	0.53
Catalyst H	2.52	1.17	2.64	0.96	0.98	0.57

3.1. *LVRPA*

The local volumetric rate of photon absorption per unit of incoming radiant flux $(_{LVRPA}/I_0)$ in function of x-coordinate at different catalyst loads describes the distribution of the photon absorption inside the reactor for each of the catalysts Catalyst A, Catalyst D and Catalyst H as depicted in Figure 2 $(_{L=1}$ cm). For each catalyst, the $_{LVRPA}/I_0$ profile shows the decrease of the absorption from the top side (Maximum value of $_{LVRPA}/I_0$) until the bottom of the reactor (minimum value of $_{LVRPA}/I_0$) and that the absorption at the top side of the reactor increases of the side of the reactor (minimum value of the reactor (minimum value

catalyst load due to the amount of surfaceexposed catalyst and the back-scattering energy absorption from the internal layer inside the reactor. Near the irradiated surface, the change $LVRPA/I_0$ with respect to x is linked to extinction coefficients; a rise in catalyst concentration and high extinction coefficients quickly extinguish the total energy of the system. These observations are in good agreement with the literature (Nchikou et al., 2021; Otálvaro-Marín et al., 2014). It is also observed that the more the catalyst loading increases the more the absorption at the reactor bottom decreases. The $LVRPA/I_0$ is negligible at the reactor bottom for C_{cat} greater than 0.3 g/L for Catalyst A and Catalyst D, the same result was found by using

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the SFM (Otálvaro-Marín et al., 2014) which was validated by comparing it to rigorous solution of

the RTE (Brandi et al., 1999; Zalazar et al., 2005).



Figure 2. The $_{LVRPA}$ per unit of light intensity vs x-axis with different C_{cat}, (L=1 cm) a) Catalyst A , b) Catalyst D, c) Catalyst H

The $LVRPA/I_0$ profile for each catalyst in Figure 3 shows a good uniformization of photon absorption for C_{cat} equal to 0.1 g/L (Figure 3 a, c, and e) but at 0.5 g/L (Figure 3 b, d, and f) of C_{cat} photon absorption decreases considerably at the inner zone of the reactor while increases significantly at the

zone around its wall. This could be explained by the fact that the cloudy effect starts which impedes the photon penetration in the inner part of the reactor due to the saturation of the absorption at the irradiated reactor surface (Colina-Márquez et al., 2010). With Catalyst H the clouding effect starts with Ccat greater than 0.5 g/L.

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Figure 3. Absorption profile per unit of light intensity at 0.1 g/L of C_{cat} (a), c), e) for Catalyst A, Catalyst D, and Catalyst H respectively) and at 0.5 g/L of C_{cat} (b), d), f) for Catalyst A, Catalyst D, and Catalyst H respectively).

3. 2. Optimal Operating Conditions in a Flat Plate Photoreactor

3. 2. 1. VRPA

Based on the quantity of radiation absorbed by the catalyst, the VRPA is crucial for assessing the quantum yield of photocatalytic reactions, it gives more significant information compared to the quantum efficiency, which depends on the incident photon flux. Additionally, it enables a suitable assessment of the activity of catalysts with various optical characteristics (Colina-Márquez et al., 2010). Figure 4 represents the VRPA per unit of light intensity $(VRPA/I_{o})$ as a function of catalyst loading with different reactor depths for the three catalysts and different reactor heights. It shows how the $VRPA/I_0$ increases exponentially and reaches a value from where it remains almost constant due to the photon saturation at the upper side of the reactor. The $VRPA/I_0$ with Catalyst D is a little higher than with Catalyst A for Ccat less than 0.15 g/L approximately for L=0.5 cm and $L=1\,\mathrm{cm}$ but above this value, the reverse is observed. For reactor height higher than unity, Catalyst A has the highest $V RP A/I_0$ (see Figure 4 a) and b)); meanwhile Catalyst H has the lowest $VRPA/I_0$ for C_{cat} less than 0.25 g/L and almost coincides with Catalyst D in terms of absorption for C_{cat} greater than 0.25 g/L (see Figure 4 c) and d)).

In the case when $L=1 \text{ cm}^{\prime}$, the $VRPA/I_0$ profiles present a weak absorption for C_{cat} less than 0.1 g/L and from 0.1 to 0.4 g/L an exponential increase; catalyst loading above 0.2, 0.3 and 0.4 g/L respectively for Catalyst D, Catalyst A and Catalyst H would be a catalyst waste since the $VRPA/I_0$ will no more increase significantly and high value of $C_{\mbox{\tiny cat}}$ produces the clouding effect. The optimum values of C_{cat} are around 0.2, 0.3, and 0.4 g/L for Catalyst D, Catalyst A, and Catalyst H, respectively, which agree with the literature (Otálvaro-Marín et al., 2014). In comparison to the catalyst Catalyst D, the catalyst Catalyst A maximum is VRPA approximately 12 % higher. A comparable pattern was observed, where Catalyst A is 19 % and 7 % more efficient than Catalyst D P-25 in the case of UV lamps used for polychromatic radiation and in the case of solar radiation using the SFM approach respectively (Brandi et al., 1999; Otálvaro-Marín et al., 2014). As it is shown in Figure 4, although the $VRPA/I_0$ increases with

 C_{cat} , it decreases with the increase of the reactor height. it drops to 80% approximately for each of the photocatalysts for reactor height varying from 0.5 to 3 cm.





Figure 4. The *VRPA* per unit of light intensity vs catalyst loading with different reactor depths for the three catalysts (a) $L = 0.5 \text{ cm}^{-1} \text{ b}$ L = 1 cm, c L = 2 cm, d $L = 3 \text{ cm}^{-1}$.

3. 2. 2. Impact of the reactor height on the VRPA

Figure 5 shows how the reactor height affects the $VRPA/I_0$ for the three catalysts. In this figure, the $VRPA/I_0$ decreases with the increase of the reactor height for each of the three catalysts. For both catalysts Catalyst A and Catalyst D, the VRPA maximum ($VRPA_{max}/I_0$) is reduced to 50% when L increases from 0.5 to 1 cm and to 30% when L increases from 1 to 1.5 cm; over 2 cm the $VRPA_{max}/I_0$ does not vary too much. The same observation is made with the catalyst Catalyst H except that the $VRPA_{max}/I_0$ decreases to 40% approximately when increasing L from 0.5 to 1 cm. These reductions

of the $V RP A_{max} / I_0$ could be explained by the fact that the more the reactor height increases the more the photon absorption uniformization decreases since the photon pathway becomes very long and it struggles to reach the inner part of the reactor. Moreover, for the catalysts Catalyst A and Catalyst D the optimum C_{cat} is about 0.15 g/L for L > 1 cm; meanwhile for Catalyst H, the optimum C_{cat} is about 0.15 g/L for L>3 cm. Thus, for high values of L, the optimum C_{cat} is very low but unfortunately working with a very small amount of Ccat implies low production of oxidizing species. Then oversizing the reactor height reduces considerably the average value of the LVRPA inside the reactor and therefore could disfavor the photocatalytic process.







L = 0.5 cm in blue line, L = 1 cm in red line L = 1.5 cm in yellow line, L = 2 cm in green line, L = 3 cm in orange line a) Catalyst A, b) Catalyst D, c) Catalyst H

4. REACTOR SIZING

4.1. Boundary Layer

Eq. (28) establishes the boundary layer of photon absorption (δ_{abs}) dependence on the catalyst, catalyst loading, and the reactor thickness. Figure 6 shows how varies δ_{abs} with respect to C_{cat} and L for the catalysts Catalyst A, Catalyst D and Catalyst H. For any value of L, δ_{abs} decreases with the increase of C_{cat} since the more C_{cat} increases the more photons penetration to the inner part of the reactor becomes difficult. For low values of C_{cat}, δ_{abs} increases linearly with

L, probably because low C_{cat} means no saturation of catalyst particles, and therefore, photons can easily move and reach the reactor bottom and then be absorbed (δ_{abs} is almost equal to L). It is worth mentioning that the more the boundary layer approaches the reactor height the more the absorption is uniform inside the reactor. Finally, for high values of C_{cat} (greater than 0.5 g/L approximately), δ_{abs} increases linearly with L until a fixed value where it remains almost constant. This is obvious since in this case, the clouding effect occurs because of the presence of a high amount of catalyst particles at the reactor wall impeding photons penetration.



Figure 6. Boundary layer of photon absorption as a function of catalyst loading and reactor height for the three catalysts.

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Figure 7 shows how the boundary layer of photon absorption varies as a function of catalyst loading (L = 1 cm). Combining Figures 4 and 7 one can find helpful details to specify the kind and amount of catalyst as well as the design thickness of a flat plate reactor. For each of the catalysts, for C_{cat} lower than 0.1 g/L, absorption takes place in almost the whole reactor but a very small value of the $_{V\!R\!P\!A}$ is produced ($_{\delta_{\rm abs}}$ is almost equal to L). For C_{cat} from 0.1 to 0.2 g/L, photon absorption occurs over 85% of the entire reactor height while in the range 0.2-0.4 g/L of C_{cat}, the VRPA maximum is obtained between 82 and 89% of the entire reactor thickness as shown in Table 2. This table summarizes the optimum catalyst loading, maximum energy absorbed, and design thickness for different reactor thicknesses (0.5, 1, 2, and 3 cm). Comparisons have been made between the three catalysts in the literature using the SFM when I = 1 cm (Otálvaro-Marín et al., 2014), and the same observations were found here. As shown in Table 2, Catalyst D is the best catalyst since it needs only 0.2 g/L of C_{cat} to produce 0.43 (1/cm) of $_{VRPA/I_0}$ over 86 % of the reactor thickness. It was also found in the literature that for a reactor height of 1 cm, the boundary laver of photon absorption is around 0.86 cm using different Catalyst D optical properties than those in this work (Acosta-Herazo et al., n. d.). Catalyst H comes in third position, it needs 0.4 g/L of C_{cat} to reach 0.43 (1/cm) of VRPA/Io even though it has the highest $\delta_{
m abs}$ (0.89). For reactor thickness equal to 0.5 cm, Catalyst A stands as the best photocatalyst and then Catalyst D.



Figure 7. Boundary layer of photon absorption as a function of catalyst loading (L = 1 cm).

For L = 3 cm, all the catalysts produce almost the same $VRPA/I_0$ at the same optimum Ccat (0.15 g/L) but Catalyst H performs as the best catalyst followed by Catalyst A since Catalyst H has the highest $\delta_{\rm abs}$ (2.61 cm) and Catalyst D the lowest (1.52 cm).

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Table 2. Optimum catalyst loading, maximum energy absorbed, and design thickness of flat plate reactor under solar radiation for different reactor heights (0.5, 1, 2, and 3 cm). The values in bracket in column 4 represent $\frac{\delta_{abs}}{\cdot} \times 100^{\circ}$.

catalyst	C _{cat-op} (g/L)	$V RP A_{max}/I_0$ (1/cm)	Suggested thickness (cm)	Reactor height (cm)
	0.5	0.88	0.4 (80%)	0.5
Catalyst D	0.2	0.43	0.86 (86%)	1
	0.15	0.22	1.43 (72%)	2
	0.15	0.15	1.52 (51%)	3
	0.5	0.93	0.43 (86%)	0.5
Catalyst A	0.3	0.48	0.82 (82%)	1
	0.15	0.24	1.64 (82%)	2
	0.15	0.16	1.94 (65%)	3
	0.5	0.73	0.47 (94%)	0.5
Catalyst H	0.4	0.43	0.89 (89%)	1
	0.25	0.2	1.85 (93%)	2
	0.15	0.14	2.61 (87%)	3

4. 2. Apparent Optical Thickness

The volumetric rate of photon absorption per unit of reactor length VRPA/H is another design parameter, it is obtained by integrating the LVRPA over the reactor width and height. It can also be found by multiplying the $_{V\!RP\!A}$ by the cross-sectional area of the reactor. In this work, the reactor width was supposed to be unity. For the optimization of the radiation absorption inside the flat plate reactor of any height, a new apparent optical thickness ζ_{app1} was introduced with the P1 approximation (see Eq. 24) as it was the case of ζ_{ann} with the SFM approach in the literature (Colina-Márquez et al., 2010). This dimensionless design parameter removes the dependence of the optimum catalyst loading on the reactor height and catalyst albedo as it is proved in Figure 8. Figure 8 a) shows how the VRPA/H per unit of light intensity () varies as a function of ζ_{app1} for different reactor heights with Catalyst D P25 catalyst properties taken from (Colina-Márquez et al., 2010). The optimum value of $\zeta_{app1,op}$ ($\zeta_{app1,op}$) remains almost the same (about 4.87) no matter the change in the reactor height. Figure 8 b) displays the variation

of the as a function of ζ_{app1} at different catalyst albedo but with a fixed absorption coefficient ($_{\kappa}$ =2.87 $c m^2 g^{-1}$) and $L^{=2}$ cm. It was found almost the same $\zeta_{app1,op}$ for different albedo, about 4.87. ζ_{app1} keeps the same concept as the ζ_{app} with the SFM approach, therefore, reactors working at the same $\zeta_{_{app1,op}}$ have the same behavior in terms of radiation absorption (Colina-Márquez et al., 2010; Otálvaro-Marín et al., 2014). To prove this affirmation, it was supposed a flat plate reactor working at $\zeta_{app1,op}$ =4.87 with 0.3 g/L of optimum Ccat and with ω =0.85. In this case, the reactor height was found about 1.37 cm using Eq. (24). The corresponding was established and for comparison purposes, its curve (curve in yellow in Figure 8 b) was drawn as a function of ζ_{app1} by taking ω =0.85 and $_{K}$ =2.87 $c m^2 q^{-1}$ and it was found that it matches the curve in black (see Figure 8 b) ω =0.85). This proves the independence of the optimum catalyst loading on the reactor height and catalyst albedo. $\zeta_{app1,op}$ values obtained were 4.62, 4.87, and 3.7 for Catalyst A, Catalyst D, and Catalyst H,

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respectively, almost the same values of the optimum ζ_{app} obtained (4.39, 4.08, and 3.77) with reactor height taken 1 cm by using the features displayed in Table 1. The best design conditions for different catalyst loadings and reactor thicknesses are obtained by keeping

 $\zeta_{app1,op}$ constant. For optimization purposes, it is recommended to use ζ_{app} when the reactor height is not near 1 cm, therefore, the same recommendation should be considered when using ζ_{app1} .



Figure 8. a) Effect of the apparent optical thicknesses ζ_{app1} on the $_{VRPA/H}$ per unit of reactor length and per unit of light intensity with a fixed scattering albedo of the photocatalyst at three different reactor heights. Catalyst D P25 catalyst properties were taken from (Colina-Márquez et al., 2010) (ω =0.88 and $_{\kappa}$ =1.74 $_{C}m^2g^{-1}$). The boxed point in the figure corresponds to $\zeta_{app1,op}$ = 4.87 and = 0.595 cm. b) Effect of the apparent optical thicknesses ζ_{app1} on the $_{VRPA/H}$ per unit of reactor length and unit of light intensity as a function of the scattering albedo of the photocatalyst. The specific mass absorption coefficient was kept constant (2.87 $_{C}m^2g^{-1}$), and ω was varied, with reactor height equal to 2 cm. The yellow line refers to data of the VRPA obtained using $\zeta_{app1,op}$ = 4.87 (the reactor height was found about 1.37 cm using Eq. (24)), ω =0.85 and $_{\kappa}$ =2.87 $_{C}m^2g^{-1}$. The boxed point in the figure corresponds to $\zeta_{app1,op}$ = 4.87 and = 0.653 cm.

5. PERSPECTIVES

In future works, our attention will be focusing on extending this present work to flat plate photocatalytic reactors with incident radiation varying along the reactor width and this will imply solving the RTE in 2D using the P1 approximation.

6. CONCLUSION

In this study, it was demonstrated that the P1 approximation is a viable alternative approach for

characterizing the radiant field in a flat plate reactor with any kind of catalyst present in a high-scattering medium. The radiant field in a flat plate photocatalytic reactor was described and optimized using it. The VRPA, the boundary layer of photon absorption, and the new apparent optical thickness were established for optimization purposes. These design parameters were used to find the optimum operating conditions employing the catalysts Catalyst D, Catalyst A, and Catalyst H, and comparisons were made between the three catalysts as well.

While the new apparent optical thickness established with the P1 approximation approach eliminates the dependence of the optimum catalyst loading on the reactor height and catalyst albedo, the $V\!RP\!A$ and the boundary layer of photon absorption help determine the optimal catalyst loading for any reactor height and reactor sizing, respectively. Findings are consistent with those obtained using the SFM and could help for scaling up. The significance of the plate information provided here for flat photoreactor design resides in its ability to remove the necessity for statistical analysis of experimental designs, which requires a major investment of time and resources. Establishing a rate equation to explain the kinetics of the photocatalytic degradation of different chemicals could also benefit from this.

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8. COMPETING INTERESTS

The author declares that he has no known competing financial interests or personal relationships that could have appeared to influence the work reported in this article.

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