The Turkish Online Journal of Design, Art and Communication

E-ISSN: 2146-5193



DOI: 10.7456/toidac

Research Article / Araștırma Makalesi

A NEW PERSPECTIVE ON CHINESE RED GLAZE

ÇİN KIRMIZISI SIRLARA YENİ BİR BAKIŞ AÇISI

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Received/Geliş tarihi: 04.07.2024

Benzerlik Orani/Similarity Ratio: %5

Revision Requested/Revizyon talebi: 09.08.2024

Last revision received/Son revizyon teslimi: 26.08.2024

Accepted/Kabul tarihi: 17.09.2024

Etik Kurul İzni/ Ethics Committee Permission: There is no element in the study that requires ethics committee approval. / Çalışmada etik kurul onayı gerektiren bir unsur bulunmamaktadır.

Citation/Attf: Ergül, B. &Başkırkan, H. (2024). A New Perspective On Chinese Red Glaze. The Turkish Online Journal of Design Art and Communication, 14 (4), 856-868. https://doi.org/10.7456/tojdac.1510219

Abstract

Glazes obtained by the reduction method, also called Chinese red or copper red in the literature, were first seen on porcelains of the Tang Dynasty (618-907) and Northern Song Dynasty (960-1126) periods in China. These glazes, which require a very difficult process to produce, are obtained by reducing the copper compounds in the recipe in the kiln. From its historical process to the present day, the surface of Chinese red glazes has been desired to be shiny and smooth, and their production has been carried out according to this principle. Within the scope of this study, it was aimed to bring a new artistic perspective to Chinese red glazes and instead of a smooth surface appearance, it was aimed to obtain a pinhole image known as a glaze defect. For this purpose, silicon carbide (SiC) was used to achieve the desired effect. Under normal conditions, the reduction process is carried out when the kiln cools down after neutral firing. However, it aims to give a different perspective to the traditional Chinese red glazes obtained by reduction by developing an alternative firing method caused by impossibility. To contribute to the language of discourse in art ceramics, a new proposal is presented with the interpretation of Chinese red glazes.

Keywords: Glaze, Chinese Red, Silicon Carbide, Copper Compounds, Reduction.

Öz

Çin kırmızısı veya literatürde bakır kırmızısı olarak da adlandırılan, indirgeme yöntemiyle elde edilen sırlar, ilk olarak Çin'de Tang Hanedanlığı (618-907) ve Kuzey Song Hanedanlığı (960-1126) dönemleri porselenleri üzerinde görülmüştür. Üretimi oldukça zorlu bir süreç gerektiren bu sırlar, reçetesinde bulunan bakır bileşiklerinin fırında indirgenmesiyle elde edilmektedir. Tarihsel sürecinden günümüze kadar, Çin kırmızısı sırların yüzeyinin parlak ve pürüzsüz olması istenmiş, bu prensibe göre üretimleri gerçekleştirilmiştir. Bu çalışma kapsamında Çin kırmızısı sırlara yeni bir sanatsal bakış açısı getirilmek istenmiş ve pürüzsüz yüzey görünümü yerine bir sır hatası olarak bilinen iğne deliği (pin hole) görüntüsünün elde edilmesi amaçlanmıştır. Bu amaçla istenilen etkiyi elde etmek için silisyum karbür (SiC) kullanılmıştır. Normal şartlarda nötr pişirimin ardından fırın soğumaya geçtiğinde indirgeme işlemi gerçekleştirilmektedir. Fakat indirgemeyle elde edilen geleneksel Çin kırmızısı sırlarına imkansızlığın doğurduğu alternatif fırınlama yöntemi geliştirilerek farklı bir bakış açısının kazandırılması hedeflenmiştir. Sanat seramiğinde söylem diline bir katkı sağlayabilmek amacıyla Çin kırmızısı sırlara getirilen yorumla yeni bir öneri sunulmuştur.

Anahtar Kelimeler: Sır, Çin Kırmızısı, Silisyum Karbür, Bakır Bileşikleri, İndirgeme.



INTRODUCTION

Chinese red glazes, or copper-red glazes as they are also called in the literature, are first known from porcelain examples made in the village of Cahangsha in Hunan province during the Tang Dynasty (618-907) and Northern Song Dynasty (960-1126) in China. Named after China, the geographical region where it is found, these glazes have names such as oxblood, sang de boeuf, flambé, and peach bloom depending on their appearance (Başkırkan & Arcasoy, 2020). Chinese red glazes have been important for ceramic art since the Tang and Northern Song Dynasties. Traditionally, these glazes were made by reducing copper compounds to produce bright, smooth, and red surfaces. However, in contemporary ceramic art, more innovative interpretations are needed for artists seeking not only flawless surfaces but also textured and organic surface effects.

Although Chinese red glazes are characterized by their glossy, smooth, and red surfaces, today's contemporary ceramic art requires more innovative interpretations for artists seeking not only flawless surfaces but also textured and organic surface effects.

This study aims to bring a new perspective to traditional Chinese red glazes and to evaluate glaze defects such as pinholes on ceramic surfaces as an aesthetic element. Using silicon carbide (SiC), these imperfections were deliberately created, and an alternative firing method was developed. This innovative approach offers new aesthetic and creative possibilities in contemporary ceramic art, aiming to bring together both historical and modern elements in ceramic art.

HISTORY OF CHINESE RED GLAZES

The use of the color copper red dates to the Eighteenth Dynasty in Egypt (ca. 1500 BC), according to Robert Brill. The appearance of the copper red color is associated with the processing of copper metal and the beginning of glass production during this period. The opaque red glass produced in Egypt at the time contained 5-10-15% copper oxide (Brill, 1970; Tichane, 1998).

The first copper-red glaze samples were obtained by the accidental use of reductive firing. Although these samples were red, they were not very successful. Trace amounts (2-3%) of copper oxide were used in the glazes of the period and this amount was insufficient for the formation of the copper red color (Tichane, 1998, p. 5). In the Far East, copper-red glazes are known from porcelain examples made in Changsha village in Hunan province during the late Tang Dynasty (618-907) and Northern Song Dynasty (960-1126) in China. (Wood, 2000, p. 169). The famous Chun (Jun) glazes of the Northern Song Dynasty often had red-purple-blue stains due to their high copper content (Figure I).

Some of the copper red splashes on early Chun ware are magnificent, but the colors grade off into plum, purple, and blue too. Furthermore, where the copper content has become too high, they also form blacks





and greens, when the high concentration of copper is oxidized during the cooling process

Figure 1. Jun ware jar, Song through Yuan Dynasties (960–1368) (Unm.edu, n.d.).

During the reign of Hsuan-Te (Xuande), the fourth emperor of the Ming Dynasty (1426-1435 AD), copper-red glazes covered the entire surface of ceramics, unlike in previous periods. This period is also considered to be the time when the most successful examples of copper-red glazes emerged (Figure 2). Studies at the Shanghai Institute of Ceramics have shown that the glaze composition used in this period was simple, but the firing process was quite complex. The red glaze has a copper oxide content of 0.5-1% compared to the transparent glaze content in blue and white decorated works. (Vaikner, 1997, p. 187). From the end of the 15th century until the reign of Kangxi, the fourth emperor of the Qing Dynasty (1662-1722), a great effort was made in Jingdezhen to develop copper-red glazes, including peach bloom, flambé, and langyao (Tichane, 1998, p. 6).



Figure 2. Copper red dish with Xuande (1426-1435) mark, Ming Dynasty (1368–1644), The Metropolitan Museum of Art (Metmuseum, n.d.).

In today's contemporary ceramics art, many artists develop traditional Chinese red glazes by adding their interpretations and using them in their works. For example; Tom Turner has developed a technique called "Chemical Reduction" in which he obtains Chinese red glazes through the reactions of chemicals added to the glaze recipe without the need for a reducing environment during firing (

Figure 3) (Turner, 2012, p. 96).





Figure 3. Chemically Reduction Oxblood, Tom Turner, 2014 (Facebook, n.d.).

Artist John Eagle uses Chinese red glazes, a type of glaze that is difficult to control, in a controlled and conscious way. For more than 40 years, the artist has been working with Chinese red glazes and has made them his signature, consciously using the glaze layer by layer to create various landscape images on the surfaces of his forms (Figure 4) (Eagle, n.d.).



Figure 4. Inspired by Landscape, John Eagle, 2023 (John Eagle Ceramics, n.d.).

The artist Greg Daly is particularly famous for his use of glaze techniques. He conducted research on Chinese red glazes in 1999 and has published extensively on special effect glazes. Inspired by nature and landscapes, he often uses Chinese red glazes and luster glazes in his work (Figure 5) (Daly, n.d.). Apart from Greg Daly, John Britt, Nigel Wood, and Robert Tichane have also published on Chinese red glazes.





Figure 5. Copper Red Vase, Greg Daly (Marianwilliamspottery, n.d.).

THE MAKING OF CHINESE RED GLAZES

Chinese red glazes are formed by the reduction of copper oxide (CuO), a coloring oxide. This oxide, which gives green and blue tones in oxidizing or neutral firings, gives red and red tones with the reduction of oxygen in the kiln.

The amount of copper oxide used in Chinese red glazes is quite low (0.3-1.0%), but strong and saturated red tones can still be obtained. As the amount of copper oxide used increases, the red color begins to lighten and turn green. In addition, the use of tin oxide (SnO_2) in the content of Chinese red glazes not only causes the formation of blue colors in the glaze but also strengthens the red color, resulting in reds called flambé, which also contain blue-green-purple color transitions on the surface (Özalp, 2011; Sevim, 2006).

Reductive firing is required for the formation of Chinese red glazes. Reductive firing is the reduction of high-value oxides to low-value by the reduction of oxygen in the firing environment. The reduction process is carried out by adding combustible materials such as sugar, naphthalene, burnt oil, kindling, and sawdust into the kiln during firing. In Chinese red glazes, the reduction process is usually carried out between 800-700°C after the kiln cools down. During reduction, the red color is formed as a result of the transformation of copper oxide into copper oxide and the colloidal dispersion of copper (Başkırkan & Arcasoy, 2020). In addition, tin (II) oxide (SnO) and iron (II) oxide (FeO) to be used in the glaze content increase the reduction potential, convert copper(I) oxide into metallic copper, and increase the brightness of the red color (Balyemez & Başkırkan, 2021).

EXPERIMENTAL STUDIES

This study aims to bring a new perspective to the use of Chinese red glazes in contemporary ceramic art by deliberately applying pinhole, a glaze defect, on Chinese red glazes known for their shiny and smooth surface.

Pinhole is a ceramic glaze defect that is commonly seen in ceramic glazes and is caused by the bursting of air bubbles on the glaze surface during firing. These air bubbles are formed during the removal of organic matter from the glaze and mud by burning. As the temperature rises during firing, the air bubbles grow and rise to the surface of the molten glaze. When the bubbles reaching the surface burst, small holes called pinholes are formed (Figure 6). The size and density of pinholes generally depend on the firing time and the thickness of the glaze applied. If the glaze is kept long enough, the bubbles can be



filled by the glaze, thus preventing the formation of pinholes. However, if the holding time is insufficient, cooling starts before the bubbles are filled and the pinhole problem arises. In the case of thick glaze applications, larger air bubbles are formed and as a result, larger pinholes are observed (Fraser, 2005, p. 132-135).

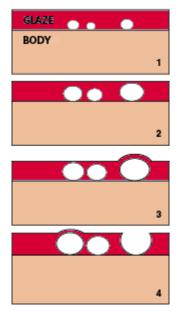


Figure 6. Pinhole formation stages (Ceramic Arts Network, n.d.).

In the personal applications, three different flambé glaze recipes were created to obtain flambé effective glazes and a total of nine different variations were obtained by making various changes in the content and ratios in these recipes. Since the pinhole effect wanted to be applied consciously and its effect wanted to be increased in the recipes created, silicon carbide (SiC) was added to the recipes to ensure this.

Silicon carbide is a compound that creates a local reduction when used at a rate of %0,2-2 in Chinese red and celadon glazes and provides a foaming and crater effect in the glaze as its rate increases in the recipe (Bloomfield, 2020). The intensity of this effect becomes more pronounced as the amount of silicon carbide added to the recipe increases. When glazes containing silicon carbide enter the melting process, the element silicon (Si) combines with oxygen (O₂) in the kiln atmosphere to form silicon dioxide (SiO₂). Following this reaction, carbon (C) reacts with oxygen to produce carbon dioxide (CO₂). The pressure created by the gases resulting from these reactions causes air bubbles to form under the glaze layer. These bubbles create pressure as they move toward the surface, causing bubbling and crater-like holes to form on the glaze surface (Avinal, 2024, p. 690).

As mentioned before, the reduction process in Chinese red glazes is normally carried out during the cooling phase of the kiln. Still, since the kiln used in this study was not suitable for this method since it was desired to make stoneware glaze at a high temperature within the scope of this study, the prepared glazes were glaze fired at 1200°C in a neutral atmosphere. The glaze-fired forms were then fired again in the enamel kiln in a reducing atmosphere (Figure 7).





Figure 7. a) Glazed forms before firing. b) Placement in the enamel kiln after glaze firing.

The glaze-fired forms were placed in the enamel kiln and the kiln was heated to 1000°C. After waiting for a while at this temperature, the kiln was allowed to cool down slowly. After the kiln was cooled down to 800°C, sugar-filled ceramic pots were placed in the kiln for reduction, and reduction was carried out in the kiln (Figure 8). The reduction process was repeated 6 times in total (Figure 9).



Figure 8. Placing the ceramic pots in the kiln.





Figure 9. Reduction process.

After the reduction process was repeated 6 times, the kiln door was closed and allowed to cool slowly. The results of the neutral firing and reduction process are presented in Table 1. with the recipes used.

Neutral Firing	No.1		Reductive Firing
	Sodium Feldspar	40	
	Quartz	26	
No.	FerroFrit 3289	14,50	
	Whiting	14	The second
	Zinc Oxide	3	
A STATE OF THE A	Kaolin	2,50	A STATE OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER
Martin States All	+	100	
Station and	Bentonite	3	State of the local division of the local div
	Red Iron Oxide	1	the state of the
	Silicon Carbide	1	ALC: NO.
	Tin Oxide	1	
	Copper Carbonate	0,40	
Neutral Firing	No.2		Reductive Firing
Concession of Concession, Name	Quartz	35	Concession of the local division of the loca
	Potash Feldspar	25	
and the second se	Kaolin	12	1.000
	FerroFrit 3289	12	and the second sec
first to the state	Dolomite	8	ALC: NOT THE REAL
and the second s	Whiting	8	
A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF	+	100	
and the second of the	Red Iron Oxide	1	Contraction of the local division of the loc
and the second s	Tin Oxide	1	and the second second
	Copper Carbonate	0,40	
Neutral Firing	No.3		Reductive Firing
	Barium Carbonate	4,04	
	Sodium Feldspar	4,62	
No. of Concession, Name	Magnesium Carbonate	4,04	
No.	Whiting	14,43	
A REAL	FerroFrit 3289	1,01	
Contraction of the local division of the loc	Potash Feldspar	41,56	ALC: NO
Concerning of	Kaolin	4,33	
Con the second second	Quartz	25,97	Successive State of State
And Andrews (1)-/	+	100	A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF A CONTRACTOR OF
and the second	Tin Oxide	0,87	and the second sec
the second second second second second second second second second second second second second second second se	Zinc Oxide	1,73	
	Ellie Ollide		
	Copper Carbonate Bentonite	0,40	

Table 1. Recipes and test results.



Neutral Firing	No.4		Reductive Firing
	Sodium Feldspar	54,50	
Name of Street	Quartz	26	
	Whiting	14	
	Zinc Oxide	3	and the second
A DECEMBER OF	Kaolin	2,50	
A CONTRACT OF A	+	100	A CONTRACT OF
	Bentonite	3	A PARTY OF A
A DESCRIPTION OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER OF THE OWNER OF THE	Red Iron Oxide	1	and the second se
the second	Silicon Carbide	1	
and a second second second second second second second second second second second second second second second	Tin Oxide	1	
	Copper Oxide	1	
Neutral Firing	No.5		Reductive Firing
The second second second second second second second second second second second second second second second se	Quartz	35	
	Potash Feldspar	37	
and the second se	Kaolin	12	and the second se
and the second sec	Dolomite	8	
A California	Whiting +	8 100	
A COMPANY OF	Red Iron Oxide	100	
Contract of	Tin Oxide	1	
All and a second second second second second second second second second second second second second second se	Copper Oxide	1	
	copper onlice	-	
Neutral Firing	No.6		Reductive Firing
	Barium Carbonate	4.04	
and the second s	Sodium Feldspar	5,63	
	Magnesium Carbonate	4,04	
A STA	Whiting	14,03	
	Potash Feldspar Kaolin	41,56	
A State of the	Quartz	4,55	
Catholic March &	Qualitz	100	
A Designed and the second second second second second second second second second second second second second s	Tin Oxide	0,87	
and a second	Zinc Oxide	1,73	
and the second s	Copper Oxide	1	and the second s
	Bentonite	1,01	
Neutral Firing	No.7	-	Reductive Firing
	Sodium Feldspar	40	
	Quartz	26	Contraction of the local division of the loc
	TF-11 Lead Frit	14,50	
A state	Whiting Zina Orida	14	
A STATE OF A STATE OF A STATE OF A STATE OF A STATE OF A STATE OF A STATE OF A STATE OF A STATE OF A STATE OF A	Zinc Oxide Kaolin	3 2,50	
Carlo Martin	+	2,30 100	Contraction of the second second
AND ADDRESS OF	Bentonite	3	
Contraction of the local division of the loc	Red Iron Oxide	1	
A DESCRIPTION OF THE OWNER	Silicon Carbide	1	
	Tin Oxide	1	
	Copper Carbonate	1	
Neutral Firing	No.8		Reductive Firing
termine .	Quartz	35	The second second second second second second second second second second second second second second second se
1000	Potash Feldspar	25	
	Kaolin	12	1 sec.
A set	TF-11 Lead Frit	12	
A CONTRACTOR OF	Dolomite Whiting	8 8	A Water and
and the second second second second second second second second second second second second second second second	whiting +	8 100	ALC: NOT THE REAL PROPERTY OF
And Barriston	Red Iron Oxide	1	
and the second se	Tin Oxide	1	
and the second s	Copper Carbonate	1	and and



Neutral Firing	No.9		Reductive Firing
	Barium Carbonate	4,04	
	Sodium Feldspar	4,62	
No. of Concession, Name	Magnesium Carbonate	4,04	
No. 1	Whiting	14,43	
	TF-11 Lead Frit	1,01	
A SHORE	Potash Feldspar	41,56	
As a second	Kaolin	4,33	All and a second
States and a	Quartz	25,97	
Contraction of the	+	100	Constant and a second
	Tin Oxide	0,87	
	Zinc Oxide	1,73	
	Copper Carbonate	1	
	Bentonite	1,01	

As a result of the tests, the silicon carbide in the recipe content was first increased up to %5 to improve the pinhole effect and copper color obtained in the glaze recipe No.7 (Table 2). Then, the proportion of copper carbonate compound in the recipe content was kept constant and copper oxide was used instead and the silicon carbide content was increased up to %5 (Table 3).

Silicon Carbide	%2	%3	%4	%5
Neutral Firing			3	
Reductive Firing				

Table 2. Tests with copper carbonate.

Table 3. Tests with copper oxide

Silicon Carbide	%2	%3	%4	%5
Neutral Firing				





Stoneware clay was used in the tests. In order to add richness to the study and to observe the reactions of the glaze as a result of interaction with different clay bodies, tests were carried out on forms shaped with red stoneware clay and porcelain clay by applying glaze code No.7 (Table 4).

Table 4. Results on different clay bodies.			
	Red Stoneware Clay	Porcelain Clay	
Neutral Firing			
Reductive Firing			

Within the scope of the study, experiments were carried out by adding silicon carbide to oxblood glaze recipes prepared with two different frits to observe the effect of silicon carbide use on other Chinese red glazes besides the Chinese red glazes called flambê (Table 5).

Neutral Firing	Oxblood Recipe F	277	Reductive Firing
	Potash Feldspar	62	
	F277 Frit	22	
and the second s	Whiting	10	A COLOR
and a state	Kaolin	4	
1 3 3 5	Quartz	2	And the Case of the
	+	100	Part of the
	Copper Carbonate	2	
Contraction of the local division of the loc	Tin Oxide	1	
ALC: NO	Bentonite	2	
	Silicon Carbide	1	
NI			
Neutral Firing	Oxblood Recipe T	FII	Reductive Firing
Neutral Firing	Potash Feldspar	FII 62	Reductive Firing
Neutral Firing			Reductive Firing
Neutral Firing	Potash Feldspar	62	
Neutral Firing	Potash Feldspar TF-11 Lead Frit	62 22	
Neutral Firing	Potash Feldspar TF-11 Lead Frit Whiting	62 22 10	Reductive Firing
Neutral Firing	Potash Feldspar TF-11 Lead Frit Whiting Kaolin	62 22 10 4	Reductive Firing
Neutral Firing	Potash Feldspar TF-11 Lead Frit Whiting Kaolin Quartz + Copper Carbonate	62 22 10 4 2	Reductive Firing
Neutral Firing	Potash Feldspar TF-11 Lead Frit Whiting Kaolin Quartz +	62 22 10 4 2 100	Reductive Firing
Neutral Firing	Potash Feldspar TF-11 Lead Frit Whiting Kaolin Quartz + Copper Carbonate	62 22 10 4 2 100 2	Reductive Firing

Table 5. Oxblood tests with silicon carbide additives.

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As a result of the tests, it was concluded that glaze recipe No.7 successfully achieved the desired effect. To observe the effect of the Chinese red glaze with pinhole appearance on large surface forms, threedimensional artistic forms were designed, and glaze application was realized (Figure 10). The glaze was applied and fired by adhering to the process followed in the experiments (Figure 11).



Figure 10. Artistic form applications.



Figure 11. Artistic form applications pinhole and copper effect details.

CONCLUSION

In this article, the types of Chinese red glazes and the variables affecting the formation of these glazes are discussed. Traditional Chinese red glazes are characterized by their glossy and flawless surface. Due to the changing aesthetic orientations in contemporary ceramic art, an artistic approach was brought to these glazes, and it was aimed to create the pinhole effect, which is considered a defect in ceramic glazes, silicon carbide was added to the glaze recipes for this purpose.

As a result of the experiments, Chinese red glaze was obtained in all recipes. In the recipes coded No.1, and No.2, the glaze shows a matte feature, but the desired flambé effects are also seen. In recipe No.3, a clear and bright red glaze very close to the glazes characterized as oxblood was obtained. In these recipes, copper carbonate was preferred instead of copper oxide.

In the second group of recipes (No.4, No.5, No.6), the ratios of the coloring oxides were equal to %1 and the ratio of the raw materials was again equal to %100 by removing the frit from the recipes. Copper oxide was used as a colorant instead of copper carbonate. The results obtained in the trials in this group are reds with more semi-matte properties and closer to burgundy. In glaze No.4, pinhole effects were observed due to the silicon carbide content. Flambé effects were again observed in the No.6 glaze.

In the third and last group of recipes (No.7, No.8, No.9), TF-11 leaded frit was added instead of FerroFrit 3289. Coloring oxides were again equal to %1 and copper carbonate was used again as colorant. In the



experiment coded No.7, it was observed that the pinhole effect seen in the previous group increased considerably. Unlike Chinese red, the interaction of silicon carbide and leaded frit resulted in the formation of very dense bubbles on the surface, and an intense copper effect was observed in them. In the application coded No.8, a semi-matte surface was obtained and very slight flambé effects were observed. No.9 was the most successful application in terms of flambé glazes. It is the closest example to flambé glazes with its brightness and turquoise color transitions on the surface.

Due to the pinhole effect observed in the recipe coded No.7, it was decided to improve this recipe and to increase the effect, the silicon carbide value in the recipe was increased up to %5 and the effects were observed. It was observed that as the silicon carbide value increased, the glaze surface deteriorated and swelled and the form lost its formal properties. Experiments were carried out with both copper carbonate and copper oxide with a fixed ratio, and it was found that the copper appearance on the surface varied depending on the copper compound.

While in traditional Chinese red applications, the reduction process is usually carried out in gas kilns, an alternative method was developed in this study due to technical limitations. In this method, ceramic forms were first fired in a neutral atmosphere in an electric kiln and then an enamel kiln was used for the reduction process. After the forms were placed in the enamel kiln, they were heated to soften the glaze and the reduction process was carried out there. This innovative approach has overcome technical limitations and achieved both high-firing stoneware glazes and the desired reduction effects.

Compared to traditional Chinese red glazes, the copper effect and the pinholes on the surface, which are characteristics of this glaze that can be described as "defective", break the molds of traditional glaze understanding and create textured glazes that are very popular in contemporary ceramics today.

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