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MICROBIAL FUEL CELL POTENTIAL IN AUGMENTING NIGERIAS' ENERGY MIX

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ABSTRACT: Nigeria generates less than 10000MW of electricity. There is a pressing demand for the increase in power generation for domestic and industrial application. Energy mix is a solution to generate more power. Different technologies harness different energy stored in different element. The microbial fuel cell (MFC) harnesses the energy from organic matter. In this study, three single chamber MFCs and an organic battery was built and organic waste matter used as an energy source to power multiple LED indicator bulbs functioning as road traffic lights. The average power generated by the MFCs and microbial fuel battery (MFB) during the period under investigation was 3.27V and 8.15V daily. Materials selected for the electrodes were cylindrical graphite rods and aluminium plates, while the organic matter used as electrolyte was a mixture of cow dung and water in the ration of 4:1 by weight. Tests and experiments indicated the MFC can be used in a redesigned road traffic light and safety signs to function from power generated from the MFC. Results indicated the use of organic waste matter in the MFC can serve to augment Nigeria's energy mix while simultaneously managing organic waste matter generated and improving the state of the economy.

Keywords: Microbial Fuel Cell, Mediatorless, Organic Matter, Power, Voltage, Current, LED bulbs, Traffic Lights, Energy Mix

1. INTRODUCTION

There is global concern about the negative effects on the environment from the continued use of fossil fuels for power generation; hence identification of renewable energy sources and the development of technology's used in harnessing such energy for power generation is of great importance to scientist and engineers globally [4,8]. Nigeria, a sub-Sahara African country with a population of over 180 million has electric power generation capacity of 10000MW and transmits about 4000MW. It has been identified that economic viability of a nation is dependent on the electric power generated, transmitted and utilised [9].

Electric power can be generated via coal, oil and natural gas; hydro-power, water turbine; nuclear power; solar-wind or water wave turbine; solar thermal generator and solar voltaic generator; however, Nigeria generates power from gas-fired and hydro-electric turbines for bulk generation. The amount of electric power generated from these sources is insufficient to meet domestic, industrial and social demand; hence other energy sources need to be harnessed to improve the power generation in Nigeria [10].

Waste is regarded as unwanted material which is disposed of or cannot be reused after primary use [17]. The amount of waste generated annually in Nigeria exceeds 30 million tonnes and over 50% is organic in nature. Organic matter can be used to generate electricity with the use

of MFC. Though the electric power generated is low, it however finds applications in wastewater treatment, environment sensors for geographical weather monitoring, bioremediation and hydrogen production. Waste generated is improperly managed in Nigeria, with more that 80% not collected for recycling or proper disposal. Failure to properly manage waste generated leads to environmental dilapidation as well as a threat to public health safety [18].

Social energy demand such as energy required for functionality of road traffic lights, road safety signs which helps check road user is often times not available. Due to the current challenges of inability to meet the energy requirement for development and its sustenance such as that required for the powering of road traffic lights and road safety signs, the Federal, State and Local government have resulted to using solar power system to meet this energy demand; however, the solar power system is under threat of theft as a result of the high cost of procurement and the high demand, and the environment as it dependent on the availability of sunlight (insolation). This leads to unchecked road traffic, use of human capital in road traffic control, thereby reducing human productivity. An alternative clean energy source that is cheap readily available and suitable is required to meet the social energy demand. The microbial fuel cell can be utilized in road traffic while simultaneously helping to manage the organic waste generated.

The microbial fuel cell is a device which is used to harness the stored chemical energy in organic matter by converting it directly to electrical energy. There are two types of MFC i.e. mediator and mediatorless MFC [3,19]. Power generated from the MFC is in microwatts but currently finds application in geographical environmental sensors. It can also find application for different social energy demand, for example, in the redesign of road traffic light and safety signs that requires micropower for operation. The MFC has an energy conversion rate of machine or device, hence, there is therefore the need to harness the chemical energy stored in organic waste matter and convert it to electrical energy in the MFC to augment the energy mix for increased power generation while simultaneously efficiently and effectively managing the organic waste generated.

While the current power generated in Nigeria is below the demand, distribution companies resolve to distributing power on an intermittent basis leading to economic stagnation. A standalone road traffic or street light powered solely from the electrical energy harnessed from a microbial fuel battery built from waste materials and agricultural waste (cow dung) will reduce the current load on the infrastructure in the power sector as there will be no direct connection with the nation's power grid. This will enable the rerouting of the power utilized for operating road traffic or street light to other production industries and technologies for economic growth and development.

The MFC technology is a maturing industry with positive prospects with regards to economic development. With further development, MFCs have the potential to produce hydrogen for fuel cells, desalinate sea water, and provide sustainable energy sources for remote areas.

2. METHODS

The materials and method applied in the execution of this work is tabulated and explained below.

Three single chamber microbial fuel cells labelled I, II and III respectively were built and connected in series. Scrap aluminium plates were collected from a manufacturing warehouse, cut and folded into multiple rectangular plates which were used as the anode electrode. Cylindrical graphite rods were purchased from a local dealer and were used as the cathode electrode in the each MFC. 8, 6 and 4 electrodes each of aluminium plates were together connected in series respectively thereby increasing the surface area. 8, 6 and 4 graphite rods were also together connected in series with the same motive of increasing the surface area. The ratio of aluminium electrode to graphite electrode used for the three cells was 2:1. The series connections of the various respective aluminium and graphite electrodes were placed in three separate plastic containers and prevented from touching each other with the use of a cut out rubber separator. Cow dung was collected from an abattoir and charged into each MFC. Functionality of the MFCs was tested with an LED indicator bulb. From experimental results, an organic fuel battery was built and used to power a multiple LED indicator bulbs arranged in the form of a road traffic light. The voltage and current output data was measured with a multimeter and the pH level of the organic matter was measured using a pen-type pH meter. The setup was monitored for a period of 28 days.

S/N	Material/Equipment	Function		
1	72 polyethylene (PET) bottles	Each PET bottle functioned as one MFC chamber		
2	Electrodes (Aluminium plates and	The electrode is to establish the presence of a potential		
	cylindrical graphite rods)	difference which the microbes transfer the released		
		electrons to.		
3	Organic matter (cow dung + blended	The organic matter is the source of fuel in the MFC.		
	fruits)			
4	Flexible copper wires	Connected between both electrodes, its function is to		
		transfer the electrons deposited at the anode electrode		
		through the equipment or machine to be powered and		
		then through to the cathode electrode.		
5	LED bulbs	Used to test the functionality of the MFC.		
6	Pen-type pH meter	Used to obtain the pH value of the organic matter.		
7	Electrical multimeter	Used to obtain voltage and current parameters of the		
		organic matter.		

Table 1. Material used and their function	Table 1.	Material	used ar	nd their	function
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3. EXPERIMENTAL

Mathematical formulae were used to obtain the values of parameters such as power and power density as expressed in (3) and (4) respectively.

Area of aluminium electrodes $(A_a) = 2 * 1 * b$ (1)

Area of graphite electrode (A_g) = $2\pi r(r + h)$ (2)

Power (P) = voltage (V) * current (I) i.e. P = VI (3)

Power density (Pd) = (Power (W)) / (Total surface area (mm2)) (4)

Volume of containers (V) = length (l) * breadth (b) * height (h) (5)

3.1 Series and Parallel Connection of MFCs

The microbial fuel cell is similar to other electrical cell. The MFC obeys the principles of electricity and therefore may be connected in same way other electrical cells are connected. Figure 1 shows the series connection of the MFCs. Equation 6 and equation 7 shows the mathematical relationship of the total voltage output obtained from the resulting series

connection. Equation 8 shows the mathematical relationship of the total current output $(MFC_{t.current})$ obtained from the resulting series connection.

Figure 2 show the parallel connection of the MFCs. Equation 9 and equation 10 shows the mathematical relationship of the total voltage output obtained from the resulting parallel connection. Equation 11 show the mathematical relationship of the total current output (MFC_{T.current}) obtained from the resulting parallel connection.

3.1.1 Series Connection



Figure 1. Series connection of the MFCs

Total MFC current output (MFC_{t.current}) =
$$\frac{1}{MFCt.current} = \frac{1}{MFCI_1} + \frac{1}{MFCI_2} + \frac{1}{MFCI_3}$$
 (7)

Total current output (MFC_{T.current}) =
$$\frac{(MFCI2 X MFCI3) + (MFCI1 X MFCI3) + (MFCI1 X MFCI2)}{(MFCI1 X MFCI2 X MFCI3)}$$
(8)

3.1.2 Parallel Connection



Figure 2. Parallel connection of MFCs

Total voltage output (MFC_{t.volt})

$$\frac{1}{MFCt.volt} = \frac{1}{MFC1.v} + \frac{1}{MFC2.v} + \frac{1}{MFC3.v}$$
(9)

$$MFC_{t.volt} = \frac{(MFC2.v X MFC3.v) + (MFC1.v X MFC3.v) + (MFC1.v X MFC2.v)}{(MFC1.v X MFC2.v X MFC3.v)}$$
(10)

3.2 Microbial Fuel Battery (MFB) Specification

In this research work, the fabricated fuel battery specification was obtained from the data collected. This is presented below. Table 2 show the volume of electrolyte in each MFC of the MFB

Number of stacked MFC = 6

Number of single cells in each microbial fuel battery = 12

3.2.1 Volume Calculation

The plastic container used for each single cell in the MFB was of the same type and dimension. Plastic containers used were a hollow rectangular shape having dimensions of 65mm by 65mm by 130mm.

Volume of single cell container = 65mm * 65mm * 130mm

= 549250 mm²

Volume of single cell electrolyte (slurry) = 65mm * 65mm * 120mm

= 507000 mm²

S/No	Microbial Fuel Cell	Volume of Electrolyte (mm ³)
1	Cell 1	507000
2	Cell 2	507000
3	Cell 3	507000
4	Cell 4	507000
5	Cell 5	507000
6	Cell 6	507000
7	Cell 7	507000
8	Cell 8	507000
9	Cell 9	507000
10	Cell 10	507000
11	Cell 11	507000
12	Cell 12	507000
	Total	6084000

Table 2. Volume of electrolyte in the MFC of the MFB

3.2.2 Electrode Specification

Total area of aluminium electrode $(mm^2) = 74496mm^2$

Total area of graphite electrode $(mm^2) = 35152.42mm^2$

Ratio of graphite to aluminium electrode = 1:2

4. **RESULTS**

Table 3. Readings from MFC 1.

Sample	The cathode is graphite and the anode aluminium. The electrolyte is cow dung mixed with water.								
Description	Average area of graphite electrode1.91 square cm								
and	Total area of graphite electrode15.26 square cm								
Parameters	Averag	e area of alu	30.	30.94 square cm					
	Total a	rea of alumir	247	247.5 square cm					
	Mass of	f dung	4kg	4kg					
	Mass of	f water adde	1kg	5					
	Total M	lass	5kg						
Day	pН	Volt (V)	Current (mA)	Power (mW)	Power Density (mW/m	nm ²) LED Bulb Status			
1	5.42	0.73	10.87	7.94	0.106	ON			
2	5.57	0.72	10.53	7.58	0.101	ON			
3	5.71	0.71	10.46	7.42	0.099	ON			
4	5.96	0.65	8.63	5.61	0.075	ON			
5	6.03	0.67	6.12	4.1	0.055	ON			
6	6.14	0.68	6.2	4.22	0.057	ON			
7	6.38	0.65	5.87	3.82	0.051	ON			
8	6.41	0.65	4.81	3.13	0.042	ON			
9	6.59	0.65	4.39	2.85	0.038	ON			
10	6.71	0.6	4.2	2.52	0.034	ON			
11	9.39	0.6	7.26	4.36	0.058	ON			
12	9.17	0.55	6.38	3.51	0.047	ON			
13	8.92	0.54	6.83	3.69	0.049	ON			
14	8.83	0.53	6.67	3.54	0.047	ON			
15	8.55	0.51	6.11	3.12	0.042	ON			
16	8.31	0.5	6.53	3.27	0.043	ON			
17	8.16	0.49	5.7	2.79	0.037	ON			
18	7.97	0.48	5.4	2.59	0.035	ON			
19	7.68	0.41	5.23	2.14	0.029	ON			
20	7.42	0.41	5.59	2.29	0.031	ON			
21	7.31	0.4	5.43	2.17	0.029	ON			
22	7.25	0.39	4.32	1.68	0.023	ON			
23	7.18	0.36	4.88	1.76	0.024	ON			
24	7.14	0.35	5.21	1.82	0.024	ON			
25	7.11	0.32	3.63	1.16	0.016	ON			
26	7.09	0.32	3.72	1.19	0.016	ON			
27	7.06	0.27	3.93	1.06	0.014	ON			
28	7.04	0.26	3.2	0.83	0.011	ON			



Figure 3. Graph of pH, voltage and current readings from Table 2.

	The cathode is graphite and the anode aluminium. The electrolyte is cow dung mixed with water							
	Average	area of graph		2.29 square cm				
	Total area of graphite electrode						13.75 square cm	
Sample	Average	area of alumi	nium electrode			32.5 squa	32.5 square cm	
Description	Total area	a of aluminiu	m electrode			195 square cm		
allu Doromotors	Mass of d	lung		4kg				
Mass of water added							1kg	
	Total Mass						5kg	
Day	pH Volt (V) Current (mA) Power (mW) Power Dem					Density	LED Bulb	
					(mW	//mm ²)	Status	
1	5.53	0.7	9.82	6.31	0	.110	ON	
2	5.61	0.69	8.29	5.72	0	.099	ON	
3	5.73	0.67	7.02	4.7	0	.081	ON	
4	5.89	0.67	6.92	4.64	0	.080	ON	
5	6.01	0.67	5.73	3.84	0	.066	ON	
6	6.23	0.65	4.82	3.13	0	0.054 ON		
7	6.47	0.65	5.88	3.82	0	0.066 ON		
8	6.65	0.64	5.61	3.59	0	0.062 ON		
9	6.86 0.64 5.83 3.59 0.062 0		ON					
10	9.13	0.63	4.83	3.04	0	.052	ON	
11	9.47	0.61	8.15	4.97	0	.086	ON	
12	9.25	0.61	6.56	4.00	0	.069	ON	
13	9.11	0.6	6.24	3.74	0	.064	ON	
14	9.03	0.59	5.18	3.06	0	.053	ON	
15	8.97	0.55	6.84	3.76	0	.065	ON	
16	8.91	0.53	6.4	3.39	0	.058	ON	
17	8.63	0.52	5.66	2.94	0	.051	ON	
18	8.42	0.48	6.2	2.98	0	.051	ON	
19	8.27	0.48	5.88	2.82	0	.049	ON	
20	7.94	0.46	5 5.11 2.35 0.041 ON		ON			
21	7.68	0.46	4.8	2.21	0	.038	ON	
22	7.31	0.41	4.75	1.95	0	.034	ON	
23	7.23	0.41	5.29	2.17	0	.037	ON	
24	7.21	0.37	5.63	2.08	0	.036	ON	
25	7.20	0.36	4.52	1.63	0	.028	ON	
26	7.19	0.32	4.4	1.41	0	.024	ON	
27	7.16	0.3	4.57	1.37	0	.024	ON	
28	7.12	0.3	3.22	0.97	0	.018	ON	

Table 4. Readings from MFC 2.



Table 4 show the maximum voltage obtained was 0.7V with a corresponding maximum current of 9.82mA. The minimum voltage and minimum current obtained were 0.3V and 3.22mA

respectively. The voltage drop was 0.4V during the period under investigation. The average voltage and average current and average power obtained were 0.53V, 5.86mA and 3.22mW respectively. Figure 4 show the curve pattern of pH, voltage and current data obtained from the experiment conducted. The graph shows the current having a fluctuating but decreasing trend. The voltage showed a steady and gradual decrease. The pH showed a steady and gradual increase until a sudden climb followed by a gradual decrease due to the change in the organic decomposition stage.

	The cat	The cathode is graphite and the anode aluminium. The electrolyte is cow dung mixed with							
	water.	water.							
	Average area of graphite electrode2.19 square cm								
Sample	Total ar	ea of graphite	e electrode		8.77 cm				
Description	Average	e area of alum	inium electrode		29.5 squa	29.5 square cm			
and	Total ar	Total area of aluminium electrode118 square cm							
Parameters	Mass of	Mass of dung 4kg							
	Mass of	water added	1kg	1kg					
	Total M	lass	5kg	5kg					
Day	pH	Volt (V)	Current	Power (mW)	Power Density	LED Bulb			
· ·	•		(mA)		(mW/mm ²)	Status			
1	5.48	0.69	9.96	6.87	0.190	ON			
2	5.66	0.67	9.76	6.54	0.180	ON			
3	5.93	0.67	9.53	6.39	0.180	ON			
4	6.27	0.67	8.82	5.91	0.160	ON			
5	6.41	0.64	7.53	4.82	0.130	ON			
6	6.54	0.63	6.63	4.18	0.110	ON			
7	6.62	0.62	6.43	3.99	0.110	ON			
8	6.73	0.58	5.32	3.09	0.085	ON			
9	6.88	0.57	6.01	3.43	0.094	ON			
10	8.76	0.57	6.15	3.51	0.096	ON			
11	9.31	0.52	8.63	4.49	0.120	ON			
12	9.11	0.52	8.15	4.24	0.120	ON			
13	8.86	0.51	6.78	3.48	0.096	ON			
14	8.77	0.51	6.16	3.14	0.086	ON			
15	8.63	0.51	6.05	3.09	0.085	ON			
16	8.52	0.5	5.58	2.79	0.077	ON			
17	8.44	0.49	6.11	2.99	0.082	ON			
18	8.19	0.47	5.23	2.46	0.068	ON			
19	7.91	0.47	5.21	2.45	0.067	ON			
20	7.83	0.45	5.09	2.29	0.063	ON			
21	7.67	0.43	4.93	2.13	0.059	ON			
22	7.48	0.42	4.51	1.89	0.052	ON			
23	7.39	0.41	4.31	1.77	0.049	ON			
24	7.31	0.41	3.88	1.59	0.044	ON			
25	7.27	0.41	3.57	1.46	0.040	ON			
26	7.22	0.38	3.31	1.26	0.035	ON			
27	7.17	0.35	3.26	1.14	0.031	ON			
28	7.13	0.34	3.08	1.05	0.029	ON			

Table 5. Readings from MFC 3.



Figure 5. Graph of pH, voltage and current readings from Table 4.

Table 5 show the maximum and minimum voltage obtained were 0.69V and 0.34V respectively. The voltage drop in the cell during the period under investigation was 0.35V. The average voltage, average current and average power obtained was 0.51V, 7.08mA and 3.30mW respectively. Figure 5 shows the curve patterns exhibited by the pH, voltage and current data obtained. The voltage curve pattern shows a gradual and steady decrease. A gradual and steady increase of pH value was observed with a sudden climb followed by a gradual and steady decrease during the period under investigation. The curve pattern exhibited by the current showed a decrease followed by a sudden climb which was then accompanied by a gradual fall. The graph shows one point of intersection between the pH and current curves. The point of intersection occurred at a value of 6.59.

From Table 3, Table 4 and Table 5, the surface area of electrode had an effect on the output of the MFC. The result indicated that the MFC with the largest surface area had the highest voltage drop of 0.47V. This could be as a result of the microbes having more electrode surface area to attach themselves to, thereby transferring more energy.

From Figure 3, Figure 4 and Figure 5 exhibited similar curve patterns indicating that the organic matter undergone through the decomposition process and also the current output is dependent on the stages of these processes.

	The cathorwater.	The cathode is graphite and the anode aluminium. The electrolyte is cow dung mixed with water.								
Sample	Average a	area of graphite electr	2.1 square cm							
Description	Total area	Total area of graphite electrode 37.78 s								
and	Average a	area of aluminium ele	31.14 square cm							
Parameters	Total area	a of aluminium electr	560.5 sc	560.5 square cm						
	Total Mas	ss of dung	12kg	12kg						
	Total Mas	ss of water added	3kg							
Day	Volt (V)	Current (mA)	Power (mW)	Power De	nsity	LED Bulb				
				(mW/m	m ²)	Status				
1	2.06	9.71	20.00	0.082		ON				
2	2.01	9.59	19.28	0.150		ON				
3	1.93	8.46	16.33	0.097		ON				
4	1.92	9.98	19.16	0.110		ON				
5	1.73	6.45	11.16	0.066	ON					
6	1.81	9.85	17.83	0.110		ON				
7	1.61	7.89	12.7	0.075		ON				
8	1.90	7.95	15.11	0.089		ON				
9	1.60	6.12	9.79	0.058		ON				
10	1.23	1.23 4.98 6.13 0.036 OI								

Table 6. Readings from MFC 1, MFC 2 and MFC 3 connected in series.



Table 6 shows the maximum voltage obtained was 2.06V. This showed that when MFCs are connected in series, the voltage increased, implying that MFCs obey the principles of electricity. The minimum voltage obtained was 0.89V. The voltage drop over the period under investigation was 1.17V. The average voltage and average current obtained were 1.21V and 6.13mA respectively. Figure 6 shows the different curve patterns obtained for the voltage, current, power and power density respectively. The curve pattern shown by the voltage reveals a gradual and steady decline with little fluctuations which then seemed to flatten out at the latter end of its curve. The power and current curve patterns have similar trend. Various points of intersection can be seen to occur in both the curve pattern of power and current having values between 3.84 and 2.23. Comparing results from Table 3, Table 4, Table 5 and Table 6, it can be seen that the voltages from MFC1, MFC2 and MFC3, gave a total combined output when connected in series considering resistance as would be expected indicating that the organic fuel cell obeys the principle of electricity.

5. DISCUSSION

Results indicate that the MFC can be used in road traffic applications. Road traffic light and indicator signs help maintain safety for road users by checking their speed, awareness and consideration of other road users. The use of MFC in enabling the continuous functionality of these road traffic light and indicator signs will help reduce road accident and the probability of it occurring. Other equipment such as reading lamps, traffic signal wand and flash lights can be designed to function on the power generated from the MFC. This will help reduce the amount of energy required and used in producing chemical cells and batteries for same purpose. The

saved energy can be redirected for other energy requirements. The MFC can also be used to meet some of the energy requirement in tourist events such as "Light Carnivals". This attracts tourists, brings in revenue as well as saves costs in using other energy source to meet same demand.

6. CONCLUSION

The current threats to public health and safety from organic waste matter as well as the economic threat as a result of the insufficient energy supply leads to economic woes. The MFC can be used to manage the organic waste matter making it an attractive energy source in augmenting Nigeria's energy mix while simultaneously helping improve the economy of Nigeria. Cow dung was investigated to examine its potential as a source for generating electrical energy with a single chamber MFC and the result indicated it is a viable resource for power generation. Results indicated that the MFC generated more power when it the organic matter used as the energy source was acidic. The surface area of electrodes also affected the power output of the system. The use of organic waste as an energy resource in the MFC makes organic waste an attractive energy resource which is readily available, hence, stand-alone road traffic lights can be redesigned to operate on the energy output from MFC, thereby making all road traffic lights run on renewable energy from organic source. Moreover, reading lamps, geographical sensors etc. can also be redesign to run on the energy produced from single chamber MFC. The MFC can help attract tourist by using it in "Light Carnivals" as well as in the decoration of trees and flowers in open parks thereby giving the environment an attracting sight. The microbial fuel cell technology is the most efficient and effective means of organic waste treatment while profiting from it.

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