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Environmental Research & Technology

#### Lime stabilization of clayey landfill base liners: Consolidation behavior and hydraulic properties

Tanay KARADEMİR<sup>\*</sup>

Department of Civil Engineering, İstanbul Bilgi University, İstanbul, Türkiye

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#### ABSTRACT

Lime soil treatment is a chemical process in which lime (quicklime, hydrated lime or lime slurry) is mixed with the in place subgrade soil and a chemical reaction takes place. The lime reacts with the clay particles in the soil to create a cementitious matrix. The design of landfill base liners including a clay layer as a fluid barrier (i.e. water-resistant impervious layer) requires a neat engineering approach considering, in particular, consolidation behavior as well as hydraulic properties of the clay contained. For sake of safe and stable design of such baseliners under the landfills, the reduction of consolidation settlement in clay when subjected to the accumulated waste load (i.e. superposed action) during operation as well as the accomplishment for ensuring the waterproof of those composite base liners (comprised of multiple different layers) not to allow the penetration of the contaminated water - produced as a result of exothermic reactions occurring in the waste body in landfills - by enabling enhanced isolation from the natural ground is of importance. In light of this, in order to address those two most important design concerns (i.e. Consolidation and Hydraulic Properties of Clay) as well as to in an attempt to develop an enhanced clay layer system for landfill baseliners that has a greater bearing capacity (i.e. load resistance), and hence, more robust against settlements, and additionally, possessing improved hydraulic properties by being relatively more water-resistant and greatly impermeable, a series of consolidation tests were performed in the laboratory at different lime contents (lime/lime-clay mixture proportions by weight: 0%, 10%, and 20%) on clay specimens to investigate the stabilization and improvement of clayey landfill baseliners. Lime treatment on clay specimens has shown as a result of the experimental program that the strength of clay against loading improves, and further, exhibits less vertical deformation (settlement) under the load owing to an increase in lime content. Moreover, the clay becomes highly impermeable and displays substantially larger water-resistant properties because of increased lime mass proportion in clayey soil. The findings of the experimental program demonstrate that lime stabilization of the clayey soils in landfill baseliners will benefit the bearing capacity and the imperviousness (water tightness) engineering design properties as compared to standard composite multi-layered landfill baseliner systems.

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#### \*Corresponding author.

\*E-mail address: tanay.karademir@bilgi.edu.tr

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#### INTRODUCTION AND LITERATURE REVIEW

Lime stabilization is a chemical process established through physical mixing with the in-place soil along with the addition of a limited amount of water to facilitate (i.e. catalyze) and accelerate the chemical reaction for the formation of clay-lime cementitious matrix as well as to prevent dust emission during construction. In this perspective, over decades, lime has been used as a modifier as well as a binder for fine grained soils such as clay. Further, the treatment of clays with high plasticity results in a decrease of the plasticity index, and thus, the clay becomes possessing less affinity with water to exhibit substantial volume changes (dilation and/or expansion) which may cause the endangerment of the stability of the infrastructure constructed using that clay layer. Additionally, the pozzolanic action in fine grained soils accompanies strength increment in soil. Moreover, lime also provides binding action even for course grained soils.

The improvement of the engineering design properties of soil up to desired values to satisfy the project requirements is called soil stabilization for which the principal objective is to increase load carrying resistance (i.e. bearing capacity) as well as to enhance hydraulic properties of the soil layers designed in infrastructural projects in order to fulfill the constructional purpose. Landfills are one of the commonly constructed geo-environmental infrastructure applications throughout the world, particularly in the developed countries to protect the environment from the damaging impact of the contaminants that exist in the stacked and stored industrial and/or municipal waste bodies. To this end, base, side slope and cover liners of the landfill applications requires a neat engineering approach in design and development to accomplish a proper performance in terms of sufficient bearing capacity and superior hydraulic properties by enabling improved compressive strength against accumulated waste load and by providing enhanced imperviousness against infiltration of contaminated water leaking out from the waste. Within this scope, lime treatment of clay layers, essential part (component) of multi-layered composite landfill liners, leads to building and establishing a strengthened and highly water-resistant cementitious matrix in clay so as that the desired bearing capacity and hydraulic properties are achieved in the landfill liners. Furthermore, the main advantages of lime stabilization method in comparison to the other techniques including mechanical (i.e. compaction) and chemical (i.e. cement stabilization) stabilization methods are; (i) soil becomes more workable, (ii) strength is substantially improved, (iii) compressive strength is increased as high as 60 times, (iv) it is effective for most soil types. On the other hand, the disadvantages of lime stabilization could be described as follows; (i) lime is produced by burning of limestone in kilns, therefore it is harmful for the environment, (ii) the burning of limestone requires more cost, (iii) it is not effective for gravelly soils [1, 2].

Over the decades, several soil stabilization techniques including physical stabilization, chemical stabilization and mechanical stabilization have been used by the engineers in order to reinforce soil by improving mechanical strength characteristics (i.e. consolidation behavior of clayey soils) as well as by enhancing hydraulic properties (i.e. hydraulic conductivity or imperviousness characteristics) [1]. To this end, the lime-treatment process (i.e. being a chemical stabilization method) comprised mainly of chemical reactions in between lime and soil particles such that an improvement for the performance of soil layer is achieved by controlling volume change, and also, by increasing strength. The mineralogical properties of the soils, as per stabilization, determines their degree of reactivity with lime, and hence, the ultimate strength that the stabilized layers will develop. Owing to pozzolanic action in fine grained soils such as clays, the strength increment (i.e. gaining resistance against loads) is particularly expected in clayey soils so that the stabilization of the soil layer is accomplished.

In the lime-treatment process, there are two stages of the soil-lime chemical reaction. The initial stage - categorized as immediate or short-term treatment - develops within a few hours or days after the lime is admixed with soil in which three primary chemical reactions, namely; cation exchange, flocculation-agglomeration and carbonation are observed. The latter stage - categorized as long-term treatment - requires several months or years for completion in which a principal chemical reaction, namely as a pozzolanic reaction is detected [2, 3]. During lime-treatment process, the increase in soil workability as a result of drying of wet soil is attributed to immediate treatment and the increase in soil strength and durability is associated with long-term treatment [4]. As such, the calcium ions (Ca<sup>+2</sup>) and the hydroxide anions (OH<sup>-</sup>) are produced as a result of the addition of lime into the soil multiphase system including water in pore space. Then, during the process of cation exchange in soil, bivalent calcium ions (Ca<sup>+2</sup>) are replaced by monovalent cations. The Ca<sup>+2</sup> ions link to the soil minerals possessing negative charges in the encapsulation of diffused double layer, and thus, resulting in a reduction of the repulsion forces leading to the shrinkage of thickness of the diffused double layer (i.e. water layer) around clay particles so as that the bonds between the soil particles are strengthened. The reduction in the thickness of diffused double layer results in the proximity (i.e. closeness) of clay particles, thereby the soil texture changes and the multiphase soil matrix becomes densified causing a strength increment in the soil. The remaining hydroxide anions (OH-) in the solution existing in the voids (i.e. pore spaces) of soil media catalyze and induce an increment in alkalinity. This physiochemical process is called flocculation and agglomeration which is the main



Figure 1. (a) Clay (Bentonite). (b) Lime.

factor of the increase in the strength of soil or the enhancement of the bearing capacity to exhibit better performance and greater resistance against loadings [5]. As such, the mineral particles re-arrange in the flocculated/aggregated structure, giving rise to an intra-aggregate porosity and an apparent change in texture, with clay particles clumping together into larger-sized aggregates [6]. The above-described reactions are commonly identified as short-term reactions and they reduce the soil plasticity index (PI) and its water affinity [7]. The swelling potential decreases while hydraulic conductivity usually increases [8]. In the long-term, pozzolanic reactions develop between calcium ions and the silica or alumina of the lattices of clay minerals, improving strength and compressibility of the soil [6–8].

The clayey soils mostly have soft and sensitive nature that is related to and controlled by depositional and post-depositional factors [9]. As such, the main depositional factor is the inter-particle flocculation that creates an open micro-structure in such soils. As a consequence of different environmental depositional conditions, the soft, sensitive nature clay deposits located worldwide are characterized by different mineralogical and mechanical characteristics [10]. In this regard, lime is a stabilizing agent capable of improving properties of fine-grained soils to obtain proper hydraulic and mechanical characteristics for earthen structures (e.g., dikes, road embankments) [11]. Lime stabilization is acknowledged as an environmentally sound and cost-effective application because costs of high-quality material from quarries and of disposal of the unsuitable in situ soil are eliminated, and therefore, it is widely applied worldwide [12]. In this regard, the soil stabilization utilizing lime has also been practically applied in the field for important infrastructural projects in Turkey including roadway constructions conducted by General Directorate of Highways (i.e. KGM) in Marmara Region (Tekirdag), Central Anatolia (Konya). To this end, this paper will extend the understanding regarding the effect of lime stabilization on the consolidation behavior, mechanical characteristics and hydraulic properties of soils.

#### MATERIAL PROPERTIES AND TESTING PROGRAM

In this section, the experimental method followed in the laboratory for performing the comprehensive testing program will be explained along with the information provided on the engineering properties of the testing materials as well as the description given regarding the testing device and the experimental procedures conducted for the preparation of test specimens, installation of the equipment and the placement of the specimen in the experimental set-up.

#### Materials

The clayey soil selected to be used throughout the entire experimental program was the natural calcium Bentonite (i.e. has a very high proportion of exchangeable calcium whereas including a very low proportion of exchangeable sodium) (Fig. 1a) that is industrial clay for which the physical, chemical and index properties are suitable for the engineering standards. The lime (Fig. 1b) admixed into clay as a stabilizing agent was slaked lime of which the chemical composition is comprised mainly of Calcium (CaO) that transforms to a white powder and constitutes Ca(OH)<sub>2</sub> by reacting with water vapor in the air, thereby leading to obtain white amorphous material as shown in Figure 1b.

#### **Testing Equipment**

A floating ring cell consolidometer (UTS – 0300 Oedometer) in which the ring containing the soil sample is unrestrained in the container was utilized in the laboratory to perform consolidation tests on clayey soil – lime admixture



Figure 2. (a) Automated consolidation device. (b) Close up view of consolidation device.

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Samples	Test specimen I	Test specimen II	Test specimen III
Mass proportions	100% Clay –	90% Clay –	%80 Clay
	0% Lime	10% Lime	20% Lime

specimens (Fig. 2). A linear variable differential transformer (LVDT) was employed in the system to measure vertical displacement in evaluating the deformations developing in the soil specimen under the application of load during the tests. The displacement readings are logged into a computer through a multi-channel data logger such that the communication with the computer is enabled using RS-232 ports.

#### **Testing Program**

In this study, a series of consolidation tests [13] were performed in the laboratory at different lime contents (lime: lime-clay-mixture proportions by dry weight: 0%, 10%, and 20%) on the test specimens prepared in the laboratory to investigate the lime-stabilization (Table 1). Those different lime contents were mixed with clayey soil specimens such that the required weights of slaked lime were properly and completely admixed with batches of air-dried clay soil specimens to achieve lime-soil admixtures containing 0%, 10%, 20% lime contents on a dry weight basis. Further, the reconstituted consolidation test specimens were prepared at a dry unit weight of 13 kN/m<sup>3</sup> with a optimum moisture content of 30%. The results of consolidation tests conducted were used to determine the engineering consolidation parameters including; coefficient of consolidation (c,), compression index (C<sub>1</sub>), recompression index (C<sub>1</sub>) for evaluating strength-deformation characteristics as well as to identify an important hydraulic property such as coefficient of permeability (k) for assessing hydraulic characteristics. The test specimens (cylindrical: 6.0 cm in diameter and 2.5 cm in height) were prepared by admixing lime into clay at different dry mass proportions, and then, by enclosing in a stiff metal ring and placing between two porous stones in a cylindrical container filled with water. Later, A metal load platen mounted on top of the upper porous stone transmits the applied vertical stress (vertical total stress) to the soil sample. Both the metal platen and the upper porous stone can move vertically inside the ring as the soil settles under the applied vertical stress.



Figure 3. Deformation versus square root of time curves.

Incremental loads are applied to the platen, and the settlement of the soil at various fixed times under each load increment is measured by the transducer (LVDT) described in Testing Equipment Section. Each test continued six days with sequential incremental loads of 5, 10, 20, 40, 60, and 80 kgf, successively. As such, each load increment was doubled and allowed to remain on the soil until the change in settlement is negligible and the excess porewater pressure developed under the current load increment has dissipated. For many soils, this usually occurs within 24 hours [14]. Therefore, the vertical displacements (i.e. deformations) occurred in the specimen at every load increment were measured and recorded automatically in the computer over 24 hours (i.e. 1 day) at time intervals of logged data as follows: 0.05, 0.1, 0.2, 0.25, 0.5, 1, 2, 4, 8, 15, 30, 60, 120, 240, 480, 960, 1440 minutes, successively. Further, it should be emphasized that the ratio of the load increment to the previous load called as the load increment ratio (LIR) was selected as two conventionally. Similarly, the measurement time interval for reading the deformations developed in the specimen under loading were also selected as two purposefully to examine consolidation behavior in a comprehensive manner.

#### RESULTS

The experimental findings of consolidation tests performed at different lime contents (0%, 10%, and 20%) on the test specimens prepared in the laboratory to investigate the lime-stabilization will be presented in this Section along with supplementary theoretical descriptions and technical



**Figure 4**. Void ratio (e) versus effective stress, logarithmic scale (Log  $\sigma$ ') curves.

explanations. One purpose for changing lime content (i.e. dry mass percentage) in clay soil is to examine and unveil the mass proportion of lime content required to change the characteristics of the clayey soils, and thus, dominating the consolidation behavior and engineering properties.

#### **Deformation Readings with Time**

Three different loading conditions (20, 40 and 60 kgf) among distinct loads ranging 5 kgf up to 80 kgf incrementally applied during the course of each consolidation test were selected representatively to show the deformation readings with time, that is to say, the developed displacements measured in the tests with time for different clay – lime admixtures. As such, the generated displacement versus square root of time curves based on experimental measurements are presented in Figure 3 in the sequence of incremental loads successively.

#### Volumetric Compression and Loading

Soil is a multiphase material that consists of solid soil particles and void space filled by fluid and/or occupied by air exhangeably. The relative proportion of the volume of void space with respect to the volume of the solid soil particles is called void ratio (e) [15]. This engineering parameter is used very commonly to examine the change in volume of soil (i.e. volumetric compression) because of loading. To this end, the change in void ratio of the clay–lime admixtures (at different dry weight proportions) measured in the consolidation tests due to load increment, that is to say, an increase in the effective stress (applied load/specimen area) are presented in Figure 4. The compression and the recompression stages are evident in the curves such that the soil specimens exhibited elastoplastic behavior. That is, some part of the volumetric compression under the load (i.e. settlement under loading) is recoverable quantized through an engineering design parameter called recompression index (C), while the other part is permanent quantized through a consolidation property called compression index (C<sub>2</sub>). Additionally, the unloading-reloading cycle has been performed in the tests of the 90% Clay-10% Lime and the 80% Clay-20% Lime specimens to clearly detect and evidently reveal recompression stage, and thus, accurately determine the important consolidation parameter; recompression index (C).

Both  $C_c$  and  $C_r$  consolidation parameters (from Fig. 4) helps us extend the understanding on the estimation of the amount of consolidation settlement the soil will undergo due to application of overburden in the field, and thus, the bearing capacity of the soil under loading. As such, the load-deformation behavior of the clayey soil is assessed through these important engineering parameters;  $C_c$  and  $C_r$ . Further, the coefficient of consolidation ( $c_v$ ) (from Fig. 3) aids in the evaluation of the rate of consolidation settlement, and thus, time characteristics of the soil subjected

Engineering Design Parameters (Consolidation Strength, Deformation, Hydraulic Properties)						
<u>Clay - Lime Admixture</u>	Compression Index (Cc), []	Recompression Index (Cr), []	Coefficient of Consolidation (Cv), [cm <sup>2</sup> /min]	Coefficient of Permeability (k), [cm/min]		
100% Clay - 0% Lime	1.439E-02	3.364E-03	0.19781	3.920E-04		
90% Clay - 10% Lime	5.122E-03	4.787E-05	0.12036	1.151E-04		
80% Clay - 20% Lime	3.085E-05	7.776E-06	0.03565	2.229E-06		

Table 2. Consolidation strength and deformation as well as hydraulic properties

to loading in the field as leading to comprehend strength performance of the clayey soil under loading.

### Determination of Consolidation Engineering Design Parameters

The following theoretical methods were carried over to detemine the engineering design parameters including consolidation strength-deformation-time properties:

(i) The compression index (Equation 1) that is the average slope of compression stage in the e versus  $log(\sigma')$  plots – was determined using the void ratio and effective stress curves as presented in Figure 4 for different clay – lime admixtures.

$$(C_c = \Delta e / (\log \sigma'_2 / \sigma'_1)) \tag{1}$$

Where:

 $\Delta e$ : Change in void ratio

 $\log(\sigma'_2 / \sigma'_1)$ : Change in vertical effective stress

(ii) Similarly, the recompression index (Equation 2) that is the average slope of recompression stage in the e versus  $log(\sigma')$  plots – was determined using the void ratio and effective stress curves as presented in Figure 4 for different clay – lime admixtures.

$$(C_r = \Delta e / (\log \sigma'_2 / \sigma'_1))$$
 (2)  
Where:

 $\Delta$ e: Change in void ratio

 $\log(\sigma'_2 / \sigma'_1)$ : Change in vertical effective stress

(iii) The coefficient of consolidation ( $c_v$ ) was determined by applying "*Root time method*" (*Taylor's method*) on displacement versus square root of time curves as presented in Figure 3 at distinct loads for different clay – lime admixtures. According to Taylor's method, the cv can be calculated as follows:

$$c_{v} = \frac{T_{v} (H/2)^{2}}{t_{90}}$$
(3)

Where:

T<sub>y</sub>: Time factor

H: The specimen height at the beginning of the test (i.e. 2.5 cm)

 $t_{90}$ : Time required for 90% of consolidation has occurred

#### **Determination of Hydraulic Properties**

The hydraulic conductivity characteristics (i.e. coefficient of permeability, k) of clayey soils are associated with the time rate of consolidation properties (i.e. coefficient of consolidation,  $c_v$ ) through Terzaghi's one dimensional consolidation theory as follows:

$$k = c_v . m_v . \gamma_w \tag{4}$$

Where:

c<sub>v</sub>: Coefficient of consolidation

my: Coefficient of volume change or compressibility

 $\gamma_{w}$ : Unit weight of water

Using Equation 4, the important engineering parameter that is coefficient of permeability (k) employed to assess hydraulic properties of clayey soils was calculated for different clay – lime admixtures.

### COMPARATIVE ANALYSIS ON EXPERIMENTAL FINDINGS

The engineering design parameters for clayey soils aiding in quantification and qualification of consolidation strength, deformation and hydraulic properties are listed in Table 2 to explore, and thus, to evaluate their values as well as to gauge the changes in clayey soil as a result of adding lime content on a purpose for stabilization. Substantial variations in the values of compression index (C<sub>2</sub>), recompression index (C<sub>2</sub>), coefficient of consolidation (c<sub>2</sub>) and coefficient of permeability (k) have been investigated positively such that the consolidation strength-deformation characteristics have been enhanced as well as the hydraulic properties of the clayey soil have been improved owing to the contribution (favorable influence) of lime addition to the soil.

Moreover, a comparative analysis was performed on the computed values of engineering design parameters regarding consolidation strength-deformation characteristics  $(C_c, C_r, c_v)$  assisting in evaluating load bearing capacity of clayey soil (i.e. load resistance performance), and additionally, regarding hydraulic properties (k) assisting in assessing imperviousness features of clayey soil layer in the field



Figure 5. Comparative analysis on engineering design parameters.

(Fig. 5). The decrease in both compression index ( $C_c$ ) and recompression index ( $C_r$ ) (Fig. 5a, b, respectively) with an increase in lime content shows that the strength of clay becomes greater exhibiting smaller amount of deformation (consolidation settlement) against loading, and hence, displaying higher bearing capacity (greater load-carrying resistance). This is attributed to the lime treatment of the clay resulting in stabilization of soft soil through possible reactions including base-exchange, coagulation, and flocculation, reduction in thickness of water film around clay particles, cementing action and carbonation. As such, fine platy-like clay particles react with lime, and thus, get flocculated or aggregated into larger particle groups that are moderately stable even under subsequent soaking.

Since the bentonite clay used throughout the experimental program possesses inherent plastic constitution, the higher the agglomeration, the larger the strength increment was investigated compared to the other soil types such as sandy or silty soils. Therefore, lime treatment on clay specimens has shown as a result of the experimental program that the strength of clay against loading enhances, and further, exhibits less vertical deformation (settlement) under the load owing to an increase in lime content.

As per hydraulic properties, the values of coefficient of permeability (k) (Fig. 5d) have decreased such that the clay has become highly impermeable and displayed substantially improved water-resistant properties because of the increased lime content of the clayey soil. On the other hand, the coefficient of consolidation (c\_) (Fig. 5c) - showing quantitatively the time rate of consolidation deformation process (i.e. time rate of consolidation settlement in the field) - has decreased with an increase in lime content in soil. This results in extension of time duration required for the completion of consolidation settlement process of the clayey soil when subjected to loading that could be overcome by inserting wick drains in the constrcution area to facilitate and accelerate the expulsion (i.e. escape) of pore water from void space so that the progress of consolidation settlement has been eased and sped up to normal level. Similar results regarding the change in consolidation design parameters (Fig. 5) owing to the addition of lime into fine grained soils were also shown by the Reference [16] such that there observed a general increase in hydraulic conductivity (k), an evident decrease in consolidation engineering properties including coefficient of consolidation  $(c_y)$ , compression as well as recompresion indices (C<sub>c</sub>, C<sub>r</sub>, respectively). Such finding for lime-treated soils is attributed to the aggregation of clay particles due to lime addition [6, 11, 12]. Further, as previously investigated and revealed by the Reference [6, 8], the reduction in hydraulic conductivity (i.e. k) as well as the enhancement in compressive strength properties against loading (i.e. C, C) are related and governed by the growth of pozzolanic products that are capable of partially filling the inter-aggregate porosity. As such, the significant reduction in compressibility of the tested clayey soil owing

to the lime-treatment as evidently detected from the decrease in the values of  $C_c$ ,  $C_r$  (Fig. 5a, b) confirms the eaerlier findings of the References [7, 8,]. Moreover, as already indicated by the References [12, 16] and verified through the experimental results of the testing program as reported in this paper, the clayey soil – lime admixtures exhibit similar behavior analogous to over-consolidated soils attributed to and associated with the cementation characteristics induced by pozzolanic reactions being able to create a more rigid structure. Therefore, low deformations in the range of pressure typical of geo-environmental earthworks including landfills, embankments could be deemed to be ensured when soil–lime mixtures are selected to be utilized in their construction [11, 12].

The effectiveness of lime treatment can evidently be observed from the improvement of the consolidation engineering properties as shown in Figure 5. This type of enhancements in the consolidation design parameters were also reported by the References [17, 18]. In particular, for the case of soils characterized by high water contents (i.e. sensitive clayey soils), a proper design of soil-lime mixtures is based on a preliminary investigation of the soil to be used to verify its suitability to lime treatment as earlier indicated by the References [18-20]. The extensive laboratory testing program carried out on the soil-lime admixtures in the current study revealed the effect of mix proportion taking into account the specific in situ construction procedures and curing conditions in order to verify the effectiveness of quicklime treatment on a typical soft, sensitive, clayey soil for assessment at a laboratory scale.

#### CONCLUSION

On a purpose to address the two most important design concerns (i.e. Consolidation and Hydraulic Properties of Clay) as well as to in an attempt to develop an enhanced clay layer barrier system for landfill base liners that has enhanced bearing capacity (i.e. load carrying resistance), and hence, being robust, stable and durable against settlements, and additionaly, possesing improved hydraulic properties by being relatively more water-resistant and highly and favorably impermeable, a series of consolidation tests were performed in the laboratory at different lime contents on clay specimens to investigate the stabilization and improvement of clayey landfill base liners. The test results showed that the detected values of C, C, k decreased approximately 1/10 of the original value of the pure clayey soil, while the measured values of  $c_v$  decreased roughly 1/4 of the original magnitude as a result of an increase in lime content up to 20% by dry weight. As such, the findings of the experimental program demonstrates that lime stabilization of the clayey soils in landfill base liners will benefit the bearing capacity and the imperviousness (watertightness) engineering design properties as compared to standard composite multi-layered 9

landfill base liner systems. For sake of safe and stable design of base liners, containing clay layer barriers, under the landfills, the reduction of consolidation settlement in clay by enabling more stability, and thus, larger bearing capacity (i.e. greater load-carrying resistance) when subjected to the accumulated waste load (i.e. superposed action) during landfill operation could be achieved by lime stabilization. Furthermore, the accomplishment for ensuring the waterproof of those composite base liners (comprised of multiple different layers) not to allow the penetration of the contaminated water - produced as a result of exothermic reactions occurring in the waste body in landfills - by enabling enhanced isolation from natural ground could be realized in design. To this end, lime-treatment, filling voids, leads to reduction of porosity, and consequently, the reduction of permeability and the improvement of the imperviousness of clay layer. Further, the increase in bonding between soil grains as a result of lime-treatment produces mechanical strength increase, and hence, the enhancement of bearing capacity to demonstrate higher load-resistant performance of clayey soil layers in the field when employed in geo-environmental infrastructural projects.

#### DATA AVAILABILITY STATEMENT

The author confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

#### **CONFLICT OF INTEREST**

The author declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

#### ETHICS

There are no ethical issues with the publication of this manuscript.

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**Research Article** 

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Environmental Research & Technology

#### Investigating ambient air quality of a shooting range during official national competitions

S. Yeşer ASLANOĞLU<sup>\*1</sup>, Fatma ÖZTÜRK<sup>2</sup>, Gülen GÜLLÜ<sup>1</sup>

<sup>1</sup>Department of Environmental Engineering, Hacettepe University, Ankara, Türkiye <sup>2</sup>Department of Environmental Engineering, Bolu Abant İzzet Baysal University, Bolu, Türkiye

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#### ABSTRACT

Shooting is among nine sports branches that formed the first modern Olympic Games in Athens in 1896. A professional shooting athlete shoots millions of bullets throughout their sports life to commune with their gun dedicatedly. The number of simultaneous shots per unit time depends on the capacity of the range. It can enormously increase when a national match is a case. Shooting can cause gunshot residue exposure, including lead, other elements, and their by-products accumulate in ambient air and reveal significant health risks. This study aims to find the levels of PM<sub>10</sub> and its chemical composition during official threeday 50 m 22-cal competitions in May 2016, Mersin, Turkey. To this end, PM<sub>10</sub> samples were collected on quartz-fiber filters and analyzed for elements by ICP-MS and carbonaceous material by Thermal-Optical EC/OC analyzer. The total PM<sub>10</sub> mass concentration average is 28.7±7.3 µg/m³ within the indoor threshold values of different countries. The ambient mass concentrations of PM<sub>10</sub>, OC, EC, TC, Cd, and Pb were higher during once pistol matches instead of rifle matches. Although Pb values did not exceed the indoor limits for shooting ranges, it has the highest concentration among the analyzed elements. Additionally, Cr poses cancer risk potential. Except for Zn, Sr, and Cu, all the measured parameters have higher calculated emission factor (EF) values during pistol shots. To our best knowledge, this study reports the airborne mass concentrations of EC, OC, and TC from indoor shooting ranges and investigates indoor air quality for shooting sport for the first time.

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#### INTRODUCTION

Air pollution, which accounts for 1 in 8 deaths in 2012 according to WHO [1], is one of the most significant environmental issues in terms of human health. People are prone to high exposure risk from air pollution in indoor and outdoor environments, even at low concentrations [2]. Indoor air quality (IAQ) has been paid attention to by researchers due to the time spent in indoor environments. IAQ has been considerably influencing on well-being and productivity of people, while indoor air pollutants are increasing the risks for a variety of diseases. Jenkins et al. [3] reported that people spend 87% of their time indoors while only 6% outdoors and 7% in transit. The performed

\*Corresponding author.

\*E-mail address: yaslanoglu@hacettepe.edu.tr



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studies starting in the middle of the 1970s put forward that indoor concentrations of several pollutants were significantly higher than their corresponding outdoor concentrations [4].

Considerable IAQ studies on schools [5-8], elderly care centers [9, 10], homes [11-15] hospitals and nursing homes [9, 16, 17], and offices [18, 19]. On the other hand, researchers recently paid attention to the IAQ of environments used for physical exercise and sports [20-25]. Like other indoor places, construction materials, maintenance, and ventilation types form IAQ in sports centers. Moreover, higher human occupancy and the type of acts performed in the centers make them peculiar. During exercise, the respiratory ventilation per minute rises, which leads to inhalation of more air and pollutants existing in the air. Consequently, metabolic reactions to physical exercise open the human body to an elevated amount of pollutants [26]. The nasal particle-filtering system is not used during training since air is inhaled through the mouth. This process causes an increase in airflow velocity, which results in the movement of pollutants to most parts of the respiratory system and produces more risk to human health [26]. According to Ramos et al. [24], people who conduct physical exercise in polluted environments put their health at risk. Shooting is a bit tricky at this point. It is well known that a shooter's heart rate is considerably lower during shooting than in daily life. They use abdominal breathing; additionally, they inhale and exhale by their nose during aiming and triggering, not their mouth.

IAQ studies performed in sports centers revealed that occupants expose to various air pollutants, including particulate matter (PM), combustion-related emissions such as carbon monoxide (CO) and nitrogen dioxide (NO<sub>2</sub>), carbon dioxide (CO<sub>2</sub>). Additionally, biological pollutants such as dust mites, molds, fungus, and bacteria; volatile organic compounds (VOCs), for instance, formaldehyde and benzene; inorganic chemicals, for example, chlorinated compounds (mainly in swimming pools); heavy metals such as lead and mercury; and asbestos, which is primarily due to the building materials [27-29]. These studies focus on fitness centers [24, 30, 31] and gymnasiums and sports facilities in educational premises [32–35]. Andrade and Dominski [20] reviewed the studies performed on IAQ of places used for sports. Authors reported that the gymnasium, fitness and sports centers, and ice-skating rinks, were the most investigated places in the reviewed studies. Moreover, CO, NO<sub>2</sub>, and PM were the IAQ parameters primarily measured in these indoor environments. On the other hand, the literature on the level and composition of emissions from firing ranges as indoor sport and recreational activity is scarce except for a few studies [36-44]. Shooting at firing ranges has become very popular among people as a recreational activity in many countries. 16,000-18,000 indoor firing ranges in the United States alone and 20 million people nationwide exercise target shooting for leisure [45]. Recreational, in other words, private sector pistol shooting ranges are about 30 thousand currently in Turkey [46]. Despite this, ranges for licensed athletes are very rare. The only world-cup standard shooting range in Turkey is located in Mersin, Erdemli. Including Erdemli shooting range, there are about 70 shooting ranges in Turkey in different cities and, these shooting ranges are operated by the Turkish Ministry of Youth and Sports [47]. A large number of these are outdoor shotgun ranges also used for recreational purposes. Only a few indoor shooting ranges are used by licensed shooting athletes, primarily for training. Operational liabilities, including ventilation, cleaning, transportation of athletes, belong to the provincial directorates. In order to reduce operating costs, unfortunately, electricity expenses such as ventilation are the first items to be reduced.

Indoor firing ranges are enclosed facilities, which have a unique operation. Improper use and design of indoor firing ranges could lead to adverse effects on human health even though military and civilian personnel prefer them for their controlled environment to outdoor counterparts. Well documented in the literature that mainly metals and gaseous compounds increase to high concentrations in the air and floor during shooting activities [38, 42, 48]. Not only major combustion gases such as carbon dioxide (CO<sub>2</sub>) and water vapor (H<sub>2</sub>O) but also carbon monoxide (CO), hydrogen cyanide (HCN), ammonia (NH<sub>3</sub>), nitrous oxides (NO<sub>1</sub>), sulfur dioxide (SO<sub>2</sub>), and hydrogen chloride (HCl) are released during shooting [49, 50]. In addition, particulate matter (PM) consists of soot and metals, for instance, lead (Pb), copper (Cu), zinc (Zn), and iron (Fe), along with trace quantities of chromium (Cr) and molybdenum (Mo) are emitted to the indoor environment as a result of shooting activities [49]. Elevated concentrations of these airborne compounds in indoor environments cause serious health issues in occupants of indoor firing ranges.

Precisely at this point, we need to explain deeper the mechanism when a shooter pulls the trigger. A professional shooting athlete shoots millions of bullets, cartridges, or pellets throughout their shooting life, approximately starting at 13. Ammunition differs among shooting branches as lead pellets for air guns, cartridges for shotguns (rifle), and bullets for rifled guns. As in our case, in 50 m and 25 m competitions, .22-cal bullets are suitable for specialized rifles and pistols. According to the gun type, when the shooter pulls the trigger, the firing pin drops to the bullet jacket, bullet core leaves from the jacket. Due to the rifling inside the gun, the core aerodynamically travels through the shooting line till it hits the target. If the firing pin hits the jacket from the center, it is called center-fire. Otherwise, if it hits from the side part, it is called rim-fire. Compared to the center-fire bullet, less gunpowder and bullet



Figure 1. Schematic description of the shooting range.

materials are used in the rim-firing system. As the explosion is more efficient between jacket and core, emission rates and residues are reduced in rim-fire bullets. RWS R 50° rim-fire bullets are officially used at Turkish Shooting & Hunting Federation's competitions [51, 52].

In order to explain the health effects of shooting, Laidlaw et al. [45] reviewed thirty-six articles published in the literature and evaluated the shooters' blood lead levels (BLLs) at firing ranges. Researchers revealed that all BLL measurements exceeded the reference level of 5 µg/ dL recommended by the U.S. Centers for Disease Control and Prevention/National Institute of Occupational Safety and Health (CDC/NIOSH). There is sufficient evidence that BLLs <10 µg/dL are caused to essential tremors, hypertension, cardiovascular-related mortality. Additionally, electrocardiography abnormalities and decreased kidney glomerular filtration rate among adult men and women while that  $<5 \mu g/dL$  leads to decreased fetal growth for an adult woman. Moreover, there is "sufficient evidence" that BLL <5 µg/dL caused several problems in children, such as declined academic achievement and intelligence quotient (IQ), reduced perinatal growth [53]. Most studies focused on the Pb because projectiles and primers contain a considerable Pb, and a huge Pb becomes airborne during shooting activities [54, 55]. International Agency for Research on Cancer (IARC) [56] classified Pb as class II (B) carcinogens, that is, possible human carcinogens. Similar to Pb, other metals emitted

during shooting activities are associated with adverse health effects. Arsenic, Ni, Cd, and Cr were classified as class I carcinogenic contaminants by the IARC, while Zn, Cu, and Mn were classified as non-carcinogenic contaminants [57]. Residential exposure to a low level of Cd is related to renal toxicity, osteoporosis, and bone fractures [58]. People exposed to As by inhalation depicted an excess risk of lung cancer [59]. Moreover, upon inhalation, Zn may destroy plasmid DNA [60]. Based on epidemiological studies, it has been revealed that elemental carbon (EC) is correlated with cardiovascular and respiratory hospitalizations [61], preterm birth [62], and mortality [63]. Likewise, studies that measured organic carbon (OC) have found associations between respiratory outcomes and OC [64] and associations between cardiovascular outcomes and OC [65].

24-hr indoor PM<sub>10</sub> samples were collected in three-day official competitions in this study. Collected samples were analyzed for elements and elemental and organic carbon (EC and OC, respectively). To our best knowledge, this is the first study reporting the airborne concentrations of elements, EC, and OC in a firing range in our country. This study is also unique since no indoor EC, and OC data from shooting activities have yet been produced in the literature. Another critical point is that the Mersin/Erdemli shooting range is neither commercial, recreational, nor military. So, occupants are licensed shooting athletes, their families, official coaches, referees, and audiences.

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Parameter	RSD (%) (n=3)	DL (µg/cm²) (n=7)	Sunset Lab. PES (µg/cm²) (n=5)		This study PES (µg/cm²) (n=5)		
			Avg	SD	Avg	SD	
OC	4.60	0.31	16.75	0.94	13.37	1.33	
EC	5.02	0.06	2.05	0.20	1.88	0.34	
TC	4.65	0.34	18.80	1.14	15.26	1.67	

Table 1. Figures of merit of EC/OC analysis

#### MATERIALS AND METHODS

#### Sample Collection

24-hr PM<sub>10</sub> samples were collected in this study during 50 m-range 22-cal rifle and pistol competitions, organized between 12 and 15 May 2016 in Erdemli (Mersin, Turkey) by the Turkish Shooting and Hunting Federation to investigate the air quality from firearms. Figure 1 depicts the schematic of the shooting range and the place where the Tecora Skypost PM<sub>10</sub> sampler, which is working by the EN 12341:2014 norm, was located. According to the statutes, rules, and regulations of the International Shooting Sport Federation (ISSF), athletes cannot go beyond the 5 m distance at the shooting line, where only the referees can cross this border. Entry and exit of the athletes to the shooting area can only be performed under the supervision of referees, and all the supporters and coaches should obey this rule of ISSF during match and training times. In addition, even if referees should keep quiet during all shooting competitions except for the final match. Otherwise, supporters are invited to out of the range by the referees. Since a high-volume sampler produces an appreciable degree of noise during sampling, it was placed 10 m behind and 5 m above the fire area during the matches. Three match samples and one blank sample were collected on the pre-fired Whatman quartz fiber filter during the shooting activities. Quartz filters were pre-conditioned at 25°C and 25% relative humidity for one day before sampling, weighted and stored at -18°C until sampling. The sampler was operated only for 5 min at the sampling flow rate of 34 L/sec for the blank sample. All the samples were kept in the Petri slides in the freezer till analysis. Before analysis, the samples were pre-conditioned for one day under the same conditions stated previously, and PM load was determined by subtracting the tare of the filter. Rifle matches took place on the first and second day of the competitions, while pistol matches were last.

#### **EC/OC Analysis**

A 1.5 cm<sup>2</sup> punch was cut from the collected filters and analyzed utilizing Sunset Lab. (Oregon, USA) thermal-optical transmission EC/OC analyzer for elemental, organic, and total carbon (EC, OC, and TC=EC + OC). National Institute of Occupational Safety and Health (NIOSH) 870 protocol was followed during the analysis [66]. The details of the EC/OC analysis were provided in Öztürk and Keleş [67]. Briefly, the analyzer uses thermal, optical, and chemical principles to determine the carbon content of the filter samples. Firstly, an inert atmosphere is created by purging 100% He inside the oven, and OC formed during this step is converted to CO<sub>2</sub>. Then, EC is transformed to CO, under oxidizing medium by purging a mixture of gas composed of 10% O2 and 90% He (vol/vol). Afterward, generated CO<sub>2</sub> is reduced to CH<sub>4</sub>, which is detected by a flame ionization detector (FID). The performance of FID is checked at the end of each run by injecting a fixed volume of methane (5% CH<sub>4</sub> plus 95% He, vol/vol) as an internal standard. As a part of the quality assurance and quality control (QA/QC) protocol, a known amount of sucrose solution was spiked over the pre-fired blank filters. Analysis was conducted under the same conditions as the filter samples. The measured sucrose concentrations have deviated only 0.10% from the standard sucrose solution based on n=14 repeated measurements. In addition, the instrument was operated without putting any sample at the beginning of each analysis day. The average OC levels determined in the instrument blanks were less than 0.02 µg/cm<sup>2</sup> while no EC was detected.

The detection limit (DL) and precision as relative standard deviation (RSD) of EC/OC measurements were also calculated in this study. DL values (three times the standard deviation of the blank filter measurements) were calculated based on the repeated blank measurements, and RSD values were estimated by analyzing the performance evaluation standard (PES) provided by the Sunset Lab. Moreover, the accuracy of the EC/OC measurements was calculated based on the data provided by Sunset Lab. for PES measurements. Table 1 summarizes figures of merit of EC/OC analysis.

As given in Table 1, the percent RSD values were almost  $\leq$ 5, indicating that repeatability of the analysis is acceptable. DL values of measurements were 0.31, 0.06 and 0.34  $\mu$ g/cm<sup>2</sup> for OC, EC and TC, respectively. The accuracy of the measurements was found to be 20, 8, and 18%, respectively (Table 1). The EC/OC results provided in this paper were corrected for filter blank. Since no carbonate carbon (CC) peak was detected during analysis, CC correction was not performed.

		Step				
	Ι	II	III	IV		
Ramp time (min)	3	2	2	1		
Hold time (min)	5	10	30	10		
Temperature (°C)	140	160	200	50		
Pressure (bar)	30	30	35	25		
Power (%)	75	85	90	0		

Table 2. Figures of merit of EC/OC analysis

#### **Elemental Analysis**

The rest of the filters from EC/OC analysis were digested in a mix of high purity acids (5 mL HNO<sub>3</sub> + 1 mL HF + 0.5 mL  $H_2O_2$  + 1 mL de-ionized (DI) water using Berghof (speed wave-2, Germany) Microwave Digestion Oven. The steps of the digestion program are tabulated in Table 2. Acid and field filter blank were also digested along with the filter samples and treated similarly. After digestion, all the samples and blanks were diluted to 50 mL with DI water, transferred to HDPE bottles, and kept in the refrigerator till analysis. After micro-wave digestion only, samples that showed visible residues of soot carbon were filtered through 0.45 µm pore size Millipore brand mixed esters of cellulose filter (Sartorius AG).

Agilent 7700 Model Inductively Coupled Plasma Mass Spectrometry (ICP-MS) (California, USA) was employed in this study to perform the trace element analysis of the samples. Samples were analyzed for 15 elements using ICP- MS (Be, Cd, Co, Cr, Cu, Hg, Mn, Ni, As, Ba, Pb, Se, Sr, V, and Zn). 100-ppb germanium and terbium were used as internal standards during the analysis of samples. Agilent internal standard mix for ICP-MS systems (part number 5188-6525) calibrated the instrument. Calibration standards, blanks, and samples were spiked with this internal standard to overcome instrumental and sample-related variations. Internal standard element recovery was monitored closely by following the procedure described in EPA Method 200.8, section 9.4.3 [68]. Recoveries tried to be kept between 70% and 120%. In addition, the method detection limit (MDL) of the analysis was also calculated based on the same method as described in section 9.2.4 [68]. Moreover, instrument detection limit (IDL) values were estimated following the procedure provided in the same method under section 9.2.5. The accuracy of the measurements was also monitored closely as a part of the quality assurance and quality control (QA/QC) procedure. To this end, Standard Reference Material (SRM) 2783 obtained from the National Institute of Standard and Technologies (NIST) was digested by following the procedure used to extract filter samples. The obtained results were compared with the certificate sheet of the SRM. The precision of the measurements was also estimated based on the relative standard deviation of repeated SRM analysis. The values corresponding to SRM measurements, precision, MDL, IDL, and recovery for the measured parameters are summarized in Table 3 below.

Except for the Cd, the recoveries were >70% for all of the elements. Thus, the measured concentrations were not cor-

	SRM 2783 Certified values (mg/kg)		Obtained values (mg/kg)		Precision (%)	IDL (µg/kg)	MDL (µg/kg)	Recovery (%)	
Parameter	Avg	SD	Avg	SD				Avg	SD
As	7.0	1.6	5.8	0.3	17.1	4.2	112	79.4	2.82
Ba									
Be									
Cd	7.3	3.7	7.5	0.5	3.1	2.5	7.5	57.8	2.27
Со						6.1	163.2	75.6	2.5
Cr	80	22	54.5	4.1	24.7	32.2	125.6	79.6	5.02
Cu						20	122.5	73.3	2.24
Hg						7.5	22.6	74.9	6.05
Mn						89.8	172.2	72.8	2.51
Ni						135.5	406.5	77.2	3.46
Pb	85.9	7.2	64.6	11.7	31.9	20.8	33.3	87.0	6.83
Se									
Sr									
V						34.9	122.8	78.3	2.22
Zn						90.6	941.4	78.1	5.19

 Table 3. Figures of merit of ICPMS measurements

Parameter	Unit	1 <sup>st</sup> -day rifle	2 <sup>nd</sup> -day rifle	3 <sup>rd</sup> -day pistol	Avg	SD
PM <sub>10</sub>	µg/m³	25.4	21.9	38.8	28.7	7.3
OC	μg/m³	4.02	3.72	6.47	4.74	1.23
EC	µg/m³	0.197	0.194	0.349	0.247	0.072
TC	μg/m³	4.24	3.92	6.82	4.99	1.30
As	ng/m <sup>3</sup>	0.334	0.432	0.251	0.339	0.074
Ba	ng/m <sup>3</sup>	21.9	13.8	19	18.2	3.4
Be	ng/m <sup>3</sup>	0.026	0.035	0.032	0.031	0.004
Cd	ng/m <sup>3</sup>	0.28	0.117	0.495	0.297	0.155
Cr	ng/m <sup>3</sup>	1.06	3.39	2.13	2.19	0.95
Cu	ng/m <sup>3</sup>	3.18	1.51	1.27	1.99	0.85
Mn	ng/m <sup>3</sup>	1.81	2.55	2.24	2.20	0.30
Ni	ng/m <sup>3</sup>	2.18	3.03	2.27	2.49	0.38
Pb	ng/m <sup>3</sup>	221	110	259	197	63
Se	ng/m³	1.34	2.81	1.6	1.92	0.64
Sr	ng/m³	0.266	1.349	0.548	0.721	0.459
V	ng/m³	6.35	6.7	5.86	6.30	0.34
Zn	ng/m <sup>3</sup>	4.72	4.55	1.15	3.47	1.64

Table 4. The average concentration of parameters measured in PM10 samples during the competitions

rected with the recovery values. If IDL values <0 for the parameter of concern, the corresponding value was replaced with one-third of the associated MDL value. Moreover, if the measured concentration of the parameter is less than both IDL and MDL, the analyte concentration was replaced with half its corresponding MDL value to use the data in the statistical analysis. Furthermore, the metal levels reported in this study were field blank corrected.

#### **RESULTS AND DISCUSSION**

The concentration of parameters measured during 3-day competitions is summarized in Table 4, including the average (Avg) and standard deviation (SD) of measured concentrations for the whole field study. On the third day, when the pistol athletes were competing, almost two times higher  $PM_{10}$  mass concentration was measured compared to the previous two days. The  $PM_{10}$  mass concentrations for three days were measured as 25.4, 21.9, and 38.8 µg/m<sup>3</sup>, respectively. The highest concentration was measured for OC (4.02 µg/m<sup>3</sup>), while the minimum concentration was obtained for Be (0.026 ng/m<sup>3</sup>) on the first day when rifle matches took place. The maximum and minimum concentrations were obtained for the same parameters on the second and third days of the competitions, as tabulated in Table 4.

It is also worthy to note that the highest  $PM_{10}$ , OC, EC, TC, Cd, and Pb levels were measured in the samples while the pistol was being used for shooting in the third-day competitions. Grabinski et al. [48] revealed that PM mass emissions

when the shooters used rifle is about an order of magnitude lower than the emissions released from pistols. This situation can be attributed to the greater barrel diameter and shorter barrel length used in pistols.

During the matches, the measured PM<sub>10</sub> mass concentration was found as 25.4, 21.9, and 38.8 µg/m<sup>3</sup>, respectively, for the first, second, and third day, while the average of the whole event was 28.7±7.3 µg/m<sup>3</sup>. The measured values were below the American Society of Heating, Refrigerating and Air-Conditioning Engineers (ASHRAE) and England indoor PM<sub>10</sub> limit, 75 and 50 µg/m<sup>3</sup>, respectively. Hong Kong's first level value for indoor  $PM_{10}$  is 20 µg/m<sup>3</sup> [69]. Table 4 revealed that during the matches, this limit value was exceeded. Wingfors et al. [38] reported that most of the particles released from indoor firing ranges fall to nanoparticle size regime, which is more critical in terms of human health point of view since these particles are capable of penetrating deep into the human respiratory tract by inhalation [70]. Consequently, the mass concentration and the size distribution of PM are significant for indoor shooters.

The measured  $PM_{10}$  concentrations in this study were comparable with the ones reported by Orru et al. [71]. The average PM concentration in one of the shooting ranges in which pistols were used is 28.9 µg/m<sup>3</sup>. Researchers collected size-segregated PM samples at indoor military shooting ranges and analyzed the collected samples in terms of metals, including Pb, Cu, Ni, and Zn. In contrast to PM<sub>10</sub> levels, significantly higher metal concentrations were reported in this military shooting range. Grabinski et al. [48] reported Cu and Zn concentration in a ventilated, indoor firing range, where PM samples were collected in cases both pistol and rifle used for firing. The airborne mass concentration of Cu was reported to vary from  $\leq 1$  to  $16 \,\mu\text{g/m}^3$ , which is considerably higher than one measured in this study (average of the competition 1.99 ng/ m<sup>3</sup>). In addition, researchers found that Zn levels ranged from 0.5 to 1.8  $\mu\text{g/m}^3$ , which is again much higher than the competition average value (Table 4).

Indoor EC and OC levels were published in some studies in the literature. For example, Na and Cocker [72] determined concentrations of these two parameters inside the 20 residential settings and one local high school in CA, USA. In the homes with smokers, OC levels ranged from 9.3 $\pm$ 1.6 to 229 $\pm$ 67.6 µg/m<sup>3</sup> while EC concentrations were between 1.7±0.7 and 4.0±1.5 µg/m<sup>3</sup> in PM<sub>25</sub> samples. Authors claimed that there are no indoor EC sources while OC considerably contributed to indoor PM<sub>2,5</sub> levels. In a similar study, Ho et al. [73] collected  $PM_{25}$  samples inside the five buildings near roadsides in Hong Kong. The average inside OC and EC levels were found as  $11.3\pm5.5$  and  $4.8\pm3.4$  µg/m<sup>3</sup>. The significant sources determining the indoor concentrations of these pollutants were attributed to the penetration of outdoor pollution. Seleventi et al. [74] performed PM<sub>25</sub> sampling inside an apartment in Athens. The researchers reported the indoor average OC and EC levels as 9.6 and 1.9  $\mu g/m^3$ , respectively. The indoor source of OC was thought to be several activities performed by the residents, such as smoking, cooking, and cleaning. However, all these studies were conducted inside the residential places, and reported values cannot be compared to those measured in a shooting range. To our best knowledge, there is only one study in the literature reporting black carbon (BC) concentration for indoor air during sportive activities. Bisht et al. [75] monitored the indoor air quality for stadiums during the 19th Common Wealth Games (CWG) at Delhi (India). It is good to mention that BC determination relies on optical methods, and thermal-optical methods determine EC. Although co-located measurements showed that BC data could be 20% higher than EC levels [76], EC can be used as a surrogate measure of BC [77] In Bisht et al. [75], BC levels inside the three sports venues during CWG were reported to range about from 12 to 14  $\mu$ g/m<sup>3</sup>. Unfortunately, there is no information about the nature of sports competitions provided in the study. Researchers revealed that BC showed a positive correlation with CO, a pollutant released from incomplete combustion. Once the EC data generated in the shooting range were compared with the BC levels reported by Bisht et al. [75], it can be concluded that BC data was at least 50 times higher than EC measured in the shooting range.

Among the elements analyzed in the  $PM_{10}$  samples collected during the competitions and listed in Table 4, Cr is paying attention. Cr presents in the ambient atmosphere as Cr (III) and Cr (VI). Cr (III) is essential in trace levels for the proper functioning of living organisms. On the other hand, Cr (VI) is known as a pulmonary carcinogen by the International Agency for Research on Cancer and US Toxicology Program [78]. Indoor exposure to Cr (VI) is related to elevated lung and nasal cancer risk [79]. US Environmental Protection Agency listed Cr (VI) compounds as one of the 18 core Hazardous Air Pollutants (HAPs) [80]. It has been estimated that one in a million-cancer risk threshold for Cr (VI) is 0.083 ng/m<sup>3</sup>. The average Cr (VI) to total Cr ratios ranged from 1 to 30% for the ambient air [81]. By taking a ratio of 15% on average, Cr (VI) values for this study can be estimated to vary from 0.159 to 0.509 ng/m<sup>3</sup> for the first and second day, respectively. Consequently, the emitted Cr during the competitions has cancer risk potential.

Once Table 4 is evaluated in terms of elements, it can be seen that the highest level was measured for Pb compared to other metals. The average Pb concentration for the whole study was 197±63 ng/m<sup>3</sup>. Several guidelines regulate occupational Pb exposure in the world. For example, OSHA (Occupational Safety and Health Administration) (USA) set 30 µg of Pb per cubic meter of air as an action level for indoor environments. In addition, the time-weighted average (exposure over an eight-hour average) for Pb was set to 50 µg/m<sup>3</sup> in the same regulation. Furthermore, NIOSH recommended exposure limit (8-hour average) for Pb is 50  $\mu$ g/m<sup>3</sup> while increasing Pb exposure to 100  $\mu$ g/m<sup>3</sup> indicates the level that is Immediately Dangerous to Life and Health (IDLH) [82]. When these limit values were compared with the data generated in our study, none of the days' Pb exposure limits were exceeded. However, it should be kept in mind that the samples collected in this study for 24-hr long. Consequently, the measured Pb levels that we reported in this study were smoothed out.

The shooters are exposed to Pb from three different sources during the shooting activity. The first source is ammunition primer, composed of lead styphnate, which initiates the mercury fulminate explosion and lead azide propellant, released to the ambient air upon firing. The second one is burning propellant in the cartridge, which vaporizes the Pb due to extreme temperatures as high as 1100 °F. The last is associated with the dust and lead oxide fumes, which are emitted when the bullet hits to target [83]. Once the Pb is released to the indoor environment, the occupants inside the firing range are exposed to this metal through dermal contact [84], ingestion [85], and inhalation [84]. The adverse health impacts of elevated Pb exposure are well documented in the literature [36]. International Agency for Research on Cancer (IARC) declared inorganic lead as a probable human carcinogen (group 2A) [86]. Conversely, limited evidence has been found in human studies. In contrast, there is adequate data on the carcinogenicity of Pb in experimental animal studies [56]. Gulson et al. [87] suggested using non-lead primers to reduce the uptake of lead by recreational shooters.

Emission factors (EF) for the measured parameters were also calculated by dividing the mass concentrations of the parameters by the number of shots each day. Table 5 below summarizes the EF values for the corresponding parameters for rifle (average of first two days) and pistol (third day).

Once the normalized values were compared for rifle and pistol, it was found that Cd mass emission per bullet from pistol shooting is about four times higher than the one measured for the rifle. Similarly, about three times higher EF values were calculated for  $PM_{10}$ , TC, OC, EC, and Pb when pistol matches took place instead of the rifle. Estimated EF values for Mn, Cr, Ni, V, and Ba are about 1.5 times higher for pistol emissions than for the rifle. Zinc was the only pollutant that had a higher EF value in rifle emissions. EF values calculated for Sr and Cu are comparable.

In addition to the parameters listed herein, ammunition during shooting activities may release other stressors to the indoor environment. For example, nitrogen  $(NO_x)$  oxides are among these pollutants and irritate the eyes and respiratory system. Carbon monoxide (CO), another pollutant released into the indoor atmosphere due to firing, is known to reduce the ability of blood to carry oxygen and leads to headaches and nausea. Moreover, polycyclic aromatic hydrocarbons (PAHs), hydrogen cy-anide (HCN), ammonia (NH<sub>3</sub>), sulfur dioxide (SO<sub>2</sub>), and hydrogen chloride (HCl) are other air pollutants released to indoor air upon firing. The literature has well reported the detrimental health impacts associated with these pollutants on humans and the environment.

#### Limitations

Since the competition was officially three days, PM<sub>10</sub> samples, including two in rifle and one in pistol, were collected during this study, which prevents us from making a comprehensive assessment about the differences or similarities in the chemical compositions of the collected samples. In addition, 24-hr PM<sub>10</sub> samples were collected during the study through the matches were performed from 09.00 am to 08.00 pm. Shooting range opening and closing times are spread over a wider range. After the scheduled matches, shooters may perform small training shots in order to adjust guns, shooting position, and other equipment revealed to prolonged shooting hours. In regular training conditions, samples may have been collected in shorter time windows. However, it should be noted that this is a well-attended national organization. Also, there is no other example in the literature on a measurement related to sports-shooting competition cases. Additionally, the ratio of Cr (VI) to Cr was used in this study based on the ambient PM data, and no value was found for the indoor environments. Consequently, one should consider these limitations while interpreting the generated data in this study.

$(1 M_{10}^2, 00, 10 \text{ cm} \text{ rcm} \text{ mg/bullet and elements m pg/bullet)}$					
Parameter	Rifle	Pistol			
PM <sub>10</sub>	116	321			
TC	20	56			
OC	19	54			
EC	0.959	2.890			
Be	0.149	0.267			
Cd	0.972	4.087			
As	1.88	2.07			
Sr	3.96	4.52			
Se	10	13			
Mn	11	18			
Cr	11	18			
Cu	11	11			
Ni	13	19			
Zn	23	10			
V	32	48			
Ba	87	157			
РЬ	810	2134			

## **Table 5.** Calculated emission factor (EF) values for $PM_{10}$ , carbonaceous materials, and elements for rifle and pistol shots ( $PM_{10}$ , OC, EC & TC in ng/bullet and elements in pg/bullet)

#### CONCLUSION

This study was conducted during a three-day-long official national shooting competition. To our best knowledge, we report first-time EC, OC, and TC concentrations associated with the indoor firing ranges. Another critical point is that the Mersin/Erdemli shooting range is neither commercial, recreational, nor military. So, occupants are licensed shooting athletes, their families, official coaches, referees, and audiences. It was demonstrated here that shooting activities produce a considerable amount of particulate matter, carbonaceous material, and many toxic elements associated with it. Lead was the most dominant metal component of a PM that we measured during the competitions regardless of pistol or rifle used. However, its concentration did not exceed the permissible levels for indoor firing ranges.

Another crucial point is that lead-free bullets have reduced precision and accuracy at the shot point on the target. On the other hand, in terms of athlete and environmental health, green or "lead-free" bullets should be encouraged to use during indoor shooting to reduce the risk of Pb exposure. Among the elements analyzed, Cr was one of the elements that have cancer risk potential. Additionally, there are some technical differences between a .22-cal rifle and pistol. The pistol bore length is shorter, and the bore radius is larger than the rifle. Also, the rifle has a higher bullet core release speed than the pistol can cause more pollutant accumulation close to the shooter in pistol matches. It should also be considered that pollutants can accumulate in the environment for three days and repeatedly be resuspended from the ground. Ventilation is another critical point that should not be negligible. It should be kept out of the operational cost savings. Indoor air quality in terms of PM and all these pollutants should be closely monitored at indoor firing ranges to take more proactive actions against these pollutants. Furthermore, the operator should regularly check the ventilation system of indoor air firing ranges to ensure acceptable air quality. Personal sampling provides a more accurate evaluation of human exposure during shooting activities. Consequently, personal sampling and indoor air quality monitoring should be coupled to understand better the impact of indoor air firing ranges on the occupants.

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#### DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

#### CONFLICT OF INTEREST

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

#### ETHICS

There are no ethical issues with the publication of this manuscript.

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## Application of an airlift internal circulation membrane bioreactor for the treatment of textile wastewater

#### Abdulkadir ÇAĞLAK<sup>®</sup>, Nouha Bakaraki TURAN<sup>®</sup>, Hanife SARI ERKAN<sup>®</sup>, Güleda ÖNKAL ENGİN<sup>®</sup>

Department of Environmental Engineering, Yıldız Technical University Civil Engineering Faculty, İstanbul, Türkiye

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#### ABSTRACT

A large amount of water is used in the textile industry during the finishing and dyeing processes leading to the production of what is known as textile wastewaters. Textile wastewater is highly rich in COD and color and is characterized by relatively low biodegradability. This study aimed to investigate the treatability of reactive red dye-rich textile wastewater with the application of an airlift internal circulation membrane bioreactor (AIC-MBR). Experimental results demonstrated that high removal efficiencies of COD, NH<sub>3</sub>-N, and reactive red up to 99.70%, 97.83%, and 97.23%, respectively, can be achieved using the AIC-MBR system. Besides, EPS and SMP analyses reflected an SMP polysaccharide (PS) and protein (PN) membrane rejection that reached 88% and 72.6%, respectively. Finally, the capillary suction time measurement highlighted a good dewatering capacity of the sludge with a low membrane fouling tendency at the end of the operating period.

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#### INTRODUCTION

Textile industry is recognized as one of the most wastewater producing manufacturing sectors. Textile wastewater is colored with a composition, which varies according to the type of fiber and chemicals used, the techniques applied, and the machines operated [1]. It is known that approximately 6900 additives and 8000 dying agents used in the sector increase the organic and inorganic pollution load of textile wastewater [2]. Therefore, it is characterized by a high total dissolved solid (TDS), COD/BOD5, salt and color. Color is usually studied as an important parameter in textile wastewater because if not removed effectively, it reduces light scarcity in the receiving water affecting in turn aquatic organisms [3]. Different types of dyes are used in the textile industry classified as reactive, dispersive, basic, acidic, azoic, direct and sulfuric dyes. Azoic, sulfuric and dispersive dyes are easily removed from effluents as they are insoluble in water. However, highly soluble dyes, namely direct, basic, acidic and reactive dyes, are hardly removed by conventional separation and treatment methods. Furthermore, the most problematic dye group used in the textile industry is reactive dyes. Because these dyes may form complexes with heavy metals such as nickel, copper and chromium, thus,

\*Corresponding author.

\*E-mail address: nahnoouha87@hotmail.com



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posing a further risk when released into the environment [4]. Thus, different treatment processes including physical [5], chemical [6], electrochemical [7], and biological [8] are applied before its discharge to the environment.

Membrane bioreactors (MBRs) are proven to ensure higher effluent quality due to the complete retention of contaminants by the microbial community. However, membrane fouling is one of the limitations that affect MBR operation, increasing energy demand and reducing membrane permeability. Different reactor configurations have been developed to reduce this disadvantage of membrane bioreactors. An example of these is airlift membrane bioreactor which is characterized by a simple configuration [9]. The bioreactor is divided into a fully gassed riser and two ungassed downcomers. The density difference between them will result in the liquid circulation, which will positively affect the membrane fouling [10]. Airlift membrane bioreactors were successful in reducing the fouling processes by the rising air bubbles that generate shear stress and remove, in turn, the deposited particle along the membrane surface [11]. Besides, airlift membrane bioreactor was proved to reduce membrane fouling enhancing a better filtration performance with a lower energy consumption [12–14].

In this study, an airlift internal circulation membrane bioreactor (AIC-MBR) was tested for the treatability of Reactive Red dye-rich textile wastewater. Three different concentrations of the reactive dye were tested in the AIC-MBR. Operational parameters such as COD, NH<sub>3</sub>-N and EPS/SMP were monitored throughout the study. Besides, the removal of reactive red dye was investigated.

#### MATERIALS AND METHODS

### Airlift Membrane Bioreactor System and Operating Conditions

An airlift membrane bioreactor was used in this study for the treatment of synthetic textile wastewater. The reactor design is shown in Figure 1. The reactor was made of plexiglass (32 cm x 9 cm) with an effective volume of 1.5 L. Two vertical plates, having a height of 27.7 cm and placed 4.8 cm apart, were mounted inside reactor creating a riser and two downcomers. Each plate was perforated creating a hole opening of 1.8 cm. The air diffuser is located at the bottom of the reactor supplying air inside the riser. The holes on each plate ensured the circulation of air bubbles between the riser and downcomers. A hollow fiber membrane module having a pore size of 0.2  $\mu$ m and an effective area of 0.0170 m<sup>2</sup> was used in the AIC-MBR system. The membrane was made of polyvinylidene fluoride (PVDF)-based microporous membrane containing a small quantity of polyethersulfone (PES). The system was operated continuously based on the Archimedes theory where the same volume filtered from the membrane module was fed with synthetic wastewater to the reactor.



**Figure 1**. Schematic representation of the AIC-MBR system, air compressor (1), air diffusor (2), hollow fibre membrane module (3), manometer (4), peristaltic pump (5), feed tank (6), permeate tank, computer (7).

Table 1. Operating conditions of the AIC-	MBR
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Operating parameter	Value
рН	8.48
Dissolved oxygen, mg/L	8.65
Temperature, °C	18.28
SRT, days	20
HRT, hours	24
F/M (mg COD/mg MLSS.day)	0.26
Lorg, mg COD/L.day	1.095
Effective membrane area (m <sup>2</sup> )	0.017
Net flux (LMH)	3.70

<b>Table 2.</b> Operational remotes in the more with	Table 2. (	Operational	Periods in	the AIC-MBR
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Periods	Concentration of reactive red in feed (mg/L)	Duration (day)
1	0	1-13
2	10	14-34
3	20	35-66
4	40	67–95

AIC-MBR was operated at a steady-state condition under a sludge retention time of 20 days and an organic loading rate of 1.095 L/m<sup>2</sup>.d. The operating conditions are summarized in Table 1. Oxygen was provided continuously through an air diffuser. The trans-membrane pressure (TMP) was measured using a pressure gauge.

The concentration of the reactive red dye was increased every 30 days as 10, 20 and 40 mg/L, in order to monitor the treatability of three different concentrations of the colorant. The operational periods are shown in Table 2.

### Characterization of Textile Synthetic Wastewater and Inoculum

Real textile wastewater is characterized by a high concentration of biochemical oxygen demand (BOD) and chemical oxygen demand (COD), in addition to high values of chlorides, nitrates, suspended solids and metals [15]. However, synthetic wastewater is usually preferred because of its simplicity in evaluating results. The recipe used in this study was previously used by Yurtsever et al. (2016) as shown in Table 3 [16]. About 1000 mg/L of glucose as a source of biodegradable carbon source and different inorganics were added to meet the characteristics of real textile wastewater. Activated sludge seeded to the reactor was taken from a municipal wastewater treatment plant in İstanbul having an MLSS concentration of 5800 mg/L.

#### Sampling and Analyses Procedure

Chemical oxygen demand (COD), ammonium-nitrate NH<sub>2</sub>-N, mixed liquor suspended solids (MLSS) and mixed liquor volatile suspended solids (MLVSS) were measured three days per week based on the Standard Methods [17]. COD was analyzed in both the filtrate and the supernatant taken from the activated sludge. The dissolved oxygen (DO) and pH were monitored daily in the bioreactors using a WTW Multiline P4 multimeter (CellOx 325 DO probe and SenTix 41 pH probe). Extracellular polymeric substances (EPS) and soluble microbial products (SMP) with their protein and polysaccharide fractions were analyzed once per week from both the activated sludge and filtrate samples. EPS and SMP analyses were performed following the formaldehyde extraction method [18] where the protein (PN) and polysaccharide (PS) fractions were tested using the Lowry [19] and phenol-sulphuric acid methods [20]. The capillary suction time (CST) and sludge volume index (SVI) were analyzed every 15 days. CST was measured using a capillary suction timer (Triton type

 
 Table 3. Composition of the synthetic textile wastewater used in this study [16]

Added chemical	mg/L		
$\overline{C_{6}H_{12}O_{6}H_{2}O}$	1000		
NaHCO <sub>3</sub>	1000		
NH <sub>4</sub> Cl	230		
K <sub>2</sub> HPO <sub>4</sub>	37		
KH <sub>2</sub> PO <sub>4</sub>	67		
CaCl <sub>2</sub> .2H <sub>2</sub> O	4		
MgCl <sub>2</sub> .H <sub>2</sub> O	3.4		
FeSO <sub>4</sub> .7H <sub>2</sub> O	5.92		
MnSO <sub>4</sub> .2H <sub>2</sub> O	0.4289		
ZnSO±.7H <sub>2</sub> O	0.1053		
Na <sub>2</sub> SO <sub>3</sub>	0.2811		
NiSO <sub>4</sub> .6H <sub>2</sub> O	0.1		
CoCl <sub>2</sub>	0.5457		
Reactive red	10, 20 and 40		

304M) and a standard filter paper obtained from Triton. EPS, SMP and CST analysis were measured at steady-state conditions. Reactive red color treatability was monitored by analyzing its concentration in both the filtrate and the supernatant from the activated sludge. Reactive red was measured at a wavelength of 455 nm using Hach Lange DR 5000 spectrophotometer.

#### **RESULTS AND DISCUSSION**

#### **Evaluation of AIC-MBR System Performance**

Figure 2 shows the variations in the MLSS and MLVSS concentrations in addition to MLVSS/MLSS ratio. The



Figure 2. Variation of MLSS and MLVSS concentration and ratio.



Figure 3. Variation of membrane flux and TMP values throughout AIC-MBR operation.



Figure 4. COD removal efficiencies in the supernatant and permeate over time.

initial MLSS concentration was 5800 mg/L which fluctuated until it reached its steady-state conditions after about 20 days. MLSS was then kept stable between 4000 to 5000 mg/L for the remaining 80 days of operation. Similarly, the MLVSS and MLVSS/MLSS ratio changed in the range of 4000 to 3000 mg/L and 75.5 to 85.5%, respectively. This high MLVSS/MLSS ratio reflected the microbial activity in the reactor. An SRT value of 20 days was kept by drawing a daily sludge volume of 75 mL from the AIC-MBR. The SRT in aerobic MBRs is usually preferred to be between 20 and 50 days, which confirms the suitability of the operating SRT in this study [21].

The average membrane flux was about  $3.70 \text{ L/m}^2$ .h (LMH) where the transmembrane pressure (TMP) decreased from zero to -0.2 bar at the end of the operating

period (Fig. 3). Membrane flux and TMP were directly affected by the aeration rate influencing in turn membrane fouling. In this study, no back-washing or chemical washing was applied to the membrane. Therefore, the minimal reduction occurred in TMP is acceptable.

#### COD Removal Efficiency by AIC-MBR

COD removal efficiency was measured in both the permeate and the supernatant during the operating period of the reactor. As can be seen from Figure 4, the COD removal efficiency in the supernatant fluctuates in the range of 93.02–98.91% while in the permeate it registered a removal efficiency of about 98.62 to 99.70%. Thus, COD removal efficiency was higher in the permeate then in the supernatant noting that both demonstrated great removal efficiencies. This is in correspondence with Lee



Figure 5. NH<sub>3</sub>-N removal efficiencies over time.



Figure 6. Color removal efficiencies over time.

et al. (2003) who found out that the COD removal in the permeate was higher than the supernatant [22]. Besides, Yurtsever et al. (2016) reported in their study that COD concentrations of 54±14 mg/L and 47±12 mg/L were recorded for the supernatant and permeate, respectively [16]. Berube et al., (2010) reported that the COD removal efficiency in a conventional activated sludge process may typically reach about 95%, however, this value may increase to reach a range of 96 to 99% in a membrane bioreactor process [23]. This can be explained by the fact the colloids and soluble compounds can be attached to the suspended solids which can be retained in MBR system giving a better COD removal and particle-free effluent [24, 25]. The average removal efficiency of 90% was obtained in another study that used airlift external circulation membrane bioreactor (AEC-MBR) for the treatment of toilet wastewater [26].

#### NH<sub>3</sub>-N Removal Efficiency by AIC-MBR

The laboratory prepared synthetic textile wastewater had a NH<sub>4</sub><sup>+</sup> concentration of 77.4 mg/L. The removal efficiency of NH<sub>3</sub>-N in the permeate from an AIC-MBR was studied and the experimental results are presented in Figure 5. NH<sub>3</sub>-N removal efficiency changed between 95.73 and 97.83% during the operating period. However, an important decrease in NH,-N removal was observed after the addition of reactive red color to the reactor which affected the performance of the microorganisms. The adaptation time of microorganisms to any new component entering the reactor medium may differ in COD values, which may explain the reason behind the sudden decrease observed on the day of dye addition [27]. The removal efficiency of 96.6% was obtained using a draft tube MBR without carriers for the treatment of mixed wastewater [28]. The COD/TN ratio with the dissolved

	Ghaly et al. (2014) [32]	Kehinde and Aziz (2014) [33]	Tavangar et al. (2019) [34]	Bhuvaneswari et al. (2016) [35]	Yurtsever et al. (2020) [36]
рН	6–10	6.95-11.8	7.03±0.01	8.6-9.2	8-9.5
Color (Pt-Co)	50-2500	50-2500	2100±5	N/A	500-1250
COD (mg/L)	150-12000	150-30000	2690±10	3880-4400	700-1250
TSS (mg/L)	15-8000	15-8000	280±2	550-650	200-450

Table 4. Typical characteristics of textile effluents



**Figure** 7. Variance in the protein and polysaccharide concentrations of EPS (a), polysaccharide (b) and protein (c) concentrations in SMP in both supernatant and permeate.

oxygen level was found to affect the simultaneous nitrification and denitrification in an airlift membrane bioreactor. A COD/TN level in the range of 4.77 to 10.04 leads to a nitrogen removal that exceeds 70% which is comparable to the results in this study taking into consideration that COD/TN is about 12.92 [29].

#### Reactive Red Removal Efficiency by AIC-MBR

Reactive red was used in this study as the type of colorant to be removed. As mentioned earlier, three different concentrations of reactive red were added subsequently to the reactor. The color removal efficiency was monitored in both permeate and supernatant and the obtained results are shown in Figure 6. The removal efficiency in the supernatant varied between 83.25 and 94.47% where it increased to 89.29 and 97.23% in the permeate. A slight decrease occurred after the addition of the third concentration of reactive red of 40 mg/L. Color removal was performed using different methods in the literature. The removal efficiency varied according to the initial concentration and the treatment process applied [30]. In an anaerobic/aerobic sequential batch reactor system, 20 mg/L of reactive black 5, reactive blue 19 and reactive blue 5 was treated and the removal efficiencies were found to be 63, 64 and 66%, respectively [31]. Color may be removed through different processes through cleavage of the chemical bonds or adsorption to the microbial flocs. The removal occurred in supernatant reflects the role of the microbial activity in reducing or adsorbing the dye molecules [30]. Besides, the higher removal efficiencies obtained in the permeate is in close relationship with the membrane used which blocks the passage of any particle, as mentioned earlier.

The composition of textile industry wastewater varies from factory to factory and from country to country, depending on the process, the equipment used, the type of fabric produced and the chemicals applied [30]. Table 4 shows the real textile wastewater characteristics reported from different sources and countries. As seen from Table 4, the pH value of real textile wastewater varies in a wide range between 5.5–11.8. This wide pH or COD variations may lead to negative effects in MBR operation, especially on active biomass, making it impossible to achieve a stable biological treatment in terms of pollutant removal and membrane fouling.

#### Variation in EPS and SMP Components

Extracellular polymeric substances (EPS) are divided into bound EPS and soluble EPS [mainly named soluble microbial products (SMP)]. Major constituents of EPS consist mainly of proteins and polysaccharides [37]. The results regarding the variation in EPS and SMP concentrations are presented in Figures 7, respectively. An overall evaluation of both plots in Figure 7a shows a general increasing trend in EPS and SMP level with time noting that EPS values were higher than SMP ones. The increase in these values may be attributed to the increase in MLSS concentration in the reactor. Besides, the increasing trend may be related to the increase in the TMP levels discussed above and thus to membrane biofouling [38]. It was stated by researchers that EPS concentration and characteristics are directly affected by multiple parameters such as the sludge composition, sludge retention time (SRT) and aeration [39, 40]. The comparison between the polysaccharide and protein fractions of EPS and SMP reflected a superiority of polysaccharide levels in both EPS and SMP. Polysaccharide fractions usually affect fouling in membrane bioreactors, as reported by Yigit et al. [41]. SMP polysaccharide (PS) and protein fractions (PN) were measured in both the supernatant and permeate. The corresponding results are presented in Figure 7b and 7c. It can be seen clearly than PS and PN of SMP shows higher and increasing values in the supernatant than in the permeate. The PS rejection of SMP by the hollow membrane was found to be 0.88% at 28<sup>th</sup> day, while the PS rejections by the membrane increased in the next operation period and were determined between 64 and 88%. On the other hand, PN rejection of SMP was found 31.4% at 28th day, while after this point, the rejection rates were found to vary between 49.7 and 72.6%. The obtained results can be explained by the membrane filterability which blocks the soluble part of EPS that may be adhered to microbial flocs or other particles in the sludge [38]. However, the minimal part of the SMP appearing in the permeate is caused by the membrane permeability that may enhance the passage of the soluble and non-adhesive part of the SMP [42].

#### Variation in CST Levels

The capillary suction time (CST) of the sludge was measured in order to test the dewatering properties of the activated sludge. As can be seen from Figure 8, CST values showed fluctuations between 5.5 and 9.6 s with a minimal increase at the end of the operating period. Additionally, an increase observed in the CST may be related to the increase occurred in the polysaccharide values of EPS. However, CST values were still in an acceptable range reflecting a good dewatering capacity of the sludge [43].

On the  $69^{th}$  day of operation, CST value decreased from 6.9 s to 5.5 s and then increased to 9.6 s on the  $90^{th}$  day.



Figure 8. CST values obtained from activated sludge samples over time.

These results could be related to the concentrations of PN and PS of SMP, as the CST values presented a similar trend with SMP concentrations of supernatant. It was reported by Zhang et al. (2015) [44] that there is a significant relationship between SMP and membrane fouling and SMPs are the main soluble components in a gel layer and cake layer on the membrane surface and pores. Higher specific filtration resistance occurred in the gel layer due to the adsorption of SMP [45].

#### CONCLUSION

The application of an AIC-MBR demonstrated quite high COD,  $NH_3$ -N and reactive red removal efficiencies up to 99.70%, 97.83%, and 97.23%, respectively. Moreover, the proposed system ensures satisfactory results even with increased initial concentrations of the reactive dye. Additionally, a relatively low membrane fouling tendency was observed in this technology based on the results from EPS, SMP and CST. The overall results indicated that the AIC-MBR can be applied for the treatment of real textile wastewater, after necessary preliminary tests were carried out.

#### DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

#### **CONFLICT OF INTEREST**

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

#### ETHICS

There are no ethical issues with the publication of this manuscript.

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# Optimization of the effect of copper electrodes on the removal efficiency of 4-clorophenol from aqueous solution by electrocoagulation

Gülizar KURTOĞLU AKKAYA<sup>\*1</sup>0, Muhammed Kamil ÖDEN<sup>2</sup>0

<sup>1</sup>Department Environmental Engineering, Necmettin Erbakan University, Konya, Türkiye <sup>2</sup>Department of Environmental Protection Technology, Selcuk University Sarayönü Vocational High School, Konya, Türkiye

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#### ABSTRACT

In this study, the investigation of 4-clorophenol (CP) removal from aqueous solutions using copper electrodes by electrocoagulation (EC) process was done. The effects of various experimental parameters such as pH, current density and exposure time, which affect the EC process, on 4-CP removal were investigated. To optimize the process, response surface methodology (RSM) Box Behnken Design was used by MINITAB program, a series of experimental sets were obtained and carried out. Afterward, 4-CP removal was analyzed and calculated. Results were entered into the MINITAB program as a response. At the end of the optimization, optimum operating conditions were determined as 74 mA/cm<sup>2</sup>, 45 min, 4.24 for current density, exposure times and pH, respectively. When the results were evaluated, approximately 92% phenol removal efficiencies were obtained. Additionally, according to the model results, it was understood that the factors with the greatest effect on 4-CP removal were the exposure time and current density and these had a linear effect, but the pH value did not have a significant effect.

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# INTRODUCTION

With the developing technology and increasing population, the reequipments of people have changed and increased. Uncontrollable increase in production and consumption has caused rapid depletion of resources and increased pollution. As a consequence, it increases the consumption of water, which is one of the most important natural resources and rapidly causes pollution of water resources.

Water is necessary for developing countries [1, 2]. Agricultural irrigation uses water resources the most with

70% of the total water consumption. This is followed by electricity generation, steam-based electricity generation and cooling waters [1].

The wastewaters are identified as waters, which are used in various processes in people activities and in some industries, are discharged to the receiving environments. They disrupt the ecological balance in the receiving environment where they are discharged or they cause great destruction. For this reason, they must be treated before been discharged into the receiving water environments and must obtain the legal discharge limit values [3].

\*Corresponding author.

\*E-mail address: ka.gulizar@gmail.com



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Wastewaters are treated using conventional processes such as biological processes [4], physico-chemical processes [5], as well as developing technologies, namely membrane filtration [6] and adsorption [7], advanced oxidation processes [8]. Advanced oxidation processes require strong oxidizers, which makes wastewater treatment safe and cost-effective. Biological processes, on the other hand, require tightly controlled conditions with long retention times. Chemical processes require extensive chemical addition, which not only increases the cost of the process but also complicates downstream processes with an increased risk of secondary contamination. Membrane filtration and adsorption methods alone cannot effectively treat wastewater unless integrated with extensive pretreatment processes, especially in the treatment of phenolic compounds. Therefore, wastewater treatment research using electrochemical processes is highly attractive [9, 10].

Recently, electrochemical treatment methods such as electrooxidation and electrocoagulation (EC) have drawn attention due to their environmental importance and economic efficiency [11, 12]. High removal efficiencies have been achieved for treating different wastewaters and drinking water using the EC process [3]. EC is a complex process in which sacrificial electrodes are used to form ions that act as coagulants in wastewater, including many removal mechanisms such as adsorption, co-precipitation, chemical oxidation-reduction and flotation [13]. Metal hydroxides are formed by the anodic dissolution and hydrolysis of metal anodes, such as aluminum and iron [14].

There are many pollutants in wastewater that needs to be treated, depending on the variety of activities. Pollutants such as color, hardness, dissolved oxygen, detergent types, heavy metals, oil-grease, polycyclic aromatic hydrocarbons (PAH) and volatile organic compounds and similar toxic and/or carcinogenic pollutants, nitrogenous compounds, biological hazards and phenols are available in water and wastewater [15].

Phenolic compounds have significant effects on the environment and people health [16]. They are easily taken into the body by respiratory, oral and dermal routes. Phenol and phenol forms are contained in products such as nylon, epoxy resins, surfactants, synthetic detergents, plasticizers, antioxidants, phenolic resins, cyclohexanol, aspirin, dyes, wood preservatives, drugs, fungicides, gasoline additive, inhibitors, explosives and pesticides. Phenol causes excessive irritation and corrosion if it comes into contact with the skin and other tissues. Absorption of phenol can cause cyanosis, shock, weakness, liver and kidney damage, coma and death [17, 18].

The industry wastewaters are one of the sources of industrial phenolic pollutants. There are wastewaters containing high amounts of phenol in the world and in Turkey. These are mainly found in iron and steel factories, coke ovens,



Figure 1. Schematic representation of EC in the laboratory.

petrochemical facilities, pharmaceutical factories, socks factories, plastic factories, paper industry, photographic industry, explosive industry, paint, pharmaceutical and resin production facilities [19, 20].

Besides phenol which is a common research pollutant model, 4-chlorophenol (4-CP) is a highly toxic chemical compound which is difficult to degrade and is commonly found in industrial wastewater. Because of these properties, 4-CP was chosen as the target pollutant for this study. The treatment of it by EC was investigated. Here, the effects of operational parameters that significantly affect the EC process such as initial pH, current density and electrolysis time on 4-CP removal efficiency in a reactor using monopolar parallel connected copper electrodes were investigated.

# **EXPERIMENT**

# Material and Methods

Stock solutions of 4-CP were prepared by dissolving 1 g of analytical reagent grade (Merck, Germany) in 1 L of distilled water without pH adjustment. Experiments were conducted on 4-CP solutions with 50 mg/L concentration prepared during the experiment. 2 mg/L NaCl was added to increase the ionic strength of the solution. Each run volume was 250 ml. In Figure 1, a schematic representation of the laboratory scale system used in the EC process is given. The EC reactor, made of Plexiglas, has a diameter of 19 cm and a height of 15 cm. In the study, the copper plate electrodes were used on the anode and cathode (6 cm wide, 11 cm high and 0.2 cm wall thickness) and the distance between the electrodes was 2 cm. The total effective electrode areas were determined as 24 cm<sup>2</sup>. The reactor was supplied with a direct current power supply. A homogeneous mixture of the solution was provided by means of a magnetic stirrer at 150 rpm. The electrodes were washed 2 times before each run. After run was completed, samples were waited for a 1-hour settling period. Supernatants of samples were collected and the remaining 4-CP concentration was determined according to the Direct photometric method (Standard Methods 5530-D) [21]. The method

1 0	-			
Factors	Factor	-1	0	1
Initial pH	$X_1$	4	7	10
Exposure time (min)	$X_2$	5	25	45
Current density (mA/cm <sup>2</sup> )	X,	10	42	74

**Table 1.** Experiment design input and factors

based on the spectrophotometric analysis of the developed color resulting from the reaction of 4-CP with 4-aminoantipyrine. The absorbance of colored samples was measured at 500 nm by the Hach UV/Vis DR5000 spectrophotometer after 15 min of reaction. To find the 4-CP concentration, 11 standard solutions (0.5, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10 mg/L) were prepared and a calibration curve was drawn. After the analysis, the absorbances of 4-CP were read by spectrophotometer and put in the calibration equation. Afterwards, the 4-CP concentrations were calculated. 4-CP removal efficiency were obtained using equation 1.

$$R\% = \frac{100(C_0 - C_1)}{C} \tag{1}$$

where  $C_0$  and  $C_1$  are initial and final concentration, respectively.

# **Experimental Study and System Design**

4-CP removal by EC process was optimized using the Response-Surface Method (RSM). The current density, exposure time and pH affecting the EC process were optimized using RSM. The Box-Behnken design model included in Minitab 17 (Trial version) software is used to determine the individual and combined effects of operating parameters on pollutant removal efficiency. In this study, experiments for 4-CP removal were designed by the Box-Behnken design. As shown in Table 1, the three-factor Box-Behnken response surface was used in the optimization and investigation of the process variables, namely initial pH (4-10), exposure time (5-45 mins), current density (10-74 mA/cm<sup>2</sup>). With the runs created in the program, 15 batch EC processes were conducted and the effects of initial pH (X<sub>1</sub>), current density (X<sub>2</sub>) and exposure time (X<sub>3</sub>) changes on 4-CP removal efficiency were determined. Depending on the responses, the relationships between the criteria affecting the removal efficiency were determined.

The effects of the selected independent variables and their interactions on responses according to BBD were described using the second order polynomial equation 2.

$$Y_{i} = \beta_{0} + \sum_{i}^{k} \beta_{i} x_{i} + \sum_{i=1}^{k} \sum_{j \ge i}^{k} \beta_{ij} x_{i} x_{j} + c$$
(2)

where  $\beta_0$ ,  $\beta_i$ ,  $\beta_{ii}$ , and  $\beta_{ij}$  are constant, linear, quadratic, and cross-factor interaction coefficients, respectively;  $X_i$  and  $X_j$  represent the independent variables;  $Y_i$  is the predicted response; and k and C are the number of factors and the residual terms, respectively [22].

Run order	рН	Current density (mA/cm²)	Exposure time (min)	Efficiency (%)
1	4	10	25	1
2	10	10	25	7
3	4	74	25	44
4	10	74	25	73
5	4	42	5	1
6	10	42	5	4
7	4	42	45	83
8	10	42	45	41
9	7	10	5	1
10	7	74	5	1
11	7	10	45	11
12	7	74	45	81
13	7	42	25	44
14	7	42	25	50
15	7	42	25	48

#### Table 2. Experimental results

#### **RESULTS AND DISCUSSION**

Response Analysis and Interpretation using BBD Three factors in the three-level Box-Behnken response surface design (BBD) is used to optimize and examine the effect of process variables. Table 1 shows the BBD matrix of the independent variables and the response values. Then, the experimental results for 4-CP removal efficiency (Table 2) were entered to software Minitab 17 (trial version). The calculated response functions correspond to the experimental data for Y (Efficiency)  $R^2$ =92.51.

Regression Equation in Uncoded Units

The adequacy of the quadratic model and the quality of the correlation between parameters and responses were examined by analysis of variance (ANOVA) at 95% confidence interval (CI). For a term to be meaningful in ANOVA, its p-value must be less than 0.05. The p-value is used as a tool to check the significance of each factor [23]. The ANOVA table of the factors is given in Table 3. When the Table 3a, b is examined, it is seen that the factors that have the greatest effect on 4-CP removal, exposure time and current density factors have a linear effect, and the p-values are 0.007 and 0.003. However, the p-value of the pH is 0.935 and it does not have a significant effect.

Generally, the coefficient of determination (R-squareadj) explains the total variability taken into account by the independent variables in the regression model. In

			Analysis of varianc	e	
Source	DF	Adj SS	Adj MS	F-value	P-value
Model	9	12443.7	1382.63	6.87	0.024
Linear	3	9575.7	3191.91	15.85	0.005
X1	1	1.5	1.46	0.01	0.935
X2	1	4027.4	4027.42	20	0.007
X3	1	5546.8	5546.85	27.55	0.003
Square	3	999.8	333.27	1.66	0.29
X1*X1	1	40	40	0.2	0.674
X2*X2	1	571.1	571.14	2.84	0.153
X3*X3	1	495.4	495.37	2.46	0.178
2-way interaction	3	1868.1	622.71	3.09	0.128
X1*X2	1	129	128.98	0.64	0.46
X1*X3	1	515.9	515.91	2.56	0.17
X2*X3	1	1223.2	1223.24	6.07	0.057
Error	5	1006.8	201.36		
Lack-of-fit	3	986.9	328.95	32.96	0.03
Pure error	2	20	9.98		
Total	14	13450.5			

Table 3a. ANOVA tabl	e for 4-CP removal	(analysis of	f variance)
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Table 3b. ANOVA table for 4-CP removal (continued-coded coefficients)

		Coded coefficients					
Term	Effect	Coef	SE coef	T-value	P-value	VIF	
Constant		47.14	8.19	5.75	0.002		
X1	-0.85	-0.43	5.02	-0.09	0.935	1	
X2	44.87	22.44	5.02	4.47	0.007	1	
X3	52.66	26.33	5.02	5.25	0.003	1	
X1*X1	-6.58	-3.29	7.38	-0.45	0.674	1,01	
X2*X2	-24.87	-12.44	7.38	-1.68	0.153	1,01	
X3*X3	-23.17	-11.58	7.38	-1.57	0.178	1,01	
X1*X2	11.36	5.68	7.1	0.8	0.46	1	
X1*X3	-22.71	-11.36	7.1	-1.6	0.17	1	

# Table 3c. ANOVA table for 4-CP removal (continued-model summary)

		Model summary	
S	R-sq	R-sq(adj)	R-sq(pred)
14,1902	92.51%	79.04%	0.00%

this model, the Adj. R<sup>2</sup> sufficient to explain the variability were found to be 79.04% (Tablo 3c).

The residuals are the differences between the observed and the predicted values of the data. It is a diagnostic measure used when evaluating the quality of a model, is known as errors. Residues are important when determining the quality of a model. The residuals are normally distributed. Figure 2 is shown residual plots. The normal probability plot shows



Figure 2. Residual plots for efficiency.

that the data are normally distributed and that variables affect the response. In this model, outliers were not found in the data. The fitted values against the residual values show that the variance is constant and there is a non-linear relationship. The histogram proves that the data is not skewed and that there are no outliers. Residuals by order of data indicate systematic effects on data due to time or order of data collection.

#### **Effects of Operating Parameters**

4-CP was removed from the water by floc formation due to oxidation of the anode. This was a three-step process with oxidation of the anode, adsorption/replacement of contaminants, and precipitation of aggregated mass of coagulant with 4-CP. During the initial processing, the anode material changed from coagulant to its hydroxides (insoluble in water). In the second step, 4-CP was adsorbed on the coagulant surface. In the third step, the flocs are precipitated and then removed from the water by filtration. Experiments were conducted at pH 4.0–10.0, under both acidic and alkaline conditions. 4-CP removal was observed in both acidic and basic environments, confirming the hydroxide formation in both cases. The electrochemical reactions of hydroxide formation for copper electrodes are shown below.

Reactions occurring on copper electrodes [24].

Anodic electrochemical dissolution:

$$Cu_{(s)} \to Cu_{(aq)}^{2+} + 2e^{-}$$
 (4)

Oxidation of copper ions:

$$Cu_{(aq)}^{2+} + 2H_{(aq)}^{+} + \frac{1}{4}O_{2(g)} \to Cu_{(aq)}^{2+} + \frac{1}{2}H_2O(I)$$
(5)  
The hydrolysis reaction:

$$Cu_{(aq)}^{2+} + 2H_2O(I) \to Cu(OH)_{2(s)} + 2H_{(aq)}^+$$
(6)

The cathodic electrochemical reaction:

$$2H_{(aq)}^+ + 2e^- \to H_{2(g)} \tag{7}$$

Overall reaction:

$$2Cu_{(s)} + 3H_2O(I) + \frac{1}{2}O_{2(g)} \to 2Cu(OH)_{2(s)} + H_{2(g)}$$
(8)

The Cu<sup>2+</sup> and Cu<sup>3+</sup> ions hydrate and hydrolyses to form monomeric and polymeric species: Cu(OH)<sub>2</sub><sup>+</sup>, CuOH<sup>2+</sup>, Cu<sub>2</sub>(OH)<sub>2</sub><sup>4+</sup>, Cu(OH)<sub>4</sub><sup>-</sup>, Cu(H<sub>2</sub>O)<sub>2</sub><sup>+</sup>, Cu(H<sub>2</sub>O)<sub>5</sub>OH<sup>2+</sup>, Cu(H<sub>2</sub>O)<sub>4</sub>(OH)<sub>2</sub><sup>+</sup> etc. [25].

Electrical conductivity in metals is a result of the movement of electrically charged particles. Atoms of metal elements are characterized by the presence of valence electrons. It is 'free electrons' that allow metals to conduct an electric current. Copper has high conductivity ( $6.30 \times 10^7$  S/m at 20°C) and less resistivity ( $1.59 \times 10^{-8} \Omega$  m at 20°C). °C). Energy transfer is strong when there is little resistance. For this reason, it was effective in the removal of 4-CP.



Figure 3. Contour plots and 3D plots.



Figure 4. Process optimization for 4-CP removal.

The pH of the EC process is one of the most fundamental factors for controlling the removal of contaminants [26, 27]. To examine the pH effect in the EC, experiments were conducted at various pH values covering acidic, neutral and basic conditions (range 4-10 pH). Maximum 4-CP removal occurred at pH 4-7. The applied current density, on the other hand, directly affects the coagulant dosage rate and bubble generation rate, the current density, which determines the size and growth of the flocs, and affects the removal efficiency of any electrochemical process [28]. Considering the results in Table 2, the increase in the applied current density increased the 4-CP removal, too. The current density produced divalent ion Cu<sup>2+</sup> and removed it by reacting with 4-CP. Applying the current density for a long time provided ion increase and increased the removal efficiency. 4-CP removal efficiency was 44% in Run 12, while it was 83% in Run 13. After the 25<sup>th</sup> minute of the exposure time, there was an increase in the removal efficiency.

The combined effects of these 3 factors affecting the EC process were evaluated. For this, the Contour Graph in RSM was used to examine the relationship between a response variable and two predictive variables. In a contour plot, the values of the two predictive variables are represented on the x and y axes, and the values of the response variable are represented by darksome regions called contours. In this study, two-dimensional (2D) contour graphs were drawn representing the interaction effects of pH-Current density, pH-Exposure Time and Current Density-Exposure time on 4-CP removal efficiencies.

In the graphs (Fig. 3), the darker regions show the high removal efficiency achieved by the interaction of the two factors pH and current density. When the contour graph is examined, it is seen that the current density partially interacts with pH and affects the removal efficiency. However, exposure time and current density significantly interact with each other. With the increase of theirs, it was increased to release of copper ions. Thus, easier precipitation of pollutants is ensured and 4-CP removal efficiency is increased. It is indisputable that current density and exposure time greatly affect 4-CP removal with copper electrodes.

Exposure time is one of the most important parameters in the EC. In the EC process, the sufficient time is required for the copper electrodes to dissolve, produce hydroxide, and complete the coagulation. It is seen that the removal efficiency increases with the increase in the study time. This trend can be explained by the greater current supplied to the electrodes, resulting in a greater dosage of coagulant into the water [29].

#### **BBD** Optimization

The model was finalized with the multi-response numerical optimization technique. According to the BBD results in Figure 4, the maximum 4-CP removal efficiency was obtained under the following operating conditions at a pH of 4.24, an exposure time of 45 min, and a current density of 74 mA/cm<sup>2</sup>. Under these optimal conditions, the 4-CP removal efficiency was 92.20%. It can easily be deduced from this graph that while current density and exposure time have a positive effect, pH removal has not much effect.

As can be seen in Table 4, phenol types removal from real and synthetic wastewater were investigated. The phenol removal efficiencies vary between 100% and 59%. The electrodes used such as aluminum, iron, zinc, steel and metals enriched with different substances. In this study, 4-CP removal was demonstrated using monopolar parallel connected copper electrodes. With the study, approximately 92% phenol removal was achieved. In this study, better efficiency was obtained than many studies in the literature. However, it will be more important to work with real wastewater and achieve the same efficiency in the field.

Wastewater type	Treatment technology	% Phenol removal efficiency	Initial [phenol] (mg L <sup>-1</sup> )	Reference
Phenol wastewater	Persulfate enhanced electrochemical oxidation	97	500	[30]
	[EC/PS oxidation (CuFe <sub>2</sub> O <sub>4</sub> /ACF cathode and			
	RuO <sub>2</sub> /Ti anode, adding PS)]			
High salinity waters	Electrochemical advanced oxidation	>70	50	[31]
Synthetic phenol solution	Ti/Pt and MMO (Mixed Metal Oxide) electrodes	84	0.5	[32]
Pistachio processing wastewater (PPW)	The combined electrochemical-assisted fungal	88.7	3205	[33]
	treatment process			
Synthetic phenol solution	Electrochemical filtration carbon membrane (ECM)	91.65	50	[34]
Synthetic phenol solution	Bioelectrochemical Technologies [oxygen-diffused	59	200	[35]
	microbial electrochemical systems (MESs)]			
The simulated and the real wastewater	EC (Zn anode/stainless steel cathode)	84.2-72.3	327-740	[36]
(olive mill)				
Real refinery wastewater	EC (Al electrode)	100	3	[18]
Synthetic phenol solution	EC (Al and Fe electrode)	94.72-98	5	[37]
Petroleum refinery wastewater	electrochemical oxidation by using boron doped	98.74	192.9	[29]
	diamond anode (BDD)			
Paper mill effluents	EC (Al and Fe electrode)	98-93	0.535	[38]
Synthetic phenol solution	Electrochemical oxidation (ruthenium mixed	99.7	200	[39]
	metal oxide electrode)			
Olive mill wastewater (OMW)	EC (Al electrode)	91	2420	[40]
Synthetic 4-chlorophenol solution	Biochar-load particle electrodes	99.93	500	[41]
Synthetic 4-chlorophenol solution	Two Pd/graphene gas-diffusion cathodes and one	94.9%	-	[42]
	Ti/IrO2/RuO2 anode			
Paper industry wastewater (4-chlorophenol)	Al electrode	100%	0.28	[43]
Paper mill waste water (4-chlorophenol)	Stainless steel (316 L) electrode	99.4%	0.56	[44]
Synthetic phenol solution	EC	92	50	This study

Table 4. Some studies in literature about removal of phenol types from wastewaters

Besides EC, other treatment methods are used efficiently in 4-CP removal [45, 46]. For example, 4-CP removal performances were investigated by synthesizing different carbon adsorbents and 4-CP absorbability was observed as 90% [47]. In a different study, it was proven that 4-CP can be removed 40% from the solution by using composite TiO2 material [48]. Also, over 80% 4-CP removal was achieved with magnetic activated carbon [49]. It is observed from this study that the removal of 4-CP is effective with 92%.

# CONCLUSION

In this study, the effect of initial pH, current density and electrolysis time on 4-CP removal efficiency in a reactor using parallel connected copper electrodes was investigated. In the optimization study using the RSM, the maximum phenol removal was found as a function of the current density, reaction time and initial pH value affecting the EC process. The ANOVA results showed the fit of the second-order regression with the experimental data. As a result of this study, it has been determined and optimized that 4-CP is effectively removed from the wastewater with the copper electrodes.

# DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

# CONFLICT OF INTEREST

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

# ETHICS

There are no ethical issues with the publication of this manuscript.

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**Research Article** 

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# Life cycle comparison of passenger air and rail transportation

Levent BİLGİLİ<sup>1</sup><sup>®</sup>, Afsin Y. CETİNKAYA<sup>2</sup><sup>®</sup>, S. Levent KUZU<sup>2</sup><sup>®</sup>

<sup>1</sup>Department of Naval Architecture and Marine Engineering, Bandırma Onyedi Eylül University, Balıkesir, Türkiye <sup>2</sup>Department of Environmental Engineering, Yıldız Technical University Faculty of Civil Engineering, İstanbul, Türkiye

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#### ABSTRACT

Air transportation has an undisputed speed advantage among all other modes. On the other hand, it is known that the environmental metrics of aviation is quite unsatisfactory compared to other transportation types due to its fuel characteristics and the amount of consumed fuel. However, it would be a wrong choice to rely solely on operational processes to make a true comparison. For this reason, a Life Cycle Assessment (LCA) model should be generated by taking into account processes such as production except the operation process and the calculations should be performed with a comprehensive and holistic perspective. In this study, the environmental impacts of air and rail transport types are compared from the life cycle perspective. For this purpose, first, the emissions in the case of one passenger per one km (pkm) transportation by air and rail were calculated. Then, taking into account the production and disposal processes of the aircraft and passenger trains, the LCA cycle was completed and total emissions were calculated. SimaPro version 9.0.0.49 package program and 1.09 version of ReCiPe 2008 method were used for LCA calculations. With the help of the program, emissions generated during both production and one pkm transportation processes of an aircraft, high-speed and normal train were estimated. Accordingly, the greenhouse gas produced one pkm in air transport was 126.8 g in terms of carbon dioxide equivalent (CO<sub>2eq</sub>), while CO<sub>2eq</sub> was 0.3 and 0.31 g for high-speed trains and regular trains, respectively. Considering the production processes, 2072.1, 28.72, and 19.07 t of greenhouse gases are produced, respectively for these three transportation modes.

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# INTRODUCTION

In today's society, environmental aspects are important in the transportation sector. The society will presumably change their choice in the transport sector due to changes in the energy/environmental situation in the future [1]. Strategic decisions concerning the development of the transport sector must be based on solid facts concerning both the transport infrastructure and the operation. With the increase in  $CO_2$  emissions, the effects of global warming and climate change, which have accelerated in recent years, have become more evident [2–4]. Understanding the importance of  $CO_2$  provides a new perspective on the discussion of air and rail transport uses [5]. The two transportation modes differ among themselves in terms of supply chain, costs, and energy use. In the literature, each trans-

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\*Corresponding author.

\*E-mail address: skuzu@yildiz.edu.tr



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portation mode was examined individually and has not been compared to any other transportation mode [6-8]. It is important to examine the passenger air and the rail transportation and to compare them with each other.

The LCA method is used to identify, report, and manage environmental impacts at different stages of the life cycle, starting with the acquisition of raw materials that are used in the production of a product or service, including all relevant production, shipment, use by the consumer and disposal as waste [9-12]. LCA evaluates the entire life cycle of a product or service and their connections with each other. As a result, any environmental impact that may arise in all processes from "cradle to grave" of the product or service being evaluated [13, 14]. The railway system represents one of the most resource-efficient answer to the ever-growing demand for transport service. Development trends for the following years project a substantial increase in this sector. Environmental effects caused by railway transport services have been rarely inspected systematically and existing studies focus on single typologies of environmental aspects, like energy consumption and air emissions [15, 16]. A wide summary of LCA based studies showing the transportation sector was provided by Andreoni et al. [17].

In today's world, the aviation industry is becoming more and more important. While it is important in terms of the benefits it creates, it also contributes especially to air pollution and global warming, a problem that the whole world needs to find a solution [18]. Significant contribution to atmospheric pollution was determined in Istanbul Atatürk Airport in Turkey [19]. Some parameters such as NO<sub>x</sub> exceeded the air quality threshold value within airport area. The resources used in aviation sector activities directly affect the lives of living things, cause climate change, and create global externalities. The aircraft industry continues to grow as predicted, so reducing emissions is critical. Aviation is responsible for almost 12% of transport related global CO<sub>2</sub> emissions, having approximately 500 Mt annual CO<sub>2</sub> emission [20]. Global passenger and freight air transportation is expected to grow annually at 3-5% per year rate over the next decades [20].

Therefore, this study evaluates the environmental impacts of the railway and the airline transportation through a life cycle analysis, considering  $CO_2$  emissions from both operations and infrastructure construction. A full LCA was conducted to compare modes of transport in terms of their environmental impact.

# MATERIALS AND METHODS

LCA is a method that started to be developed about 50 years ago and is now widely used by many industries. LCA examines the environmental impact of a product or service (or, with widespread use, product system) in a holistic way.



Figure 1. System boundaries.

Unlike traditional methods, the production, transportation, operation, maintenance, and disposal processes of a product system are simultaneously analysed and the results are correlated and then evaluated. This allows a comparison to be made as to which process has a better or worse environmental impact. In addition, thanks to the LCA, a wide variety of previously defined environmental impact categories can be calculated. LCA can also be used to estimate environmental costs [21]. There are many package programs available to make LCA calculations. There are also various methods in these programs. In this study, 9.0.0.49 version of SimaPro package program and 1.09 version of ReCiPe 2008 method were utilised.

In this study, LCA was chosen to make a comparison between product systems and has a holistic perspective on the environmental effects of the product system. The reason for choosing the ReCiPe 2008 method is that it offers up-todate and comprehensive results. The emissions produced by air transport and regular and high-speed rail transport per 1 pkm, which means the transport of 1 passenger for 1 km, are calculated in this study.

In the calculations of the air transport, the production and operation data of 1 aircraft were used. In the production data; the material consumed during the production, energy and water, and the transportation of the materials by land and rail were considered. In the production data of high-speed and regular train; the production material, the amount of electrical energy and light oil used in production, and the disposal processes of the high-speed train were evaluated. Figure 1 presents the system boundaries for the upstream (manufacturing) processes.

Carbon dioxide ( $CO_2$ -biogenic and fossil), carbon monoxide (CO-biogenic and fossil), dinitrogen monoxide ( $N_2O$ ), waste heat, methane ( $CH_4$ -biogenic and fossil), nitrogen

Table 1. Social cost factors (€/kg pollutant)

Pollutant type	Cost factor
CO <sub>2</sub>	0.0566
CO	0.0958
N <sub>2</sub> O	15
$CH_4$	1.75
NO <sub>x</sub>	34.7
NMVOC	2.1
PM <sub>2.5</sub>	79.5
SO <sub>2</sub>	24.9

oxides (NO<sub>x</sub>), non-methane volatile organic compounds (NMVOC), particulate matter (PM<sub>2.5</sub>), sulphur dioxide (SO<sub>2</sub>) and water vapour (H<sub>2</sub>O) were calculated in the LCA. Of these gases, CO<sub>2</sub>, N<sub>2</sub>O, CH<sub>4</sub>, and H<sub>2</sub>O are the main contributors of global warming [22]. There are findings that waste heat and PM<sub>2.5</sub> may contribute to global warming [23–27]. In addition, NO<sub>x</sub>, PM<sub>2.5</sub> and CO have adverse effects on human health and the environment [28]. Unlike the others, CO<sub>2</sub>, CO, and CH<sub>4</sub> emissions were examined under both biogenic and fossil emission headings. Biogenic emissions are defined as part of the carbon cycle. Therefore, the net value of biogenic emissions is zero, i.e. the amount of emissions produced and used by nature through photosynthesis is equal [29].

In addition to the emission amounts, the social (or environmental) cost, which expresses the cost spent to reduce the environmental damage caused by the emissions, is also included in the calculations. In this way, the economic, as well as environmental, burden of the emissions can be calculated. Social cost factors are taken from Zevenhoven and Beyene [23] and presented in Table 1.

# **RESULTS AND DISCUSSIONS**

Results obtained as a result of LCA calculations are presented in Figure 2. While Figure 2 includes only the emissions released during the transport process, Figure 3 presents the values obtained as a result of taking the production process into account.

Emissions and waste heat generated during the transport process are quite low for 1 pkm. Figure 2 shows that air transport has higher fossil  $CO_2$  emissions than rail transport. Air transport generates 126.54 g of fossil  $CO_2$  per 1 pkm, while this value is calculated as 0.11 g for rail transport. N<sub>2</sub>O and fossil CH<sub>4</sub> production per 1 pkm is higher in railway transportation. Similarly, water vapour production is higher for the rail transportation. However, due to the abundance of  $CO_2$  production, air transport is more unfavourable than the rail transport in terms of greenhouse gas (GHGs) production potential and contribution to global warming. When other emissions are analysed, it



Figure 2. Transport emissions of air and rail modes.



Figure 3. Total emissions of air and rail modes.

is observed that there is a balanced distribution. For example, air transport is more dominant in the production of biogenic  $CO_2$ , fossil CO,  $NO_x$ , and NMVOC, while rail transport produces more biogenic CO,  $PM_{2.5}$ , and  $SO_2$ . The values for high-speed and regular train emissions are generally close to each other. Only waste heat is produced much more by high-speed train.

Figure 3 includes the calculations results of production processes to the transportation emissions shown in Figure 2. Air transport is obviously ahead in fossil CO<sub>2</sub> production, when the production processes are also considered. Although rail transport (especially the high-speed train) produces most of other GHGs (N<sub>2</sub>O, CH<sub>4</sub> and H<sub>2</sub>O), it is possible to admit that the negative contribution of air transport to global warming is higher than that of rail transport. A relatively balanced distribution is observed in other gases. While rail transport is ahead of air transport in the production of fossil CO, waste heat, NO, PM<sub>2 5</sub>, and SO<sub>2</sub>, air transport surpassed the rail transport in NMVOC production. Since the data in Figure 3 includes the production values of one plane and train, the pkm unit is no more applicable. Since the production values are very high, the values in Figure 2 have decreased to a negligible level.

Social costs caused from emissions were also calculated by multiplying the social cost factors in Table 1 with the values in Figure 3. Accordingly, the total costs of air transport, high-speed train and regular train were calculated as  $\notin$  355,109,  $\notin$  288,059, and  $\notin$  167,898, respectively. According to these values, air transport performed the worst for the environmental costs.

In one of the studies conducted in 2016, environmental performance comparison of airline and high-speed rail transportation was performed [30]. According to the findings of that study, although the general belief is that the high-speed train operations on the same route are more environmentally friendly than the airline operations, the environmentally friendliness of high-speed train operations generally depends on the source from which the energy is obtained, so a definite conclusion cannot be reached. Results of the present study show that the use of the railroad is clearly more environmentally friendly. On the other hand, if a more comprehensive LCA calculation based on country-based energy production studies is made, changes in results are very likely.

In another study conducted in the same year, a comparison was made between a high-speed train operating in Australia and short-distance air transport [26].  $CO_2$  production per 1 pkm in 2026 was estimated as 104.5 g and 30.3 g for air transport and high-speed rail transport, respectively. These values include the production, maintenance, and operation of the aircraft and the production, maintenance, operation, and disposal processes of the train. According to the results of the study, using highspeed trains instead of airways in short-distance travels in Australia will provide a great reduction in GHG production [26]. In our study,  $CO_2$  values calculated per pkm for air transport and high-speed train are 126.54 g and 0.1 g, respectively. The common point of both studies is that the high-speed train is more environmentally friendly than the air transport.

#### CONCLUSIONS

Transportation is one of the principal needs of modern people. Different modes of transportation have been developed to meet different demands. These modes of transport have superior and weak characteristics in various areas. Especially in recent years, with the foundation of the concept of sustainability into daily life, the environmental performance of different transport modes has been questioned extensively. LCA is one of the most frequently used methods in these comparisons due to its holistic perspective.

In this study, air and rail modes for 1 pkm transportation were compared with LCA method in terms of environmental performance. Air transport seems to be ahead of the rail transport, especially in GHG production. Air transportation produces 126.54 g of fossil CO, per 1 pkm, while the high-speed train produces 0.1 g and the regular train produces 0.12 g of fossil CO<sub>2</sub>. Similar results are obtained when the production process is also considered. When the production of one aircraft and the transportation per 1 pkm are examined together, it is seen that 2035.82 t CO<sub>2</sub> is produced. The same values are 19.31 t and 13.73 t for high-speed and regular train. In addition, it is seen that air transportation produces higher social costs compared to rail transport. On the other hand, while air transportation produces more CO, NO, and NMVOC during transportation phase, rail transportation modes cause more of them in terms of total emissions. Besides, rail transportation modes are generally responsible for more emissions.

According to these results, it is concluded that air transportation produced worse results for GHGs and thus, for climate change. However, rail transportation is not as environmentally friendly as it is thought when compared to air transport. On the other hand, despite the comparison of the modes of transportation made within the scope of the LCA, more detailed data on the production, operation, maintenance, and disposal processes are required in order to reach more precise conclusions and judgments. The question of 'which mode gives the best solution for the environment' is not easy to answer and because the results of the study provides a limited outcome for a certain comment, it is of great importance to support further studies with a more comprehensive database.

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# DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

# **CONFLICT OF INTEREST**

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

# ETHICS

There are no ethical issues with the publication of this manuscript.

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# Extraction of some heavy metal ions from aquatic solution by banana peel-based biosorbents

# Doğu RAMAZANOĞLU<sup>\*1</sup><sup>©</sup>, Zaman Adnan MOHAMMED<sup>1</sup><sup>©</sup>, Khalid Ali MAHER<sup>2</sup><sup>©</sup>

<sup>1</sup>Zakho University, Scientific Research Center, Zakho, Iraq <sup>2</sup>Department of Chemistry, Zakho University, Zakho, Iraq

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#### ABSTRACT

In this work, economic, natural, and eco-friendly biosorbent want to be advised instead of traditional methods. For that reason, banana peel-based biosorbents were been done to remove  $Cu^{2+}$ ,  $Ni^{2+}$ , and  $Co^{2+}$  ions from the solution by batch method. Moreover, how their natural assets like density, water intake, solubility, and heavy metal extraction performance were affected by their starch and oil contents had been determined. Flame atomic absorption spectroscopy (FAAS) having been using for the analysis of biosorbents uprooting performance. The density of the biosorbents had worked according to ASTM D 792 standards. As a result, the untreated banana peel-based biosorbents had transported 4.87 mg of  $Co^{2+}$  ions, 4.73 mg of  $Ni^{2+}$  ions, and 4.29 mg of  $Cu^{2+}$  ions from 25 ml 7 ppm of each heavy metal ions solution with 175 rpm agitation speed at 24°C during half-hour.

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# INTRODUCTION

Processing of heavy metals had been beginning since antics time without awareness of their toxicity. The usage amount of heavy metal-containing coals that causes air pollution also develop industrialization. Water pollution is the thinnest point for damage to the food chain for that reason, the studies about extraction of organic and inorganic contaminants from wastewater are so valuable [1–8]. Numerous organic and inorganic studies have been done and suggested for wastewater treatment methods like as, electrochemical treatment, ion exchange, reverse osmosis, and chemical precipitate. These traditional methods have low performance, are expensive, and are unsuitable for large-scale processing [9–12]. Hence, an alternative method has been desiring.

Nowadays, the biosorbents have been shined like an alternative method instead of these traditional ones. Biosorption is a natural potential of biomaterials to the extraction of heavy metal ions from water metabolically or physiochemically [13]. Non-living biomass like shrimp, bark, crab shell, etc.; were used as traditional biosorbent [14]. In this study, banana peels have operated as an alternative biosorbent because of their abundance and low cost [15, 16]. As a biosorbent, they have the potential for reducing toxic metal ions like Cu<sup>2+</sup>, Ni<sup>2+</sup>, and Co<sup>2+</sup> from urban environ-

\*Corresponding author.

\*E-mail address: doguramazanoglu@hotmail.com



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mental problems as well as metal-contaminated industrial effluent treatment expenses.

Banana is one of the most enjoyed fruits around the world. Although, its peels do not have specific usage. For that reason, the banana peels were tested as adsorbents of metal ions. This biomass has polymeric groups such as proteins, pectin lignin, cellulose, and hemicellulose which take a role in the adsorption of metal ions [17].

Besides these, it has other chemical groups like carboxylic acid, phosphate, and hydroxyl group attach to metals ions [18]. Figure 1 shows the chemical groups of banana peels; (a) carboxyl, (b) hydroxyl, (c) phosphate, and metal ions. In this study, 6 different biosorbents were supplied from banana peels to extract  $Cu^{2+}$ ,  $Ni^{2+}$ , and  $Co^{2+}$  ions from the aquatic. Lastly, how their content affects their extraction performance during 30 minutes of contact time at room temperature investigated.

# MATERIALS AND METHODS

#### Materials

Corn and wheat starch, apple cider vinegar (4–5% acetic acid), sunflower oil, Banana has bought from a local grocery store in Zaho. Cobalt (II) Acetate  $C_4H_6CoO_4$ , Nickel (II) Nitrate Hexahydrate Ni(NO<sub>3</sub>)<sub>2</sub>.6H<sub>2</sub>O, and Copper (II) chloride dihydrate CuCl<sub>2</sub>.2H<sub>2</sub>O has provided from (Merck, Germany).

#### Methods

Am Scope brand microscope has trained for taking surface images at the biology lab at Zakho university. Atomic absorption spectroscopy (FAAS) that (PerkinElmer, Turkey) brand has applied for a measured left of the concentration of solution after treatment with biosorbents in the scientific research center. Banana (Musa) peels having dried for two days at  $103\pm2^{\circ}$ C. Dried peels had been granulated by a coffee pulverizing machine.

#### **Preparation of Biosorbents**

The component of the banana peel-based biosorbents cornstarch, banana, sunflower oil having weighed and placed in a 500 ml beaker as given in Table 1. 25 ml of pure water summed to the beaker. Later, 3 ml of acetic acid was added and stirred to break the long-chain molecules of the starch. After, 1–2 ml of sunflower oil was combined as a plasticizer to re-crystallize to depolymerized polymer chains. It having mixed and fired at 75°C until gelation proceeds. Then, they left in the oven for 45 minutes at  $103\pm2^{\circ}$ C as shown in Figure 2.

#### Solubility Test (%)

The solubility test had found in 1992 by Gontard et al. [17]. Samples have taken in the oven for 24 hours at  $103\pm2^{\circ}$ C.



**Figure 1**. Interactions between the metal ions (M<sup>2+</sup>) and the chemical groups present in the banana peel; (a) carboxyl, (b) hydroxyl, and (c) phosphate groups [10].

B1	B2	B3	<b>B4</b>	B5	B6
1g.B/	1.5g.B/	2g.B/	1g.B/	1.5g.B/	2g.B/
2g.S	1.5g.S	1g.S	2g.S	1.5g.S	1g.S
0.5 ml	0.5 ml	0.5 ml	1 ml	1 ml	1 ml
Oil	Oil	Oil	Oil	Oil	Oil

Then, they were weighed (Wi) and washing with 50 ml of purified water at 175 rpm for a day. After the rinsing step, they have balanced again. Finally, the solubility of the biosorbents had determined according to the equation given in (1).

$$S=(Wi-Wf)/Wix100$$
 (1)

Wi: Initial mass; Wf: Final mass.

#### Water Intake (%)

The water intake (%) values of biosorbents kept in pure water for 24 hours have determined according to the formula is given in (2).

$$SW = [(Mw-Md)/Md]x100$$
<sup>(2)</sup>

Md=Sample initial weight (g); Mw=The weight of the sample after immersion in water (g); SW=water uptake rate (%).

# Density Test (g/cm<sup>3</sup>)

The Air-dry weights of the samples were found and recorded. Then, they were immersed in water and weighed. The density of those samples had calculated concerning the equation in (3).

Density 
$$(g/cm^3) = [Ma/Mw]$$
 (3)

Here, Ma=weight of the sample in the air (g). Mw=The weight of the sample in water (g) is given.

#### **Batch Experiment**

CuCI<sub>2</sub>.2H<sub>2</sub>O, Ni(NO<sub>3</sub>)<sub>2</sub>.6H<sub>2</sub>O, and C<sub>4</sub>H<sub>6</sub>CoO<sub>4</sub> were used for preparing 25 ml 7 ppm of Cu<sup>2+</sup>, Co<sup>2+</sup>, and Ni<sup>+2</sup> heavy metal ions solutions. Then, 25 mg of biosorbent was separated for the adsorption of each solution. Adsorption circumstances have adjusted to 175 rpm of stir speed, at 24°C for 30



Figure 2. Preparation of biosorbents.



**Figure 3.** Banana peels-based biosorbents. Banana peels, starch, and oil content ratio of biosorbent (1/2/0.5) for B1, (1.5/1.5/0.5) for B2, (2/1/0.5) for B3, (1/2/1) for B4, (1.5/1.5/1) for B5, (2/1/1) for B6 respectively.

minutes. The performances have been obtained from the equation in (4).

$$Q = [(Co-Ce).V]/m$$
(4)

Where; Co (mg/L) is an initial concentration, and Ce (mg/L) is the final concentration. V is the volume in a liter (L) unit, and m is biosorption in gram (g) units.

# Morphology of Biosorbents

The surface photographs of biosorbents having taken by microscope in Figure 3. The biosorbents have different ingredients that make appearing not the same.

# **RESULTS AND DISCUSSION**

# Water Intake Analysis of Biosorbents

Banana peels-based biosorbents solubilities are calculated according to the equation in (1). Also, water intake values are shown in Figure 4.

The bar diagram trades with the water intake capacity of banana peel-based biosorbents. When, water intake was being decreased from B1 to B3 with an increasing quantity of banana peels contents [18, 19]. While the starch amount was raising biosorbents, the water adsorption capacity was diminishing. The chart has separated into



**Figure 4**. Water-intake of banana peels-based biosorbents. Banana peels, starch, and oil content ratio of biosorbent (1/2/0.5) for B1, (1.5/1.5/0.5) for B2, (2/1/0.5) for B3, (1/2/1) for B4, (1.5/1.5/1) for B5, (2/1/1) for B6 respectively.



**Figure 5**. Solubility of banana peel-based biosorbents. Banana peels, starch, and oil content ratio of biosorbent (1/2/0.5) for B1, (1.5/1.5/0.5) for B2, (2/1/0.5) for B3, (1/2/1) for B4, (1.5/1.5/1) for B5, (2/1/1) for B6 respectively.

two groups referring to their oil contents. The first group of biosorbents had B1, B2, and B3. The second group had B4, B5, and B6.

#### Water Solubility Analysis of Biosorbents

The solubility of banana peel-based biosorbent was measured and given in Figure 5. The increasing amount of starch in biosorbent causes more solubility. However, the increased oil contents for B4, B5, and B6 show us how to decrease the solubility percentage. Hence, the B4 was the least water-soluble banana peel-based biosorbent with a 13.0% value and the B3 had the highest solubility value with 90.0%. There is just one difference between B1, B2, and B4, B5 that double oil content which decreased to solubility nearly 50%. On the other hand, B3 and B6 nearly lose the solubility ratio 10 percent. It showed how starch content negatively affects the solubility of biosorbents.

#### **Density of Biosorbents**

The bar chart has given in Figure 6 deals with the density of banana peel-based biosorbents. While the density of biosorbents was diminishing, starch content was rising. The elevated sunflower oil causes a slight uptick in mass [18–21]. The frequency of biosorbent had been directly equivalent to banana barks content.



**Figure 6**. Density of banana peel-based biosorbents. Banana peels, starch, and oil content ratio of biosorbent (1/2/0.5) for B1, (1.5/1.5/0.5) for B2, (2/1/0.5) for B3, (1/2/1) for B4, (1.5/1.5/1) for B5, (2/1/1) for B6 respectively.



**Figure 7**. Heavy metal ions adsorption capacity of banana peels-based biosorbents; Banana peels, starch, and oil content ratio of biosorbent (1/2/0.5) for B1, (1.5/1.5/0.5) for B2, (2/1/0.5) for B3, (1/2/1) for B4, (1.5/1.5/1) for B5, (2/1/1) for B6 respectively.



Figure 8. Interaction between water and oil molecules [22].

#### **Adsorption Studies of Biosorbents**

The adsorption studies of banana peel-based biosorbents having evaluated at the same condition for all heavy metal ions. The adsorption amount of Co<sup>2+</sup>, Ni<sup>2+</sup>, and Cu<sup>2+</sup> heavy metal ions have given in Figure 7.

It has been obtained from the chart that untreated biosorbent has more adsorption capacity. Also, an expanded volume of biomass content boosted the sorption capacity. Generally, biosorbents had double the amount of sunflower oil ingredients that have less adsorption capacity than others. That might be explained better by the interaction between water and oil molecules in Figure 8.

Oil has non-polar molecules. As we know, the electronegativity of non-polar molecules atoms is equal. But water has polar molecules that mean their electronegativity is not stable. Because of these polarity distinctions, might be cause for the adsorption capacity diminished. The other reason why untreated biosorbent performed better adsorption than other bio-composites is because of the chemical group's abundancy like carboxyl, hydroxyl, and phosphate groups present on banana peels which includes less from others.

The untreated biosorbent; B0 was the highest biosorption among them with the values of 4.87 mg/g, 4.73 mg/g and 4.51 mg/g for  $\text{Co}^{2+}$ ,  $\text{Ni}^{2+}$ , and  $\text{Cu}^{2+}$  heavy metals ions respectively during half hour, at 24°C and 175 rpm agitation speed.

# CONCLUSION

- 1. In the water intake analysis, the water intake amount decreases with increasing oil content. When the amount of banana peels content has increased, intake of water has diminished. Moreover, the surged starch content was inversely proportional to the water intake of banana peel-based biosorbent.
- 2. In the water solubility study, the raised sunflower oil content having decreased the solubility of all biosorbents. The B3 was observed as the highest soluble one with value of 90% among its groups. The B4 was recognized as the least soluble with value of 13.0%. The increasing amount of biomass enhances the solubility of biosorbents. Moreover, expanded biomass inversely proportions with a starch content have not been saved in a bioform. That is why the diminished starch amount grows the solubility.
- 3. In the density analysis, the booming oil content improves density for all biosorbents. Biosorbent that has the highest densities in its group was B6 determined as 1.03 g/cm<sup>3</sup>. The biosorbent B1 was found as lowest one with the value of 0.99 g/cm<sup>3</sup>.
- 4. In the adsorption step, untreated Banana peels has the highest Co<sup>2+</sup>, Ni<sup>2+</sup> and Cu<sup>2+</sup> ions adsorption among others. Moreover, the increased sunflower oil reduced the heavy metal ion adsorption. Additionally, decreasing the amount of starch content also lessened adsorption capacity. Oppositely, the raised amount of biomass improves the adsorption performance. The density of all biosorbents was directly proportional to adsorption capacities.

According to results, banana peels can be used as biosorbent directly instead of other expensive water treatment methods. Further studies can be about to determination of the optimum condition of biosorbents and their carbonized form.

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# DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

# **CONFLICT OF INTEREST**

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

# ETHICS

There are no ethical issues with the publication of this manuscript.

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# Developing a GMDH-type neural network model for spatial prediction of NO<sub>x</sub>: A case study of Çerkezköy, Tekirdağ

Can Burak ÖZKAL<sup>\*1</sup><sup>(</sup>), Özkan ARSLAN<sup>2</sup>

<sup>1</sup>Department of Environmental Engineering, Tekirdağ Namık Kemal University, Tekirdağ, Türkiye <sup>2</sup>Department of Electronics and Communication Engineering, Tekirdağ Namık Kemal University, Tekirdağ, Türkiye

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#### ABSTRACT

Air pollution-induced issues involve public health, environmental, agricultural and socio-economic aspects. Therefore, decision-makers need low-cost, efficient tools with high spatiotemporal representation for monitoring air pollutants around urban areas and sensitive regions. Air pollution forecasting models with different time steps and forecast lengths are used as an alternative and support to traditional air quality monitoring stations (AQMS). In recent decades, given their eligibility to reconcile the relationship between parameters of complex systems, artificial neural networks have acquired the utmost importance in the field of air pollution forecasting. In this study, different machine learning regression methods are used to establish a mathematical relationship between air pollutants and meteorological factors from four AQMS (A-D) located between Çerkezköy and Süleymanpaşa, Tekirdağ. The model input variables included air pollutants and meteorological parameters. All developed models were used with the intent to provide instantaneous prediction of the air pollutant parameter NOx within the AQMS and across different stations. In the GMDH (group method of data handling)-type neural network method (namely the self-organizing deep learning approach), a five hidden layer structure consisting of a maximum of five neurons was preferred and, choice of layers and neurons were made in a way to minimize the error. In all models developed, the data were divided into a training (80%) and a testing set (20%). Based on R<sup>2</sup>, RMSE, and MAE values of all developed models, GMDH provided superior results regarding the NOx prediction within AQMS (reaching 0.94, 10.95, and 6.65, respectively for station A) and between different AQMS. The GMDH model yielded NOx prediction of station B by using station A input variables (without using NOx data as model input) with R<sup>2</sup>, RMSE and MAE values 0.80, 10.88, 7.31 respectively. The GMDH model is found suitable for being employed to fill in the gaps of air pollution records within and across-AQMS.

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\*Corresponding author.

\*E-mail address: cbozkal@nku.edu.tr



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# INTRODUCTION

Air pollutants can be categorized as primary (SO<sub>2</sub>, NO<sub>2</sub>, CO, hydrocarbons, particle matter) and secondary (ozone) pollutants. NOx can be represented by the most commonly found NO<sub>2</sub> form generated as a result of reaction between combusted hydrocarbons due to industrial and traffic emissions and atmospheric oxygen. NO, are responsible for the formation of tropospheric ozone (bad ozone). Nitrogen oxides likewise SO<sub>2</sub>, ammonia and volatile organic carbons are responsible for the formation of PM<sub>2.5</sub>. and have been subject a to numerous research on the basis of monitoring, prediction [1, 2] and mitigation methodologies development [3]. The spread and transport of air pollutants that are released into the atmosphere is influenced by weather and climatic factors [4, 5]. Interaction of meteorological factors, mainly wind speed and wind direction with the air pollutants bring along variations in spatial distribution, spatiotemporal variation of air pollutants. Numerous research have demonstrated regional transport of air pollutants interpretation with the main meteorological factors; Humidity, temperature, pressure, wind direction and wind speed, dew point [6].

Recently, a city air quality determination study has gained importance. Thus, the periodic and spatial dimensions of pollution necessitate accurate determination. In order to overcome observed deficiencies, higher maintenance and repair expenses in air quality monitoring stations (AQMS) and provide supplementary data, solutions including increasing prevalence of AQMS, use of mobile AQMS and low cost sensors (LCS) are widely applied [7, 8]. Up to date, research have emphasized the modeling approaches for temporal and spatial prediction [9] and forecasting of air pollution [10].

Monitoring of air quality and air pollution parameters is performed by static, mobile AQMS and establishment of low cost sensors (LCS). Regulatory pollutants (carbon monoxide, nitrogen oxides, ozone and particulate matter are measured by certified reference instruments at static AQMS. Those stations and sensors are large and expensive, also necessitate strict calibration and maintenance routines in order to provide high quality data and comparability between different region and stations [7, 8, 11]. Sensors and low cost sensors are able to monitor a range of air pollutants but mostly they are unable to meet the Air Quality Directive - Data quality objectives criteria and under effect of chemical interferences and environmental conditions [7, 12]. According to framework and legal requirements described in air quality directive 2008/50/EC for ambient air quality assessment and management, the reference measurement methods are applied in the stable AQMS in Europe. The data provided by LCS are usually less accurate than AQMS [11]. The Air Quality Directive also pave the way for alternative and supplementary techniques such as air quality models for air quality and air pollution management.

Recent research accordingly have hypothesized to estimate air pollutants concentrations through their association/ interpretation with meteorological parameters [13], landscape data, environmental information [14] and other measured air pollutants [15, 16]. Underlie the spatiotemporal correlations between air pollutants emission and diffusion mechanisms, mechanistic or a non-linear model must be considering those correlation and/or able to realize non-linear mechanism of air pollutants spread, diffusion, transport and interaction under environmental, meteorological and atmospheric conditions [16].

Air quality management includes monitoring and timely application of foreseen preventions related to extreme air quality scenarios. Therefore, short and mid-term forecasting of air pollution and air quality indexes became the focus of numerous research [2]. There are many gaps encountered in air pollution data series of stable AQMS, during their observation period [17]. The research method for predictive determination of air pollution parameters and air quality mainly based on statistical or deterministic approaches [16]. The emission and diffusion of pollutant is related both with interaction of pollutants and the meteorological factors. As a result it is prerequisite to use sufficient number of meteorological and air quality parameters in an air pollution prediction model. Similar approach was applied via Pearson correlation, support vector regression (SVR) with or without principal component analysis, for the purpose of decision making on keeping the most correlated pollutant parameters in the data set [9, 16]. According to recent literature findings, modelling approaches including artificial neural network (ANN) methods were found eligible to be employed to fill in the gaps of air pollution records by deep learning based prediction of PM<sub>25</sub>, LSTM based estimation of air pollutant concentrations that cannot be directly measured by the air quality monitor and ANN based forecasting of the spatial-temporal profile of pollutant concentrations and air quality determinants in specific cases of power outages, negative and wrong records of pollutants [11, 13, 16].

Most generally, machine learning (ML) methods and mostly ANN application are proposed to forecast the spatial-temporal pollutant concentration profiles during extreme scenarios be it; power outage, maintenance, sensor repairmen and replacement, negative and faulty pollutant records. Also recent research have reported successful application of various modelling techniques developed with intent to eliminate high number of monitoring stations requirement and become an alternative for advanced representation of air pollution spatial variability; where Feed forward neural networks and Long short-term memory deep learning techniques were applied with intent to provide data at currently unmonitored locations [11, 18].

Traditional linear models or deterministic models descriptive of chemical dispersion and transportation remain limited as a result of the high degree of non-linearity between different air pollutants and weather conditions [2]. As it was reported in recent literature, the nonlinear mechanism of atmospheric phenomenon can be realized by ANN [17, 19] and excellent prediction performance can be achieved [16]. For the very reason air pollutant parameters prediction using ANN is found superior to multi-linear regression [17, 20].

Conventional recurrent neural networks (RNN) and long short-term memory (LSTM) is suitable option to be applied on time series thereby applied in various research topics from various disciplines [11]. These methods mostly conform with the cases where the values of concern are related to their previous situation like traffic flow prediction, air pollution prediction, solar irradiation scenarios [11]. For the case of interpolation and extrapolation based estimation, model inputs are selected from a group of monitoring stations. Model training is performed by historical data of limited number of station with LSTM method, therefore the model was proposed to be suitable in small cities where only a few monitoring stations are established [11]. Besides the reported high performance of LSTM application in research related to air pollution, concentration prediction is the common field of research area [16, 21].

Recently, soft computation artificial intelligence (AI) techniques have been used successfully in the prediction of air pollution parameters such as NOx. Because of the many parameters that affect NOx, the results predicted from empirical models do not match well with the measured results. Therefore, it is necessary to develop models that provide more accurate prediction of NOx under various air pollution and meteorological conditions. Group Method of Data Handling (GMDH), an AI-based method, is a self-organizing technique that can be used to solve complex problems in nonlinear system with large degrees of complexity. The GMDH technique, which is a multi-layered structure, uses only neurons that can provide the most effective and accurate results unlike traditional machine learning methods. Thus, it is ensured that the most efficient input variables are used instead of using all input variables for the predictive model output. In this regard, the GMDH approach requires less data training compared to classical ANN methods and facilitates the interpretation of model input and output parameters. In air pollution prediction applications, it is very important to determine the pollutant (emission) sources and to determine the prevailing wind direction [19]. Demonstrated through preliminary research on AQMS of Tekirdağ and literature findings, deficiencies are observed in the AQMS data, with measurements either not being performed or not being shared on numerous occasions during the course of the year [13, 18, 22]. In order to overcome these shortcomings, GMDH-type neural networks using a non-linear structure can be preferred for the estimation of the NOx parameter.

In this study, we hypothesize the employment of GM-DH-type neural network as an alternative technique in prediction (forecast length equal to 0) of  $NO_x$  air pollutant concentrations within the AQMS and spatial prediction of air pollutant concentration between different stations. The developed model provided prediction of NOx within a certain station by using data from a total number of four different weather stations and taking one station as reference. It has been observed that the GMDH-type neural network model minimizes the error rates under certain NOx prediction states within a station and between different stations.

# MATERIALS AND METHODS

#### Location of Interest

Tekirdağ is located at European part of Turkey, in the Thrace region surrounded by Marmara Sea, Greece and Bulgaria. The city is located within the Ergene basin and is the center of population growth (predominated by the industrial development in the eastern part of the city, Çerkezköy and Çorlu districts) and air pollution as a result of heating, traffic and industrial activity. Tekirdağ constitutes the western border of İstanbul and the northern border of the Marmara Sea. Süleymanpaşa is the biggest district of Tekirdağ. There many organized industrial zones (OIZ) located at Çorlu, Çerkezköy, Kapaklı, Velimeşe, Ergene districts, as indicated in Figure 1. Tekirdağ host over 1100 factories (with the frequency of occurrence; textile, paper, packaging, chemical and metal industries respectively). In Tekirdağ, there a total of 14 organized industrial zones, while 5 of them established around Corlu district and its immediate surroundings. 4 of the organized industrial zones (OIZ) are lined up in the west-east direction along Corlu, the Velimeşe OIZ is located between Cerkezköy and Corlu (North south direction) and more than 500 facilities operate. Çorlu is the area where the new settlement is located and the traffic is concentrated while Çerkezkoy host an OIZ under which more than 270 facilities operate and it one of Turkey's largest OIZs [23, 24]. The topographical properties of the location of interest (as indicated in Figure 1) can be described as land appearance in the form of wavy plains and is uneven, with low to mid slope values. Corlu is under the influence of a transition type climate where Black Sea, Mediterranean and continental climate characteristics are encountered together. Cold air masses descending from the north and humid-warm air currents coming from the south, the Mediterranean and the Aegean affect the climate structure of the region. Typically, the wind blows at Tekirdağ dominantly from the directions of NNE-NE and rarely from the directions of SW-SSW [24].

#### **Air Quality Monitoring Stations**

In this study, urban and industrial AQMS located on the Çerkezköy-Çorlu-Tekirdağ line were selected as



Figure 1. AQMS locations for stations; A: Çerkezköy, B: Çorlu, C: Çorlu Center, D: Tekirdağ center.

data source. Air quality of Çorlu is monitored through 2 AQMS; an urban AQMS located in town center (as a result of its distance from industrial activity) and an industrial AQMS located between town center and Çerkezköy (Fig. 1). For the case of Tekirdağ (Suleymanpasa center city) air quality is monitored at two different stations. The exact location, coordinates and bird flight (BF) distance between selected stations in Tekirdağ-Suleymanpasa, Çorlu and Çerkezköy are described in Figure 1 and AQMS distances between stations are given in Table 1. Between studied dates, northern winds were prevalent around station A. For station B and C, the direction of wind could be described as spreaded over a wider range (as a result of E-SE winds prevalence), as it is demonstrated in Figure 2. Station B has distinctly strong winds where the average is above the meteorological upper limits reported for mild winds. Those strong winds are reported to trigger air pollutants transport and dispersal mechanisms [16]. Air pollutants at regions where mountain-valley and land-sea breezes cycles are dominant

BF Distances between AQMS of interest			
AQMS # (stations separated by-)	Distance in km		
Ed A-B	≈21		
A-C	≈23		
A-D	≈54		
B-C	≈3		
B-D	≈34		
C-D	≈31		

Letter Code/Name of AQMS. A: Çerkezköy; B: Çorlu; C: Çorlu Center; D: Tekirdağ center.

wind systems, are not easily transported from emission sources and accumulate. Also, WS over 5 m/s may create unstable weather conditions which increase PM and SO<sub>2</sub>, NO<sub>x</sub> distribution [4, 17].

#### **Data Acquisition**

The air pollutant and meteorological data source is the official Air Quality Monitoring Network website of the Ministry of Environment and Urbanization. Daily average values were used and taken as references all along the study.  $PM_{10}$ ,  $SO_2$ ,  $NO_x$  data as air pollutant parameters and air temperature (T), wind direction (WD), wind speed (WS), relative humidity (RH) and Air pressure (AP) were used as meteorological model variables for the period between December 2017 – December 2018.

It can be figured out from the correlation coefficients (depicted in Figure 4) of 0.79, 0.26, 0.43 and 0.91 for stations A, B, C, and D between targets output NOx and input variable PM, the PM parameter will be effective in NOx prediction for all stations. The PM<sub>10</sub> and SO<sub>2</sub> are amongst the sole parameters continuously measured at each AQMS in common. Therefore, the choice of using PM10 and SO<sub>2</sub> as model inputs for NO<sub>x</sub> prediction could be reasoned based on theory; the relation between air pollutant parameters as a result of their complex and interacting formation mechanisms. Nitrogen oxides likewise SO<sub>2</sub>, ammonia and volatile organic carbons are responsible for the formation of particulate matter.

In the Nox prediction model, 174-day air pollution and meteorological data were used, which were recorded for 1 year. The statistical distributions of these data are shown in Figure 2 and can be summarized as;  $PM_{10}$ ,  $SO_2$  and  $NO_x$  data of station D shows a wider distribution compared to station A-

C. The distribution of annual temperature data did not differentiate between stations. Higher average wind-speed values were measured at Station C. A substantial difference of wind direction distribution was not observed at the stations. Average RH were reported to be the highest and lowest for station A and B respectively. Average AP values were distinctively higher at station D.

#### **Regression Methods**

Various machine learning regression methods are used to establish a mathematical relationship between the inputs which are air pollutants ( $PM_{10}$ ,  $NO_x$  and  $SO_2$ ) and meteorological parameters (T, WD, WS, RH and AP) and the target output. The predicted output is calculated by training the data in all models for different air quality measurement stations. In this study, various regression methods that provide a relationship between target  $NO_x$  and inputs are given below.

#### **Linear Regression**

Linear regression is one of the simplest methods that provide a mathematical relationship between the input parameters and the target output. It is often preferred because of the simple and convenient mathematical structure. In this regression method, the mathematical equation of the target based on the inputs is obtained with a slope and intercept value. The relationship between the target NO<sub>x</sub> and the input variables is expressed by linear regression as:

$$Y = a_0 + a_1 x_1 + a_2 x_2 + \dots + a_n x_n \tag{1}$$

where *Y* is output,  $x_1, x_2, ..., x_n$  are input variables and  $a_1, a_2, ..., a_n$  are coefficients obtained from the model.

## **Random Forest Regression**

Random forest (RF) regression is one of the machine learning models that can be effective in predictive analysis under conditions where the output and input parameters are in a non-linear relationship. In this method, which reduces over fitting in model training, the predictions of all decision trees are combined to obtain more accurate and stable results. The forest tree diversity increases the robustness of the model obtained by regression [25].

#### Multilayer Perceptron Regression

Multilayer perceptron (MLP) is an artificial neural network method frequently used in regression. MLP method, which is also considered as the early stage of deep learning, consists of an input, multiple hidden and an output layer. In MLP neural networks, the first layer contains the input parameters and the output layer makes a prediction about the input. Hidden layers are used as a computational tool between the input and output layers [26].

MLP regression method is often used in supervised learning applications that involve the training phase for certain target and input parameters are used in all hidden layer neurons. In this method, which is based on the training of inputs and output, a model based on correlation between input and output is learned. In the training phase, the parameters and weight coefficients of the model that minimize the error are obtained.



**Figure 2**. Boxplot data summary of  $PM_{10}$ , SO<sub>2</sub>, NO<sub>x</sub>, T, WD, WS, RH and AP parameters for stations A-WD: Wind direction (degrees), WS: Wind speed (m/s), RH: relative humidity (%), AP: Atmospheric pressure (hPa).



Figure 3. General structure of GMDH-type neural network.

#### GMDH-type Neural Network Regression

Group Method of Data Handling (GMDH), which is a self-organizing network model that behaves according to the input data, is preferred in regression analysis applications, unlike MLP artificial neural network models where the inputs are used on all neurons in the hidden layer [27]. The use of input parameters in all neurons may cause over fitting and performance degradation in regression models. In addition, there are difficulties and shortcomings in adjusting the bias and weight coefficients for a small number of datasets.

GMDH-type neural networks are one of the best methods for model estimation in complex structured problems. This neural network model is a multi-layered structure and uses only neurons that can provide the most effective and accurate results. Each layer consists of independent neurons and these neurons are used in pairs. In this network model, a quadratic polynomial function is used as the activation function. The neurons in the hidden layers work independently and neuron outputs which minimize the error rate are used. Thus, a multilayer neural network model consisting of optimal layers and neurons is designed instead of using all neurons in the layers [28]. Figure 3 shows the general structure of GMDH-type neural network.

GMDH-type neural networks are defined as a relationship between input and output parameters expressed in the form of a stepwise complex Kolmogorov-Gabor polynomial function. This relationship is expressed as a nonlinear form of the Kolmogorov-Gabor function [29].

$$\bar{y} = \alpha_0 + \sum_{i=1}^n \alpha_i x_i + \sum_{i=1}^n \sum_{j=1}^n \alpha_{ij} x_i x_j + \sum_{i=1}^n \sum_{j=1}^n \sum_{k=1}^n \alpha_{ijk} x_i x_j x_k + \cdots$$
(2)

where  $\overline{y}$  and  $\alpha$  are predicted output and the coefficients of the quadratic polynomial, respectively. The number *n* is the degree of polynomial function and *i*, *j*, *k*  $\in$  (1, 2, ..., *n*). In this study, the number *n* was chosen as 2. The polynomial operation is performed in three steps for *i*={0, 1 and 2}.

The Kolmogorov-Gabor polynomial, which has a nonlinear structure, is expressed in the form of a quadratic polynomial consisting of two variables as follows:

$$\bar{y} = G(x_i, x_j) = \alpha_0 + \alpha_1 x_i + \alpha_2 x_j + \alpha_3 x_i x_j + \alpha_4 x_i^2 + \alpha_5 x_j^2 \quad (3)$$

The GMDH-type neural network estimates the output for each set of input parameters ( $x_i$  and  $x_i$ ) and is used to estimate the  $\alpha_i$  coefficients that minimize the mean squared error between the predicted and the actual output. This process is called self-organization of models, and neurons with minimum error calculated by the least squares method are selected.

In the GMDG-type neural network model, the coefficient vector of the quadratic polynomial is calculated and the neurons that increase the error are eliminated. The objective function (OF), which is a selection criterion, is used for elimination process and OF is expressed as:

$$0F = \frac{1}{n} \sum_{i=1}^{n} (y_{pre} - y_{mea})^2$$
(4)

where  $y_{pre}$ ,  $y_{mea}$  and n are the predicted, measured values and total number of dataset, respectively.

#### **Evaluation of Models**

The performance of a regression model is evaluated by calculating the error rate of the predicted output obtained by the model. In addition, the fit of the regression line to the data set is also used as a criterion in model evaluation. The correlation coefficient ( $R^2$ ), root-mean square error (RMSE) and mean absolute error (MAE) are used to calculate between the predicted and actual values. The  $R^2$  value is between 0 and 1, and a larger value indicates a better fit between the predicted and the actual values. The  $R^2$  is a good measure to determine how well the model fits the dependent variables and is expressed as:

$$R^{2} = 1 - \frac{\sum_{i=1}^{n} (y_{mea} - y_{pre})^{2}}{\sum_{i=1}^{n} (y_{mea} - y_{m})^{2}}$$
(5)

The RMSE is calculated as the sum of the square of the error by subtracting the predicted from the actual value, then divided by the total number of data and the square root is taken. The RMSE, which is widely used in the evaluation of models, is expressed as follows:

$$RMSE = \sqrt{\frac{\sum_{i=1}^{n} (y_{mea} - y_{pre})^2}{n}}$$
(6)

The mean absolute error (MAE), a measure similar to the mean squared error (MSE), is defined as the sum of the absolute value of the error and is expressed mathematically as follows:

$$MAE = \frac{1}{n} \sum_{i=1}^{n} \left| \mathbf{y}_{\text{mea}} - \mathbf{y}_{\text{pre}} \right| \tag{7}$$

where  $y_{mea}$ ,  $y_{pre}$ ,  $y_m$  and n represent the measured, predicted, the average of measured values and the total number of dataset, respectively.

#### Model Development Setup

Many empirical models have been developed to estimate the air pollution parameter  $NO_x$  values using air pollution and meteorological parameters. Recent research have focused on instant prediction of target pollutant parameter value [13]. Machine learning regression methods are extensively used to derive empirical equations. However, the application of ANN models in larger spatial dimensions would bring along a significant decline in the model's performance for places far away from the station (data of which was used for model training), as was hypothesized in recent studies [9, 11, 30]. This study aims to develop empirical models based on not only pollution conditions, but also depend on meteorological conditions for NOx prediction. In this study, different regression methods such as linear, RF, MLP, and GMDH-type NN were analyzed for the NOx prediction model. The influences of different parameters, including air pollutants (PM<sub>10</sub>, SO<sub>2</sub>) and meteorological parameters (T, WD, WS, RH, and AP), on the prediction of NOx within station and across different stations were investigated. In the experimental setup, the parameters of the regression methods were set as follows: the number of forest trees was taken as 500 for RF, five hidden layer structures consisting of 20 neurons was established for MLP and 0.01 learning rate, rectified linear unit activation function, Adam optimization were chosen. In the GMDH-type neural network, a five hidden layer structure consisting of a maximum of five neurons was preferred and, layers and neurons that minimized the error were used. In all developed models, the data were split into a training (80%) and a testing set (20%). R<sup>2</sup>, RMSE, and MAE values were calculated to obtain the most effective and accurate empirical model that can be used in the prediction of  $NO_x$ .

# **RESULTS AND DISCUSSIONS**

In this study, pollution and meteorological parameters were used as input variables in different regression methods for the NO<sub>x</sub> prediction model and it was aimed to obtain a model that provides the best prediction. Figure 3 shows the correlation between the parameters used as input variables and the output NO<sub>x</sub>, and the correlation values between these parameters. As can be seen from Figure 4, it has been observed that target output NO<sub>x</sub> has a high correlation with PM and SO<sub>2</sub> parameters at all stations. It can be said that the obtained correlation results are compatible with the literature. A similar hypothesis was proposed in recent research in the Marmara region; were correlations between SO<sub>2</sub> and PM<sub>10</sub> values reported for residential areas (with solid fuel use) were higher than it is determined for industrial areas [31].

There are correlation coefficients of 0.79, 0.26, 0.43 and 0.91 for stations A, B, C, and D between targets output  $NO_x$  and input variable PM, respectively. These results show that the PM parameter will be effective in prediction the output  $NO_x$  for stations A and D. Similarly, the correlation between output NOx and  $SO_2$  is 0.83, 0.73, 0.11 and 0.89 for stations A, B, C, and D, respectively. It is concluded that the  $SO_2$  parameter will have a negative effect on the prediction of the  $NO_x$  parameter for station C. In addition, it can be said that there is no high correlation between meteorological parameters and  $NO_x$  for all stations. However, the use of meteorological parameters in the prediction of target output  $NO_x$  is important to obtain a more robust model.



Figure 4. Pearson's correlation coefficients of air pollutant and meteorological parameters for stations A, B, C, D.

#### The Prediction Approach

Due to the non-linear relationship between  $NO_x$  and input parameters, linear regression methods were found ineligible for modelling. All the cells in the layers are used in MLP neural network-based modelling, resulting in an excessive training problem and a recession in model performance. RF modelling is one of the machine learning models that can be efficient in predictive analysis even in non-linear relations, have lower regression and higher error rates compared to the GMDH method. GMDH neural network prefer the most appropriate cells and pathways that minimize the error rate in the estimation.

As a result the GMDH method performed better for estimating  $NO_x$  as compared to other methods. In this method, the number of hidden layers and the number of cells in those layers are optimally obtained based on the input parameters. The  $NO_x$  prediction within a certain station by the GMDH-type neural network model resulted with regression coefficients (Table 2) ranging from 0.87 to 0.94. For the case of using meteorological and air pollution data of station A as model inputs; the regression coefficient of  $NO_x$  prediction by the GMDH-type models were found to be 0.85, 0.54, and 0.65 for station B, C, D (as shown in Table 4), respectively.

In another research, it was emphasized to use ANN algorithms for predicting hourly concentrations of  $O_3$ ,  $NO_2$ ,  $PM_{10}$ ,  $PM_{2.5}$ ,  $SO_2$ , CO with correlation coefficient ( $R^2$ ) between measured and predicted values and root-mean-

Station	Machine learning techniques											
	Linear regression			Random forest regression			MLP regression			GMDH regression		
	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE
A	0.85	15.16	10.72	0.91	11.61	8.03	0.82	16.32	10.76	0.94	10.95	6.65
В	0.59	16.52	8.18	0.52	17.92	9.21	0.57	32.48	20.69	0.87	8.99	6.10
С	0.46	12.96	10.00	0.77	8.40	5.68	0.64	22.57	16.75	0.88	7.19	5.44
D	0.89	17.42	13.26	0.84	21.45	15.23	0.83	22.08	17.03	0.93	18.68	13.60

**Table 2.** Prediction of each station's NOX for input parameters (PM<sub>10</sub>, SO<sub>2</sub>, T, WD, WS, RH and AP)



**Figure 5**. The prediction of  $NO_x$  in station A. (a) structure of double hidden GMDH layers; (b) actual and predicted NOX by GMDH model.

square (RMSE) values of 0.87 and 59.5 respectively [13]. It was reported by another research group that, using the WS, WD, and temperature as input variables, and ANN, AN-FIS models have provided SO<sub>2</sub> prediction with R<sup>2</sup> values between 0.20 and 0.50 [32], and in another study with R<sup>2</sup> >0.70 [33]. Recent research have proposed using both the meteorological factors and air pollutant parameters as input variables and reported that the ANN model produced a PM<sub>2.5</sub> prediction with R<sup>2</sup>>0.92 [34]. Another study revealed that the use of NO<sub>x</sub> and meteorological parameters as input variables and the ANFIS model provided O<sub>3</sub> predictions with R<sup>2</sup>>0.94 [35]. Based on findings of a recent study, using weather factors and air visibility as input variables is feasible for the ANFIS model and CO-NO<sub>2</sub>, PM<sub>10</sub>, SO<sub>2</sub> – O<sub>3</sub> were predicted with R<sup>2</sup> between 0.65–0.89 [10].

#### Prediction of NO<sub>v</sub> Parameter Within a Station

In this study, the  $NO_x$  prediction was obtained by various regression methods using the  $PM_{10}$ ,  $SO_2$ , T, WD, WS, RH, and AP of each station as input factors. Table 2 shows the calculated R<sup>2</sup>, RMSE, and MAE values of linear, RF, MLP

and GMDH-type regression algorithms for all stations. Regarding  $R^2$ , RMSE and MAE values, the prediction of  $NO_x$ within a certain station by the GMDH-type neural network have provided better results.

 $NO_x$  prediction at station A has been provided by the GM-DH-type neural network with the design demonstrated in Figure 5a. This designed network structure consists of an input layer with 7 neurons, two hidden layers with 5 and 3 neurons, respectively, and an output layer with a single neuron. In this network structure, neuron outputs that minimize the error rate between predicted and actual output were selected. As a result, at station A the NO<sub>x</sub> was predicted with R<sup>2</sup>=0.94, RMSE=10.95, and MAE=6.65, (Fig. 5b). Similarly, prediction of NO<sub>x</sub> in stations B, C, and D by the GMDH-type neural network model have ended up with regression coefficients of 0.87, 0.88, and 0.93, respectively.

The prediction of  $NO_x$  is formulated using the optimal neuron outputs of the network structure shown in Figure 5. Each polynomial equation obtained with active neuron outputs and finally the  $NO_x$  prediction equation is given in Table 3. Relative humidity (RH), one of the input layer pa-

NO <sub>X</sub> III Station A	
Equation	No.
$\overline{Y_1} = 54.76 + 3.06 \text{ SO}_2 - 37.49 \text{ WS} - 0.02 \text{ SO}_2^2 + 6.47 \text{ WS}^2 - 0.17 \text{ SO}_2 \cdot \text{WS}$	
$Y_2 = 15.54 + 2.58 \text{ SO}_2 - 0.56 \text{ T} - 0.02 \text{ SO}_2^2 - 0.003 \text{ T}^2 + 0.04 \text{ SO}_2 \cdot \text{T}$	
$Y_3 = [3.99 - 0.0008 \text{ SO}_2 - 0.0081 \text{ AP}] \cdot 10^4$	
$Y_4 = -6.09 + 0.77 Y_1 + 0.69 Y_3 + 0.014 Y_1^2 + 0.004 Y_3^2 - 0.02 Y_1 \cdot Y_3$	
$Y_5 = -6.96 + 0.75 Y_1 + 0.8 Y_2 + 0.02 Y_1^2 + 0.009 Y_2^2 - 0.04 Y_1 \cdot Y_2$	
$NO_{x} = 0.26 - 0.43 Y_{4} + 1.39 Y_{5} + 0.04 Y_{4}^{2} + 0.03 Y_{5}^{2} - 0.08 Y_{4} \cdot Y_{5}$	(8)

**Table 3.** Parameters and coefficients used in neuron equations for the prediction of NO<sub>v</sub> in station A

Table 4. Prediction of B, C and D station's NOX for input parameters (PM<sub>10</sub>, SO<sub>2</sub>, NO<sub>3</sub>, T, WD, WS, RH and AP) of station A

Station		Machine learning techniques												
	Linear regression			Random forest regression			MLP regression			GMDH regression				
	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE		
В	0.56	17.25	9.31	0.63	15.83	8.44	0.55	17.36	9.69	0.85	9.45	6.16		
С	0.11	16.66	9.78	0.22	15.54	9.54	0.08	16.92	10.30	0.54	13.24	8.19		
D	0.29	44.85	33.44	0.38	41.94	28.62	0.25	46.25	32.66	0.65	38.63	29.15		

rameters, is deactivated because it increases the error rate in the NO<sub>x</sub> prediction model. The output of PM<sub>10</sub> and WD polynomial pair in the first hidden layer, and the output of SO<sub>2</sub> and WD polynomial pair and Y1 output in the second hidden layer are not included in the model because they increase the error rate in the NO<sub>x</sub> prediction. Thus, the NO<sub>x</sub> prediction is modelled as a polynomial function using the selected optimal neuron outputs.

#### **Prediction of NOX Across Stations**

The NOX prediction of stations located at different distances and directions were performed using station A data as input parameters via various machine learning regression methods. Table 4 shows the  $NO_x$  prediction results of stations B, C, and D. The  $NO_x$  values are also used as input data of station A which is accepted as a source station. The results show that the GMDH-type neural network provides higher R<sup>2</sup>, lower RMSE, and MAE values than other methods for all stations. It has been observed that the prediction performance of the proposed model is more successful at station B than at other stations. This result is due to the fact that reference station A, whose data is used as input, is closer to station B than to other stations.

The GMDH-type neural network model designed for the prediction of  $NO_x$  values at station B using data from station A and the regression fit line of this model are shown in Figure 6. In the double hidden layer model designed in Figure 6a, the input parameters WD and RH were deacti-

vated because they increased the error between the predicted and actual output. The  $PM_{10}$  and  $NO_x$  polynomial pair output from the data of station A in the first hidden layer and output polynomial pair  $Y_1$  and  $Y_2$  in the second hidden layer were not used because they adversely affected the  $NO_x$  prediction of station B.

The  $NO_x$  prediction at station B is modelled as a polynomial function via the remaining optimal neuron outputs and the fit line of model is shown in Figure 5b. The model equations were obtained by using the optimal neuron outputs of the GMDH-type neural network. The design is demonstrated in Figure 5a, and the equations are given in Table 5. As a result, Eq. (9) is obtained for the case that neuron outputs are used as polynomial pairs in the  $NO_x$  prediction model of station B.

# Prediction of NOX Across Stations without Using NO<sub>x</sub> as Input Variable

The  $NO_x$  prediction across stations that are located at different distances and directions was performed using meteorological and air pollution data of station A data (the  $NO_x$ is excluded) as input parameters via linear, RF, MLP and GMDH-type neural network regression methods. The R<sup>2</sup>, RMSE and MAE results of these regression methods including all stations were listed in Table 6. According to the analysis of  $NO_x$  prediction results of stations B, C and D (without using the  $NO_x$  data of station A, that was taken as the reference station). It is apparent that GMDH-type



**Figure 6**. The prediction of  $NO_x$  in station B using  $NO_x$  of station A. (a) structure of double hidden GMDH layers; (b) actual and predicted  $NO_x$  by GMDH model.

Table 5. Parameters and coefficients used in neuron equations for the prediction of  $NO_x$  in station B using NOX of station A

Equation	No.
$Y_1 = 15.43 + 1.08 \text{ NO}_x + 0.76 \text{ T} - 0.03 \text{ NO}_x^2 - 0.01 \text{ T}^2 - 0.03 \text{ NO}_x \cdot \text{T}$	
$Y_2 = 21.5 - 0.52 \text{ SO}_2 + 1.003 \text{ NO}_x + 0.0002 \text{ SO}_2^2 - 0.005 \text{ NO}_x^2 + 0.04 \text{ SO}_2 \cdot \text{NO}_x$	
$Y_3 = 4.51 + 1.11 \text{ NO}_{x} + 16.59 \text{ WS} - 0.004 \text{ NO}_{x}^2 - 3.64 \text{ WS}^2 - 0.11 \text{ NO}_{x} \cdot \text{WS}$	
$Y_4 = [-1.05 + 0.0004 \text{ NO}_x + 0.0021 \text{ AP}] \cdot 10^4$	
$Y_5 = 0.27 + 0.94 Y_1 + 0.05 Y_3 - 0.06 Y_1^2 - 0.05 Y_3^2 + 0.11 Y_1 \cdot Y_3$	
$Y_6 = 7.03 - 0.52 Y_2 + 1.17 Y_4 + 0.006 Y_2^2 - 0.02 Y_4^2 + 0.01 Y_2 \cdot Y_4$	
$NO_{x} = -1.08 + 0.45 Y_{5} + 0.59 Y_{6} - 0.01 Y_{5}^{2} - 0.02 Y_{6}^{2} + 0.03 Y_{5} \cdot Y_{6}$	(9)

Table 6. Prediction of NO<sub>x</sub> for station B, C and D station's by using (PM<sub>10</sub>, SO<sub>2</sub>, T, WD, WS, RH and AP) input parameters of station A

Station		Machine learning techniques											
	Linear regression			Random forest regression			MLP regression			GMDH regression			
	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE	<b>R</b> <sup>2</sup>	RMSE	MAE	
В	0.49	18.58	9.01	0.54	17.64	9.68	0.50	18.27	10.19	0.80	10.88	7.31	
С	0.17	16.04	9.71	0.23	15.50	9.63	0.16	16.14	10.16	0.48	13.69	8.51	
D	0.34	43.38	32.45	0.38	41.91	29.35	0.20	47.62	35.10	0.71	35.90	26.13	

neural network has higher R<sup>2</sup> values; 0.80, 0.48 and 0.71 for stations B, C, and D compared to other methods, respectively. Therefore, empirical models for prediction of  $NO_x$  were obtained via GMDH-type neural networks.

The GMDH-type neural network model designed for the prediction of  $NO_x$  values at station B using station A data (not including  $NO_x$ ) and the regression fit line of this model are shown in Figure 7. In the double hidden



**Figure 7**. The prediction of  $NO_x$  in station B without  $NO_x$  of station A. (a) structure of double hidden GMDH layers; (b) actual and predicted  $NO_y$  by GMDH model.

**Table 7.** Parameters and coefficients used in neuron equations for the prediction of  $NO_x$  in station B without  $NO_x$  of station A

Equation	No.
$Y_1 = 13.51 + 1.47 \text{ PM}_{10} + 0.28 \text{ T} - 0.003 \text{ PM}_{10}^2 + 0.01 \text{ T}^2 - 0.06 \text{ PM}_{10} \cdot \text{T}$	
$Y_2 = 49.94 + 1.49 \text{ SO}_2 - 11.5 \text{ WS} - 0.01 \text{ SO}_2^2 + 0.79 \text{ WS}^2 + 0.009 \text{ SO}_2 \cdot \text{WS}^2$	
$Y_3 = 307.35 - 1.73 \text{ PM}_{10} - 6.48 \text{ RH} - 0.0002 \text{ PM}_{10}^2 + 0.04 \text{RH}^2 + 0.03 \text{ PM}_{10} \cdot \text{RH}$	
$Y_4 = 27.84 - 0.16 \text{ PM}_{10} + 0.1 \text{ WD} - 0.005 \text{ PM}_{10}^2 - 0.0008 \text{ WD}^2 + 0.007 \text{ PM}_{10} \cdot \text{WD}_{10}$	
$Y_5 = -1.58 + 1.12 Y_1 - 0.1 Y_3 - 0.01 Y_1^2 - 0.007 Y_3^2 + 0.02 Y_1 \cdot Y_3$	
$Y_6 = 1.61 + 1.42 Y_2 - 0.5 Y_4 + 0.003 Y_2^2 + 0.01 Y_4^2 - 0.02 Y_2 \cdot Y_4$	
$NO_{x} = -13.96 + 0.82 Y_{5} + 0.8 Y_{6} - 0.02 Y_{5}^{2} - 0.02 Y_{6}^{2} + 0.04 Y_{5} \cdot Y_{6}$	(10)

layer model shown in Figure 7a, the AP input parameter was not used because it increases the error between the predicted and actual  $NO_x$  values. The T and WD polynomial pair output was not used in the first hidden layer. In the second hidden layer, as they may pose an adverse effect on  $NO_x$  prediction, the polynomial pair output and the polynomial output formed by the Y<sub>3</sub> output were not used.

The NO<sub>x</sub> prediction of station B is modelled in terms of polynomial function by using the optimal neuron outputs and the data fit line is shown in Figure 7b. The equations of the model obtained by using the optimal neuron outputs of the GMDH-type neural network, are given in Table 7. And the network structure was demonstrated in Figure 7a. As a result, Eq. (10) is obtained for the case that the neuron outputs are used as polynomial pairs in the NO<sub>x</sub> prediction model of station B.

The results show that the predicted NO<sub>v</sub> output at stations B, C and D varies depending on the distance and direction from the source station A. The stations B, C and D are located 21, 23 and 54 km from source station A, respectively. The correlation coefficient R<sup>2</sup> of the GMDH model proposed for stations B, C and D is 0.85, 0.54 and 0.65, respectively when source station A NO<sub>2</sub> parameter is used. The station B, which is closest to the source station, has the highest correlation coefficient. Similarly, when the source station A station NO<sub>x</sub> parameter is not used, the correlation coefficient R<sup>2</sup> of the GMDH model proposed for the stations B, C and D is 0.80, 0.48 and 0.71, respectively (prediction accuracy negatively affected for station D). Although the station D is at the farthest distance from source station A, the correlation coefficient results of the prediction model are higher than station C. That can be interpreted to the influence of local air pollution sources (point and/or linear
sources like industry and traffic) on measured NOX values rather than the level of effect ascribed to near surrounding environment. In order to carry out an ascendant evaluation and sort out such entangled issues, air pollution trajectory and dispersion model and their outputs can be used as helpful tools and useful source of information for specific periods of time [4, 36].

The developed model provided prediction of  $NO_x$  within a certain station by using data from a total number of four different weather stations and taking one station as reference. For the specific cases of  $NO_x$  prediction within a station and across different stations, the GMDH like neural network model results have been increased as a means of enhancing accuracy and minimizing error rates (lower error rates obtained).

#### CONCLUSIONS

This study presents a case study of application of machine learning algorithms to predict  $NO_x$  concentrations using both air pollution and meteorological parameters. To accurately predict the  $NO_x$  parameter, data of a certain station and across different stations were compiled and new models were derived. The GMDH model produced a precise prediction of  $NO_x$  within stations and across stations (station to station) with/without using  $NO_x$  as the model input variable. The key findings can be emphasized below:

- (1) The results show that meteorological parameters significantly affect the  $NO_x$  air pollution parameter and that the effects of meteorological parameters change with distance between stations.
- (2) The proposed empirical models provide a rapid assessment of air quality and the prediction of  $NO_x$  with an acceptable range of accuracy ( $R^2$ ={0.94, 0.85, 0.80} within station A, B via source station data and B via source station not including  $NO_x$ ).
- (3) Results obtained through GMDH models exhibit a high degree of accuracy for  $NO_x$  prediction values and significantly outperform conventional methods. The proposed model provides the opportunity to evaluate the effect of each input parameter on the model output. It has been observed that relative humidity (RH) increases the error rate and is disabled in the derivation of empirical models.
- (4) The proposed GMDH-type neural network model uses air pollution parameters (PM<sub>10</sub> and SO<sub>2</sub>) and meteorological parameters (T, WD, WS, RH and AP) as inputs to estimate NO<sub>x</sub> air pollution values.
- (5) The obtained model is region specific, but for a wide range of spatial representations and validity, data can be gathered from multiple stations at certain distance.
- (6) As a future scope, GMDH-type neural network and

proposed approach can be used to support decision makers and engineers in planning stages including but not limited to optimization of total number and spatial distribution of AQMS to be set-up in a specific region.

#### DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

#### **CONFLICT OF INTEREST**

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

#### ETHICS

There are no ethical issues with the publication of this manuscript.

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#### **Research Article**

# Morphogenesis, physico-chemical properties, mineralogical composition and nature of parent materials of some alluvial soils of the Lower Niger River plain, Nigeria

Achimota DICKSON<sup>1</sup>, Joseph ARULEBA<sup>2</sup>, Joseph Oyinbrakemi TATE<sup>\*1</sup>

<sup>1</sup>Department of Soil Science, Niger Delta University Faculty of Agriculture, Yenagoa, Bayelsa State, Nigeria <sup>2</sup>Department of Soil Science, Ekiti State University, Ado Ekiti, Ekiti State, Nigeria

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#### ABSTRACT

Nine pedons of alluvial origin in the Lower Niger River floodplains of Nigeria were examined for morphogenesis, physicochemical properties, mineralogical composition, and heterogeneity of the parent materials. The soils were stratified with redoximorphic features observed in the different layers reaching A-horizon with subsurface gravization. Soils that received annual alluvial enrichment were found to be structurally weak while others were moderately strong. Soil characteristics showed varying degrees of heterogeneity with source of parent materials and degrees of hydromorphism, moulding morphogenesis and gleization as major soil-forming processes. Silt loam was the predominant soil texture except ELM3 and TFN3 dominated by sandy loam and loamy sand textures. Soils were strongly acid to neutral [pH (H,O), 4.94-7.00], having very low to medium organic matter (0.13-4.02 %), and low to very high K (0.1-2.13 cmol kg<sup>-1</sup>). Quartz dominated the identified mineral phases followed by kaolinite, indicating the dominance of low activity clays and low ferromagnesian minerals presence. The presence of several K-bearing minerals in the pedons (micas and feldspars) suggest that the K pool could naturally be replenished. Textural diversity between the different SMUs is ascribed to different sources of the water-borne sediments and the flow rate of the floodwater at the time of deposition of the parent materials. Organic carbon distribution patterns indicated stratification and heterogeneity of parent materials. Wetness, flooding, and soil chemical and physical fertility were major constraints to increased and sustainable crop production in the Lower Niger River floodplain soils.

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#### INTRODUCTION

Floodplain soils, worldwide, are very useful for agricultural production as they constitute a huge reserve of available nutrients for utilization by crop plants [1]. The agricultural potentials of alluvial soils however, have not been fully exploited because of lack of understanding of their physical and chemical properties and the changes they undergo under intensive cultivation [2]. For instance, greyish colouration at lower depths have been associated with ground water influence or poor drainage [3, 4] while [5] associated clear and smooth horizon boundaries with some diffuse, wavy,

\*Corresponding author.

\*E-mail address: j.o.tate@ndu.edu.ng



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Figure 1. Map showing soil mapping units for the study.

and gradual boundaries to recentness of the soils and young soil development or rejuvenation processes. Two dominant factors appeared to condition kaolinization process under tropical conditions which included composition of the soil solution in the weathering environment and mineralogical composition of the weathering soil or parent rock. However, warm soil temperatures are believed to cause marked dissociation of soil water leading to a build-up of hydrogen ions or lowering of the pH of the soil solution. Under such condition, hydrolytic or H-weathering of silicates to kaolinite progresses rapidly [6]. Texture, organic carbon distribution and clay mineralogy are features commonly used as pointers to the homogeneity or otherwise of the parent materials. Again, the very fine sand and fine sand proportions are used to indicate lithologic discontinuities. Though, it is believed that the floodplain soils of Bayelsa state have high agricultural potentials, current information and knowledge on the dominant soil forming processes, soil characteristics including mineralogical composition, and the homogeneity of parent materials or otherwise are inadequate. For instance, [7] characterization and classification of alluvial soils of the Nun River floodplains did not supply information on the morphogenesis and/or mineralogical status of Bayelsa State soils. Similarly, [8] classification of Ogochie River floodplain soils

in Imo State, Nigeria did not provide sufficient morphogenetic information that could be resourceful as an indicator for proper policy formulation by government. Hence, the efficient management of the soils for increase and sustainable crop production is constrained. The situation is further exacerbated by smallholder farmers' poor management practice of low to sub-optimal use of fertilizer inputs [9-11] that neither mitigate the nutrient mining process nor adequately guarantee the restoration of the fertility status [12]. With the present drive of the world towards food security, the state cannot be left behind and government is working towards agricultural intensification in various parts of the state which cannot be achieved with the present level of information and knowledge on the soils. The aim of this study, therefore, is to examine the morphogenetic features, physico-chemical, mineralogical properties and nature of the parent materials of soils of some selected communities earmarked by the Bayelsa State government for agricultural intensification.

#### MATERIALS AND METHODS

#### **Description of the Study Areas**

This study was carried out in Bayelsa State in the Niger Delta region, Southern Nigeria. The study locations lie



**Figure 2**. Scattered plot showing the distribution of soil pH, Available P, Total N, Organic C and C/N ratio of Elemebiri, Odoni and Trofani soils.

between latitude 05° 22' 03.9" N and 04° 59' 08.9" N and longitude 006° 30' 21.1" E and 006° 06' 54.1" E. The Niger River traverses Nigeria in a North-western to Southern direction with the attendant sediment load ensuring that the delta platform ends up as flat terrain, making it a unique geologic environment. The Niger River flows southward and breaks up into two- the Forcados and Nun Rivers in Bayelsa State, the Nun River, running north and south down the middle of the Bayelsa State, which remains the most direct tributary of the Niger while Forcados River demarcates the western borders of the state. The Elemebiri community by the Lower Niger River, Odoni by Nun River and Trofani by the Forcados River (Figure 1) were chosen for the study due to the proposed agricultural intensification.

#### Soil Sampling and Analyses

Following a detailed soil survey conducted on agricultural lands from Elemebiri (ELM), Odoni (ODN) and Trofani (TFN) using rigid grids, the SMUs were examined for morphogenetic features, physico-chemical properties, mineralogical composition and heterogeneity of parent materials. The designation of the soil mapping units (SMUs) were ELM1, ELM2 and ELM3 for Elemebiri, ODN1, ODN2, and ODN3 for Odoni soils and TFN1, TFN2 and TFN3 for Trofani, representing levee crest, levee slope and floodplain, soil profiles described following the procedures prescribed by the USDA Soil Taxonomy [13] and the World Resource Base. Three representative soil pedons were dug per location, one each on the levee crest, levee slope and flood plain or recent alluvial soils in the channel of the present active river. The soils were morphologically described in-situ and samples collected

from the different horizons for physico-chemical properties following standard procedures. Using the geographic positioning system (GPS), coordinates of each SMU boundaries and profile pit locations were taken during the field survey. The soil samples collected were air-dried, crushed and sieved to pass through a 2 mm mesh. Soil analyses were carried out in the Green River Project laboratory of the Nigerian Agip Oil Company and Zadell laboratory, Port Harcourt, Nigeria. Standard laboratory methods were used to determine the physical and chemical properties of the soil samples. Soil particle size analysis was determined using [14] method, popularly known as hydrometer method. Soil pH both in water and CaCl2 (1:2 ratio) was determined using glass electrode pH meter and electrical conductivity determined using conductivity meter [15]. Organic carbon was determined using the modified dichromate oxidation method of Walkley-Black as described by [15] and the values obtained multiplied by 1.724 to obtain organic matter, total N was determined using macro-kjeldahl digestion-distillation method as described by [16] and available P by Bray P-1 method [17]. Exchangeable acidity was extracted with 1M KCl and determined by titration with NaOH solution using phenolphthalein indicator [18] and exchangeable Al with 0.01 M HCl [19]. Exchangeable cations were extracted with neutral normal ammonium acetate solution as described by [15] and potassium and sodium in the extract measured by flame photometry and calcium and magnesium by atomic absorption spectrophotometry. Cation exchange capacity (CEC) was by the summation method [20]. The mineralogical study on the clay fraction of the soils was carried out based on random powder analysis using A PANalytical X'Pert Pro instrument (XRD) [21].



**Figure 3**. Scattered plot showing the distribution of Exchangeable bases and particle distribution of Elemebiri, Odoni and Trofani soils.



Figure 4. X-ray diffractogram of Elemebiri, Trofani and Odoni soils.

#### **RESULTS AND DISCUSSIONS**

#### **Morphological Properties**

Morphological characteristics of the Elemebiri, Odoni and Trofani soils are presented on Tables 1, 2, and 3, respectively. The soils were generally deep, no much variation in soil colour as most of the SMUs have Hues of 10 YR except some SMUs in Odoni dominated by hues of 7.5 YR. At the 18–31cm depth of the ELM3, few, medium, distinct 2.5 YR 3/3 mottles were observed and at the 38–69 cm depth of TFN3, many, medium, distinct, 10 YR 3/3 mottles were observed. The zone of water saturation during the flood season was at the soil surface for ELM3, ODN3 and TFN3, about 40 cm from the mineral soil surface for ELM2, ODN2 and TFN2, and at about 118 cm for ELM1, 117cm for ODN1 and 140cm for TFN1 (Tables 1, 2, and 3). All the studied profiles were considered hydromorphic because of the presence of mottles except ELM3 and TFN3. Since there was no perched water table below the 18–31cm depth of ELM3 and the 38–69 cm depth of TFN3, which were the observed mottled layers and no mottles occurred below them, the mottling in these layers was ascribed to

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Horizon	Depth (cm)	Soil colour	Mott	les	Texture	Structure	Consi	istence	Concr.	Boundary
			Colour	Pattern			Moist	Wet		
			ш	LM1 (Aquic Dy.	strudepts/Fluvic (	Cambisol)				
Ap	0-8	10YR3/2			fsil	Cr to WSAB	fr	Ss, Sp		cs
Ap2	8-21	10YR3/3			fsilcl	WSAB	sfi	Ss, Sp		cs
B1	21 - 34	10YR3/4			fsilcl	SAB	ımfi	Ss, Sp		cs
B2	34-65	10YR3/4			fsil	SAB	ımfi	Ss, Sp	С	cs
C1	65-90	10YR4/4			fsil	SAB	sfi	Ss, Sp		cs
C2	90-118	10YR4/4			fsilcl	SAB	sfi	Ss, Sp		cs
C3	118 - 150	10YR5/4	7.5YR5/6	F2D	fsil	SAB	sfi	Ss, Sp		cs
C4	150 - 200	10YR5/3	5YR4/6	C2D	fsil	SAB	sfi	Ss, Sp		
			[	ELM2 (Typic Ep	iaquepts/Fluvic C	'ambisol)				
Ap	0 - 11	10YR2/2			fsil	Cr to WSAB	fr	ss, sp		cs
Ap2	11 - 19	10YR3/3			fsilcl	SAB	ımfi	ss, sp		cs
B1	19–32	10YR3/2			fsilcl	SAB	ımfi	ss, sp		cs
B2	32-42	10YR4/3			fsil	SAB	sfi	ss, sp		cs
B3	42-57	10YR5/3	7.5YR4/3		fsil	SAB	sfi	ss, sp		cs
B4	57-88	10YR5/2	7.5YR3/4	M2D	fsilcl	SAB	sfi	ss, sp		cs
C1	88 - 106	10YR5/2	7.5YR4/6	M2D	fsil	SAB	sfi	ss, sp		cs
C2	106 - 190	10YR5/1	7.5YR4/6	M2D	fsil	SAB	sfi	ss, sp		
			ELN	A3 (Eutric Udifli	wents/Haplic-Flu	wic Fluvisol)				
А	0 - 18	10YR3/6			fsl	Cr	vfr	ns, np		cs
Ap1	18-31	10YR3/3	2.5YR3/3	F2D	fsl	Sg	fr	ns, np		c
Ap2	31 - 44	10YR6/4	I	I	ls	Sg	fr	ns, np		c
C1	44-68	10YR4/6	I	I	sl	Sg	fr	ns, np		cs
C2	68-81	10YR5/4	I	I	sl	Sg	fr	ns, np		cs
C3	81-123	10YR3/4	I	I	ls	VWSAB	sfr	ns, np		cs
C4	123-160	7.5YR4/4	I	I	ls	SAB	sfr	ns, np		ac
C5	160 - 200	7.5YR2.5/3	I	I	ls	VWSAB	fr	ns, np		
Mottle pattern- inent).; colour: fsil=fine silt loa	The first letter denote YR=yellowish red; sti m, sl=sandy loam, ls=	s abundance (F=few; C ructure: Cr=crumbly, loamy sand, fsl=fine sa	C=common; M=ma VWSAB=very weak udy loam, fsicl=fine	ny); The center nu : sub angular bloc s silty clay loam, C	mber denotes size ky, WSAB=weak s consistence: ns=noi	(1=fine; 2=medium; 3= ub angular blocky, SA a sticky, np=non plasti	-coarse); The sec B=sub angular l c, ss=slightly stic	cond letter denot blocky, crumb=c cky, sp=slightly p	es contrast (D=dis rumbly, sg=single lastic, fr=friable, v	tinct; P- prom- grain; Texture: fr=very friable,

AlonAlonAlonMoitMoitWet $\gamma$	Horizon	Depth (cm)	Soil colour	Mot	tles	Texture	Structure	Consi	stence	Concr.	Boundary
App         0-23         557R3/3         Abite (minic Lambisol)         Asp         c           Ap2         23-30         757R4/4         551         VWSAB         51         Sasp         c         c           B1         30-63         757R4/4         551         VMSAB         51         ssp         c         c           B2         30-63         757R4/3         57R4/4         M3P         sif         SAB         mf         ssp         c </th <th></th> <th></th> <th></th> <th>Colour</th> <th>Pattern</th> <th></th> <th></th> <th>Moist</th> <th>Wet</th> <th></th> <th></th>				Colour	Pattern			Moist	Wet		
$\phi_{1}$ $0-2$ $757R_{3}/2$ $778R_{3}/2$					DN1 (Humic D	ystrudepts/Fluvic	Cambisol)				
Ap2         23-30 $73YR4/4$ fsi         with         ssip         -         -         c           Bi         30-63 $73YR4/4$ 51 $8AB$ mit         ssip         -         -         64           Bi         117-160 $75YR4/4$ sil $8AB$ mit         ssip         -         -         64           Bi         117-160 $75YR4/6$ $5YR4/4$ M3D         sil $SAB$ mit         ssip         -         -         64           C         100-200 $75YR4/6$ $5YR4/4$ M3D         sil         SAB         mit         ssip         -         -         64           A         20-40         10YR5/4 $5YR4/4$ M3D         sil         C         F         f         <	Ap	0-23	7.5YR3/2			fsil	VWSAB	sfi	ss, sp	I	cs
B1 $30-63$ $757R44$ fsi         fsi         SAB         mf         ssp $-$ dw           B2 $63-117$ $757R44$ $57R44$ $M3D$ $si$ $MB$ $mf$ $ssp$ $ dw$ B3 $117-160$ $757R446$ $57R44$ $M3D$ $si$ $MB$ $mf$ $ssp$ $ dw$ C $10-200$ $757R46$ $57R44$ $M3D$ $si$ $SAB$ $mf$ $ssp$ $ dw$ A $2-20$ $107R51$ $57R44$ $M3D$ $si$ $SAB$ $mf$ $ssp$ $ dw$ A $20-40$ $107R51$ $57R44$ $M3D$ $si$ $SAB$ $mf$ $ssp$ $ dw$ B1 $40-110$ $107R51$ $57R44$ $N3D$ $si$ $SAB$ $si$ $ssp$ $  dw$ B2 $10-10$ $107R51$ $57R44$ $N3D$ $si$ <td< td=""><td>Ap2</td><td>23-30</td><td>7.5YR4/4</td><td></td><td></td><td>fsil</td><td>WSAB</td><td>sfi</td><td>ss, sp</td><td>I</td><td>cs</td></td<>	Ap2	23-30	7.5YR4/4			fsil	WSAB	sfi	ss, sp	I	cs
B2 $63-117$ $7.57R4/6$ $5YR4/4$ $M3D$ $sil$ $SAB$ $nfi$ $ss.gp$ $z$ $dw$ B3 $117-160$ $7.5YR4/3$ $5YR4/4$ $M3D$ $sil$ $SAB$ $nfi$ $ss.gp$ $z$ $dw$ C $160-200$ $7.5YR4/3$ $5YR4/4$ $M3P$ $sil$ $SAB$ $nfi$ $ss.gp$ $z$ $dw$ AP $0-20$ $10YR3/4$ $5YR4/6$ $M3P$ $sil$ $SAB$ $nfi$ $ss.gp$ $z$ $dw$ AP $0-20$ $10YR3/4$ $5YR4/4$ $M3D$ $sil$ $SAB$ $nfi$ $ss.gp$ $z$ $dw$ AP $2-0-40$ $10YR3/4$ $5YR4/4$ $N3D$ $sil$ $SAB$ $sil$ $sign$ $z$ $z$ $z$ B1 $4-11-180$ $75YR3/4$ $5YR4/4$ $N3D$ $sil$ $SAB$ $z$ $z$ $z$ B2 $11-180$ $75YR4/4$ $YR4/4$ <td>B1</td> <td>30-63</td> <td>7.5YR4/4</td> <td></td> <td></td> <td>fsil</td> <td>SAB</td> <td>mfi</td> <td>ss, sp</td> <td>I</td> <td>dw</td>	B1	30-63	7.5YR4/4			fsil	SAB	mfi	ss, sp	I	dw
B3         117-160 $7.57R4/3$ $57R4/4$ $M3D$ $sil$ $SAB$ $mf$ $s.s.p$ $ dw$ C         160-200 $7.57R4/6$ $5YR4/6$ $M3P$ $sl$ $SAB$ $mf$ $s.s.p$ $ dw$ AP         0-200 $7.57R4/6$ $5YR4/6$ $M3P$ $sl$ $SAB$ $mf$ $s.s.p$ $ dw$ AP         0-200 $10YR3/3$ $5YR6/4$ $F2D$ $sl$ $F$ $B$ </td <td>B2</td> <td>63-117</td> <td>7.5YR4/6</td> <td></td> <td></td> <td>sil</td> <td>SAB</td> <td>mfi</td> <td>ss, sp</td> <td>I</td> <td>dw</td>	B2	63-117	7.5YR4/6			sil	SAB	mfi	ss, sp	I	dw
C         160-200 $75YR4/6$ $N3P$ stil         SAB         mfi         s.s.p         -           AP         0-20         10YYB3/4 $-20N2$ 10YYB3/4 $-51$ $Cr$ $fr$ $ns.p$ $ cs$ AP         0-20         10YYB3/4 $5YR4/4$ $F3D$ $fr$ $fr$ $ns.p$ $ cs$ B1         40-110         10YYS3 $5YR4/4$ $F2D$ $sl$ $sh$ $fr$ $ns.p$ $ cs$ B2         110-141 $75YR5/2$ $5YR4/4$ $N3D$ $sl$ $sh$ $fr$ $ns.p$ $ cs$ B3         141-180 $75YR5/2$ $5YR4/4$ $N3D$ $sl$ $sh$ $fr$ $r$ $r$ A $-5$ $180-200$ $10YR5/1$ $5YR4/4$ $N3D$ $sl$ $sh$ $fr$ $r$ $r$ A $-5$ $5YR4/4$ $N3D$ $sl$ $SAB$ $sl$ $ssp$ $ r$	B3	117-160	7.5YR4/3	5YR4/4	M3D	sil	SAB	mfi	ss, sp	I	dw
Ap $0-20$ IOTR314         ISTR312 (Typic Epiaqueps/Fluvic Cambra)           Ap $0-20$ IOTR314 $fil$ Cr         fr $ns, np$ $ cs$ Bit $40-110$ IOTR53         SYR64         F2D         sil         Cr         fr $ns, np$ $ cs$ Bit $40-110$ IOTR53         SYR64         F2D         sil         SAB         sin $s_5$ $cs$ $cs$ Bit $40-110$ IOTR53         SYR64         F2D         sil         SAB $sff$ $s_5$ $cs$ $cs$ Bit $20-40$ IOTR51         SYR34         M3P $slf$ SAB $sff$ $s_5$ $cs$ Bit $5-11$ $7.5YR44$ M3P $slf$ SAB $sff$ $s_5$ $cs$ $cs$ Ap $-5-11$ $7.5YR44$ M3P $slf$ $SAB$ $sff$ $ns, np$ $cs$ $dth$ Ap $-5-11$ $7.5YR44$ $SYB$	С	160 - 200	7.5YR4/6	5YR4/6	M3P	sil	SAB	mfi	ss, sp	I	
Ap         0-20         10YB3/4         fsil         Cr         fr         ns.np         -         -         r         r         r         ns.np         -         -         r         r         r         r         ns.np         r         r         r         r         r         ns.np         r         r         r         r         r         r         ns.np         r					ODN2 (Typic Ef	viaquepts/Fluvic C	ambisol)				
A         20-40         IOTS/A         I         WSAB         sfi         ss.sp         -         cs           B1         40-110         10YTS/3         5YTS/4         FYD         sil         SAB         sfi         ss.sp         -         cs           B2         110-141         7.5YTS/4         5YTS/4         FYD         sil         SAB         sfi         ss.sp         -         cs           B3         141-180         7.5YTS/2         5YTS/4         M3D         sil         SAB         sfi         ss.sp         -         cs           C         180-200         10YTS/1         5YTS/4         M3D         sil         SAB         sfi         ss.sp         -         cs           A         0-5         7.5YTS/2         5YTS/4         M3P         sil         SAB         sfi         ss.sp         -         cs           A         0-5         7.5YTS/4         5YTS/4         M3P         sil         NSAB         sfi         ss.sp         -         cs         df           A         0-5         7.5YTS/4         5YTS/4         5YTS/4         5YTS/4         SYTS/4         SYTS/4         SYTS/4         SYTS/4         SYTS/4	Ap	0-20	10YR3/4			fsil	Cr	fr	us, np	I	cs
B1 $40-110$ $10YR5/3$ $5YR6/4$ $F2D$ $sil$ $SAB$ $si$ $ss,sp$ $ cs$ B2 $110-141$ $7.5YR5/4$ $5YR4/4$ $M3D$ $sil$ $SAB$ $Si$ $ss,sp$ $ cs$ B3 $110-141$ $7.5YR5/4$ $5YR3/4$ $M3D$ $sil$ $SAB$ $Si$ $ss,sp$ $ cs$ B3 $110-180$ $7.5YR5/2$ $5YR3/4$ $M3P$ $sil$ $SAB$ $si$ $ss,sp$ $ cs$ A $0-5$ $7.5YR3/4$ $5YR3/4$ $M3P$ $sil$ $SAB$ $sil$ $ss,sp$ $ cs$ A $0-5$ $7.5YR4/4$ $5YR5/4$ $5YR5/4$ $SYR6/4$ $SYR6/4$ $SIR$ $sil$ $ss,sp$ $ cs$ A $0-5$ $7.5YR4/4$ $5YR5/4$ $SYR6/4$ $SYR6/4$ $SIR         sil ss,sp  cs           A         0-5 5YR$	А	20 - 40	10YR5/4			1	WSAB	sfi	ss, sp	I	cs
B2         110-141         7.5YR5/4         5YR4/4         M3D         sil         SAB         Sf         s,s,p         -         div           B3         141-180         7.5YR5/2         5YR3/4         M3P         sil         SAB         Sf         s,sp         -         div           B3         141-180         7.5YR5/2         5YR3/4         M3P         sil         SAB         sf         s,sp         -         div           C         180-200         10YR5/1         5YR3/4         M3P         sil         SAB         sf         s,sp         -         cs           A         0-5         7.5YR3/2         5YR3/4         SYR5/1         Kil         WSAB         sf         s,sp         -         dg           AP1         5-11         7.5YR4/4         5YR6/1         C2D         sil         WSAB         sf         s,sp         -         dg           AP2         11-25         7.5YR4/4         5YR6/1         C2D         sil         SAB         mf         ms,mp         -         dg           AP2         11-25         7.5YR4/4         5YR6/1         C2D         sil         SAB         mf         ms,mp         - <t< td=""><td>B1</td><td>40-110</td><td>10YR5/3</td><td>5YR6/4</td><td>F2D</td><td>sil</td><td>SAB</td><td>sfi</td><td>ss, sp</td><td>I</td><td>cs</td></t<>	B1	40-110	10YR5/3	5YR6/4	F2D	sil	SAB	sfi	ss, sp	I	cs
B3         141-180         7.5YR5/2         5YR3/4         M3P         stil         SAB         sfi         ss, sp         -         ct         ct           C         180-200         10YR5/1         5YR3/4         M3P         stil         SAB         sfi         s, sp         -         ct         ct           A         0-5         7.5YR3/2         5YR3/4         M3P         stil         SAB         sfi         s, sp         -         ct         ds           A         0-5         7.5YR4/4         5YR5/3         C2D         stil         WSAB         sfi         s, sp         -         dg           Ap1         5-11         7.5YR4/4         5YR5/1         C2D         stil         WSAB         sfi         s, sp         -         dg           Ap2         11-25         7.5YR4/4         5YR6/1         SYB         stil         SAB         m6         m5, mp         -         dg           B1         25-41         7.5YR4/4         5YR6/1         C2D         stil         SAB         mf         m5, mp         -         dg           B1         25-41         7.5YR4/4         5YR4/4         M3P         cl         SAB	B2	110-141	7.5YR5/4	5YR4/4	M3D	sil	SAB	Sfi	ss, sp	I	dw
C $180-200$ $10YRS/I$ $5YR3/4$ $M3P$ $sil$ $SAB$ $sfi$ $ss,sp$ $-$ A $0-5$ $7.5YR3/2$ $ODN3$ ( $Fluvaquentic Epiaquepts/Fluvic Cambisol$ ) $r$ $r$ A $0-5$ $7.5YR3/2$ $DDN3$ ( $Fluvaquentic Epiaquepts/Fluvic Cambisol$ ) $r$ $r$ Apl $5-11$ $7.5YR4/4$ $5YR6/1$ $5YR6/1$ $sil$ $WSAB$ $sfi$ $ss,sp$ $ dg$ Apl $5-11$ $7.5YR4/4$ $5YR6/1$ $C2D$ $sil$ $WSAB$ $sfi$ $ms,mp$ $ dg$ Ap2 $11-25$ $7.5YR4/4$ $5YR6/1$ $C2D$ $sil$ $SAB$ $mf$ $ms,mp$ $ dg$ B1 $2.5-41$ $7.5YR4/4$ $5YR6/1$ $C2D$ $sil$ $SAB$ $mf$ $ms,mp$ $ dg$ B2 $41-48$ $7.5YR4/4$ $5YR4/4$ $M3P$ $cl$ $SAB$ $mf$ $ms,mp$ $ cs$ </td <td>B3</td> <td>141 - 180</td> <td>7.5YR5/2</td> <td>5YR3/4</td> <td>M3P</td> <td>sil</td> <td>SAB</td> <td>sfi</td> <td>ss, sp</td> <td>I</td> <td>cs</td>	B3	141 - 180	7.5YR5/2	5YR3/4	M3P	sil	SAB	sfi	ss, sp	I	cs
A         0-5         7.5YR3/2         fsil         WSAB         sfi         ss.sp         -         dd           API         5-11         7.5YR3/2         22D         sil         WSAB         sfi         ss.sp         -         dd           API         5-11         7.5YR4/4         5YR5/3         C2D         sil         WSAB         sfi         ss.sp         -         dd           AP2         11-25         7.5YR4/4         5YR6/4         C2D         sil         NSAB         sfi         ms.mp         -         dd           B1         25-41         7.5YR6/2         5YR6/1         C2D         sil         SAB         mf         ms.mp         -         dd           B2         41-48         7.5YR6/2         5YR6/1         C2D         sil         SAB         mf         ms.mp         -         dd           B3         48-56         7.5YR5/4         5YR4/4         M3P         cl         SAB         mf         ms.mp         -         cs           B3         66-122         7.5YR5/6         5YR4/4         M3P         cl         SAB         mf         ms.mp         -         cs           C1         56-122 <td>C</td> <td>180 - 200</td> <td>10YR5/1</td> <td>5YR3/4</td> <td>M3P</td> <td>sil</td> <td>SAB</td> <td>sfi</td> <td>ss, sp</td> <td>I</td> <td></td>	C	180 - 200	10YR5/1	5YR3/4	M3P	sil	SAB	sfi	ss, sp	I	
A         0-5         7.5YR3/2         fail         WSAB         sfi         ussap         -         dd           Apl         5-11         7.5YR4/4         5YR5/3         C2D         sil         WSAB         Sfi         ussap         -         dd           Ap2         11-25         7.5YR4/4         5YR6/4         C2D         sil         WSAB         Sfi         ussap         -         dd           B1         25-41         7.5YR4/2         5YR6/1         C2D         sil         SAB         mf         ms, mp         -         dg           B1         25-41         7.5YR4/4         5YR6/1         C2D         sil         SAB         mf         ms, mp         -         dg           B1         25-41         7.5YR4/4         5YR6/1         C2D         sil         SAB         mf         ms, mp         -         dg           B2         41-48         7.5YR4/4         5YR4/4         M3P         cl         SAB         mf         ms, mp         -         ms           B3         48-56         7.5YR5/6         5YR4/4         M3P         cl         SAB         mf         ms, mp         -         -         -         - <td></td> <td></td> <td></td> <td>OD</td> <td>N3 (Fluvaquenti</td> <td>c Epiaquepts/Fluv</td> <td>ic Cambisol)</td> <td></td> <td></td> <td></td> <td></td>				OD	N3 (Fluvaquenti	c Epiaquepts/Fluv	ic Cambisol)				
Apl         5-11         7.5YR4/4         5YR5/3         C2D         sil         WSAB         Sil         s,sp         -         dw           Ap2         11-25         7.5YR4/4         5YR6/4         5YR6/1         C2D         sil         SAB         sfn         ms,mp         -         dg           B1         25-41         7.5YR6/2         5YR6/1         C2D         sil         SAB         mf         ms,mp         -         dg           B2         41-48         7.5YR4/4         5YR5/2         C2D         sicl         SAB         mf         ms,mp         Fe-Mn         cs           B3         48-56         7.5YR5/4         5YR4/4         M3P         cl         SAB         mf         ms,mp         -         cs           C1         56-122         7.5YR5/6         5YR4/4         M3P         1         SAB         mf         ms,mp         -         cs           C2         122-200         7.5YR4/6         5YR4/6         M3P         1         SAB         mf         ms,mp         -         cs	Α	0-5	7.5YR3/2			fsil	WSAB	sfi	ss, sp	I	dg
Ap2         11–25         7.5YR4/4         5YR6/4         C2D         sil         SAB         sfi         ms, mp         -         dg           B1         25–41         7.5YR6/2         5YR6/1         C2D         sil         SAB         mfi         ms, mp         -         dg           B2         41–48         7.5YR4/4         5YR5/2         C2D         sicl         SAB         mfi         ms, mp         Fe-Mn         cs           B3         48–56         7.5YR5/4         5YR4/4         M3P         cl         SAB         mfi         ms, mp         -         cs           C1         56–122         7.5YR5/6         5YR4/4         M3P         1         SAB         mfi         ms, mp         -         cs           C2         122–200         7.5YR4/6         5YR4/6         M3P         1         SAB         mfi         ms, mp         -         cs	Ap1	5-11	7.5YR4/4	5YR5/3	C2D	sil	WSAB	Sfi	ss, sp	I	dw
B1         25-41         7.5YR6/2         5YR6/1         C2D         sil         SAB         mfi         ms, mp         Fe-Mn         cs           B2         41-48         7.5YR4/4         5YR5/2         C2D         sicl         SAB         mfi         ms, mp         Fe-Mn         cs           B3         48-56         7.5YR5/4         5YR4/4         M3P         cl         SAB         mfi         ms, mp         Fe-Mn         cs           C1         56-122         7.5YR5/6         5YR4/4         M3P         cl         SAB         mfi         ms, mp         -         cs           C1         56-122         7.5YR5/6         5YR4/4         M3P         1         SAB         mfi         ms, mp         -         cs           C2         122-200         7.5YR4/6         5YR4/6         M3P         1         SAB         mfi         ms, mp         -         cs	Ap2	11 - 25	7.5YR4/4	5YR6/4	C2D	sil	SAB	sfi	ms, mp	I	dg
B2       41-48       7.5YR4/4       5YR5/2       C2D       sicl       SAB       mfi       ms, mp       Fe-Mn       cs         B3       48-56       7.5YR5/4       5YR4/4       M3P       cl       SAB       mfi       ms, mp       -       cs         C1       56-122       7.5YR5/6       5YR4/4       M3P       1       SAB       mfi       ms, mp       -       cs         C2       122-200       7.5YR4/6       5YR4/6       M3P       1       SAB       mfi       ms, mp       -       cs	B1	25-41	7.5YR6/2	5YR6/1	C2D	sil	SAB	mfi	ms, mp	Fe-Mn	cs
B3         48–56         7.5YR5/4         5YR4/4         M3P         cl         SAB         mfi         ms, mp         -         cs           C1         56–122         7.5YR5/6         5YR4/4         M3P         1         SAB         mfi         ms, mp         -         cs           C2         122–200         7.5YR4/6         5YR4/6         M3P         1         SAB         mfi         ms, mp         -         cs	B2	41 - 48	7.5YR4/4	5YR5/2	C2D	sicl	SAB	mfi	ms, mp	Fe-Mn	cs
CI 56-122 7.5YR5/6 5YR4/4 M3P 1 SAB mfi ms,mp - cs C2 122-200 7.5YR4/6 5YR4/6 M3P 1 SAB mfi ms,mp -	B3	48-56	7.5YR5/4	5YR4/4	M3P	cl	SAB	mfi	ms, mp	I	cs
C2 122–200 7.5YR4/6 5YR4/6 M3P 1 SAB mfi ms, mp –	C1	56-122	7.5YR5/6	5YR4/4	M3P	Ι	SAB	mfi	ms, mp	I	cs
	C2	122-200	7.5YR4/6	5YR4/6	M3P	1	SAB	mfi	ms, mp	I	

10711011	Depth (cm)	Soil colour	Mott	les	Texture	Structure	Consi	istence	Concr.	Boundary
			Colour	Pattern			Moist	Wet		
				TFN1 (Aquic Dy:	strudepts/Fluvic C	ambisol)				
Ap	0-14	10YR3/4			fsil	SAB	sfi	ss, sp		cs
А	14-31	10YR3/4			fsil	SAB	sfi	ss, sp		cs
B1	31 - 55	10YR3/4			fsil	SAB	mfi	ss, sp		cs
B2	55-140	10YR3/6			fsil	SAB	mfi	ss, sp		CS
B3	140 - 150	10YR4/2	2.5YR3/4	C2D	fsilcl	SAB	mfi	ms, mp	С	cs
С	150 - 200 +	10YR4/4	5YR4/6	M3P	fsil	SAB	mfi	ms, mp		
				TFN2 (Typic Ep	iaquepts/Fluvic Co	ambisol)				
Ap	0-11	7.5YR3/3			fls	Cr	fr	ns, np		cs
Ap2	11 - 35	10YR4/4			fsil	SAB	sfi	ss, sp	С	cs
Bl	35-44	10YR5/2			fsilcl	SAB	mfi	ss, sp		cs
B2	44-70	10YR5/3	5YR5/6	C2D	fsil	SAB	sfi	ss, sp		cs
C1	70-126	10YR5/2	5YR3/3	M3P	fsil	SAB	sfi	ss, sp		dg
C2	126-200+	10YR6/2	5YR3/3	M3P	I	SAB	sfi	ss, sp		
				TFN3 (Aquic Ui	difluvents/Haplic	Fluvisol)				
Α	0 - 13	10YR3/3			fsl	Μ	sfi	ss, sp		dw
Ap1	13-23	10YR3/4			sl	Gr	fr	ns, np		dw
Ap2	23–38	10YR5/4			sl	Gr	fr	ns, np		cs
CI	38-52	10YR3/3	10YR3/3	M2D	sl	Gr	fr	ns, np		CS
C2	52-69	10YR4/6	10YR3/3	M2D	sl	Gr	fr	ns, np		cs
C3	69–83	10YR3/3			ls	Sg	I	ns, np		cs
C4	83-200+	10YR6/3			S	Sg	I	ns, np		

SMU				Pe	rcentage min	eral in the soi	l			
	Quartz	Kaolinite	Vermiculite	Biotite	Muscovite	Microline	Albite	Anorthite	Chlorite	Zeolite
ELM1	46.1	15.1	nd	nd	11.2	13.6	14	nd	nd	nd
ELM2	41.2	20	nd	nd	11.3	18.3	9.2	nd	nd	nd
ELM3	51.1	9.9	0.4	nd	9.1	nd	12.4	17.1	nd	nd
ODN1	57.6	8.3	0.1	nd	5.8	13.5	6.7	8	nd	nd
ODN2	60.9	4	0.1	nd	5.8	11.9	6.4	10.9	nd	nd
ODN3	48	15.4	nd	nd	8.6	14.4	7.2	6.4	nd	nd
TFN1	41.4	19.2	0.6	1.8	9.5	14.7	12.8	nd	nd	nd
TFN2	46.2	22.6	0.4	1.1	8.7	12.9	8.2	nd	nd	nd
TFN3	51.8	8	0.2	2.5	6	20.3	11.1	nd	nd	nd

Table 4. Percentage distribution of clay mineral types in the soil mapping units

nd: Not detectable.

the parent materials of the specific layers which indicated heterogeneity in parent materials of the specific SMUs. No saturation zone was found in any of the profiles during the dry season. All the profiles therefore were considered to be influenced by an annual cycle of wet/dry soil moisture regimes which qualified as 'udic' moisture regimes [13]. The drainage condition of the profiles was dictated by the rainy season and the annual Niger River floods. Subsurface greyization (gleys) was a notable morphometric feature in the levee slope soils (ELM2, ODN2 and TFN2) and ODN3 of the flood plain. Regular decrease of organic carbon down the profiles with hydromorphism within the surface 50 cm was observed for ELM2, ODN2 and TFN2 and within 5 cm for ODN3. The presence of mottles in these soils indicated fluctuation of ground water table while the presence of gleys in the lower horizons of ELM2, ODN2 and TFN2 indicated perennial hydromorphism as noted previously by [5]. Greyish colouration at lower depths is associated with ground water influence or poor drainage [3] and/or such soil layer is subject to groundwater influence in most part of the year [4]. The pedons showed mostly clear and smooth horizon boundaries with some diffuse, wavy, and gradual boundaries, reflecting the recentness of the soils and young soil development or rejuvenation processes [5] and different seasons of deposition of the parent materials. Moreover, no clay movement from A to B horizon was noticed confirming young soil development or rejuvenation processes. Similarly, cutans (clay skins) were absent on ped surfaces as there was no illuvial accumulation of clays in the pedons. The presence of many mica flakes in the profiles corroborated the incipient nature of the soils. Surface and subsurface texture of the SMUs was fairly uniform, dominated by silt loam and silty clay loam except ELM3 and TFN3 dominated by sandy loam and loamy sand. The ELM3 and TFN3 SMUs, of recent alluvial soils in the channels of present active rivers received annual alluvial materials enrichment from the yearly floods and were structurally weak. Structural development in the other SMUs was moderately strong.

#### **Physico-Chemical Properties**

The texture of the soils was dominantly silt loam followed by silty clay loam and loam except ELM3 and TFN3, dominated by loamy sand and sandy loam (Figure 3). The dominance of sand in ELM3 and TFN3 (Figure 3) indicated that the SMUs have high infiltration rate and low water holding capacity with possibility of moisture stress during dry months [22, 23]. The clay distribution within ELM1, ELM2, ODN1, ODN2, ODN3, TFN1 and TFN2 SMUs was irregular. Reported [24] irregular distribution of clay within the subsoil of three pedons, characteristic of cambic horizon. Though the distribution of silt/clay ratio was also irregular with depth, silt/clay ratios generally increased with increase in silt content and vice versa. Higher silt/clay ratio in the surface layers reflected annual alluvial enrichment of the surface through deposition by the annual floods. Report [25] indicated that soils with silt/clay ratios below 0.15 indicated that such soils are of old parent material, while those above 0.15 are of young parent materials with low degree of weathering. Recorded [24] silt/clay ratio of <1.00 in Southern Guinea Savanna soils in Nigeria and concluded that the soils have undergone ferralitic pedogenesis. All the SMUs recorded silt/clay ratios far above unity confirming that the soils are young with weatherable minerals and have not gone through ferralitic pedogenesis.

The SMUs were moderately acidic to neutral, pH (water) ranging from 5.31 to 7.00 for Elemebiri soils, 5.33–6.70 for Odoni soils and 5.30–6.80 for Trofani soils, respectively (Figure 2). The pH of the soils generally increased with soil depth due to less H+ ions released from organic matter decomposition as organic matter decreased irregularly (Figure 2) with increase in depth [26]. Reported [27] pH of 6.0 to 7.0 as the optimum pH for most agricultural crops while [28] and [29] gave 5.5 to 7.0 as the preferred range for most crops. Among the SMUs, the surface layers of ELM1 and ELM2 fall below the FAO preferred pH range as reported by [28]. This is an indication that the SMUs need some form

of soil amendments. Attributed [5] increase in soil pH with depth to ferrolysis which is acidification of topsoil caused by continual displacement of bases by ferrous ion during the reduction phase associated with annual flooding. The study area is prone to high rainfall and flooding therefore, there is possibility of ferrolysis. Usually,  $\Delta pH$  value is used to estimate the presence of negatively charged clay colloids in soils [30]. Positive  $\Delta pH$  values were obtained for all the soils indicating that the soils were all negatively charged.

Organic matter content of the soils, generally, was low to moderate, ranging from 0.19-3.88%, 0.13-4.02% and 0.37-2.76% for the Elemebiri, Odoni and Trofani soils, respectively (Figure 2). Generally, organic carbon and indeed organic matter levels decreased irregularly with soil depth which agreed with the reports of previous authors in Nigeria [31, 32] and [30] in Ethiopia. Also reported [5] organic C decrease with soil depth for Bangladeshi soils and low organic C content was attributed to rapid decomposition of organic matter under hyperthermic temperature regime. For the soils under consideration, low organic matter concentration was attributed to low biomass return to the soils owing to short fallow periods coupled with the cultural practice of bush burning which destroys organic materials. It is necessary to note that organic matter mineralization rate in the soils is high due to high temperatures and heavy rainfall as the SMUs belong to the iso-hyperthermic soil temperature regime. Low N values was traced to high rate of organic matter decomposition and mineralization as well as leaching, coupled with intermittent flooding and drying which is known to favour N loss through nitrification-denitrification processes [29]. Reported [33] that soils with less than 0.07% total N have limited N mineralization potential, whereas those having values greater than 0.15% would be expected to mineralize sufficient amount of N during the succeeding crop cycle. Based on this, the surface layers of ELM1, ELM2, ODN2, ODN3, TFN1, and TFN3 are likely to have reasonable mineralization potential while the mineralization potential of ELM3, ODN1 and TFN2 was low.

Available P distribution apparently decreased with depth in ELM and ODN with an inconsistent distribution pattern in TFN soils. Available P concentration in ELM and ODN agreed with the results of [34] in a study of soils along a toposequence in Ethiopia that reported available P showing an increasing trend down topographic position and a decreasing trend with depth which they attributed to increase in clay content and decrease in soil organic matter content. Whilst the inconsistency in distribution of available P in TFN with an expected distribution pattern of organic matter in the same SMU, the result, however, could be attributed to the differences in source of parent material. Similar variations were observed in ELM soils where organic matter was seen to be inconsistent with available P distribution. Moreover, the distribution of P within the profiles showed no regular pattern of decrease which agreed with the findings of [35]. This could be due to P fixing capacity and the slow release by the soils as a result of the relatively high level of iron and aluminum oxides in the soils. Low P availability in tropical soils can as well be attributed to the nature of the chemical forms of soil P and the high content of oxides of Fe and Al which are associated with high P fixation.

Exchangeable K varied from 0.18-1.81 cmol kg<sup>-1</sup> in Elemebiri, 0.10-2.13 cmol kg<sup>-1</sup> in Odoni and 0.14-1.88 cmol kg<sup>-1</sup> in the Trofani soils (Figure 3). Also, the ECEC values were low, ranging from 1.49-6.11 cmol kg<sup>-1</sup> in Elemebiri soils, 2.47-8.06 cmol kg-1 in Odoni soils, and 2.79-6.37 cmol kg-1 in Trofani Exchangeable Ca2+ dominated the exchange complex of the SMUs followed by Mg<sup>2+</sup>. Cation ratios are helpful in identifying soil structural problems. In the SMUs, Ca<sup>2+</sup>/Mg<sup>2+</sup> ratio of most of the layers was above unity. Reported [5] Ca<sup>2+</sup>/Mg<sup>2+</sup> ratios of less than unity in Bangladeshi soils, attributing this development to loss of Ca<sup>2+</sup> due to gleization. Reported [26] that Ca2+/Mg2+ ratio in soils decreases with increasing maturity. The low Ca2+/Mg2+ ratios recorded in the SMUs was rather ascribed to the inherently low concentration of ferromagnesian minerals that supply Ca and to a lesser extent, possible loss of Ca by gleization as noted previously by [5]. Low exchangeable bases in soils (Ca, Mg, K and Na) have been attributed to acidifying properties of organic matter, high aluminium concentration and leaching loss of exchangeable bases [36]. The low exchangeable Ca and Mg in these soils was attributed to the inherently low concentration of ferromagnesian minerals, low nutrient retentive capacity, high exchangeable Al and leaching losses due to the high rainfall. Based on the categorisation of [28], it is obvious that K in most layers of the pedons was medium to very high.

The exchange acidity of 45% of the soils was 2.0 cmol kg<sup>-1</sup> and above suggesting that 45% of the soils were slightly to strongly acidic [8], Odoni soils having higher total exchange acidity.

#### MINERALOGICAL COMPOSITION

Table 4 shows the variation in the composition of clay mineral assemblage in the pedons. Ten different mineral phases were identified in the nine locations, kaolinite, quartz, muscovite  $(KAl_2(AlSi_3O_{10})(OH,F)_2$  and albite  $(NaAlSi_3O_8)$  identified in all the locations. Quartz was the dominant mineral in the locations while kaolinite was easily detectable and dominated the clay fraction. Microline was detected in all the locations except ELM3. The occurrence of vermiculite and biotite was low, vermiculite, detected only in ELM1, ELM2 and ODN3 and biotite, in Trofani soils (TFN1, TFN2 and TFN3) only. The X-ray diffractograms of the pedons are presented in Figure 4. The dominance of silt-sized quartz fraction in the clay mineral assemblage further confirmed the assertion that the soils were recent in origin. Kaolinite, muscovite, microline (KAlSi<sub>3</sub>O<sub>8</sub>) and albite of the

plagioclase group were also prominent (Table 4). Anorthite [Ca (Al<sub>2</sub>Si<sub>2</sub>)O<sub>8</sub>], another mineral of the plagioclase group was present in ELM3 out of the nine locations. There was also vermiculite, an interstratified clay mineral, present in six (ELM3, ODN1, ODN2, TFN1, TFN2, and TFN3) SMUs though not in large quantities and biotite [K(Mg.Fe)<sub>2</sub>(Al-Si<sub>2</sub>O<sub>10</sub>)(OH,F)<sub>2</sub>] in TFN1, TFN2, and TFN3. These findings corroborate an earlier report by [37]. It was further reported that kaolinite, quartz, mica, vermiculite and interstratified or mixed layer silicates were present in the clay fractions of Mbiama-Kaiama soils of the same meander belts region of Bayelsa State. Similarly [38], identified quartz, kaolinite, illite, smectite, vermiculite and interstratified types as common minerals in the silt-clay fraction of floodplain soils in Kogi State. In the [39] characterization of clays in Odukpani, South-eastern Nigeria reported that Kaolinite is the dominant clay mineral.

Another major observation in the clay mineralogy of the soils was the near absence of ferromagnesian minerals in the soils. The only minerals of the ferromagnesian family were vermiculite and biotite (Table 4) whose quantities were low. This may account for the low levels of basic cations such as Mg and Ca in the soils. Consequently, Mg:K ratio in the soils was low which placed most of the pedons in the marginally suitable class for oil palm production. However, the presence of several K bearing minerals in the pedons such as the micas (biotite and muscovite) and the feldspars suggest that the K pool in the pedons could naturally be replenished. This agreed with the medium to high K concentration results recorded in most of the pedons.

Among the silicate clay minerals, kaolinite was dominant (Table 4) which agreed with findings of [39]. Kaolinization, therefore, is the dominant clay forming process in these pedons hence the dominance of kaolinite among silicate clays in all the pedons. Vermiculite and biotite, the only ferromagnesian minerals detected were low and were not detected in most of the pedons which could be evidence that parent materials are heterogenous. The near absence of ferromagnesian minerals in the clay mineralogical composition of the soils and the dominance of kaolinite implied that the soils are dominated by low activity clays (LAC) and can easily be eroded. Consequently, split application of recommended fertilizers rates is suggested to avert leaching loss of nutrients when applying fertilizer. In addition, cultural practices such as bush burning that destroy soil organic matter should be avoided to maintain organic matter levels in the soils.

#### Soil Properties and Nature of the Parent Materials

Texture (Figure 3), organic carbon distribution (Figure 2) and clay mineralogy (Table 4 and Figure 4) are features commonly used as indicators of the homogeneity or otherwise of the parent materials. The very fine sand and fine sand proportions are used to indicate lithologic discontinuities with-

in individual soil profiles because these two size fractions form the greatest percentage of the sand fraction, and are assumed to be made up of quartz and other resistant primary minerals. Though sand was not separated into fine, medium and coarse in this study, the distribution pattern of sand in some of the SMUs presented a lot about the parent materials. Figure 3 indicated significant change and differences in the distribution pattern of sand in the different horizons of ELM3, ODN1, ODN2, TFN2, TFN3, and to a lesser extent, ODN3 and TFN1. Similar observation was made in the distributional pattern of silt and in particular clay in the soils. The textural diversity observed between the different SMUs was ascribed to differences in the sources of the water-borne sediments and flow rate of the flood water at the time of deposition of the parent materials. Whereas the parent materials of ELM3 and TFN3 were deposited during the period of high flood as they are recent alluvial soils from the channels of present active Niger and Forcados Rivers and dominantly constituted by sand-sized particles (sandy loam, loamy sand and sand), other profiles were dominated by silt loam, silty clay loam and loam. The finer soil particles, in suspension, probably were transported for longer period of time over greater distances and deposited at low flood period when there was less turbulence than the case of ELM3 and TFN3, deposited during the high flood water under high current. Reported [24] higher amount of silt in JG3 profile of Southern Guinea Savannah of Nigeria and linked it to the seasonal depositional effect of the seasonal stream and the Suleja water reservoir inundating the JG3 area. The 0-11 cm layer of ELM2 seems to be of different parent material from the rest of the horizons as the proportion of sand in this layer was 79% as compared to other horizons with 40% or less sand. The high concentration of sand in this layer cannot be ascribed to clay eluviation or surface erosion. The dominance of sand in the two bottom layers of ODN2 also showed that these layers were of different parent material (Figure 3).

Organic carbon distribution pattern in the soils (Figure 2) indicated stratification. Irregular decrease in organic matter content with depth was consistent with the properties of fluvents [13]. The organic C distribution pattern in ELM1, ELM2, ELM3, ODN2, ODN3, TFN1, TFN2 and TFN3 did not suggest uniform parent materials with the observed abrupt increase in organic C in some horizons down the profile of some SMUs. As shown on Figure 2, organic C abruptly increased from 0.7% in the 90-118 cm layer to 1.07% in the 150-200 cm layer of ELM1, 0.11% in the 42-57 cm layer to 0.16% in the 88–106 cm layer of ELM2, 0.2% in the 21-37 cm layer to 0.73% in the 79-149 cm layer of ODN2, 0.31% in the 25-41 cm layer to 0.99% in the 41-48 cm layer and 0.31% in the 48-56 cm layer to 0.86% in the 122-200 cm layer of ODN3, 0.21% in the 55-140 cm layer to 1.04% in the 150-200 cm layer of TFN2 and 0.30% in the 38-52 cm layer to 0.68% in the 52-69 cm layer of TFN3 which probably indicated heterogeneity.

#### CONCLUSIONS

The floodplain soils of the Lower Niger River and the two major tributaries (Nun and Forcados) showed some degree of differences in morphological, physical, chemical and mineralogical characteristics, as well as heterogeneity of parent materials. The source of parent materials and degree of hydromorphism, being the major factors moulding morphogenesis. Parent materials were of mixed origin and the soils were at an initial stage of development. Seasonal inundation by the flood water and dryness in the dry season set the stage for alternate oxidation and reduction, providing the most distinguishing feature of the pedochemical environment. Subsurface gravization was a notable morphometric feature and gleization, a major soil forming process. The SMUs were dominated by quartz and low activity clays, kaolinization, being the dominant clay mineral forming process. The study, however, found out that flooding, wetness and soil fertility are major constraints to agricultural intensification that must be addressed should government desire achievable results for sustainable crop production. The findings therefore, provides a robust and reliable guide for the Bayelsa State government to draw effective agricultural policies not only for sustainable crop production ventures within these affected areas but for the state at large as soils of the entire state are likely to have similar parent materials. Further morphological and mineralogical studies are therefore recommended on a larger scale so as to provide both State and the Federal governments the needed information for proper agricultural intensification programmes.

#### DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

#### CONFLICT OF INTEREST

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

#### **ETHICS**

There are no ethical issues with the publication of this manuscript.

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### Waste management practices towards low carbon cities

#### Ümmü Ayca BİLGİ\*<sup>®</sup>, Ece Ümmü DEVECİ<sup>®</sup>

Department of Environmental Engineering, Niğde Ömer Halisdemir University, Niğde, Türkiye

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#### ABSTRACT

Urbanization has increased rapidly in the world. It uses an intensive resource consumption for urbanization and the expanding economy. The consequences of unconscious consumption of these resources, which constitute the thermodynamic system of cities, are solid wastes, wastewater and air pollution. As a result of improper management of these wastes in cities, it increases the effect of climate change by producing greenhouse gas emissions directly or indirectly. The most innovative approach in controlling these effects is "low-carbon city" studies. Turkey should give priority to low-carbon city applications that comply with the European Green Deal and transition to a green economy. In this context, with this study, applications and solutions were examined together to become a low-carbon city in Turkey. Considering policies and commitments, the roadmap is drawn in this way, the study will be a reference for many cities.

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#### INTRODUCTION

It has become a living area with intense immigration and a crowded population with the increase in the need for manpower of the industry and the service sector [1]. Cities have the capacity to support their population. When this capacity is exceeded, various problems occur for the city and its inhabitants. The growth of cities is giving way to inefficient layouts that consume large amounts of energy and water, technologies and significant amounts of waste problems [2]. In cities that do not have waste management, many problems arise such as leaving the wastes in various areas with wild storage, decreasing the air quality due to the fossil fuels used in the city, and climate change due to the increase in greenhouse gases [3]. In order to find solutions to these problems, new urban concepts emerged which have been on the agenda in recent years and on which studies have increased [4].

- a. Sustainable cities,
- b. Ecological cities,
- c. Low carbon cities,
- d. Livable cities,
- e. Smart cities,
- f. Digital cities.

Although these terms and their meanings are different, their goals are the same. Each approach aims to find solutions to the city's problems.

\*Corresponding author.

<sup>\*</sup>E-mail address: aycabilgii@gmail.com



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As centers of population, industry, transport and infrastructure, cities are responsible for 70% of global CO<sub>2</sub> emissions. With the gradual acceleration of urbanization and industrialization processes, a series of environmental regulations are needed to reduce greenhouse gas emissions in world cities [5]. The biggest cause of global warming has been identified as the increase in carbon dioxide emissions. This situation has drawn attention to carbon emissions and a low-carbon city strategy has been developed as an effective reduction method [6]. Low carbon city; It is a concept that will reduce the carbon footprint by providing the increasing energy need in cities from renewable energy sources and help increase the quality of life with high efficiency [4]. Low carbon city; it is a concept that focuses on the amount of carbon in order to prevent climate change, reduce greenhouse gas emissions and continue the carbon cycle in the atmosphere. This focal point is included in the solutions of the city's problems, and the contribution of the solutions to the carbon cycle is important. The concept of a low-carbon city differs from other definitions in this respect.

#### The Latest Status of Waste Management

According to the IPCC 1.5 degree report, they stated that sectors such as energy, transportation, food and waste should switch to low carbon practices in order to take the right steps in the fight against climate change. With the increase in the urban population, the growth in every area causes an increase in the amount of waste produced [7]. Integrated waste management has an important place in the formation of low-carbon cities. This has economic, environmental and social consequences. Inaccuracies and deficiencies in the management of waste negatively affect the environment and constitute 3% of the total greenhouse gas emissions in the atmosphere [8]. Waste management; it is to minimize the effect of the elimination of wastes generated in the system on the environment and economy [9]. The waste management hierarchy has been accepted as in Figure 1 as the opposite of the waste hierarchy that was given and taught before [10]. According to this figure; the importance of waste management hierarchy to prevent waste; by devoting the largest part of the pyramid to waste prevention, it supports savings and emphasizes the minimization of the materials used. In order for this pyramid to function well, first of all, the materials considered as waste should be reduced, and then the benefits that can be obtained from the waste such as reuse, recycling and recovery of the waste ones should be maximized. In the end, it is desired to minimize the amount of waste that will come to the last item by including the disposal item [11]. While in the previous strategy, most of the waste was sent to landfills as disposal, in the new strategy, minimal disposal is accepted in accordance with the zero-waste philosophy.

Turkey now accepts this newly created waste hierarchy and puts the necessary legal regulations into practice. The legislation required for waste management in Turkey is shown in Figure 2.



Figure 1. Waste management hierarchy [10].

The Waste Management Regulation published on 02.04.2015 regulates waste management. Among the purposes of this regulation; ensuring the management of wastes from generation to disposal without harming the environment and human health, reducing waste generation, reducing the use of natural resources and ensuring waste management through ways such as reuse, recycling and recovery of wastes [12].

"National Waste Management and Action Plan" was prepared in Turkey in 2015 and this plan, the current state of Turkey's current waste management and the elements that need to be improved and developed by determining the current situation were determined, and investments in waste management and 2023 targets were set [13].

The Zero Waste Regulation, which started to work in our country in 2017, entered into force in 2019. The implementation of the Zero Waste Principle, which includes protecting our resources, preventing waste, reducing the amount of waste, collecting waste separately at the source and recycling waste, is very important in terms of leaving a clean and livable world to future generations [14].

Although the regulations issued for Turkey, which tends to bury more than 65% of its wastes under the ground, show the efficient management of waste, there is a long way to go in waste management practices. Particularly, the problems experienced by municipalities in reaching the households, the increase in street collectors over time and the inefficient development of public awareness on this issue make the management of waste difficult. The Turkish economy, similar to other developing and developed economies, operates on a linear buy-use-dispose resource model that generates significant amounts of waste. It is inevitable that over the years, Turkey's economic and population growth, the increase in resource and energy use, and the waste generation are pending a solution. Turkey is facing the risk of raw material shortages to meet the production and consumption demand, together with the difficulties in waste disposal with increasing urbanization. Similar trends apply on a global scale.



Figure 2. Waste legislation.

Within the scope of the Eleventh Development Plan (2019– 2023), waste management under the title of "Livable Cities, Sustainable Environment", provided detailed information. In addition, statements are included "Public awareness will be raised in the recycling of solid wastes. Zero Waste Project practices will be expanded. Technical standards for the recovered secondary product will be developed, incentive and guidance legislation will be improved. A separate collection system of wastes will be expanded." In addition, it was stated in the Eleventh Development Plan that the domestic solid waste recycling and disposal facility projects and the transfer station projects of the local governments with insufficient financial power would be financially supported. In this context, 2023 targets are also included in the plan [15].

Greenhouse gas emissions and climate change are also mentioned in the Eleventh Development Plan. It has been stated that studies will be carried out within the framework of the Intended National Contribution for the emission control of the sectors that cause greenhouse gas emissions. It has been stated that in order to adapt to climate change, regional and city-based needs and solutions will be determined according to these needs, and Climate Change Action Plans will be prepared for this purpose [15]. Inadequate waste management; It supports global warming by producing methane gas, and the leachate flowing from wild storage areas causes pollution of underground and surface waters and endanger human health [16]. Waste production globally has been determined to have increased from 635 million tons in 1965 to 1999 million tons in 2015 and is estimated using modeling techniques to reach 3539 million tons by 2050 [17]. In the studies carried out, 1385 million tons of CO<sub>2</sub> emission value obtained from the solid waste collected in the city in 2018 alone constitutes 3.7% of the global CO<sub>2</sub> emissions of that year [17].

In Turkey, the total greenhouse gas emissions from the waste sector in 2016 are 16.2 million tons of  $CO_2$  equivalent and constitute 3.3% of the total greenhouse gas emissions. There is an increase of 45.9% in greenhouse gas emissions from waste compared to 1990. As seen in Figure 3, the main source of greenhouse gas emissions from the waste sector is solid waste disposal and wastewater discharge and treatment. In 2016, 72.2% of the greenhouse gas emissions from waste were caused by solid waste disposal and 27.7% by wastewater discharge and treatment [18].



Figure 3. Greenhouse gas emissions from the waste sector, 1990–2016 [18].

Depending on the population growth, waste disposal in regular waste storage and disposal sites is also increasing. As seen in Figure 4, approximately 28.5 million tons of waste was dumped in the landfill in 2015 and 68% of this amount was stored in managed landfills [18].

As the amount of waste sent to the landfills increases, the amount of waste stored in the wild decreases and the greenhouse gas emission rate from the landfills decreases [18]. Solid waste has many negative environmental effects, including increasing greenhouse gas emissions and nitrogen pollution [17]. The rapid increase in the amount of waste and its impact on climate change show that waste management will become more important in the future [19]. Intelligent waste management by efficient application and energy saving; It prevents the formation of gases such as carbon dioxide in the air that can cause air pollution and global warming, making the city and nature more livable and cleaner [4].

# Mitigation and Adaptation in Climate Change and Waste Management

In order to contribute to global efforts in the fight against climate change in Turkey, changes and transformations are observed in legal, institutional and policy frameworks. This transformation begins with the establishment of the Climate Change Coordination Board (CCCB) in 2001, with the responsibility of coordinating Turkey's efforts on climate change. In 2013, its name was changed to the Climate Change and Air Management Coordination Board (CCAMCB), and its scope and responsibilities were expanded. The Board is responsible for the coordination of the activities under the United Nations Framework Convention on Climate Change (UNFCCC) and the protocols subject to these agreements and the relevant national policy and legislation [18]. Turkey created the National Climate Change Strategy in 2010 and the National Climate Change Action Plan in 2011. The Regulation on the Monitoring of Greenhouse Gas Emissions was published in the Official Gazette dated 17 May 2014 and numbered 29003 and entered into force. Apart from these, it has been a side too and contributed to international agreements and protocols. The signed or signed agreements/protocols are listed in Table 1 chronologically.

Studies to be carried out to reduce greenhouse gas emissions from waste in Turkey are included in the Intended Nationally Determined Contribution document submitted in 2015 for the Paris Climate Agreement. This document is a document put forward by our becoming a party to the Paris Climate Agreement. In addition, the practices made for the reduction committed by Turkey on behalf of the waste sector are given below. In order for these commitments to be fulfilled and everything not to remain on the writing, it is necessary to carry out various applications. It is clearly seen that Turkey has determined strategies beyond the tasks it has undertaken with it becoming a party to the agreement.

Another important development for Turkey is that it ceased to be a party to the Paris Climate Agreement on 5 October 2021 and was ratified with parliamentary approval, which was signed by many countries in 2015. In this way, while paving the way for green development, the number of tasks to be done has increased while the way for additional budgets has been opened due to the implementation and implementation of many climate-compatible studies. These increases will enable the opening of new business lines.

With the approval of the Paris Climate Agreement, it is important that the public correctly understands this agreement. Therefore, in the fight against climate change, the adaptation of people's lifestyles to the climate comes first. For this, municipalities take the first place in adapting living spaces to climate. Efficient collection and recycling of waste have an important role in reducing a city's



Figure 4. Annual waste disposal at solid waste disposal sites, 1990–2015 [18].

Table	1. Agreements/	protocols related	d to the reducti	ion of greenho	ouse gas emissions	in Turkey	/ [20	1
	<i>(</i> )				<i>(</i> )			

Agreement/Protocol	Acceptance - Effective date	Turkey's Situation
Montreal Protocol on Substances that Deplete the Ozone Layer	Accepted D.: 1987 Effective D.: 1 January 2019	It became a party on 19 December 1991 and accepted all the amendments introduced by the Protocol.
United Nations Framework Convention on Climate Change (UNFCCC)	Accepted D.: 3-14 June 1992 Effective D.: 21 Mart 1994	Joined on 24 May 2004. Turkey is the only country within the scope of Annex-1 that does not have a transition economy and whose "special conditions" were accepted by the resolutions of the Conference of the Parties.
Kyoto Protocol	Accepted D.: 1997 Effective D.: 16 February 2005	It became a party on 26 August 2009. Turkey, which was not a party to the UNFCCC when the Protocol was adopted, was not included in the Annex-B list of the Protocol. Therefore, there is no quantified emission limitation/reduction commitment.
Paris Climate Agreement	Accepted D.: 5 October 2016 Effective D.: 4 November 2016	It was signed on April 22, 2016, but it was not a party. The National Intended Statement of Contribution was submitted on 30 September 2015. According to Turkey's national contribution statement, a reduction of 18% to 21% is foreseen in 2030 compared to the increase in greenhouse gas emissions compared to the reference scenario.
Paris Climate Agreement	Approval D.: 5 October 2021	It was unanimously approved by the parliament. A zero emission target has been set for 2053.

carbon footprint and  $CO_2$  emissions. For this reason, it is necessary for municipalities to carry out many studies and to prepare training programs in order to increase public awareness. It is necessary to carry out intensive practices in order to convey environmental awareness within social activities, to create public service announcements about recycling and to ensure that everything is not at the educational level. Giving priority to encouraging ways in practices has an important place in ensuring public participation.

The items regarding the waste sector of the Intended Nationally Determined Contribution, which Turkey has committed with our becoming a party to the Paris Climate Agreement, are examined in detail below.

#### · Sending solid wastes to landfills

There were 15 landfills in Turkey until 2003, and there were 89 facilities in 2019. According to the data of the Ministry of Environment and Urbanization for 2019, the ratio of the population served by landfill facilities to the total municipal population is 82%. This situation has been determined by increasing the proportion of the population provided with waste disposal service to 100% by 2023 as a near target, and by spreading the zero-waste management system as a distant target by 2050. In this way, the landfills will be completed gradually, wild landfills will be prevented and projects will be put into effect upon sending less waste [21].

• Reuse, recycling and other processes to obtain secondary raw materials, to be used as an energy source or to be disposed of the waste The zero-waste management system, which started with the separation of wastes at the source, has made Waste Derived Fuel (WDF) facilities widespread in the waste industry. Waste-derived fuel (WDF) is an alternative solid fuel type that takes the remaining non-recyclable materials as raw materials after the recyclable materials (plastic, glass, metal, etc.) of domestic or industrial solid wastes are separated. End-of-life vehicle tires, wastepaper, waste oils, wood waste, treatment sludge, plastics are WDF raw materials. When WDF and coal are compared, WDF; It is advantageous because it is cleaner, more energy efficient, environmentally friendly and less dusty [22].

• Reducing the amount of biodegradable waste to be stored in 2015 to 75% by weight of the total amount of biodegradable waste produced in 2005, 50% in 2018 and 35% in 2025. Accordingly, the establishment of pre-treatment facilities in order to reduce the amount of biodegradable waste to be stored in 2025 to 35%,

The final product of organic wastes, which is subjected to microbial degradation (decomposition) under aerobic or anaerobic conditions, is called compost [23]. Green garden wastes (leaves, grass, pruning wastes, wood, bark) resulting from the works of the parks, gardens and afforestation units of the municipalities can be converted into compost with the appropriate composting method. Social responsibility projects on compost production have started to become widespread in Turkey [24].

- Ensuring energy recovery from waste by subjecting wastes to processes such as material recovery, bio-drying, biomethanization, compost, advanced thermal processes or incineration,
- Realization of methane recovery from landfill gas originating from regular and irregular landfills,

Methane gases, which cause an increase in greenhouse gas emissions 25 times more than carbon dioxide, are converted into electrical energy by burning after they are collected from the gas collection channels opened in the landfills [25]. Between January and February 2021, 96,358 megawatt-hours of electricity were produced by recovering methane from landfill gas. 52% of this amount was obtained from Odayeri Landfill Gas Power Plant and Seymen Landfill Gas Power Generation Facility [26].

• The industrial symbiosis approach that allows the wastes from industry to be used as alternative raw materials or fuel in another sector, and the waste from one sector to be the raw material of another sector,

Industrial Symbiosis is the union formed by turning the by-product or output of one of the industrial facilities into the raw material of another. Iskenderun Bay (Adana, Mersin, Osmaniye, Iskenderun) Industrial Symbiosis Project; It was carried out by the Technology Development Foundation of Turkey (TTGV) between 2011–2014 [27]. In this project, industrial production on animal feed production from fruit pulp, energy production from agricultural and animal waste, bioremediation product production from cottonseed waste, electricity production from waste oil, granule production from end-of-life tires, lead recovery from scrap batteries and the use of slag from iron and steel production in road construction, symbiosis studies were carried out [28].

 Carrying out appropriate studies for the evaluation of wastes from livestock and poultry farms,

Animal waste is seen as an ideal source for biogas (65% CH4, 35% CO<sub>2</sub>) production. The obtained biogas is an important energy input for electricity and heat production. In addition, manure, which is a by-product, is used in agriculture [29]. Establishing facilities such as biogas and biodiesel for the evaluation of animal waste in the world creates a sustainable cycle by preventing the damage of wastes to the environment and human health [30]. In addition, the establishment of meat and bone meal production facilities for the evaluation of animal waste and by-products occurring in slaughterhouses is among the studies evaluated [31].

Rehabilitation of irregular landfills and ensuring the disposal of wastes in landfills,

Rehabilitation (rehabilitation) by closing the areas where irregular storage is made has been legalized with the "Regulation on Regular Storage of Wastes", which was published in the Official Gazette dated 26.03.2010 and numbered 27533 [32].

In order to fulfill the commitments made regarding the waste sector, the above-mentioned studies are carried out and success is achieved in many areas. It is envisaged that the commitments in all titles will be completed by 2030. In addition to all these studies, strategies are being developed by further studies on waste.

The Paris Climate Agreement, which was signed in 2015 on the mitigation, adaptation and financing of climate change within the scope of the United Nations Framework Convention on Climate Change, entered into force in 2016. Among the countries that did not ratify the agreement, such as Eritrea, Iran, Iraq, Libya, Turkey and Yemen, Turkey was the 191<sup>st</sup> country to sign the agreement as of 2021. Paris Climate Agreement; It sheds light on what should be done to reduce greenhouse gases that cause global warming and climate change in the world. It aims to limit global warming to 1.5°C. It presents a perspective that aims to reduce greenhouse gas emissions from coal, oil or all other fossil fuels. The living now sees the problem of climate change. For a solution, it is necessary to switch to climate change adaptation policies. The difference of the Paris Climate Agreement from other agreements; The country's need to align its policies, economy and industry with climate change. The Ministry of Commerce, the Ministry of Industry and Technology, metropolitan municipalities and local governments will take quick steps, which will ensure that Turkey is less affected by the provisions of the Green Agreement and the Paris Climate Agreement. A resource is created to transfer money from developed countries to poor countries in order to reduce greenhouse gas emissions and adapt to climate change. An annual budget of around 100 billion dollars is envisaged for this resource, which is called the green climate fund.

#### Waste Management for Low Carbon Cities

Data collection and analysis are a priority for creating low-carbon city strategies [33]. In order to find solutions to the problems, there must be data belonging to the problems. Finding the most appropriate solution over these data is also possible with data analysis. Data analysis is based on organizing the collected data with various methods and eliminating unnecessary, redundant data that will affect the accuracy of the result. When the data analysis is done correctly, it will highlight the appropriate solution [34]. A few of the waste management practices made in Europe in order to achieve the goal of becoming a low-carbon city are listed below.

CO<sub>2</sub> emissions; mostly from the use of fossil fuels or from industrial processes. Renewable energy sources should be used to prevent CO<sub>2</sub> formation. For industrial processes where we cannot prevent CO<sub>2</sub> formation, it is necessary to prevent CO<sub>2</sub> emissions and to ensure CO<sub>2</sub> recovery. Carbon dioxide capture and storage processes are applied to the places where carbon dioxide is produced the most and consist of three stages. These are handling, transport and storage. The process of separating CO<sub>2</sub> from the flue gas formed during the capture phase is carried out. In the transportation phase, there are two options as pipeline or ship transportation. The storage stage is the process of injecting carbon dioxide into an underground rock formation by compressing it. In the storage phase, there are three alternatives: geological storage, oceanic storage and mineral carbonization [35].

One of the most used methods to reduce carbon emissions in the waste sector is biogas plants. Established biogas facilities both meet the city's heat and natural gas needs as a renewable resource and provide fertilizer support to be used in agricultural lands. This method prevents the landfills from exceeding their capacity, thus contributing to the storage of less waste and making fewer landfills and economically contributing to the country [36].

Cities that want to reduce their carbon emissions through waste management focus on new projects by making use of technology. The use of smart waste management systems and the collection of waste with electric vehicles reduce carbon emissions in both the transportation and waste management sectors. Routes created automatically according to the occupancy rate of the waste bins prevent unnecessary vehicle circulation and carbon emissions in the city [11]. The implementations on the so-called Climate Street in Amsterdam demonstrate the contribution of waste management to the low-carbon city. The collection of waste from the city with electrical waste separation vehicles prevents the carbon emissions emitted by the waste vehicles [37]. In addition, another exemplary application seen on the same street is energy-saving waste bins working with solar energy. By compressing the waste with solar energy, it has more waste capacity than a normal waste bin. Thus, the filling time of the waste bins is extended and the working time of the waste vehicles is shortened [38].

In Songdo city of South Korea, the system where waste is sent directly from waste bins to the recycling facility, energy generation or underground waste facility for incineration with pneumatic tube systems is one of the applications where waste management and carbon emissions are minimized. There are no waste and waste vehicles in the city [39].

The Relationship of Waste Management with Carbon Tax Another method applied to reduce carbon emissions and raise awareness is the carbon tax. It is a consumption tax per ton of coal, per barrel of oil or emissions. Although it is a method that has been applied globally for years, it was first put into effect by Finland in 1990 [40].

According to the Evaluation of Market-Based Emission Reduction Policy Options in Turkey Final Report published by the Ministry of Environment and Urbanization in 2017, priority areas in emission reduction were determined and studies to be carried out on these areas were compiled. It has been stated that the energy, process emissions and transportation sectors will be subject to tax at the first stage, while the agriculture and waste sectors will be exempted from the tax at the first stage. It is stated that it is exempted due to the complexity of emission measurement in the waste and land-use sectors and the need to further develop appropriate methodologies [41].

#### CONCLUSIONS

The increase in human needs with industrialization in cities has led to population density in cities. The fact that the cities are not ready for such a population leads to the fact that every person living in the city cannot be provided with the same quality of life. Green areas are given up for housing rights and trees are given up for road construction. The lack of adequate infrastructure reveals the most concrete evidence of environmental pollution in rapidly growing cities. Industrialization, which is not controlled against the environment and human health, has ceased to be a problem only for cities. In recent years, the whole world has started to see how big problems climate change has caused and will cause. It is understood that the absorption of carbon emissions should be provided by human-made solutions. Reducing carbon emissions, which is the starting point of low-carbon cities, can be achieved by reaching all cities. The new regulations that emerged with the Paris Climate Agreement, where the green economy is related to waste management, and these new regulations need to be pointed out in order to reach low-carbon cities.

The legal regulations made by the government in the waste sector in Turkey were examined and the signed/approved agreements/contracts on carbon reduction were listed and it was revealed how much of the commitments were made. It is seen that feasibility studies and legal regulations regarding the reduction of carbon emissions are about to be completed. It is seen that the practices that still need to be done in the waste sector for the reduction at the level committed with the Paris Climate Agreement do not cover 100% of the population in the country. Carbon reduction methods for the waste sector and the work that can be done on behalf of other sectors are explained with examples from abroad. While determining its strategies for the future, Turkey needs to closely follow and realize the changes in the world. This work; forms the basis for many studies on low-carbon cities.

#### DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

#### CONFLICT OF INTEREST

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

#### ETHICS

There are no ethical issues with the publication of this manuscript.

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## The effect of atmospheric deposition on potassium accumulation in several tree species as a biomonitor

Kaan IŞINKARALAR<sup>\*1</sup>, Ramazan ERDEM<sup>2</sup>

<sup>1</sup>Department of Environmental Engineering, Kastamonu University, Kastamonu, Türkiye <sup>2</sup>Kastamonu University Forestry and Forestry Products, Araç Rafet Vergili Vocational School, Kastamonu, Türkiye

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#### ABSTRACT

Minimizing air, water, and soil pollution are very important for a sustainable environment. Particularly, ensuring the continuity of soil fertility without deteriorating the soil structure is very important. This objective can be achieved only by determining the physical, biological, and chemical properties of atmospheric deposition and taking the required measures in agricultural lands. Trees and plants reflect the soil quality and especially they take both beneficial and harmful materials in their bodies owing to Saharan dust and using fossil fuel. Among them, nutrient elements have specific importance since it was determined that many factors including texture, irrigation method, organic matter, lime concentration, plant species and age, pH, and ion balance play effective roles in the growth or degradation of plants' productivity. Being one of the major nutrient elements taken by plants, potassium (K<sup>+</sup>) is of vital importance for trees and plants. Its concentration, which varies depending on the species of plant, is influenced by the mutual interaction between tree development and environmental/genetic factors. The scope of this study was to evaluate and rank the contribution of atmospheric potassium (K<sup>+</sup>) deposition flows to organs of *Robinia pseudoacacia* L., *Cupressus arizonica* G., and *Platanus orientalis* L. trees as biomonitors.

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#### INTRODUCTION

Rapidly increasing world population and consequent increase in urbanization, industrialization, and unawares use of agricultural lands resulted in deterioration of many forests and fertile agricultural lands [1–3]. Minimizing air, water, and soil pollution is very important for a sustainable environment [4]. Particularly, ensuring the continuity of the urban environment without deteriorating the air and soil structure is very important for a sustainable environment [5]. Although the quality of the soil belongs to the region where it is located, atmospheric precipitation plays a role in global cases [6]. This is dust transport, emissions from nearby inert sources, and acid rain, which is formed as a result of the reactions of pollutants released due to the use of fossil fuels and their effects on plants [7]. Atmospheric potassium deposition can affect the physical, biological, and chemical properties of soil and take the required measures in agricultural lands via wet, dry, and total deposition [8]. Thus, plants need fundamental

\*Corresponding author.

\*E-mail address: kisinkaralar@kastamonu.edu.tr



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nutrient elements from the soil to grow and to maintain their metabolic activities. The potassium ( $K^+$ ) is one of the most abundant elements that represent the amount of potassium defined in a spatial and temporal [9]. It is a nutrient influencing many biochemical and physiological processes [10]. It founds in dust and rainfall that likely comes from terrestrial sources that can quickly be re-deposited or be transported for large distances [11].

The K<sup>+</sup> is a fundamental nutrient allowing plants to survive various biotic and abiotic stresses (diseases, insects, drought, salinity, cold, frost, and flood) [12]. It is important for developmental crops and activation enzymes [13]. It was stated in previous studies that each plant has various concentration values and different levels of need for K element that is the vital impact for enzyme activation, protein synthesis, photosynthesis, energy transfer, stoma movement, osmoregulation, phloem transfer, and cation-anion balance [14]. The most well-known source of K in the soil minerals such as feldspar and mica although it is found in dust and rainfall likely comes from terrestrial sources [15]. The effects of differences in presence of K in different plants on species' anatomy, morphology, and plant metabolism couldn't be explained yet and they are still debated [16]. It was shown that the presence of K allowed the trees to overcome the stress conditions. In previous studies, it was reported that trees and plants primarily need mineral nutrients to overcome biotic and abiotic stress conditions [17–19]. The presence of K in fertilizers used in herbal production increases the quality of fertilizer. It was determined in many studies that the second-most important nutrient for plant development is potassium, following nitrogen. Previous studies showed that plant fertilizer has a positive correlation with crop development and quality. However, potassium deficiency can also be seen in some cases. Potassium deficiency is observed especially in acidic soils with ample water and high salinity levels [20]. Several negativities are observed in the development of plants and trees in such soils. Intake of potassium by plants and trees occurs via roots and the intake levels vary [21]. On the contrary other soil nutrients, K is available in the soil as only a K<sup>+</sup>, which supports the plant productivity and environmental services, depending on the deficiency and abundance of nutrients [22]. Besides that, the optimum benefit can be obtained from the fertilization process, which is performed by having accurate knowledge about the chemical and physical properties [23]. For this purpose, many studies were carried out to determine the productivity levels of different regions and soils, to foreknow the potential nutritional problems, and to increase the crop quality [24]. The K<sup>+</sup> is one of the elements used by the plants in the soil at a higher concentration in comparison to the others [25]. Although there are many studies about the accumulation, transport, and levels of The K<sup>+</sup> is one of the elements used by the plants in the soil at higher concentrations of potassium, they could

not calculate the amount of potassium that passes from the atmosphere to the soil [26]. For the determination of the available potassium amount, inputs (atmospheric deposition, plant residue, and animal manures, commercial fertilizers, minerals, etc.) and outputs (removal of plant, leaching, erosion, fixation, etc.) of potassium sources should be well known as ecosystem services [27, 28].

Plants and trees are great bioindicators of atmospheric metals deposition including trace and toxic metals due to their effective adsorption capacity of them [29, 30]. The main sources of potassium and other elements are atmospheric inputs as wet, dry, and total deposition [31]. There are also some studies on the atmospheric trace metals transport and deposition on plants from anthropogenic sources [32-35]. The amount of potassium is increased with acid rain and precipitation because trace metals and other pollutants oxides form chemical compounds on terrestrial ecosystems [36]. Biomonitoring with plants ensures inexpensive knowledge on the composition and quantity of the deposition of trace and toxic metals [37]. In this study, the atmospheric potassium deposition was examined in the organs of trees for years. The K<sup>+</sup> concentrations were inspected in the rings of 3 different trees growing in the Kocaeli industrial zone via years and organs. The organs of trees used for this purpose Robinia pseudoacacia L., Cupressus arizonica G., and Platanus orientalis L. were selected because they are widespread on the terrestrial ecosystem for resistance to air pollution. The concentrations of K<sup>+</sup> were analyzed in the outer bark, inner bark, and wood fractions of trees. It was aimed to determine how the K<sup>+</sup> concentration in the annual rings of trees changed over the years.

#### MATERIALS AND METHODS

#### **Study Area and Sampling Site**

The study area is located around the organized zone of Kocaeli city, Türkiye. Significant factors of air pollutants are released from many industrial activities and fuel combustion. Air pollutants may contain heavy metals and toxic elements including macronutrients. Organs (wood, inner bark, and outer bark) of four dominant tree species were used *Robinia pseudoacacia* L., *Cupressus arizonica* G., and *Platanus orientalis* L. in an industrial area and were collected no more than 4 km from the site of the organized zone. All samples were taken from the main trunk of trees. After sampling, wood, inner bark, and outer bark were urgently placed in a glass vessel and transferred to the lab for preparation of analysis.

#### **Preparation of Tree Species Sampling**

The tree rings were determined to be 30 years old (between 1991 and 2020) in three years. The species had an almost same trunk diameter and height around 1 m above the ground. All samples were rinsed with acetone (Merck, Ger-

		Species		
Organ	Robinia pseudoacacia L.	Cupressus arizonica G.	Platanus orientalis L.	F value
Wood	776.7 Aa	472 Aa	1439.2 Ba	15.8*
Inner bark	1976 Ba	783.9 Aa	6187.1 Cb	48537.8*
Outer bark	3216.8 Bb	1909.8 Ab	6458 Cb	50096.1*
F value	12.2*	36.2*	160.2*	

**Table 1.** Change of K<sup>+</sup> concentrations (ppm) based on species

\*: Significant at p<0.001. Upper and lower letters differ significantly based on the Duncan test.

many) then they were divided into groups for the age ranges. All samples taken into glass vessels were kept at 50°C for 7 days. The samples were taken as 0.5 g weighed and 6 mL of 65% nitric acid (HNO<sub>3</sub>) and 2 mL of 30% hydrogen peroxide ( $H_2O_2$ ) were added to glass vessels. According to USEPA 3052 Method, the combustion process was carried out in the microwave oven at 200°C for 15 minutes [38]. The resulting samples were made up to 50 mL with ultrapure water and potassium analyzes were made in by Inductively Coupled Plasma-Optical Emission Spectrometry (ICP-OES) with a plasma source device (SpectroBlue, Spectro). Analytical grade chemicals were used for the research.

#### **Statistical Analyses**

All measurements were repeated as in triplicate. Analysis of variance (ANOVA) and Duncan test was conducted to identify the significance of atmospheric potassium deposition in species by using the SPSS 22.0 statistical package program for Windows.

#### RESULTS

The biomonitoring organ of plants was chosen outer bark, inner bark, and wood of a *Robinia pseudoacacia* L., *Cupressus arizonica* G., and *Platanus orientalis* L. which, due to its widely used and readily available in local terrestrial ecosystem. It has been proven that it can provide information on the presence of the K<sup>+</sup> element in Table 1.

According to the results of variance analysis (ANOVA) that the change in the concentration of  $K^+$  element on an organ basis in all three species is statistically significant (p<0.001). Considering the Duncan test results, the lowest values are obtained in the wood and the highest values were obtained in the outer bark, the values obtained in the wood and inner bark of *Platanus orientalis* L. are in the same group, and in other species, each organ formed a separate group. Considering the Duncan test results, the inner bark and wood were in the same groups in *Robinia pseudoacacia* L. and *Cupressus arizonica* G. also, the inner bark and the outer bark of *Platanus orientalis* L. were in the same groups. It is noteworthy that the values obtained in the outer bark are many times higher than the values obtained in the inner bark and wood in all three species. The lowest value in the outer bark is obtained in *Cupressus arizonica* G. with 1909.8 ppm, the highest value is obtained in *Robinia pseudoaca-cia* L. with 3216.8 ppm, the highest value in the inner bark is obtained in *Platanus orientalis* L. with 6187.1 ppm, and the lowest value is obtained in *Cupressus arizonica* G. with 783.9 ppm. In the wood part, the lowest value is obtained in *Cupressus arizonica* G. with 783.9 ppm. In the wood part, the lowest value is obtained in *Cupressus arizonica* G. with 472 ppm, and the highest value is obtained in *Platanus orientalis* L. with 1439.2 ppm. According to these results, it can be said that the lowest values are obtained in *Platanus orientalis* L. The change in the K<sup>+</sup> concentration in woods depending on the age range and direction is given in Table 2.

When the values showing the change of K element according to the age range are examined, it is seen that the highest value in Robinia pseudoacacia L. is obtained with 3520.9 ppm in 2018–2020, the lowest value with 379.9 ppm in 1994-1996, the lowest value in Cupressus arizonica G. in 1997-1999 with 256.8 ppm, the highest value is obtained in the years 2015-2017 with 778.5 ppm, the highest value in Platanus orientalis L. with 3069.2 ppm in the years 2018-2020, and the lowest value with 982.1 ppm in the years 1997-1999. According to the analysis of variance results, it is determined that the variation of K concentration depending on the species is statistically significant at least 99.9% confidence level (p<0.001) in all age ranges. When the values are examined, it is very difficult to say that the K<sup>+</sup> concentration changes regularly based on species or year. This situation can be interpreted as the change of K<sup>+</sup> concentration in plants does not change primarily depending on the species or year, and other factors are more dominant.

#### DISCUSSION

In regions, where four seasons are observed, it was determined that the development of trees increased in parallel with several intakes of potassium (K) in their bodies [39]. Regarding this point, providing data about the nutrient accumulation in trees, tree rings and organs can give important information about the chronology of the atmospheric elements deposition in its ecosystem [40]. Although there are

		Species		
Years	Robinia pseudoacacia L.	Cupressus arizonica G.	Platanus orientalis L.	F value
2018-2020	3520.9 Cg	762.4 Ag	3069.2 Bh	34256.8*
2015-2017	610.2 Af	778.5 Bh	2093.0 Cg	44693.2*
2012-2014	562.6 Ae	773.4 Bh	1688.0 Cf	11214.3*
2009-2011	404.4 Abc	637.2 Be	1160.8 Ce	8293.5*
2006-2008	487.2 Ad	722.8 Bf	1064.6 Cc	2267.6*
2003-2005	591.6 Bf	449.0 Ad	999.9 Cb	15820.5*
2000-2002	393.6 Bab	339.6 Ac	1100.3 Cd	4085.8*
1997–1999	396.1 Bab	256.8 Ab	982.1 Ca	7985.1*
1994-1996	379.9 Ba	123.6 Aa	1169.8 Ce	6713.1*
1991–1993	420.2 Bc	125.8 Aa	1064.2 Cc	28405.3*
F Value	22130.5*	8467.0*	12896.2*	

Table 2. The K<sup>+</sup> concentration (ppm) age interval and species change of in wood

\*: Significant at p<0.001. Upper and lower letters differ significantly based on the Duncan test.

studies on the usability of accumulation in the rings of trees, there are few studies on the transfer of elements between the organs of trees [41]. The changes in the concentration of K<sup>+</sup> in organs of 3 different trees grown in the intense industrial zone in Kocaeli province by organs. They were determined by making use of Robinia pseudoacacia L., Cupressus arizonica G., and Platanus orientalis L. in the present study. It was aimed to evaluate the level of atmospheric K<sup>+</sup> accumulation in the inner bark, outer bark, and wood segments of the tree and to interpret if the K<sup>+</sup> concentrations in rings and organs varied by year. In this study, the change of potassium concentration in barks and wood of three species was found to be 783.9 ppm of Cupressus arizonica G. but to be 6187.1 ppm in that of Platanus orientalis L. in the inner bark. Similarly, the K<sup>+</sup> concentration was found to be 472 ppm in Cupressus arizonica G. but 1439.2 ppm in Platanus orientalis L. in wood. Accordingly, it can be stated that K<sup>+</sup> concentration significantly varied between the species.

In previous studies carried out on this subject, it was determined that the concentrations of many elements significantly varied by the species [42]. The accumulation of elements within the bodies of plants is closely related to the plant habitus and development [43]. Plant development is shaped by the mutual interaction between genetic structure and environmental conditions [44]. Hence, the factors influencing the genetic structure of plants directly influence the intake and accumulation of elements in plants and, since different species have different genetic structures, it is normal for species to have different levels of element accumulation [45]. Another important result achieved in this study is that the concentration of potassium significantly varied by organs in all the species. Examining all three species, the highest values were found in the outer bark and the lowest values in the inner bark.

Many studies reported that the element accumulations differed between the organs of the same plant [46]. Moreover, it was also determined that the differences between the organs could reach very high levels [47]. In previous studies, the lowest concentrations were generally found in wood and the highest ones in outer bark [48]. The reason for higher element concentration in outer barks when compared to other organs was reported to mainly be the structure of the organ, particle matters, and the contamination of these particles by elements. It was determined in studies carried out before, it was determined that the elements in air adhered to the surface of particles and enriched them in elements, and these elements adhered to the plant organs and increased the element concentrations in these organs. The rough surface of the outer bark makes it easier for particles to hold on these surfaces [49].

The atmospheric potassium deposition significantly varied between the woods forming in subsequent years and the difference can reach very high levels. For instance, the potassium concentration in the woods of Robinia pseudoacacia L. forming from 2015 to 2017 was found to be 610.2 ppm, whereas it was 3520.9 ppm for the woods forming from 2018 to 2020. This finding suggests that the transfer of the K element in wood can be very limited. Among the species examined in this study, the lowest proportion of change was observed in Platanus orientalis L. and the highest one in Robinia pseudoacacia L. on atmospheric potassium deposition. It suggests that, among these species, the species that is most suitable for monitoring the change of potassium concentration is Robinia pseudoacacia L. in an industrial area. The knowledge about the transfer of elements between the organs is, however, very limited in plants.

#### CONCLUSIONS

Plants are the main indicator of urban environmental pollution and they fulfill many ecologic, economic, and social functions in ecosystem services. Development and phenotypic characters of plants are shaped by the genetic structure and edaphic factors such as soil's structure and nutrient content and soil, as well as climatic factors such as light, temperature, and atmospheric deposition-precipitation. Hence, to understand regional pollution which spatial distribution information links to existing or future policy with regards to sustainable environmental pollution including air and soil. Atmospheric deposition influences plant development, it is necessary to determine the intake and accumulation of elements directly from the surrounding environment. Atmospheric deposition contains quantities of K<sup>+</sup> which can significantly affect the structure and development of plants. Biomonitoring the deposition of atmospheric K<sup>+</sup> concentration is the crucial method to present levels of other metals and toxic metals in the territorial environment. The result of this study provides biomonitoring of atmospheric metals deposition on plants that tell the level of their accumulation. Future research on atmospheric other metals deposition for the suitability of some commonly found species for biomonitoring should be investigated the spatial and temporal variation of industrial and high traffic areas.

#### DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

#### **CONFLICT OF INTEREST**

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

#### ETHICS

There are no ethical issues with the publication of this manuscript.

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**Research Article** 

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Environmental Research & Technology

# Use of a convolution neural network for the classification of *E. Coli* and *V. Cholara* bacteria in wastewater

Tohid IRANI<sup>1</sup>, Hamid AMIRI<sup>\*2</sup>, Sama AZADI<sup>3</sup>, Mohsen BAYAT<sup>4</sup>, Hedieh DEYHIM<sup>1</sup>

<sup>1</sup>Department of Civil Engineering, Shiraz Payam Noor University, Shiraz, Iran <sup>2</sup>Department of Civil and Environmental Engineering, Tarbiat Modares University, Tehran, Iran <sup>3</sup>Department of Civil and Environmental Engineering, Ferdowsi University, Mashhad, Iran <sup>4</sup>Department of Electronic Engineering, National University of Ireland, Maynooth, Ireland

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#### ABSTRACT

Identifying the microbial population and type of them is a crucial measure in the water and wastewater treatment processes, reuse of wastewater, and sludge treatment system. Todays, manual methods are usually used to count and detect the type of bacteria in water and sewage laboratories which mostly suffer from human errors. This study aims at presenting an accurate method based on image analysis through the convolution neural network (CNN) to classify *Escherichia coli* (*E. coli*) and *Vibrio cholera* (*V. cholera*) bacteria, in wastewater. About 9,000 Red-Green-Blue (RGB) microscopic images of the sewage sample containing the stained bacteria were used as the input datasets. The results showed that the bacteria would be classified and counted with the accuracy of 93.01% and 97.0%, respectively. While CNN performed pretty well in counting the number of bacteria for both RGB and grayscale color models, its classification performance is only satisfactory in the RGB images. The sensitivity analysis of CNN illustrated that the Gaussian noise enhancement caused to the increment in the standard deviation ( $\sigma$ ) that proportionally decreased the CNN accuracy.

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#### INTRODUCTION

Contamination identification, efficiency evaluation of the wastewater treatment system, and reusability of treated wastewater are basic principles of the microbial monitoring in the contaminated water resources with human wastewater [1–4]. In fact, according to the world health organiza-

tion (WHO) guidelines, the main goal is preventing the water resources from contaminating by any type of pathogenic bacteria. One effective solution to this issue is the indicator microorganisms' contamination tests. For instance, existing the *E. coli* bacteria in the water samples has been considered as a fecal pollution indicator due to its ease and rapid detection [5, 6]. Traditionally, the manual method is used

\*Corresponding author.

\*E-mail address: hamid64amiri@gmail.com



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to count the number of cultured colonies to determine the amount of fecal pollution in water samples [7]. This method not only is time and energy consuming, but also suffers from the human errors [8]. Large number of samples as well as small size and overlapping on colonies, and diversity and large number of colonies formed in a sample are the most common reasons to make errors in this method [8–10].

Recently, employing the convolution neural network (CNN) along with the computer processing power has been playing an increasing role in the field of deep learning [11]. Some interesting applications of this method are in the environmental science, cancer detection, medical image processing. The interesting results in these applications are provided by the possibility of employing deeper layers in comparison with other artificial neural networks (ANN) [4, 12-15]. So far, CNN in measuring the concentration of cyanobacteria in water [16], pollution of the water distribution network [17], detection of water impurities [18], classification of urban wastewater microbeads [19], etc. have been studied. Akbarian Mymand et al. (2014), studied the feasibility of using an image processing method for the count of bacteria in the mixture of Quail flora, sourdough, and kefir drinks. They compared this method with colony counter method in the different dilution proportions. The results of this study indicated a significant difference between the numbers of counted bacterial in samples with low dilution proportion [20]. Huang and Wu (2018) classified the clinical bacterial colonies with different morphologies into 18 categories by a deep neural network (DNN)-based classifier. They obtained over 90% identification and classification accuracy of each bacterium category [21]. Yurtsever and Yurtsever (2018) achieved 89% accuracy to classify microbeads in the wastewater by CNN [19]. Shaily and Kala (2020) classified various shapes of bacterial particles in 20 different categories with more than 99% accuracy [22]. According to a previous study, CNN effectively improves the detection and counting accuracy of the bacterial particles [23].

In this study, by employing the capabilities of this method, we try to obtain a very high accurate classifier to identify the aforementioned bacteria in sewage. For this purpose, we design a two-part network which exploits the K-means and new data producing methods to prepare images which followed by a CNN to classify the data. First, the input images contain three components of RGB images with the resolution of 749x1000 pixels are used as the primary datasets for both bacteria. About 150 images per bacterium are considered as an initial number for the training set, which is increased to 9,000 to have more generalized dataset. One way of overfitting prevention is increasing the data which is obtained by some specific operations, such as adding Gaussian noise, rotating images at a 900 angle, upside down flip, left right flip and so on. Then, K-means clustering algorithm is used to image segmentation and masking. Finally, the prepared images are exploited as the input dataset to the CNN, and the reliability of this method is evaluated by the sensitivity analysis.



Figure 1. Main steps diageram.

#### METHOD AND MATERIAL

#### Overview

In this study, the E. coli and V. cholera bacteria image dataset is obtained from the central water and wastewater laboratory of Ardabil University of medical science. These images are 1000×749-pixel sized in the RGB color model. The main steps of this study are shown in Figure 1. K-means clustering algorithm is employed to sparse and keep only few meaningful features in the input images as an input unit for the next network (Fig. 2). This helps to have a clear separation between features which makes the classification more accurate. Furthermore, different techniques, such as image 90° rotation, upside-down flip, left-right flip, and Gaussian noise are used to generate new input images in the datasets for training and validation of CNN. Then, the input layer in the CNN will be a multidimensional array including 1000x749x3 units that will be employed to train the network for the bacteria classification and counting. It is worth noting that the applied images are in both RGB and Grayscale models that generalizes the proposed method for different color models.

#### **Bacteria Specifications**

*E. coli* is an anaerobic Gram-negative bacterium considered as an indicator to identify fecal pollution in the water resources. It is also applied to evaluate the performance of the disinfection system in the wastewater treatment plants and reusing effluent for irrigation. According to the WHO guideline, the allowable limit of this pollutant varies depending on the type of usage. For example the maximum acceptable level of *E. coli* for recreation water use and general irrigation, recommended less than 385 and 1000 MPN/100ml, respectively [24]. The morphology of *E. coli* similar to most of the gram-negative bacteria bacillus is in rod shape (Fig. 3).



Figure 2. K-means clustering algorithm.

*V. cholera* is a Gram-negative comma shape bacterium which has been widely distributed in the water environment [25, 26]. Figure 3 shows a *V. cholera* bacterium with a length and width of 2.7-3.5  $\mu$ m and 0.36-0.4  $\mu$ m, respectively. This bacterium can be transmitted to the human intestine through fruits and vegetables irrigated with water polluted with this bacterium.

#### **Image Preprocessing**

Generally, image is a matrix that pixels are its entries that the values of these entries show the intensity of each pixel in the image. Image processing includes methods to enhance the quality of an image, and search to extract the relevant information in order to the following analysis in an algorithm. Firstly, the quality of image is improved by filtering unwanted effects, such as noises, lights, etc. Noise capturing filters are mostly categorized from the simple average and median filters to the more complicated Gaussian and adaptive filters [27]. Next step is image segmentation to keep the most meaningful information in it. There are also several approaches to image segmentation and masking, such as K-means clustering, threshold algorithm, watershed algorithm, neural network, and deep learning methods [23]. Since this study is aimed on two categories of bacteria, preprocessed images of each category imported as the input data to the CNN. In this study, the gram-negative bacteria considered as red/violet-color cells. All images used in RGB (Red, Green, and Blue) and grayscale modes and operations repeated for each mode. MATLAB 2015 Image Processing Toolbox used for CNN operation while supported with the Visual Studio 2013.



Figure 3. E. coli and V. cholera bacteria morphology.

#### K-means Clustering

The approach of clustering is partitioning a set of data to some clusters. K-means clustering labels each pixel of the image to a cluster with the nearest mean or shortest distance to the center of that cluster. In fact, it finds the optimum point of data by minimizing the following error function.

$$\mathbf{J} = \sum_{i} D^2. \, (\mathbf{X}, C_i), \tag{1}$$

where D is the distance of  $C_i$  as the center of i<sup>th</sup> cluster in the data group X. Different distance measures, such as Euclidean and Manhattan distances can be exploited in the K-means clustering [28]. Generally, K-means clustering algorithm includes four following steps:

- 1- Inserting random means within the data.
- 2- Calculating distance of each data to the means and assigning the data to the nearest group.
- 3- Calculating the new mean of each group.
- 4- Repeating the steps 2 and 3 to converge to the target.

K-means here is applied for segmenting and masking bacteria shapes within the images. All images are segmented and masked with K-means clustering algorithm accurately. Particularly, each image is divided into two separated background and foreground parts which the former is bacteria-shaped area, and the latter is black-colored area. The bacteria-shaped area contains intensity and position of each pixel which will be used as the input of CNN for both categories of bacteria. Figure 4 shows image masking in the RGB and grayscale modes.

#### Convolution Neural Network (CNN)

Convolution neural network as a kind of feed forward neural network and deep learning method uses a Perceptron network with some changes on classic operation. This network is based on four characteristics which have



Figure 4. Masked and segmented bacteria images (RGB, grayscale).

been implied from natural signals. Therefore, CNN can be applied to the process of natural signals and multidimensional matrices. This characteristic includes neurons near communications, layer common weights, using numbers of layer and polling ability [29]. CNN formed in layers includes convolution layers, polling layer and fully connected layer and sub layers. Each layer connects to prier layers base on the weight. Non-linear functions and activation functions include sigmoid function or ReLU functions applied to import sum of neurons' weight in each layer. These layers extract near specification of images or in other situation continuity of common specifications. Maximum values of vicinity choice to reduce the dimension of data and summarize extracted image specifications. Each CNN contains layers for the convolution and polling operation. Layers exploit ReLU non-linear functions. Each weight of layers calculates by back propagation method [30]. Dense layer with the fully connected layer used for this study to extract specifications and assorting the data. Finally, the classified results of the dense layer are considered as the outputs. The CNN with seven major layers and sub-layers used for this study CNN structure, see Figure 5.

Equation 2 defines CNN output layer operation function (f)  $Y_{w,h,m} = f(Y_{w,h,m}) = f\left(\sum_{i=(w-1)s+1}^{(w-1)s+k} \sum_{j=(h-1)s+1}^{N} \sum_{K=1}^{N} W_{k,m} x_{i,j,k} + b_m\right)$ , (2) where  $Y_{w,h,m}$  regards as convolution output layer with the dimension of h, w and m. Parameters bm and  $W_{k,m}$  are neurons bios and weights, respectively. The input of the network is  $x_{i,j,k}$  as (i,j)<sup>th</sup> pixel of the k<sup>th</sup> component of an RGB image.

A pooling layer as a sub-sampling layer is exploited after every convolution layer to summarize specifications and bold characteristics of the previous layer. In fact, accuracy in characteristics improves the training operation accuracy. In this regard, two possible sub-sampling strategies in this layer are maximum pooling and average pooling.

#### **RESULTS AND FINDING**

#### Images and CNN Layers

The images after noise reduction and removing unwanted lights rectified by the average filter. Then, the preprocessed images masked and segmented by the K-means clustering algorithm. In this step, 150 extracted bacteria units extracted to categorize each bacterium. Furthermore, the number of images in the primary dataset increased to train and validate the CNN. Rotation process employing 90° rotation, upside-down flip, and left-right flip, in collaboration with Gaussian noise addition used to increase the number of images in the dataset, see Fig 6 and 7. The resulted 9,000 units of data in the size of 85×85-pixel considered as the input layer in the CNN.

Input image dataset included both RGB and grayscale modes. Size of images regarded  $85\times85$  pixels while other image sizes like  $64\times64$ -pixel and  $125\times125$ -pixel had relatively poor results. In fact, the dimension of the images in the input layer for RGB and grayscale modes are subsequently  $85\times85\times3$  and  $85\times85\times1$ .

Counting algorithm is employed to count the number of bacteria in each microscopic image. In addition to the input layer, the network has three convolutional, pooling, fully connected layers. The trend of CNN operation includes epochs of training, frequency of validation and rate of learning. The algorithm chose 70.0% of the dataset for the training procedure including 6300 images per each labeled category of *E. coli* and *V. cholera*. It also selected 30.0% or 2700 images of the dataset as the validation set. Furthermore, for counting the number of bacteria, image ingredient counting algorithm used in the CNN.

#### Classification

Primary results of the CNN revealed that although the grayscale images due to the smaller size had a faster training/ validation, the accuracy of that was poor. However, the


Figure 5. CNN structure.



Figure 6. Masked and segmented images rotation.

RGB mode provided more data in the different color channels which resulted in higher accuracy in the cost of higher training/validation time. Color model of the images as the basic parts of this procedure are determinative factor for the bacteria classification. Since the RGB images contain more data, it provides better performance in the classification. Thus, training/validation in the grayscale space achieved 59.0% accuracy while RGB mode provides higher performance in the training/validation with accuracy of 93.01%. These results show the outstanding performance of the CNN in the classification of wastewater bacteria for the RGB images. Table 1 presents the training/validation results.

CNN training set exploited 6300 images for each bacterium and totally 12,600 images in the training procedure. CNN reached to the best point of the training in the 246<sup>th</sup> repetition. Results show that the CNN operation accuracy achieves 93.01% with least squares error (LSE) of 0.087 in the RGB mode. Figure 8 shows how the proposed method

#### Table 1. Training and validation results

Parameter	RGB	Grayscale
Accuracy (training - validation)	93.01%	59%
Error (training - validation)	0.087	1.9

#### Table 2. CNN counting results

Parameter	RGB	Grayscale
Accuracy	97%	97%
Error	0.03%	0.03%

meets the highest accuracy in the RGB mode classification. Last optimizing point achieved in 246<sup>th</sup> training and validation repetitions. Figure 9 illustrates the LSE convergence of the method which is saturated at the 246<sup>th</sup> iteration with the error of 0.087.



**Figure 7**. Adding Gaussian noise ( $\sigma$ =30) to images after rotation.



Figure 8. Procedure of accuracy optimizing (RGB), training, validation.

### **Counting Sample Bacteria**

Images ingredient counting algorithm used to count the number of bacteria in the CNN. This algorithm employs "Morphological opening" operations to keep each bacteria shape skeleton. Per each defined morph of bacteria up counter counted to add number to counter clipboard. Procedure is repeating for all morphs to the end. Total values of counter regarded as number of the bacteria for each microscopic image of bacteria. Figure 10 and 11 show the counting procedure of the microscopic sample image. This section of CNN operation's result achieved 97.0% of accuracy. Related results are presented in the Table 2.

### Sensitivity Analysis

Sensitivity analysis used to evaluate the effects of the images' quality on the accuracy. Different situation regarded as changes in primary image to increase numbers to 9,000 images for each category. These situations including normal images, image 90° rotation, upside-down flip images, and left-right flip images while Gaussian's noise ( $\sigma$ ) added to them. Accuracy of results evaluated for each situation. Enhancing the Gaussian noise caused to higher standard deviation ( $\sigma$ ) per pixel. Standard deviation relates directly to the histogram width or variance of the dataset. Figure 12 shows noise enhancement and histogram changes and the results of the sensitivity analysis are in the Table 3.

Figure 13 illustrates the effect of the Gaussian noise on four states of the images. The result reveals the inverse relation between the Gaussian noise and system accuracy. Generally, enhancing the gaussian noise totally decreases the CNN accuracy.

#### DISCUSSION

According to this study, microbial monitoring is used to analysis water, wastewater, and other water resources with high efficiency. Furthermore, evaluating wastewater treatment analysis efficiency is important. CNN model helps to analysis wastewater ingredients through the image processing methods. Monitoring the ingredients helps to analysis the qualities and effects of waste-



Figure 9. Procedure of error optimizing (RGB), training, validation.



Figure 10. E. coli counting processing procedure.

water secondary uses on the environment and human. In this study, CNN analysis used as the major approach to classify and count the bacteria in the wastewater. All operations in the CNN have been used in a fast and inexpensive manner while results achieved to high accuracy. Along with the computer processing improvement and in combination with the artificial intelligent development, new methods are competing as well with classic approaches and even are advance rather than classic approaches. This study used classification and counting bacteria simultaneously as a new method for the RGB and grayscale modes dataset. All operations are applicable to other types of the bacteria as well.

Using two categories of bacteria improves the CNN promptitude in comparison to high number of the bacteria categories. This point regarded to evaluate the CNN operation fast and remake it for better operation. Grayscale mode achieved poor accuracy in comparison to the RGB mode while counting results were equal. Increasing the number of images will improve the performance of the CNN. Clear images in the large datasets of bacteria are ideal cases but in common situations different sort of images in variety range of qualities are available. Hence, removing Gaussian noise will be such as overfit results with high accuracy of classification only in the high-quality image datasets. Therefore, providing a variety of different images in the dataset will be logical solution to elate the deep learning procedure.

The CNN used in this study made in seven layers and sub-layers, and the operation showed flexible performance in comparison to the pre-trained CNN, such as AlexNet, VGGNet, ResNet, etc. Pre-trained CNN uses more layer of convolution to achieve high accuracy of classification. These kinds of the convolution neural networks are not easily adaptable to the different cases of datasets and require more time and high-performance hardware to train and validate the datasets.

Result of sensitivity analysis verified the Gaussian noise has an inverse relation with the accuracy. Upside-down flip images took most effect of the Gaussian noise with poor accuracy of 58.0% while left-right flip images had most similar results to the normal images.



Figure 11. V. cholera counting processing procedure.



Figure 12. Gaussian noise histogram.

# CONCLUSION

This study used an image processing-based method through a CNN model in seven layers and sub-layers to classify and count bacteria in the RGB and grayscale images. *E. coli* and *V. cholera* bacteria considered as the most common bacteria and wastewater indicators.

From a computational viewpoint, this study verified the CNN performance for combining duties of classification and counting bacteria. The results revealed that the CNN model in the RGB mode achieved high efficiency with the accuracy of 93.01% and LSE of 0.087 to classify the bacteria. Also, counting results verified the CNN counting efficiency with accuracy of 97.0% for both RGB and grayscale images.



Figure 13. Plot of the sensitivity analysis.

The used method in this study can classify and count other bacteria which confirms the high flexibility of the model. Accuracy of 93.01% verifies high efficiency of classification in comparison with the pre-trained CNN with high number of layers and greater dataset. Moreover, all operations are faster and simple in use with low GPU system requirements.

Results of the sensitivity analysis indicated that Gaussian noise affects the performance of the CNN model. In the presence of the additive Gaussian noise ( $\sigma$ =30), upside-down flip images took most effect of that which resulted in decreasing the accuracy to 58.0%. Left-right flip and 90°-rotated images had most similar results to the normal images with the accuracy of 78.0% and 75.0%, respectively.

In principle, our study presents an inexpensive, fast, rather simple image processing-based method by employing the CNN. This method can be used to classify and count the number of bacteria in the wastewater, and in the similar approach sort the microbial. Adapting the used CNN for the classification and counting samples can be beneficial in the wastewater treatment, education, and microbial researchers.

# DATA AVAILABILITY STATEMENT

The authors confirm that the data that supports the findings of this study are available within the article. Raw data that support the finding of this study are available from the corresponding author, upon reasonable request.

### CONFLICT OF INTEREST

The authors declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

# ETHICS

There are no ethical issues with the publication of this manuscript.

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