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Original Research Article

Investigation on effects of the exhaust emission characteristics of diesel engine fuelled with mahua oil methyl esters and its blends with diesel



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ABSTRACT

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Diesel engines are deliberately known for environmental pollution caused by exhaust emissions and are liable for many health issues further more. Emissions of diesel exhaust might lead to cancer, irritate the eyes, affects the nose, throat and lungs. Biodiesel is an eco-friendly and typical alternative fuel which is produced from vegetable oils through transesterification technique. Biodiesel has its own several advantages than the conventional diesel says decreased in carbon monoxide (CO), unburnt hydrocarbon (HC) and particle matter (PM) emissions, and having fuel properties which are similar to conventional diesel for its easier use in diesel engines. The outcomes demonstrated that the production of lesser CO and HC emission using biodiesel. Nevertheless, a minor increment in NO_x emission was noticed for mahua oil methyl ester blends. Brake Specific Fuel Consumption (BSFC) for methyl ester blends was raised as compared to diesel. The combustion analysis exhibited a significant rise in combustion pressure and heat release rate with smaller ignition delay and extended combustion period. From the investigation, it could be said that the effects of mahua oil methyl esters and its blends on diesel engine when compared to conventional diesel ultimately minimizes the health effects of biodiesel exhaust exposure.

Keywords: Mahua oil methyl esters; Emission Characteristics; Diesel Exhaust

1. Introduction

Air pollution is now completely recognized as significant public health problem, liable for a growing with variety of health effects associated with air pollution have been well documented from the studies conducted in several regions of the world. Diesel engines have great utilization in comparison to gasoline engines attributable to their low operating costs, high durability and

reliability. They have a significant impact upon environmental pollution problem worldwide. Especially, diesel exhaust contains higher amount of particulate matter and NO_x emissions which can be accountable of extreme environmental and health problems [1]. The diesel combustion in an uncontrolled manner releases harmful particulate matter, increased NO_x and various cancer-causing substances.

Due to speedy increase of motorized vehicles and really restricted use of emission control technologies, transport emerges as the major source of urban air pollution, which is a crucial public health problem in many urban communities of developing countries like India. In developing countries, air pollution accounts for thousands of excess deaths and loses in productivity and billions of dollars in medical expenses annually [2]. The World Health Organization (WHO) estimated that approximately 2.4 million people die every year due to air pollution [3]. Of the toxins considered, carbon monoxide (CO) has critical impact on asthma for children of ages 1 – 18: when compared with the 1992 pollution levels, there could be a 5 - 14% increase in asthma admissions for the 1998 pollution levels. It is significant to represent for these endogenous responses when measuring the impact of pollution on health [4]. In addition, it will cause severe coughing, headaches and nausea. Diesel exhaust causes inflammation in the lungs which could worsen chronic respiratory symptoms and increase the frequency or intensity of asthma and could increase the risk of heart problems and premature death. Studies have shown that exposure to diesel exhaust emissions cause respiratory problems and lung damage, and there are solid confirmations that diesel emissions may cause cancer in humans [5, 6]. The feasible solution for the collective issues of energy demand and environmental pollution is replacement of fossil fuels with renewable fuel with reduced emissions [7]. A multi-sectional methodology, participating relevant bodies like transport, housing, industry, is required to develop and viably execute long-term policies and measures to diminish air pollution and also the associated risk to human health. Biodiesel is a realistic alternative to mineral diesel fuel; it is a fuel made up of renewable resources and has lower exhaust emissions than mineral diesel. When contrasted with fossil fuels; biodiesel also considerably reduces the quantity of carbon dioxide being released into the surroundings by diesel engine vehicles. In addition, biodiesel is an oxygenated fuel, thus it contributes to additional complete fuel burn and significantly improves the profile of emissions of a diesel engine. The more biodiesel mixed in a blend, the substantial reductions in exhaust emissions. The

literatures indicated that biodiesel will lead to 50-80% decline in carbon dioxide emissions as associated to mineral diesel [8, 9]. One of the distinctive benefits of biodiesel is that it considerably reduces air pollutants, like sulfur oxides and carbon oxides, which are related to diesel exhaust emissions and are suspected of causing cancer and other human health problems. The combustion of mineral diesel produces several harmful emissions like NO_x , CO, CO_2 , and HC [10]. Devan and Mahalakshmi [11] studied the performance, combustion and emission characteristics of a diesel engine on Methyl Ester of Paradise oil and its diesel blends. They have reported that significant reduction in smoke and unburned hydrocarbon emissions was found. Suryanarayanan et al. [12] investigated the performance and emission characteristics of methyl esters of various vegetable oils and compared with the mineral diesel. Saravanan et al. [13] directed the tests on a single cylinder diesel engine with mahua oil methyl ester blends. They revealed that the CO, HC emissions for mahua ester was diminished contrasted with diesel by 26% and 20% individually. NO_x was decreased by 4% for the ester contrasted to diesel. Nantha Gopal et al. [14] inspected the emission and combustion attributes of pongamia methyl ester on diesel engine. It is noticed that there was substantial decrease in CO, HC and smoke emissions for all PME mixtures when contrasted with diesel. NO_x emission for PME was marginally greater than that of diesel. Arunprasad et al. [15] researched the impact of injection timing and injection pressure on the performance and emission attributes of diesel engine with biodiesel blends. The outcomes demonstrated that decrease of 5.08% of brake specific fuel consumption (BSFC) was observed. The decline in 29.17%, 53.85% and 21.95% of HC, CO and smoke emissions was noticed and furthermore substantial rise in NO_x emission at full load was noticed. Nanthagopal et al. [16] considered the impacts of including zinc oxide and titanium dioxide nanoparticles to biodiesel mixes on a diesel engine performance and emission attributes. The outcomes revealed that the engine emissions of NO_x , HC and CO were diminished considerably by 29%, 40% and 40% individually. There was likewise a substantial

reduction in smoke, HC and CO emissions for higher volume proportions of biofuel excluding NO_x emission. The benefit of utilizing biodiesel in India is development in agriculture and rural economy, usage of waste land and decrease in crude oil import bill. The aim of the present investigation is to examine the exhaust emission attributes of diesel engine fueled with mahua oil methyl esters and its blends with diesel.

2. Preparation of Biodiesel and Its Blends

Biodiesel is an alternate fuel, which incorporates a correlation with sustainable development, environmental preservation, energy conservation and management. The biodiesel is obtained from mahua oil by transesterification process. The mahua oil was

chemically reacted with methanol in the presence of catalyst (NaOH) to produce methyl esters. The mixture was heated at a temperature of 65°C and reaction time of 2 hours and then it was allowed to settle down under gravity. Two layers were observed after the cooling. The upper layer was identified as methyl ester and then the methyl ester was washed with warm water to remove the catalyst present in the ester. The mahua methyl ester so prepared was blend with mineral diesel in different proportions (25%, 50%, 75%, 100%) by volume (B25, B50, B75 and B100) with help of a magnetic stirrer. The properties of prepared methyl esters and its blends with diesel measured as in line with ASTM are given in Table 1.

Table 1. Properties of Mahua methyl esters and its blends, diesel

Properties	Unit	Diesel	Mahua methyl ester	B25	B50	B75
Density (at 15°C)	kg/m^3	821	872.5	848	862	872
Kinematic viscosity (at 40°C)	mm^2/s	2.57	4.36	2.89	3.47	3.76
Calorific value	kJ/kg	42960	36915	41750	39215	37425
Cetane number	---	46	53	51.5	52.3	52.8
Flash point	$^\circ\text{C}$	65	130	103	115	121
Pour point	$^\circ\text{C}$	-18	5	-3	2	4.5

Table 2. Specifications of test engine

Parameters	Specification
Model	AV1 (Kirloskar made)
Engine type	Vertical, Single Cylinder, Water cooled, Four stroke, Direct injection, compression Ignition Engine
Maximum power	3.7 kW at 1500 rpm
Bore X Stroke	80 mm X 110 mm
Displacement volume	553 cc
Compression ratio	16.5:1
Fuel injection timing	23° before TDC
Injector: hole X diameter	3 X 0.25 mm
Dynamometer	Eddy current dynamometer

3. Experimental Setup and Procedure

The experimental investigation was carried out on a single cylinder, four stroke, naturally aspirated, water cooled constant speed, direct injection diesel engine. The main specifications of the engine test rig are given in Table 2 and photographic view of experimental set up is shown in Figure 1. The engine was loaded by an Eddy current dynamometer through computer. The engine was operated to have a constant speed of 1500 rpm.

The engine had a typical fuel injection system. Injector was given three holes and each having an orifice diameter of 0.25 mm. The injection pressure and the static injection timing as stated by the producer was 205 bar and 23° bTDC

correspondingly. The pressure sensor was used to measure the cylinder pressure. The engine speed, crank angle was measured using sensors in data-acquisition system which is connected through the computer. Air consumption was measured by means of sharp edged orifice plate and U-tube manometer.



Figure 1. Photographic view of experimental set up

Fuel consumption was measured by using burette and stop watch. Different digital thermocouples were used to measure the temperatures of various salient points. The amount of exhaust emissions like CO₂, CO, HC and NO_x were measured using exhaust gas analyzer (Make: AVL; Model: DiGas 444). The smoke density was measured using smoke meter (Make: AVL; Model: 437). The EGR set up was installed, which recirculates the exhaust gas back into the air-intake manifold. At the completion of every test, the engine was worked with diesel under idle load at a speed of 1500 rpm for ten minutes to confirm that the engine to confirm that the engine fuel system was free from any residuals of the earlier test fuels. The engine was sufficiently warmed up for each test. The whole investigation has been run at an atmospheric temperature of 298 K and a relative humidity of 70%. The test procedure is repeated for three times and the arithmetic mean values of these three readings were used for calculation and analysis purpose. In this present study, the biodiesel and its blends (Diesel, B25, B50, B75 and B100) were studied at different engine loads from 0% to 100% of rated engine load. Besides, investigational uncertainty has been considered

in this study in order to measure the error that happens owing to the decision of technique embraced at an ambient condition. The uncertainty particulars of the instruments utilized as a part of investigation are given in Table 3. The investigational uncertainty has been learned with the aid of propagating errors method based on Taylor's theorem.

The proportion uncertainties of various physical parameters such as brake thermal efficiency, brake specific fuel consumption, and emission parameters such as smoke, NO_x, HC and CO were measured with the percentage uncertainties of various instruments. Overall uncertainties were specified by the successive equation.

$$\begin{aligned} \text{Overall uncertainty} &= \text{Square root of} \\ &[(\text{Uncertainty of brake thermal efficiency})^2 + \\ &(\text{Uncertainty of brake specific fuel consumption})^2 + (\text{Uncertainty of pressure transducer})^2 + (\text{Uncertainty of crank angle encoder})^2 + (\text{Uncertainty of HC})^2 + (\text{Uncertainty of CO})^2 + (\text{Uncertainty of NO}_x)^2 + (\text{Uncertainty of smoke})^2] \\ &= \text{Square root of} [(1.5)^2 + (1.2)^2 + (0.2)^2 + (0.2)^2 \\ &+ (0.2)^2 + (0.2)^2 + (0.2)^2 + (1.0)^2] \\ &= \pm 2.21 \end{aligned}$$

Table 3 Uncertainty of the instruments at typical operational conditions

Instrument	Range	Accuracy	Uncertainties in %
Exhaust gas analyzer			
HC (ppm)	0 – 20,000	±10	±0.2
NO _x	0 – 5000	±20	±0.2
CO (%)	0 – 10	±0.03	±0.2
CO ₂ (%)	0 – 10	±0.02	±1.0
Smoke meter (HSU)	0 – 100	±0.2	±1.0
Crank angle encoder (°CA)		±1.0	±0.2
Measuring Burette (cc)	0 – 30	±0.25	±1.0
Pressure transducer (bar)	0 – 110	±1.0	±0.2
Thermocouple (°C)	0 – 1500	±1.0	±0.15
Speed sensor (rpm)	0 – 20,000	±10	±0.1

4. Results and Discussion

4.1. Brake specific fuel consumption

BSFC is the amount of fuel provided to the engine for produce the unit power output [17]. The BSFC of any fuel relies upon the density, heat energy content and viscosity. The BSFC of biodiesel mixes diminishes with consistent increment in load. The BSFC of diesel is found to slighter than mahua methyl ester mixes at all engine loading conditions. The notable explanations for rise in BSFC for mahua oil methyl ester are inferior heating value, upper

density and viscosity than diesel. The rise in BSFC with an increment in biodiesel mix can be owing to the extra fuel utilization of biodiesel by engine for keeping power output at a consistent level [18].

4.2. Cylinder pressure

The variations in cylinder pressure with crank angle for different test fuel blends shown in figure 3. Ignition delay is a significant constraint in the combustion process. Ignition delay period impacts the combustion phase in the engine cycle, the thermodynamic effectiveness and

degree of premixing of fuel vapour. The ignition delay displays an inferior value even though utilizing methyl esters mixes than diesel fuel while methyl ester has upper viscosity and lesser volatility properties [19]. The cylinder pressure for methyl ester mixes is marginally greater than that of diesel. The peak cylinder pressure relies chiefly on the initial stage combustion rate which is affected by the fuel taking part in the uncontrolled combustion stage. The peak cylinder pressure was noticed for methyl ester mixes.

The cause for the higher cylinder pressure is combustion enhancement which outcomes from its upper cetane number and oxygen content of the methyl ester mixes than diesel. The oxygen content of methyl ester blends increases the oxidation rate and supports the finishing of the initial stage of diffusion combustion process in the fuel rich zones and outcomes in upper cylinder pressure utilizing methyl ester blends [20].

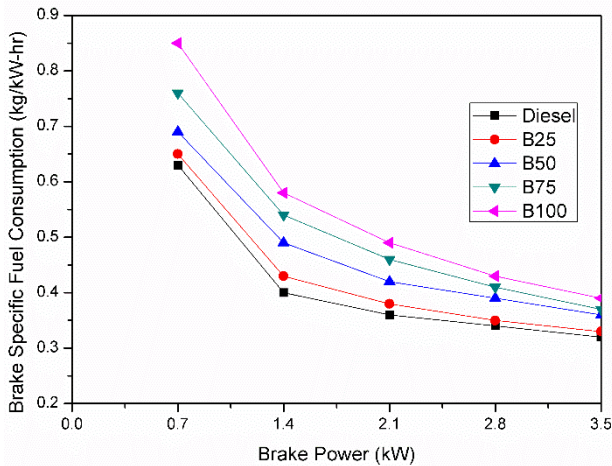


Figure 2: Variation of Brake specific fuel consumption with Brake power

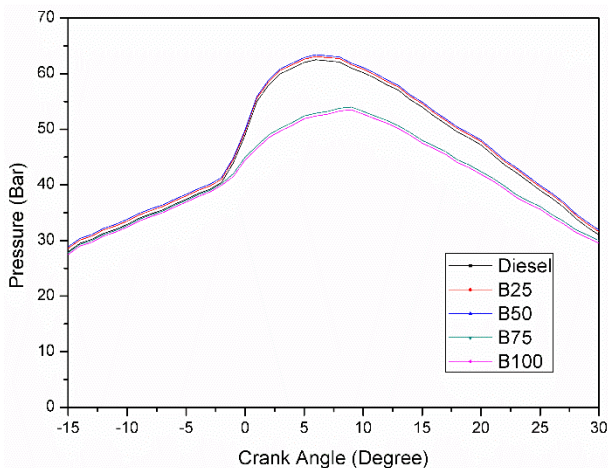


Figure 3: Variation of Cylinder Pressure with Crank angle

4.3. Heat release rate

The heat release rate gives the combustion process initiation and the quantity of fuel burnt in the premixed combustion stage [21]. Figure 4 shows the variations in heat release rate with crank angle for different test fuel blends. The heat release rate curve illustrates a lessening in heat release rate in the premixed combustion stage at the initial phases for methyl ester blends. It is owing to the shorter ignition delay and inferior heating value of methyl ester blends. The upper heat release rates for methyl ester blends could be owing to oxygen content of the methyl ester mixes which enhance the oxidation of fuel particles that prompts to the finishing of initial phase of diffusion combustion, consequently raising the heat release rate [22].

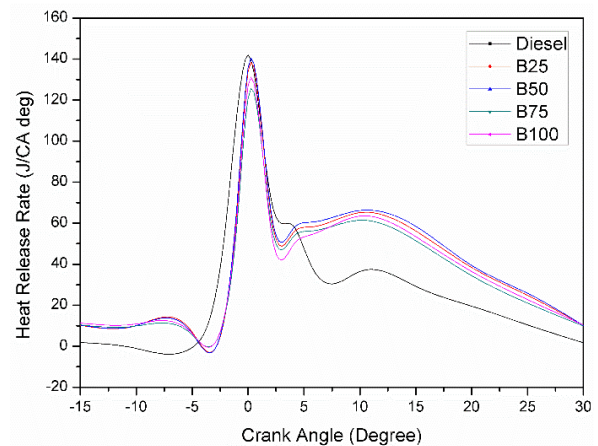


Figure 4: Variation of Heat release rate with Crank angle

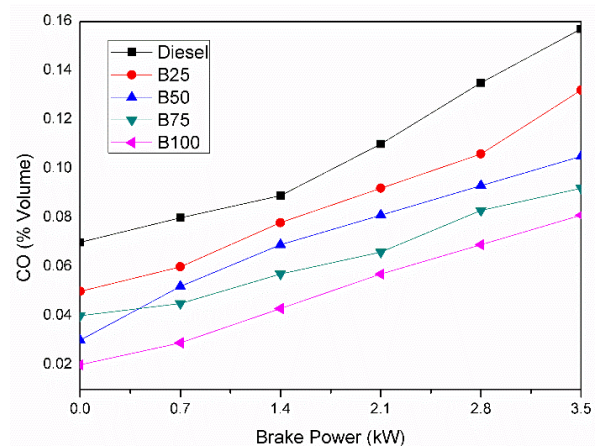


Figure 5: Variation of CO emission with Brake Power

4.4. Carbon monoxide emission

The variation of carbon monoxide (CO) emission with brake power for methyl ester blends and diesel is presented in figure 5. It is observed that the CO emissions for mahua methyl ester and its blends are lesser than that of

mineral diesel fuel. These lower CO emissions of mahua methyl ester blends could also be because of their additional complete oxidation as compared to diesel. The establishment of CO emissions is owing to unfinished combustion process and the inaccessibility of oxygen in the combustion chamber. Some of the CO formed during combustion of methyl ester blends might have converted into CO₂ by taking up the more oxygen molecule present in the methyl esters structure and thus reduces CO formation [23]. Therefore, the combustion is enhanced with the extra number of oxygen particles and complete combustion process. This reason is recognized to be main purpose for lessening of CO emission using methyl ester mixes [24]. The reduction in the concentration of CO emission is about 40% compared to diesel. The same trends were also observed in other studies [25, 26].

4.5. Carbon dioxide emission

Figure 6 depicts the carbon dioxide (CO₂) emission with brake power. The CO₂ emission plays a main part in global warming. The CO₂ emissions from a diesel engine show how efficiently the fuel is burnt inside the combustion chamber. It is seen that CO₂ increases with increasing load for all the blends of methyl esters. This is due to the occurrence of oxygen. In the case of methyl esters, additional oxygen present helps combustion compared to diesel. Hence, CO₂ emission in the exhaust is also more than that of diesel. Similar conclusions were stated by Ramadhas et al. [27] when the engine was tested with methyl ester of rubber seed oil.

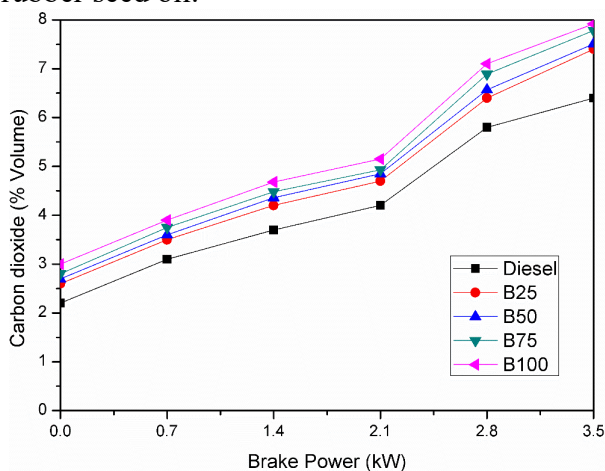


Figure 6: Variation of CO₂ emission with Brake Power

4.6. Hydrocarbons emission

The variation of hydrocarbon (HC) emission

with brake power for methyl ester blends and diesel is presented in Figure 7. HC emission is one of the organic mixtures and the outcomes of incomplete combustion process. There is a decrease in HC emission with the mahua methyl esters and its blends when compared to mineral diesel. The reduction in concentration of HC is almost 35% lesser compared to diesel. The diminution in HC emission utilizing mahua methyl esters mixes is mostly owing to the oxygen content of the methyl ester which outcomes in a cleaner and complete combustion [28].

The upper cetane number of mahua methyl ester reduces the ignition delay period and motivates the decrease in HC emission [29]. The HC emission increases with increase in engine load and decreases with increase in percentage of methyl ester in blend. The results were almost in line with the other reports [30, 31].

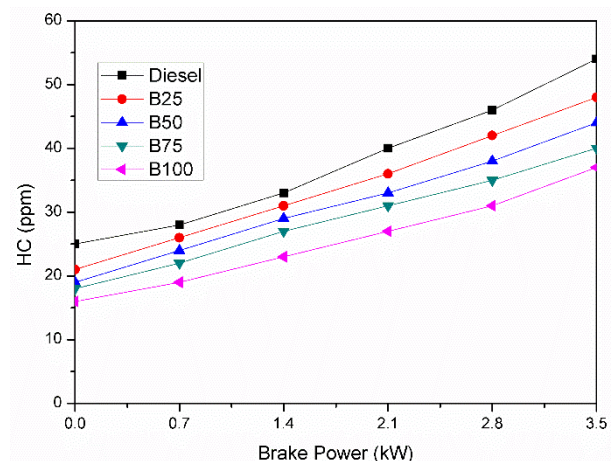


Figure 7: Variation of HC emission with Brake Power

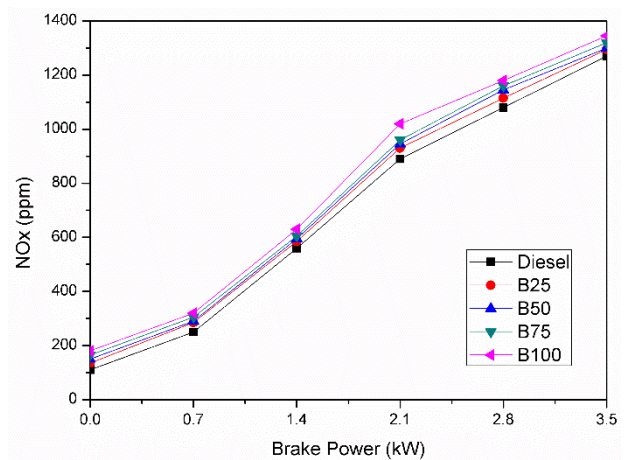


Figure 8: Variation of NO_x emission with Brake Power

4.7. NO_x emission

The variation of oxides of nitrogen (NO_x) emission with brake power for methyl ester blends and diesel is presented in figure 8. It

could be seen that the increase in proportions of mahua methyl ester with the blends found to be increase in NO_x emissions slightly when compared with mineral diesel. NO_x emissions are sensitive to adiabatic flame temperature, oxygen content and spray characteristics [32, 33]. In general, the concentration of nitrogen oxides varies linearly with engine load. The upper NO_x emission could be ascribed to the greater oxygen content of the methyl ester which prompts to a complete combustion and upper combustion temperature, thereby producing a rise in NO_x emission [34]. The NO_x emission is higher for methyl ester and its blends, but methyl ester does not create any harmful effects to human health. This is in line with other reports [35, 36].

4.8. Oxygen concentration

Figure 9 illustrates the variation of oxygen percentage with brake power for methyl ester blends and diesel. The proportion of oxygen present in the methyl esters is greater. This oxygen along with oxygen from intake air takes part in the combustion then the excess oxygen is exhausted. The oxygen proportion in the exhaust gas for the neat methyl ester blends is very low which shows that more amount of oxygen contributes in the combustion process to release heat energy. There is a minor increment in oxygen proportion with methyl esters and diesel blends. The rise of oxygen with diesel fuel operation is due to lesser combustion as compared to neat methyl esters.

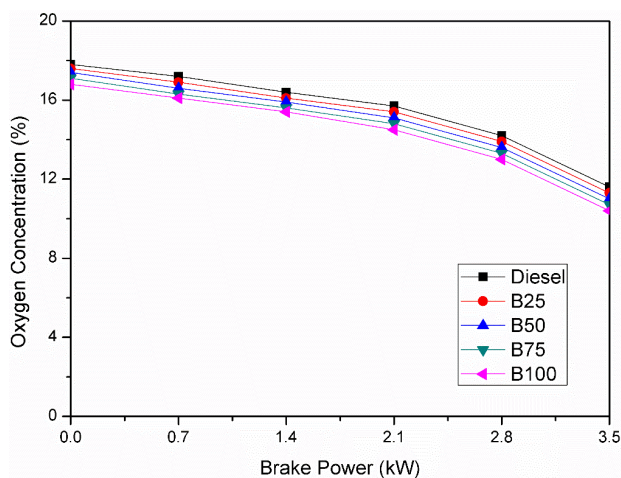


Figure 9: Variation of O_2 percentage with Brake Power

5. Conclusion

Based on the results of this study, the following detailed conclusions were drawn:

- Mahua methyl esters and its blends produced from non-edible oils are significantly less toxic to human health than mineral diesel. When using methyl ester in diesel engine applications, the pollutants of methyl ester from the engine are reduced drastically, which in turn reduces the loss of productivity and also medical expenses.

- For the engine performance, owing to the inferior heating of mahua methyl ester, the BSFC improved with incrementing methyl ester percentages in diesel fuel. Cylinder pressure incremented with smaller ignition delay owing to a greater cetane number of mahua methyl esters. It appears that the fuel-bound oxygen of methyl esters enhanced the diffusion combustion rate which outcomes in healthier combustion phasing. The heat release rate marginally improved for methyl ester blends than diesel.

- From the environmental point view, the emission of carbon monoxide (CO) and unburnt hydrocarbon (HC) were lesser except NO_x when compared with diesel. These exhaust pollutants will make harmful effects on human health and as well as on environment also.

- Utilization of methyl ester in compression ignition engines reduces greenhouse gas (GHG) emission as a result of non-edible oil production consumes part of carbon dioxide, which is produced during combustion of these fuels in the engines.

- Blends of diesel and methyl ester could also be helpful in reducing harmful gases from the engine, which in turn helps to meet out the stringent automotive emission norms and to prevent the harmful effects of pollutant emissions on environment and also on human health.

For the complete elimination of exhaust emissions from diesel engines, further researches and studies should be carried out intensely.

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