

Evaluation of biochemical and physiological responses of *salsola* spp at their natural habitats

Doğal ortamlarında yetişen *Salsola* türlerinin biyokimyasal ve fizyolojik tepkilerinin araştırılması

Sema KARAKAS^{1*}, Murat DİKİLİTAS², Mustafa ASLAN³, Ayşe Nur GÜZEL⁴

¹Harran University, Faculty of Agriculture, Department of Soil Science and Plant Nutrition, Sanliurfa, Turkey

²Harran University, Faculty of Agriculture, Department of Plant Protection, Sanliurfa, Turkey

³Harran University, Faculty of Education, Division of Biology Education Sanliurfa, Turkey

⁴Harran University, Faculty of Agriculture, Department of Plant Protection, Sanliurfa, Turkey

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Address for Correspondence:

Sema KARAKAS

e-mail:

skarakas@harran.edu.tr

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ABSTRACT

Halophytes are naturally salt-tolerant plants that are useful for soil remediation applications. Biochemical and physiological responses of *Salsola* species, namely; *S. ruthenica*, *S. dendroides* and *S. crassa* were evaluated at their natural habitats for their salt tolerance in which the biochemical responses such as peroxidase (POX, EC.1.11.1.7) and catalase (CAT, EC.1.11.1.6) enzymes, inorganic ion accumulation, proline (Pro), malondialdehyde (MDA), chlorophyll contents (Chl-*a*, Chl-*b*) were measured. The results showed that variation existed in *Salsola* spp. in their response to salinity. Although all *Salsola* spp. showed similar trends, *S. ruthenica* expressed higher activities of antioxidant enzymes along with the higher accumulation of proline, MDA contents than those of *S. dendroides* and *S. crassa* spp. Leaves of all three species exhibited high Na⁺ content while K⁺, Ca⁺⁺ and Mg⁺⁺ contents are low. *S. ruthenica* accumulated 94.5 g kg⁻¹ DW of Na⁺ ions while *S. dendroides* and *S. crassa* accumulated 82.1 g kg⁻¹ DW and 71.4 g kg⁻¹ DW Na⁺ ions, respectively. The increase in enzymatic activities and higher metabolic contents and lower MDA levels in *Salsola* spp., especially in *S. ruthenica*, suggested that these species could well be used for phytoremediation purposes. With moderate height and root-length, these plants have high potentials to be used as companion plants with glycophytes to reduce salt stress in moderately saline conditions.

Key Words: Halophytes, *Salsola* spp., Enzymes, Salinity, Proline.

ÖZ

Halofit bitkiler toprak ıslahı uygulamaları için kullanılan doğal olarak tuza-tolerant bitkilerdir. *Salsola* türlerinin (*S. ruthenica*, *S. dendroides* ve *S. crassa*) tuz toleransına karşı biyokimyasal ve fizyolojik tepkileri doğal ortamlarında peroksidaz (POX, EC.1.11.1.7) ve katalaz (CAT, EC.1.11.1.6) enzimleri, inorganik iyon birikimi, prolin (Pro), malondialdehid (MDA) sentezi, ve klorofil içerikleri (Chl-*a*, Chl-*b*) gibi parametreler ölçülerek incelenmiştir. Sonuçlar, *Salsola* türlerinin tuza tepki bakımından farklılık olduğunu göstermiştir. Bütün *Salsola* türleri benzer tepkiler vermesine rağmen, *S. ruthenica* daha yüksek MDA, prolin ve antioksidan enzim ekspresyonları ile *S. dendroides* ve *S. crassa* türlerinden ayrılmıştır. Her üç türün yaprakları yüksek düzeyde Na⁺ içerirken K⁺, Ca⁺⁺ ve Mg⁺⁺ içeriklerinde düşüş görülmüştür. *S. ruthenica* çeşidi 94.5 g kg⁻¹ KA of Na⁺ iyonu içerirken *S. dendroides* ve *S. crassa* sırasıyla, 82.1 g kg⁻¹ KA ve 71.4 g kg⁻¹ KA Na⁺ iyonu içermektedir. *Salsola* türlerinde düşük MDA ve yüksek metabolit sentezi ile enzim artışları, özellikle *S. ruthenica*'de, bu türlerin fitoremediasyon çalışmaları için rahatlıkla kullanılabilceğini göstermiştir. Orta derecede tuzlu topraklarda tuz stresini azaltmak için ortalama boy ve kök uzunluğuna sahip bu bitkilerin arkadaş bitki olarak glifikotifler ile birlikte kullanılma potansiyeli oldukça yüksektir.

Anahtar Kelimeler: Halofitler, *Salsola* spp., Enzimler, Tuzluluk, Prolin

Introduction

Soil salinity is one of the most important plant growth limiting factors in arid and semi-arid areas as it does not only decrease the crop yield but also limits the distribution and variety of crop plants. However, there are plants which thrive under moderate and high saline conditions that most of the crop plants are not able to complete their life cycle (Acosta-Motos et al., 2017; Kaya and Inan, 2017). These plants are naturally salt-adapted halophyte plants. Adaptation of halophytes to saline conditions is mostly associated with the osmoregulation adjustment that leads to the accumulation of various compounds such as free proline and sugars (Furtana et al., 2013; Meng et al., 2018). Salt tolerance mechanism in these plants is also maintained through morphological and physiological changes (Joshi et al., 2015). For example, a study carried out with *Peganum harmala* showed that this halophyte tolerated high salinity levels in which the most crop plants could not stand this plant accumulated necessary metabolites such as protein and proline to stand up to saline conditions (Dikilitas et al., 2007). In general, halophyte species (*Atriplex* spp., *Peganum* spp., *Suaeda* spp., *Salsola* spp., *Chenopodium* spp., *Portulaca* spp.) absorb the salt ions from the soil and metabolize them (Grieve and Suarez, 1997). To overcome salinity problem, several authors have been encouraged to use these biological traits to desalinate the soil via absorption of Na^+ ions through root system of halophytes in arid and semi-arid regions where low precipitations and inappropriate irrigation systems are unable to leach salts from the rhizosphere (Hasanuzzaman et al., 2014; Karakas et al., 2017). Soil phytodesalination is based on the capacity of those halophyte plants.

Halophytes are the flora of saline soils (Flowers and Colmer, 2008). These plants osmotically adjust soil salinity by accumulating Na^+ and Cl^- ions and sequestering the great majority of these ions in their vacuoles while organic solutes are accumulated in cytoplasm to prevent the adverse

effects on cell metabolism. At high salinities, however, growth is inhibited. The harmful effects of salinity may be attributed to the toxicity to metabolism of Na^+ and Cl^- ions insufficient osmotic adjustment, reduced turgor, adverse cellular water relations in the apoplast, sub-optimal levels of K^+ ions required for maintaining enzyme activities etc. (Flowers et al., 2015). Production of reactive oxygen species (ROS) and modifications in hormonal concentrations might also take great part. The capacity of defense mechanisms in such plants totally depends on the detoxification of these harmful species via antioxidant enzymes and non-enzymatic metabolites (Mittova et al., 2003).

Removal of salts by crop plants can significantly contribute to the phytoremediation process when harvested parts are not added back to the same soil. Phytoremediation is a low-cost technology that enhances plant-nutrient availability and extends the depth of ameliorated zone as well as promoting soil hydraulic properties (Qadir and Schubert, 2003; Gharaibeh et al., 2011).

Salsola species (Chenopodiaceae) contains worldwide as many as 150 species both with herbaceous and shrubby members (Willis 1973). It is a noxious bushy summer annual plant with rigid branches (Forbes and Allred, 1999). *Salsola* species have a great tolerance to water, heat and salt stress.

Halophyte plants growing near seashores have been collected since ancient times for food, for their medicinal qualities, and for their high salt contents (Tug and Yaprak, 2017). Increases in soil salinity have awakened new interest in plant species that possess inheritance in salt tolerance (Flowers et al., 2010). Scientists have studied with quite a few plant species for phytoremediation work so far (Jesus et al., 2015).

This study aimed to screen the *Salsola* spp. at their natural habitats to determine if there are any differences in terms of ion accumulation and biochemical responses among the species. Via this approach, we aimed to remediate the saline soils with the most promising species in our

future work. Therefore, this study was centered on the evaluation of these species with regards to the activities of antioxidative enzymes, lipid peroxidation, and ion contents which might play significant roles for salt tolerance.

Materials and Methods

Site location of soil samples

Soils possessing high salinity levels were collected from the Harran Plain, Akçakkale Town (36° 42'40" N- 38° 56'53" E), Turkey (Figure 1), with 5 repetition from the root zone of *Salsola* spp. which they were identified as *S. ruthenica*, *S. dendroides* and *S. crassa* according to Davis et al., (1988). *Salsola* spp. along with soil samples were then transferred to the laboratory in ice containers for mineral and biochemical analyses there after. The samples were carefully air-dried to allow sieving with a 2-mm sieve. The electrical conductivity (EC), pH, organic matter (OM), carbonate content, cation exchange capacity (CEC), exchangeable Na⁺ cations, and soluble ions (Na⁺, K⁺, Ca⁺⁺, Mg⁺⁺, Cl⁻) amounts were determined from the saturation extract mud according to Richards (1954). Detailed chemical and physical properties of the soils are presented in Table 1.

Determination of Biochemical Parameters in the Plants

Chlorophyll contents of plants were determined according to the method of Arnon (1949). A 0.5 g leaf sample was homogenized in a 5 mL acetone: water (80:20, v:v) mixture and filtered through Whatman No.1 filter paper then placed in dark tubes. Chl *a*, Chl *b* and carotenoids of the plant samples were read at UV spectrophotometer (UV - 1700, Shimadzu) at 663.5 nm, 645 nm and 470 nm, respectively against 80% acetone blank. The results were calculated as mg L⁻¹ fresh weight and expressed as mg g⁻¹ fresh weight according to the following formula.

$$\text{Chlorophyll } a \text{ (mg L}^{-1}\text{)} = 12.7 A_{663.5} - 2.69 A_{645}$$

$$\text{Chlorophyll } b \text{ (mg L}^{-1}\text{)} = 22.9 A_{645} - 2.69 A_{663.5}$$

Proline determination was made using the procedure proposed according to Bates et al (1973). Fresh leaf samples (0.5 g) were homogenized in 3% sulfosalicylic acid and the homogenate was filtered through filter paper. Then, the filtrate was mixed in a test tube with acid-ninhydrin reagent and boiled at 100°C for one hour. The reaction was terminated in an ice bath. The mixture was extracted with toluene and the absorbance of the fraction with toluene aspired from the liquid phase was read Shimadzu UV by spectrometer at 515 nm. Proline concentration was determined using calibration curve as μmol proline g⁻¹ fresh weight.

Leaf samples (0.5) from each cultivar were homogenized with an ice-cold mortar and pestle with 5 mL of 50 mmol L⁻¹ Na- phosphate buffer (pH 6.5). The homogenates were centrifuged at 10,000 g for 10 min at 4 °C. The supernatant was then used for the determination of enzymatic activities as well as protein contents.

Peroxidase enzyme activity (EC.1.11.1.7) was determined according to the method by Cvikrova et al (1994) with slight modifications (Karakas et al., 2016). For this, 100 μl extract was added to 2900 μl reaction mixture (13 mM guaiacol, 5 mM H₂O₂ and 50 mM Na- phosphate, pH 6.5). The

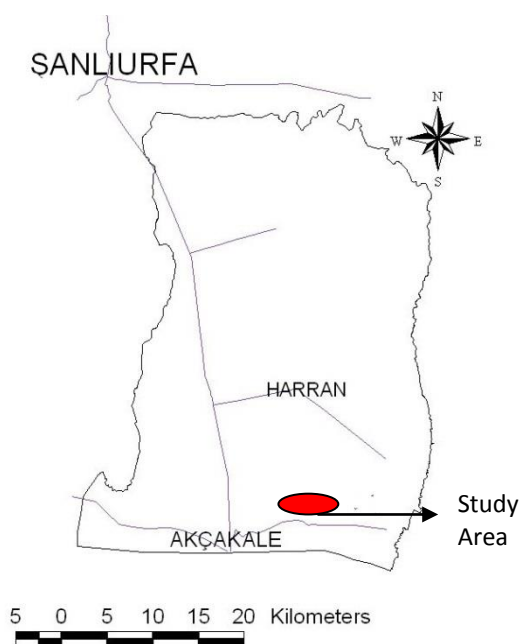


Figure 1. Study area in Harran plain
Şekil 1. Harran platosunda çalışma alanı

reaction was initiated with a H₂O₂ addition and was measured at 470 nm using a UV spectrophotometer (UV-1700, Shimadzu) at one-minute interval until 3rd minute. One unit of POX activity is defined as a change of 0.1 absorbance unit per minute at 470 nm.

Catalase enzyme activity (EC.1.11.1.6) was determined according to the method of Milosevic and Slusarenko (1996) with slight modifications (Karakas et al., 2016). For the determination enzyme activity 50 µl supernatant was added to a 2950 µl reaction mixture (10 mM H₂O₂, 50 mM Na-phosphate buffer and 4 mM Na₂EDTA) and measured at 25 °C with for 30 seconds interval at 240 nm for 1 min with a UV spectrometer (UV-1700, Shimadzu).

The malondialdehyde (MDA) content was determined according to the method of Sairam and Saxena (2000) with using slight modifications. Fresh leaf tissue sample (0.5 g) was homogenized in 0.1% trichloroacetic acid (TCA) and the homogenate was centrifuged at 10,000 g for 5 min. Thiobarbituric acid (TBA) (0.5% v/v) was added to 1 mL of the supernatant. The solution was heated at 95°C for 30 min then quickly cooled on ice. The mixture was centrifuged again at 10,000 g for 5 min and the absorbance of the clean supernatant was determined at 532 and 600 nm. Here, the MDA content of leaves is expressed as nmol g⁻¹ fresh tissue.

$$\text{MDA (nmol g}^{-1}\text{)} = \frac{\text{Extract volume (ml)} \times [(A_{532} - A_{600}) / (155 \text{ mM}^{-1} \text{ cm}^{-1})]}{\text{sample amount (g)}} \times 10^3$$

Determination of mineral contents of *Salsola* spp.

The mineral contents (Na⁺, K⁺, Ca²⁺, Mg²⁺) of leaves were determined according to the procedure of Chapman and Pratt (1961) with slight modifications (Karakas 2016). Dry plant samples (0.5) g were ground in a porcelain crucibles. The porcelain crucibles were placed into a muffle furnace, and the temperature was gradually increased up to 500 °C. The cooled ash was then dissolved in 5 mL 2 N hydrochloric acid. After 30 minutes, the volume was made to 50 mL with distilled water and supernatant was filtered

through Whatman No. 42 filter paper. Then resulting supernatant was analyzed via Inductively Coupled Plasma (ICP, Perkin Elmer).

Chloride of the plant samples was made following the method of Mohr by using K₂CrO₇ indicator. Amount of Cl was measured with the titration of AgNO₃ (Johnson and Ulrich 1959, Kacar and Inal 2008).

The amount of Na⁺ ions removed from the soil via *Salsola* spp. were calculated according to the equation made by Qadir et al. (2003):

$$S_{\text{ion-removal}} = [(S_{\text{ion-conc}})(S_{\text{DW}})/10^3]/MW_{\text{Na}}$$

Where, S_{ion-removal} is Na⁺ removal through harvested plant (mmol plant rhizosphere⁻¹), S_{ion-conc} is the Na⁺ concentration collected from plant leaves (mg kg⁻¹), S_{DW} is the plant dry weight (g plant), and MW_{Na} is molecular weight of Na⁺. Then resulting removed salt is converted kg hectare.

Statistical analysis

Data were subjected to an analysis of variance (ANOVA) at a significance level of P≤0.05 using the Duncan's Multiple Range Test (DMRT) from the SPSS software program (Version 22.0, IBM). Data are presented as a mean value ± the standard error.

Results

Determination soil physical and chemical characteristics in the study area

Soil physical and chemical characteristics, electrical conductivity (EC), pH, calcareous, organic matters, cation exchange capacity (CEC), exchangeable Na⁺, exchangeable sodium percentage (ESP), texture (sand, silt, clay), soluble ions (Na⁺, K⁺, Ca⁺⁺, Mg⁺⁺ and Cl⁻) were determined from the soil samples. The results are summarized at Table 1. According to our findings, soil EC (shows the degree of salinity), pH, calcite, ESP and CEC levels were found high while OM was found quite low. Soil texture was determined as clay.

Table 1. Soil characteristics in the study area.

Çizelge 1. Çalışma alanındaki toprak özellikleri.

Soil characteristics	Value
EC (dS m ⁻¹)	27.4
pH	7.9
calcareous (%)	25.9
OM (%)	1.1
CEC (cmol _c kg ⁻¹)	43.5
Exchangeable Na (meq 100 g ⁻¹)	14.3
ESP (%)	32.9
Sand (%)	20.5
Silt (%)	27.5
Clay (%)	52.0
Soluble Na ⁺ (meq l ⁻¹)	40.1
Soluble Ca ⁺⁺ (meq l ⁻¹)	24.3
Soluble K ⁺ (meq l ⁻¹)	1.8
Soluble Mg ⁺⁺ (meq l ⁻¹)	8.3
Soluble Cl ⁻ (meq l ⁻¹)	38.7

EC: Electrical conductivity, pH: Soil reaction, OM: Organic matter, CEC: Cation exchange capacity, ESP: Exchangeable sodium percentage.

Changes biochemical parameters in the plants

To determine the biochemical responses of *Salsola* spp. (*S. ruthenica*, *S. dendroides* and

S. crassa) chlorophyll, proline, MDA, CAT and POX levels were measured from the leaves of sampled plants.

When chlorophyll contents (Chl-*a*, Chl-*b*) and carotenoids were examined, the highest Chl *a* and Chl *b* were found in *S. crassa* as 1.3, 0.5 and 0.9 mg g⁻¹ FW, respectively, Table 2. However when proline contents were evaluated, *S. ruthenica* synthesized the highest proline content as with 7.4 μmol g⁻¹, *S. dendroides* and *S. crassa* synthesized 5.0 and 4.0 μmol g⁻¹ proline, respectively.

MDA results showed that *S. ruthenica* had the highest MDA content with 37.3 nmol g⁻¹ FW ($P<0.05$).

When antioxidant enzymes were measured, there were no significant differences among *Salsola* spp., however, *S. ruthenica* differed significantly from those of others in terms of CAT enzymes, ($P<0.05$).

Table 2. The biochemical responses of the *Salsola* species at their saline habitats.Çizelge 2. Doğal tuz ortamlarında *Salsola* türlerinin biyokimyasal tepkileri.

<i>Salsola</i> spp.	Biochemical parameters						
	Chl- <i>a</i> (mg g ⁻¹)	Chl- <i>b</i> (mg g ⁻¹)	Carotenoid (mg g ⁻¹)	Pro (μmol g ⁻¹)	MDA (nmol g ⁻¹)	CAT (ünite mg ⁻¹)	POX (ünite mg ⁻¹)
<i>S. ruthenica</i>	0.9±0.1 ^b	0.3±0.0 ^c	0.5±0.0 ^b	7.4 ±1.0 ^a	37.3±1.5 ^a	0.8±0.0 ^a	11.7±2.1 ^a
<i>S. dendroides</i>	1.0±0.0 ^b	0.4±0.1 ^b	0.7±0.1 ^b	5.0±2.0 ^b	30.0±1.1 ^b	0.4±0.0 ^b	0.9±0.6 ^a
<i>S. crassa</i>	1.3±0.1 ^a	0.5±0.1 ^a	0.9±0.0 ^a	4.0±1.0 ^b	26.7±1.5 ^b	0.4±0.0 ^b	8.0±0.5 ^a
Significance level	0.04*	0.00**	0.03*	0.00**	0.02*	0.01*	0.22 ^{NS}

Standard error was expressed in (±) number values. ^{NS} Non-significant at $P>0.05$, *Significant at $P<0.05$, **Significant at $P<0.01$. Different letters express the significant differences in columns.

Na⁺ Removal capacity of *Salsola* spp.

Salsola spp. (*S. ruthenica*, *S. dendroides* and *S. crassa*) accumulated high concentrations of salt ions their leaves. *S. ruthenica* accumulated 94.5 Na⁺ g kg⁻¹ DW leaf ash while *S. dendroides* had 82.1 g kg⁻¹ and *S. crassa* had 71.4 g kg⁻¹ Na⁺ ions. Accumulation of Na⁺ ions in leaves of *S. ruthenica* had significantly higher when compared to those of others ($P<0.05$, Table 3).

When other ions K⁺, Ca⁺⁺ were examined, *S. ruthenica* had lower concentrations than *S. dendroides* and *S. crassa*. While Mg⁺⁺ contents did not differ significantly among *Salsola* spp. It was observed that Na⁺ and K⁺ ions had an antagonistic relationship. Therefore, maintenance of salt

tolerance depends on the balance between Na⁺ and K⁺ ion contents. In our study, this level was observed in *Salsola* spp. When salt removal capacity of *Salsola* spp. was evaluated, *S. ruthenica* removed more salt as NaCl from the soil as with 787.8 kg ha⁻¹ when compared to those of *S. dendroides* and *S. crassa* as with 684.2 and 595.2 kg ha⁻¹, respectively, from the saline habitat ($P<0.05$, Table 3). This clearly showed that *Salsola* spp. had remarkable potentials in removing salts from saline areas. Only, the differences were existed among the species evidencing that *S. ruthenica* had more capacity to remove salt from the saline habitat.

Table 3. Ion contents of *Salsola* spp. and their salt removal value.Çizelge 3. *Salsola* türlerinin iyon içerikleri ve tuz uzaklaştırma değerleri.

<i>Salsola</i> spp.	Plant ion content (g kg ⁻¹)					Salt Removal _{Na} (kg ha ⁻¹)
	Na ⁺	K ⁺	Ca ⁺⁺	Mg ⁺⁺	Cl ⁻	
<i>S. ruthenica</i>	94.5±1.4 ^a	7.8±1.0 ^a	5.7±0.6 ^a	1.3±0.2 ^a	80.0±1.5 ^a	787.8±12 ^a
<i>S. dedroides</i>	82.1±1.2 ^b	11.1±1.1 ^b	6.9±0.7 ^b	1.4±0.1 ^a	65.0±2.0 ^b	684.2 ±10 ^b
<i>S. crassa</i>	71.4±2.7 ^c	12.3±1.4 ^b	8.7±0.9 ^b	1.8±0.1 ^a	60.0±2.5 ^b	595.2±22 ^c
Significance level	0.00**	0.05*	0.05*	0.10 ^{NS}	0.00**	0.00**

Standard error was expressed in (±) number values. ^{NS} Nonsignificant at $P > 0.05$, * Significant at $P < 0.05$, ** Significant at $P < 0.01$. Different letters express the significant differences in columns.

Discussion

In this study, all *Salsola* spp. synthesized CAT, POX, MDA and proline to reduce the impact of salt stress. There are distinct differences between species. Once plants were suffered from salt stress, free radicals would be accumulated, membrane permeability would be distorted and would not function properly and lose its selectivity. To prevent these occurrences, plants have to produce metabolites to remediate the physiological and biochemical disorders. For example, high concentrations of salts cause ion imbalance and hyperosmotic stress in plants (Tipirdamaz et al., 2006; Akhtar and Yun., 2017). It was previously reported that an increase in Na⁺ ion levels led to decrease in K⁺ ion concentration (Keisham et al., 2018). To ease the stress conditions, plants have to metabolize antioxidant molecules as well as accumulating toxic ions. In the case of halophytes, the accumulation of salt ions are in the prime importance when compared to the syntheses of other antioxidant molecules and enzymes. Because, halophytes are able to accumulate and metabolize salt ions rather than excluding them. Therefore, capacity of salt accumulation indicates the level of salt tolerance of the halophyte. In this study, we measured the stress-related metabolites and enzymes to see the level of stress tolerance. In our findings, *Salsola* spp. accumulated salt ions (Na⁺, Cl⁻) in their leaves on the other hand. K⁺ and Mg⁺⁺ ions contents decreased. Accumulating K⁺, Ca⁺⁺ and Mg⁺⁺ ion in leaves of *Salsola* spp. made them tolerant to salinity. In our study, *S. ruthenica* was selected as the most promising salt tolerant

candidate for remediation of saline soils. In contrast to Na⁺ and Cl⁻ ions, did not find any differences between leaf types and morphological characteristics in *Salsola* spp. suggesting that additional traits which regulate salt tolerance mechanisms may be involved in controlling Na⁺ and K⁺ homeostasis.

This and similar other attributes led researchers to suggest that halophytes could be grown in salt-affected soils to remove significant amounts of Na⁺ and Cl⁻ ions through their aerial parts (Karakas et al., 2017). It is important to note that vegetative life determines the contents of salt ions in halophyte species. Here, the vegetative life of *Atriplex* spp. is much longer than those of *Salsola* spp. Therefore, higher accumulation of ions in *Atriplex* spp. is likely to happen. For example, *Atriplex* species grown under rangeland conditions have Na⁺ ion contents in the range of 130–270 g kg⁻¹ leaf ash (Hyder, 1981) and when grown in saline soils. The species accumulated as high as 390 g kg⁻¹ Na⁺ ions in leaf ash (Malcolm et al., 1988).

Halophytes accumulating sodium salts in their shoots could be successfully used for the removal of sodium from the substrates on which they are grown, Karakas et al., (2017) estimated that *Salsola sada* L. and *Portulaca oleracea* L. were capable of removing 709 kg ha⁻¹ and 286 kg ha⁻¹, respectively. Salt accumulator plants could be very useful in saline areas.

Abiotic and biotic stress factors result in the generation of ROS such as O₂⁻ and H₂O₂ in plant cells ROS formation also accompanies normal metabolic processes in particular photosynthesis and respiration in all cellular compartments. To

ameliorate the danger posed by the presence of cellular oxidants, plant cells have evolved complex defense mechanisms (Diwan et al., 2010; Dikilitas et al., 2011). Plants possess several mechanisms that detoxify O_2^- and H_2O_2 called antioxidant systems. The primary components of antioxidant systems include non-enzymatic antioxidants (carotenoids, ascorbate, glutathione and tocopherols) and enzymes such as SOD, catalase, peroxidase (Dikilitas et al., 2011). In our experiments, we noticed that quite a high level of antioxidant enzymes and amino acids, e.g. proline were remarkably high when compared to those of *Salsola* spp. grown in low saline conditions (Heidari-Sharifabad and Mirzaie-Nodoushan., 2006; Karakas., 2013). Similar results were also obtained from those of Panahi et al. (2013) and Panahi et al. (2015) who studied *Salsola arbuscula* and *Salsola orientalis* in saline conditions.

As conclusions, these plants have been found to have significant potentials to remove salt from saline-polluted areas. These plants could grow together with glycophytes as companion plants in the same area due to its relatively short root systems compared to other halophytes in saline soils. We plan to test the performance of plants as they might have contribution to remove heavy metals and pesticide residues from the soil since they are able to synthesized high concentrations of antioxidant enzymes (CAT, POX) and low concentrations of stress molecules (MDA) when compared to those of glycophytes at the same conditions (Prasad et al., 2005; Singh et al. 2006).

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