



Chemical Composition of *Taxus baccata* L. Leaves and Male Cones Water: Methanol Extracts

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Abstract

The yew tree (*Taxus baccata*) is an ancient species in the world that has both toxic and medicinal properties. Identifying the chemical components of different parts of this tree can be useful in the better understanding of the toxicity and medicinal effect of this plant. Therefore, the chemical composition of water: methanol extracts of *T. baccata* L. leaf and male cones obtained from endemic species of Iran were characterized using GC-MS analysis. Twenty two components were identified for leaves including oleic acid (20.87%) and octadeca-9,12-dien-1-ol (17.77%) as the most abundant components, and seventeen components were identified for male cones which were 3-O-methyl-d-glucose (64.00%) and oleic acid (13.32%) as the most abundant components. Furthermore, the potential applications of some of the characterized components are discussed into the depths.

Keywords: *Taxus baccata* leaves and male cones, water: methanol extract, GC-MS, chemical composition, 3-O-methyl-d-glucose.

1. Introduction

Yew (*Taxus baccata*), is a coniferous tree or shrub which reaches a height of 15 meters [1] and widely distributed in western, central and southern Europe, northwest Africa, southwest Asia, and northern Iran [2-4]. Yew has evergreen leaves consist of several short, narrow needles with a length of 1 to 2 cm and a width of 1 to 2 mm [5]. Furthermore, the male cones appear in the axils of the leaves; which are globose with 6-14 peltate scales, each with 4-8 pollen sacs [6].

There have been many observations of poisoning due to the ingestion of yew in animals and humans, and the identification of its constituents can help to a better understanding of the toxicity and poisoning of this species [7, 8]. On the other hand, various medical characteristics have been linked to yew and embracing new therapeutic compounds for the treatment of diseases and a tendency to increase the use of natural compounds, the identification of yew compounds can

open a new window for human needs [9]. Many studies have been focused on the beneficial component extraction from the bark of the *T. baccata* which dries and destroys this slow-growing and low- population species. One of the means to protect this species is to use the re-growing parts of the plant and also finding and applying the component extraction method is of vital importance [10]. Plant species, habitat environment, and height and the method used to extract the components are effective factors in determining the type and amount of components [11, 12]. This study aimed to extract the leaf and male cones components of yew and identify its constituents to provide a better perspective about the *T. baccata* growing in Iran.

2. Materials and Methods

2.1 Plant Material

Fresh branches containing green needle leaves and male cones of the fallen yew tree due to wind or landslides were collected from Afratakhteh forests in Golestan

province, Iran, in October 2017. The plant material with a voucher specimen number 4657 was deposited to the Herbarium in the college of Agricultural and Natural Resources, Karaj Branch, Islamic Azad University, Karaj, Iran. The plant male cones were isolated from the end of leaves of the plant and they were both washed briefly with distilled water to remove dust and then dried at room temperature. They were kept in the refrigerator at 4 °C until extraction.

2.2 Extraction

All solvents used for extraction were supplied from Merck Company, Darmstadt, Germany. The water: methanol extracts of leaves and male cones were extracted sequentially in three steps. In the first step, fresh leaves and male cones samples (approximately 10 g of each) were separated and soaked in 150 mL of *n*-hexane in a 250-mL Erlenmeyer flask. The extraction was performed using the shaker technique for 3 h at 4.5 rpm speed. The extracts were filtered and the residue was eluted with 50 mL of *n*-hexane again. In the second step, the residue was processed similarly with the same amount of chloroform for the same period in a dark place at laboratory conditions. The filtered extracts were then evaporated under the laminar flow hood in a dark environment. In the third step, the residue was processed similarly with the same amount of water: methanol (1:1 v/v) for the same period and conditions and the extracts were kept dry in sealed Eppendorf tubes with aluminum sheets cover and stored in a refrigerator at 4 °C prior to chemical analysis. Water: methanol extracts were dried over anhydrous sodium sulfate before the GC-MS analysis.

Table 1. Identified chemical composition of *T. baccata* leaves water: methanol extract.

<i>t_r</i> * (min)	Compound	Class	WM-LE** (%)
6.367	Thymine	Phenolic compound	1.48
7.519	4H-pyran-4-one, 2,3-dihydro-3,5-dihydroxy-6-methyl-	Flavonoid	2.83
9.164	2,3-dihydrobenzofuran	Coumaran	1.93
9.366	2-furancarboxaldehyde, 5-(hydroxymethyl)-	n-Aldehyde	1.29
10.783	Phenol, 4-ethenyl-2-methoxy-	Phenolic compound	0.89
13.294	2-propenoic acid, 3-phenyl-, (E)-	Fatty acid	2.07
13.470	p-Propylguaiaicol	Monoterpene	0.50
13.854	3,5-dimethoxyphenol	Taxane	7.65
15.094	Acetic acid, (p-hydroxyphenyl)-	Fatty alcohol	9.67
16.692	3-(2-azidobenzyl)pyridine	Alkalooid	2.76
17.907	Benzeneacetic acid, 4-hydroxy-3-methoxy-, methyl ester	Fatty acid ester	1.62
18.130	1,7-dimethyl-4,4a,5,6-tetrahydropyrido-1H-[1,2-b]pyridazin-2(3H)-one	Others	2.09
18.955	Pluchidiol	Others	5.05
20.750	1,3-dioxane, 2-(2-bromoethyl)-	Ether	2.96
21.326	Palmitinic acid	Fatty acid	1.80
21.762	2-methyl-1-thia-cyclopentane	n-Alkane	8.87
23.744	Oleic acid	Fatty acid	20.87
23.988	Octadecanoic acid	Fatty acid	1.34
26.328	Fusaric acid	Aromatic carboxylic acid	1.39
29.908	Octadeca-9,12-dien-1-ol	Fatty alcohol	17.77
31.252	3-phenyl-1,4(E)-dodecadiene	Others	0.60
33.929	Cholest-5-en-3-ol (3.beta.)-	Steroid	1.71

**t_r*: Retention time; **WM-LE: Water: methanol leaves extract.

2.3 GC-MS Analysis

The GC-MS analysis of the resulting extracts was performed using a GC Agilent 7890A and MS Agilent 5975C mass spectrometer detector (Palo Alto, CA, USA) equipped with a HP-5MS cross-linked capillary column (30-m-long and 0.25-mm internal diameter, 0.25 μm film thickness). Helium was used as the carrier gas with a flow rate of 1 mL/min. The run time duration was 48.43 min. The program began at 60 °C for 2 min and the temperature increased at a rate of 7 °C/min up to 280 °C. It remained at this temperature for 15 min. The intrinsic energy that hits the sample in the MS system was 70 eV. The split ratio of the sample was 2:1 with a split flow of 2 mL/min. The individual compounds in the extracts were identified by their retention time relative to known compounds and further identified by comparison of their mass spectra with either the known compounds or published spectral data. Individual components were identified using Wiley 275 L and NIST05 a.L database matching, and by comparing the retention times and mass spectra of constituents with published data [13-15].

3. Results and Discussion

3.1 Leaves and Male Cones Extracts

The number and percentage of the total compounds in the leaves and male cones water: methanol extracts which were identified using GC-MS analysis, were 22 and 97.14% (Table 1), and 17 and 99.29 %, respectively (Table 2).

Table 2. Identified chemical composition of *T. baccata* male cones water: methanol extract.

<i>t_r</i> * (min)	Compound	Class	WM-MCE** (%)
6.336	Glycerol	Alcoholic sugar	0.48
9.195	2,3-dihydrobenzofuran	Coumaran	1.88
9.895	1,2-Benzenediol, 3-methoxy-	Phenolic compound	1.17
15.131	3-Hydroxyphenylacetic acid	Fatty acid	0.69
16.703	Methyl-(2-hydroxy-3-ethoxy-benzyl)ether	Ether	0.61
18.145	Mome inositol	Sugar	0.66
18.929	2-ethylthiolane	Tioalkaloeid	0.43
19.318	Propyl isopropyl ether	Aromatic ether	0.41
20.044	3,5-Heptadienal, 2-ethylidene-6-methyl-	n-Aldehyde	2.66
22.447	3-O-methyl-d-glucose	Sugar	64.00
23.723	Oleic acid	Fatty acid	13.32
23.982	Stearic acid	Fatty acid	0.77
27.028	Methyl petroselinat	Fatty acid ester	0.75
27.983	Hexadecanoic acid, 2,3-dihydroxypropyl ester	Fatty acid ester	0.64
29.908	9,12-Octadecadien-1-ol	Fatty alcohol	7.70
33.929	Cholest-5-en-3-ol (3.beta.)-	Steroid	1.10
36.259	Stigmasterol, 22,23-dihydro-	Steroid	0.73

* *t_r*: Retention time; **WM-MCE: Water: methanol male cones extract.

Table 3. Classification of the identified chemical components of the *T. baccata* leaves and male cones water: methanol extract.

No.	Chemical classes	WM-LE* (%)	WM-MCE** (%)	Sum (%)
1	Sugars	0.00	65.14	65.14
2	Acids	37.14	14.78	51.92
3	Alcohols	17.77	7.70	25.47
4	n-Alkanes	8.87	0.00	8.87
5	Taxanes	7.65	0.00	7.65
6	Ethers	2.96	2.19	5.15
7	Flavonoids	2.83	1.29	4.12
8	n-Aldehydes	1.29	2.66	3.95
9	Coumarans	1.93	1.88	3.81
10	Steroids	1.71	1.83	3.54
11	Alkaloids	2.76	0.43	3.19
12	Esters	1.62	1.39	3.01
13	Phenols	2.37	0.00	2.37
14	Terpenes	0.50	0.00	0.50
15	Others	7.74	0.00	7.74
-	Total identified	97.14	99.29	-

* WM-LE: Water: methanol leaves extract; **WM-MCE: Water: methanol male cones extract.

Through analyzing the qualitative characteristics of flavonoids in yew female cones (seed cones), the researchers showed that there were several flavonoids in methanol extract and the highest concentration of flavonoids (39.37 mg/g) was measured in ethyl acetate extract of seed cones [16]. Compared to the values obtained in leaves extract, seed cones extract had a lower concentration of flavonoids. The results of the data and identified compounds of our studies showed that the compounds 4H-pyran-4-one, 2,3-dihydro-3,5-dihydroxy-6-methyl- (called DDMP) as only flavonoids [17] accounts for only 2.83% in the water: methanol extracts of yew leaves. The identified DDMP is an anti-tumor compound. Moreover, it possesses anti-microbial, anti-oxidant, anti-inflammatory, and anti-mutagenic

characteristics [17]. The highest extraction percentage was 3-O-methyl-d-glucose by 64.00% and extracted from the male cones using water: methanol. 3-O-methyl-d-glucose is a nonmetabolizable chemical analog of glucose. Because of its metabolic stability, the early studies investigated the 3-O-methyl-d-glucose as a cellular transport, blood-brain barrier, and tissue apportionment expanse of hexoses [18]. In the more recent researches, 3-O-methyl-d-glucose have been used in cancer identification and track down [19]. Fusaric acid also known as 5-butylpicolinic acid extracted from the leaves using water: methanol as much as 1.39%. Fusaric acid has been known as a wilting agent and also proposed for various therapeutic applications such as quorum sensing inhibitors [20]. Fusaric acid is a

mycotoxin and has negative effects on mammals and also it prevents dopamine beta-hydroxylase enzyme. On the other hand, fusaric acid can prevent cell proliferation and DNA synthesis. However, it is primarily used in research and laboratories.

2-furancarboxaldehyde, 5-(hydroxymethyl)- which is a member of the class of furans, obtained from the leaves by water: methanol at 1.29%. 2-furancarboxaldehyde, 5-(hydroxymethyl)- does not exist in the fresh food, and it is naturally produced (through Millard reaction) from the processing (cooking, drying, or storing) of foods containing sugar. It is used as an index of heat treatment and deterioration in food and products such as tomato paste, honey, and fruit juices [21]. 2-furancarboxaldehyde, 5-(hydroxymethyl)- also has shown hepatoprotective effects on acute alcohol-induced liver oxidative injury in mice [22]. Additionally, 2-furancarboxaldehyde, 5-(hydroxymethyl)- possesses antioxidant properties and prevents the sickling of red blood cells [23-25]. In the yew male cones and leaves, oleic acid was the only fatty acid component which could be extracted using water: methanol (Tables 1 and 2). In this experiment, 2,3-dihydrobenzofuran, also known as coumaran, was identified in both leaves (1.93%) and male cones (1.88%) using water: methanol. 2,3-dihydrobenzofuran has shown effective results against cancer, tuberculosis, malaria, and cataracts [26]. Moreover, some plant species that are rich in 2,3-dihydrobenzofuran possess antioxidant and/or cytoprotective characteristics and insecticidal features [27, 28]. Some of the common uses of oleic acid are in the food, pharmaceutical, cosmetic, and biodiesel industries [29]. Recently, oleic acid has attracted attention due to its positive effect on human disease such as blood pressure [30].

4. Conclusion

In this study, the chemical composition of leaves and male cones extracts of *Taxus baccata* L. were sequentially extracted by water: methanol and then analyzed using the GC-MS technique. In total, identified components are classified as sugars, acids, alcohols, n-alkanes, taxanes, ethers, flavonoids, n-aldehydes, coumarans, steroids, alkaloids, esters, phenolic compounds, terpenes, etc. There were many valuable phytocomponents that are potential bioresources for phytopharmaceuticals such as 3-O-methyl-d-glucose, 4H-pyran-4-one, 2,3-dihydro-3,5-dihydroxy-6-methyl-, thymine, 2,3-dihydrobenzofuran, oleic acid, and glycerol. Alcohols were identified with almost high percentages in this experiment, the total extraction of alcohols from leaves was 17.77% and from male cones was 7.70%. The other compounds with high extraction percentage were acids, their total extraction using water: methanol solvent from leaves and male cones extracts were 37.14% and 14.78%, respectively. Furthermore, taxanes were extracted as much as 7.65%

from leaves extract, while their extraction from male cones extract was 0.00%. Also, the sugar component there was not in the leaves extract by water: methanol extract, while total sugar identification from male cones reached at 65.14%.

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Author's Contributions

Younes Shirmohammadli: Developed and performed the experiments and drafted the manuscript.

Seyyed Khalil Hosseinihashemi: Conceived of the presented experiment and supervised the findings of this work.

Abbas Jalaligoldeh: Verified the analytical methods.

Davood Efhamisizi: Aided in interpretation of the results and consulted in technical details.

Seyyed Hashem Mousavinezhad: Assisted with laboratory experiments.

Amir Lashgari: Assisted in sourcing materials and planning of the laboratory experiments.

Ethics

There are no ethical issues after the publication of this manuscript.

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